

Crystal structure of 2-acetyl-2-(4-methylbenzenesulfonamido)-*N*-phenyl-pent-4-ynamide, C₂₀H₂₀N₂O₄S

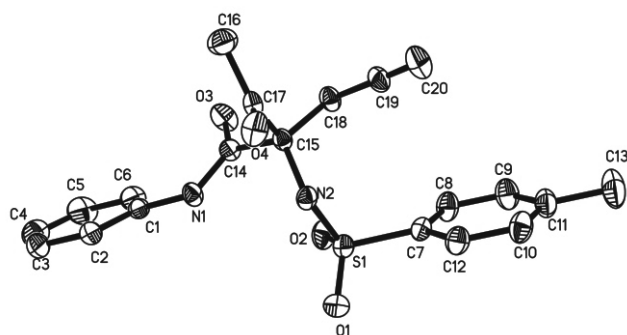
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Received March 30, 2011, accepted and available on-line November 7, 2011; CCDC no. 1267/3487



Abstract

C₂₀H₂₀N₂O₄S, monoclinic, *P*12₁/*c*1 (no. 14), *a* = 10.703(5) Å, *b* = 7.757(5) Å, *c* = 23.238(5) Å, β = 91.298(5)°, *V* = 1928.8 Å³, *Z* = 4, *R*_{gt}(*F*) = 0.048, *wR*_{ref}(*F*²) = 0.140, *T* = 293 K.

Source of material

To a solution of 2-acetyl-*N*-phenylpent-4-ynamide (215 mg, 1.0 mmol) and 4-methylbenzenesulfonamide (0.342 mg, 2.0 mmol) in CH₂Cl₂ (3.0 mL), FeCl₂ (13 mg, 0.1 mmol) was added. The reaction mixture was stirred at room temperature for 24 hours. When the starting material was consumed (monitored by TLC), the reaction mixture was quenched with 1M NH₄Cl (10 mL). The organic and aqueous layers were separated, and the aqueous layer was extracted with dichloromethane (3 times 10 mL). The combined organic layers were dried over MgSO₄ and filtered. The filtrate was concentrated *in vacuo*, and then the residue was purified by silica gel column chromatography (petroleum ether : diethyl ether = 10 : 3), to afford the product 2-acetyl-2-(4-methylbenzenesulfonamido)-*N*-phenylpent-4-ynamide (251 mg, 65 % yield). It was redissolved in a mixture of diethyl ether with ethyl acetate (10 mL, *v/v* = 3 : 1). This solution was sealed and kept at room temperature for three days. Colorless crystals of the title compound were separated and analyzed.

Experimental details

All H atoms on C atoms were generated geometrically and refined as riding atoms with *d*(C—H) = 0.96 Å (CH₃), 0.97 Å (CH₂) or 0.93 Å (CH), and *U*_{iso}(H) = 1.2 or 1.5 *U*_{eq}(C). The H atoms bonded to nitrogen atoms were located from difference Fourier map and refined with *U*_{iso}(H) = 1.5 *U*_{eq}(N).

Discussion

It is well-known that the carbon-nitrogen linkage can be formed directly from two simple carbon-hydrogen bonds (C—H) through metal-catalyzed reactions [1]. Compared with precious

metal catalysts such as Pt, Rh, Ru, Pd, Au, Ag, etc., iron is cheaper [2]. Therefore, the study of iron-catalyzed reactions involved in the carbon-nitrogen linkage has received great attention. In this work, under FeCl₂ catalyst condition, the reaction of 2-acetyl-*N*-phenylpent-4-ynamide and 4-methylbenzenesulfonamide gave the title compound 2-acetyl-2-(4-methylbenzenesulfonamido)-*N*-phenyl-pent-4-ynamide.

The title compound crystallizes with one molecule in the asymmetric unit. Geometric parameters are in the usual ranges. *d*(C14—O3) = 1.213(2) Å and *d*(C17—O4) = 1.195(2) Å are consistent with the values of the C=O double bond in carbonyl compounds. The bond length of 1.452(2) Å indicates the formation of C15—N2 single bond. There are two benzene rings in the structure, and the dihedral angles between them are 5.7°. The adjacent molecules are linked into a chain along *b* axis by intermolecular N—H⋯O and C—H⋯O hydrogen bonding interactions.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.13 × 0.18 × 0.21 mm
Wavelength:	Mo <i>K</i> _α radiation (0.71073 Å)
μ :	1.96 cm ^{−1}
Diffractionmeter, scan mode:	Rigaku RAXIS-RAPID, ω
2 θ _{max} :	58.42°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	12905, 4556
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 2979
<i>N</i> (<i>param</i>) _{refined} :	250
Programs:	SHELXS-97, SHELXL-97 [3]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4e	0.1098	0.0330	0.6473	0.068
H(3)	4e	−0.0773	−0.0845	0.7893	0.098
H(4)	4e	0.0850	0.0232	0.8455	0.079
H(5)	4e	0.2588	0.1385	0.8032	0.059
H(6)	4e	−0.0648	−0.0790	0.6907	0.092
H(8)	4e	0.6055	0.0660	0.5077	0.067
H(9)	4e	0.7578	0.1370	0.4432	0.076
H(10)	4e	0.9889	0.2666	0.5731	0.071
H(12)	4e	0.8385	0.1964	0.6379	0.061
H(13A)	4e	0.9579	0.2388	0.4205	0.139
H(13B)	4e	1.0615	0.1868	0.4660	0.139
H(13C)	4e	1.0110	0.3764	0.4640	0.139
H(16A)	4e	0.3556	0.7394	0.7000	0.097
H(16B)	4e	0.2628	0.5990	0.6760	0.097
H(16C)	4e	0.3503	0.6975	0.6340	0.097
H(18A)	4e	0.4582	0.3190	0.5522	0.053
H(18B)	4e	0.3795	0.4835	0.5665	0.053
H(20)	4e	0.7212	0.6856	0.5631	0.086
H(1A)	4e	0.372(2)	0.170(3)	0.7179(9)	0.063
H(2A)	4e	0.588(2)	0.292(3)	0.6859(8)	0.058

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
C(1)	4e	0.2007(2)	0.0993(2)	0.72107(8)	0.033(1)	0.035(1)	0.056(1)	−0.0004(8)	0.0069(8)	0.0069(8)
C(2)	4e	0.1046(2)	0.0325(3)	0.6872(1)	0.048(1)	0.049(1)	0.074(1)	−0.010(1)	0.001(1)	0.002(1)
C(3)	4e	−0.0070(3)	−0.0387(3)	0.7721(2)	0.056(2)	0.054(2)	0.137(3)	−0.010(1)	0.037(2)	0.021(2)
C(4)	4e	0.0898(2)	0.0261(3)	0.8056(1)	0.068(2)	0.052(1)	0.079(2)	0.004(1)	0.031(1)	0.018(1)
C(5)	4e	0.1936(2)	0.0950(3)	0.78039(9)	0.048(1)	0.047(1)	0.054(1)	0.0031(9)	0.0105(9)	0.0085(9)
C(6)	4e	0.0007(2)	−0.0353(3)	0.7133(1)	0.052(2)	0.057(2)	0.120(2)	−0.025(1)	0.002(2)	0.013(1)
C(7)	4e	0.7057(2)	0.1253(2)	0.57912(8)	0.035(1)	0.044(1)	0.045(1)	0.0016(8)	0.0050(8)	−0.0034(8)
C(8)	4e	0.6826(2)	0.1069(3)	0.52104(9)	0.043(1)	0.076(2)	0.049(1)	−0.008(1)	0.0022(9)	−0.009(1)
C(9)	4e	0.7742(2)	0.1494(3)	0.48249(9)	0.067(2)	0.081(2)	0.044(1)	−0.009(1)	0.011(1)	−0.006(1)
C(10)	4e	0.9117(2)	0.2265(3)	0.5598(1)	0.041(1)	0.066(1)	0.070(1)	−0.010(1)	0.008(1)	−0.012(1)
C(11)	4e	0.8896(2)	0.2098(3)	0.5010(1)	0.057(1)	0.056(1)	0.063(1)	−0.005(1)	0.023(1)	−0.004(1)
C(12)	4e	0.8219(2)	0.1849(3)	0.59859(9)	0.042(1)	0.060(1)	0.050(1)	−0.004(1)	0.0008(9)	−0.0067(9)
C(13)	4e	0.9890(3)	0.2573(4)	0.4591(1)	0.086(2)	0.100(2)	0.095(2)	−0.021(2)	0.048(2)	−0.008(2)
C(14)	4e	0.3144(2)	0.2650(2)	0.64813(7)	0.0323(9)	0.045(1)	0.0372(9)	−0.0030(8)	−0.0033(8)	0.0024(7)
C(15)	4e	0.4424(2)	0.3562(2)	0.64009(6)	0.0299(9)	0.045(1)	0.0304(8)	−0.0022(8)	−0.0003(7)	0.0050(7)
C(16)	4e	0.3439(2)	0.6497(3)	0.6719(1)	0.065(2)	0.052(1)	0.076(2)	0.008(1)	0.003(1)	−0.001(1)
C(17)	4e	0.4417(2)	0.5153(3)	0.68075(7)	0.037(1)	0.049(1)	0.0340(9)	−0.0092(8)	0.0081(8)	0.0012(8)
C(18)	4e	0.4535(2)	0.4182(3)	0.57749(7)	0.044(1)	0.059(1)	0.0302(9)	−0.0039(9)	0.0005(8)	0.0062(8)
C(19)	4e	0.5639(2)	0.5257(3)	0.57031(8)	0.054(1)	0.059(1)	0.0345(9)	0.000(1)	0.0080(9)	0.0097(9)
C(20)	4e	0.6516(3)	0.6148(4)	0.5663(1)	0.068(2)	0.080(2)	0.068(2)	−0.018(1)	0.015(1)	0.013(1)
N(1)	4e	0.3082(2)	0.1763(2)	0.69723(6)	0.0283(8)	0.057(1)	0.0406(8)	−0.0067(7)	−0.0007(6)	0.0101(7)
N(2)	4e	0.5457(1)	0.2489(2)	0.65999(6)	0.0297(8)	0.051(1)	0.0349(7)	−0.0004(7)	−0.0008(6)	0.0001(6)
O(1)	4e	0.6546(2)	−0.0248(2)	0.67373(6)	0.064(1)	0.066(1)	0.0674(9)	0.0169(8)	0.0115(8)	0.0227(8)
O(2)	4e	0.4857(1)	0.0028(2)	0.59932(6)	0.0459(9)	0.0598(9)	0.0704(9)	−0.0165(7)	0.0098(7)	−0.0109(7)
O(3)	4e	0.2287(1)	0.2875(2)	0.61393(6)	0.0409(8)	0.087(1)	0.0575(9)	−0.0157(8)	−0.0181(7)	0.0229(8)
O(4)	4e	0.5197(1)	0.5259(2)	0.71820(6)	0.0477(9)	0.080(1)	0.0472(8)	−0.0060(8)	−0.0026(7)	−0.0198(7)
S(1)	4e	0.59217(5)	0.07169(6)	0.62972(2)	0.0388(3)	0.0463(3)	0.0495(3)	−0.0003(2)	0.0075(2)	0.0032(2)

Acknowledgment. We thank the Science Foundation for Young Teachers of Northeast Normal University (grant no. 09QNJ013) for financial support.

References

1. Sun, C.-L.; Li, B.-J.; Shi, Z.-J.: Direct C-H Transformation via Iron Catalysis. *Chem. Rev.* **111** (2011) 1293-1314.
2. Wang, Z.; Zhang, Y.; Fu, H.; Jiang, Y.; Zhao, Y.: Efficient Intermolecular Iron-Catalyzed Amidation of C-H Bonds in the Presence of *N*-Bromosuccinimide. *Org. Lett.* **10** (2008) 1863-1866.
3. Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A64** (2008) 112-122.