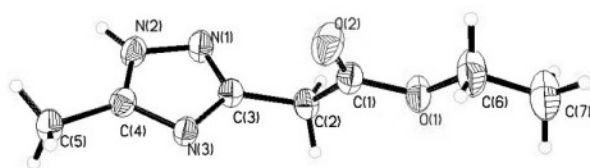


Crystal structure of ethyl 2-(5-methyl-4*H*-1,2,4-triazol-3-yl)acetate, $C_7H_{11}N_3O_2$

Guang-Zhou Zhu^I, Suo-Ping Xu^I and Zhong-Ming Yang^{*,II}^I Jiangsu Key Laboratory of Green Synthetic Chemistry for Functional Materials, Xuzhou Normal University, Xuzhou 221116, Jiangsu Province, P. R. China^{II} School of Life Sciences, Shangdong University Of Technology, Zibo 255000, Shandong Province, P. R. China

Received February 19, 2012, accepted July 19, 2012, available online November 09, 2012, CCDC no. 1267/3832



Abstract

$C_7H_{11}N_3O_2$, triclinic, $P\bar{1}$ (no. 2), $a = 5.0843(3)$ Å, $b = 7.8399(5)$ Å, $c = 12.361(1)$ Å, $\alpha = 107.030(2)^\circ$, $\beta = 95.273(1)^\circ$, $\gamma = 103.183(1)^\circ$, $V = 452.0$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.0672$, $wR_{\text{ref}}(F^2) = 0.1895$, $T = 298$ K.

Table 1. Data collection and handling.

Crystal:	colourless columns, size 0.08×0.12×0.48 mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	0.93 cm ⁻¹
Diffractionmeter, scan mode:	CCD area detector, φ and ω
$2\theta_{\text{max}}$:	50°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	2288, 1568
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1029
$N(\text{param})_{\text{refined}}$:	111
Programs:	SHELX [4]

Source of material

1,2-(5-methyl-4*H*-1,2,4-triazol-3-yl) acetic acid (0.1 mol, 14.1 g) was dissolved in ethanol (30 ml) with a reflux condenser. A catalytic amount of concentrated sulfuric acid was added at room temperature and the mixture was refluxed for another 2 hours. After the mixture was cooled in the ice bath, colourless column crystals were precipitated, filtered, washed with water for three times.

Experimental details

All H atoms were positioned geometrically (C–H = 0.96 Å for the aromatic H atoms and C–H = 0.97 Å for the aliphatic H atoms) and were refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$.

Discussion

There has been much research interest in triazole due to its biological activities [1, 2]. In this work, we report the crystal structure of the title compound, 2-(5-methyl-4*H*-1,2,4-triazol-3-yl)acetate (Fig. 1). In this compound all bond lengths are within normal ranges [2].

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	U_{iso}
H(2)	2i	−0.2035	0.6975	0.6440	0.066
H(2A)	2i	0.3252	0.5560	0.8993	0.064
H(2B)	2i	0.6027	0.6479	0.8682	0.064
H(5A)	2i	−0.0033	0.7559	0.4723	0.095
H(5B)	2i	0.2986	0.7424	0.4673	0.095
H(5C)	2i	0.2431	0.9239	0.5448	0.095
H(6A)	2i	0.7871	0.1499	0.7553	0.111
H(6B)	2i	0.5245	0.0516	0.7939	0.111
H(7A)	2i	1.0283	0.2258	0.9380	0.154
H(7B)	2i	0.9132	0.0106	0.8820	0.154
H(7C)	2i	0.7667	0.1209	0.9735	0.154

* Correspondence author (e-mail: forzayzm324@gmail.com)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
N(1)	2 <i>i</i>	0.0145(5)	0.6209(3)	0.7542(2)	0.056(2)	0.076(2)	0.060(1)	0.029(1)	0.016(1)	0.035(1)
N(2)	2 <i>i</i>	−0.0465(5)	0.6825(3)	0.6647(2)	0.047(1)	0.069(2)	0.061(1)	0.026(1)	0.009(1)	0.029(1)
N(3)	2 <i>i</i>	0.3815(4)	0.6783(3)	0.6675(2)	0.047(1)	0.054(1)	0.049(1)	0.020(1)	0.007(1)	0.022(1)
O(1)	2 <i>i</i>	0.6251(4)	0.3272(2)	0.8616(2)	0.084(2)	0.059(1)	0.056(1)	0.029(1)	−0.004(1)	0.0225(9)
O(2)	2 <i>i</i>	0.3792(5)	0.2688(3)	0.6889(2)	0.102(2)	0.072(2)	0.060(1)	0.036(1)	−0.017(1)	0.005(1)
C(4)	2 <i>i</i>	0.4711(6)	0.3703(4)	0.7852(2)	0.056(2)	0.062(2)	0.052(2)	0.020(1)	0.006(1)	0.024(1)
C(3)	2 <i>i</i>	0.4261(6)	0.5581(4)	0.8365(2)	0.061(2)	0.059(2)	0.047(1)	0.022(1)	0.011(1)	0.024(1)
C(2)	2 <i>i</i>	0.2727(6)	0.6192(3)	0.7523(2)	0.049(2)	0.052(2)	0.044(1)	0.019(1)	0.010(1)	0.019(1)
C(1)	2 <i>i</i>	0.1720(6)	0.7161(3)	0.6141(2)	0.050(2)	0.047(2)	0.050(1)	0.019(1)	0.005(1)	0.014(1)
C(5)	2 <i>i</i>	0.1781(6)	0.7913(4)	0.5158(2)	0.070(2)	0.066(2)	0.061(2)	0.020(2)	0.002(1)	0.032(1)
C(6)	2 <i>i</i>	0.692(1)	0.1518(5)	0.8199(3)	0.130(4)	0.069(2)	0.081(2)	0.050(2)	−0.009(2)	0.019(2)
C(7)	2 <i>i</i>	0.865(1)	0.1250(6)	0.9112(3)	0.138(4)	0.092(3)	0.092(3)	0.059(3)	−0.001(2)	0.034(2)

Acknowledgments. This project was funded by the Priority Academic Program Development of Jiangsu Higher Education Institutions, China.

References

1. Kiselyov, A. S.; Piatnitski, C.; Eugene, L.; Chernisheva, N. B.; Salamandra, L. K.; Semenov, V. V.: *Tetrahedron Lett.* (2009) 3809-3812.

2. Cipens, G. E.; Grinsteins, V.: *Vestis Lat Psr Zinat Akad.* (1965) 204-208.

3. Allen, F. H.; Kennard, O.; Watson, D. G.; Brammer, L.; Orpen, A. G.; Taylor, R.; *J. Chem. Soc. Perkin Trans.* **2** (1987) S1-19.

4. Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A64** (2008) 112-122.