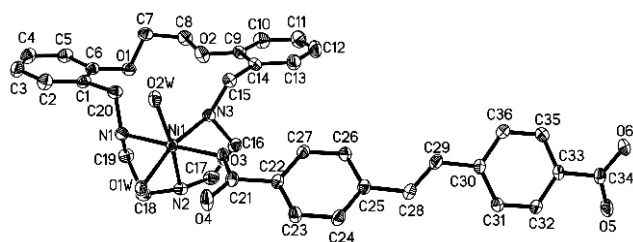


Crystal structure of diaqua(biphenylethene-4,4'-dicarboxylato- κO)-(1,12,15-triaza-3,4:9,10-dibenzo-5,8-dioxacycloheptadecane- $\kappa^3 N, N', N''$)nickel(II), $\text{Ni}(\text{H}_2\text{O})_2(\text{C}_{16}\text{H}_{10}\text{O}_4)(\text{C}_{20}\text{H}_{27}\text{N}_3\text{O}_2)$

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Received April 13, 2011, accepted and available on-line September 8, 2011; CCDC no. 1267/3519



Abstract

$\text{C}_{36}\text{H}_{41}\text{N}_3\text{NiO}_8$, monoclinic, $P12_1/c1$ (no. 14), $a = 15.193(3)$ Å, $b = 19.503(4)$ Å, $c = 11.983(3)$ Å, $\beta = 111.088(9)^\circ$, $V = 3312.9$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.047$, $wR_{\text{ref}}(F^2) = 0.100$, $T = 293$ K.

Source of material

A mixture of NiCO_3 (11.9 mg, 0.10 mmol) and biphenylethene-4,4'-dicarboxylic acid (26.8 mg, 0.10 mmol) in water was stirred for 10 min at 60 °C, then 1,12,15-triaza-3,4:9,10-dibenzo-5,8-dioxacycloheptadecane (L, 34.1 mg, 0.10 mmol) was added. After stirring for 30 min, a necessary amount of ammonia (14 mL) was added to get a clear solution. Green single crystals of the title compound were obtained by slow evaporation of the solution at ambient temperature (yield 71 %, based on NiCO_3).

Experimental details

All H atoms on C atoms were generated geometrically and refined as riding atoms with $d(\text{C}—\text{H}) = 0.97$ (CH₂) or 0.93 Å (CH), and $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$. The H atoms bonded to nitrogen and water were located from a difference Fourier map and refined with $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{O/N})$.

Discussion

The asymmetric unit of the title compound consists of one Ni(II) atom, one L ligand, one biphenylethene-4,4'-dicarboxylate ligand, and two coordinated water molecules. The Ni(II) cation is six-coordinated by three nitrogen atoms from the L ligand, one oxygen atom from the biphenylethene-4,4'-dicarboxylate ligand, and two water molecule, showing a distorted octahedral coordination. The biphenylethene-4,4'-dicarboxylate ligand coordinates to the Ni(II) center in a monodentate mode. There are intermolecular N—H...O and O—H...O hydrogen bonding between carboxylate oxygen atoms and coordinated water molecules, which connect the molecules to form a two-dimensional supra-molecular network.

Table 1. Data collection and handling.

Crystal:	green block, size 0.153 × 0.180 × 0.202 mm
Wavelength:	Mo K_α radiation (0.71069 Å)
μ :	6.43 cm ^{−1}
Diffraction, scan mode:	Rigaku RAXIS-RAPID, ω
$2\theta_{\text{max}}$:	54.96°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	31720, 7577
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 4934
$N(\text{param})_{\text{refined}}$:	454
Programs:	SHELXS-97, SHELXL-97 [1]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4e	0.4332	0.0747	0.3033	0.059
H(3)	4e	0.3248	−0.0008	0.3264	0.063
H(4)	4e	0.3334	−0.1153	0.2858	0.062
H(5)	4e	0.4527	−0.1552	0.2263	0.054
H(7A)	4e	0.5249	−0.0679	0.0288	0.055
H(7B)	4e	0.5246	−0.1459	0.0611	0.055
H(8A)	4e	0.6892	−0.1492	0.0982	0.055
H(8B)	4e	0.6286	−0.1337	−0.0366	0.055
H(10)	4e	0.7750	−0.1431	−0.0428	0.066
H(11)	4e	0.8848	−0.1209	−0.1316	0.076
H(12)	4e	0.9321	−0.0109	−0.1469	0.071
H(13)	4e	0.8773	0.0783	−0.0635	0.058
H(15A)	4e	0.7823	0.1231	0.0375	0.044
H(15B)	4e	0.6884	0.0828	0.0159	0.044
H(16A)	4e	0.9178	0.1053	0.2383	0.048
H(16B)	4e	0.9084	0.0256	0.2172	0.048
H(17A)	4e	0.8838	0.0074	0.3923	0.058
H(17B)	4e	0.9695	0.0583	0.4262	0.058
H(18A)	4e	0.7840	0.1089	0.5446	0.059
H(18B)	4e	0.8405	0.0403	0.5538	0.059
H(19A)	4e	0.7258	−0.0001	0.3757	0.056
H(19B)	4e	0.6752	0.0198	0.4649	0.056
H(20A)	4e	0.6058	0.0344	0.1780	0.048
H(20B)	4e	0.5378	0.0940	0.1799	0.048
H(23)	4e	0.9966	0.3711	0.3135	0.046
H(24)	4e	1.1160	0.3956	0.2430	0.047
H(26)	4e	1.0235	0.2451	0.0007	0.043
H(27)	4e	0.9095	0.2177	0.0770	0.041
H(28)	4e	1.2140	0.3605	0.1252	0.045
H(29)	4e	1.1156	0.2937	−0.0816	0.049
H(31)	4e	1.3528	0.3379	0.0998	0.053
H(32)	4e	1.4823	0.3216	0.0474	0.049
H(35)	4e	1.3166	0.2704	−0.2840	0.045
H(36)	4e	1.1865	0.2799	−0.2299	0.047
H(1A)	4e	0.621(2)	0.115(1)	0.365(2)	0.053
H(2A)	4e	0.888(2)	0.136(1)	0.467(2)	0.060
H(3A)	4e	0.757(2)	0.034(1)	0.193(2)	0.049

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1C)	4e	0.684(1)	0.236(2)	0.399(3)	0.059
H(1D)	4e	0.765(2)	0.263(1)	0.376(3)	0.059

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2C)	4e	0.585(2)	0.206(2)	0.175(3)	0.061
H(2D)	4e	0.622(2)	0.224(1)	0.088(2)	0.061

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Ni(1)	4e	0.74449(2)	0.15278(2)	0.27901(3)	0.0301(2)	0.0284(2)	0.0342(2)	−0.0036(1)	0.0151(1)	−0.0033(1)
C(1)	4e	0.5106(2)	0.0062(1)	0.2528(2)	0.035(1)	0.038(2)	0.038(2)	−0.009(1)	0.009(1)	0.003(1)
C(2)	4e	0.4379(2)	0.0283(2)	0.2882(3)	0.049(2)	0.040(2)	0.061(2)	−0.004(1)	0.022(2)	−0.003(1)
C(3)	4e	0.3725(2)	−0.0167(2)	0.3015(3)	0.042(2)	0.063(2)	0.057(2)	−0.004(2)	0.022(2)	0.004(2)
C(4)	4e	0.3781(2)	−0.0849(2)	0.2780(3)	0.045(2)	0.056(2)	0.055(2)	−0.014(2)	0.017(2)	0.012(2)
C(5)	4e	0.4494(2)	−0.1089(1)	0.2429(2)	0.047(2)	0.036(2)	0.047(2)	−0.009(1)	0.011(1)	0.004(1)
C(6)	4e	0.5163(2)	−0.0633(1)	0.2325(2)	0.037(1)	0.037(2)	0.032(2)	−0.004(1)	0.008(1)	0.004(1)
C(7)	4e	0.5628(2)	−0.1047(2)	0.0771(2)	0.053(2)	0.039(2)	0.042(2)	−0.009(1)	0.012(1)	−0.006(1)
C(8)	4e	0.6474(2)	−0.1161(1)	0.0444(3)	0.060(2)	0.032(2)	0.049(2)	−0.008(1)	0.023(2)	−0.005(1)
C(9)	4e	0.7602(2)	−0.0444(1)	−0.0002(2)	0.047(2)	0.039(2)	0.035(2)	−0.001(1)	0.016(1)	−0.003(1)
C(10)	4e	0.7951(2)	−0.0984(2)	−0.0471(3)	0.066(2)	0.041(2)	0.061(2)	0.001(2)	0.026(2)	−0.011(2)
C(11)	4e	0.8605(2)	−0.0849(2)	−0.1007(3)	0.063(2)	0.065(2)	0.071(2)	0.008(2)	0.034(2)	−0.022(2)
C(12)	4e	0.8897(2)	−0.0195(2)	−0.1086(3)	0.056(2)	0.069(2)	0.063(2)	−0.002(2)	0.036(2)	−0.014(2)
C(13)	4e	0.8561(2)	0.0339(2)	−0.0595(3)	0.049(2)	0.048(2)	0.053(2)	−0.004(1)	0.024(2)	−0.005(1)
C(14)	4e	0.7914(2)	0.0225(1)	−0.0045(2)	0.043(2)	0.037(2)	0.030(1)	0.002(1)	0.013(1)	−0.002(1)
C(15)	4e	0.7565(2)	0.0802(1)	0.0530(2)	0.044(2)	0.029(1)	0.040(2)	−0.001(1)	0.018(1)	0.000(1)
C(16)	4e	0.8850(2)	0.0645(1)	0.2486(2)	0.033(1)	0.040(2)	0.046(2)	0.005(1)	0.013(1)	−0.004(1)
C(17)	4e	0.9026(2)	0.0536(2)	0.3804(3)	0.042(2)	0.048(2)	0.048(2)	0.010(1)	0.006(1)	−0.000(1)
C(18)	4e	0.7998(2)	0.0733(2)	0.4985(2)	0.050(2)	0.057(2)	0.035(2)	−0.004(2)	0.009(1)	0.004(1)
C(19)	4e	0.7099(2)	0.0378(2)	0.4174(2)	0.051(2)	0.048(2)	0.040(2)	−0.009(1)	0.014(1)	0.011(1)
C(20)	4e	0.5761(2)	0.0571(1)	0.2273(2)	0.043(2)	0.037(2)	0.038(2)	−0.010(1)	0.011(1)	0.000(1)
C(21)	4e	0.8667(2)	0.2710(1)	0.2544(2)	0.032(1)	0.031(1)	0.041(2)	0.003(1)	0.017(1)	0.003(1)
C(22)	4e	0.9408(2)	0.2912(1)	0.2041(2)	0.029(1)	0.028(1)	0.037(1)	0.001(1)	0.013(1)	0.005(1)
C(23)	4e	1.0028(2)	0.3449(1)	0.2519(2)	0.044(1)	0.033(1)	0.047(2)	−0.003(1)	0.027(1)	−0.004(1)
C(24)	4e	1.0742(2)	0.3600(1)	0.2086(3)	0.035(1)	0.033(1)	0.055(2)	−0.007(1)	0.021(1)	−0.006(1)
C(25)	4e	1.0839(2)	0.3227(1)	0.1145(2)	0.029(1)	0.035(1)	0.038(2)	0.004(1)	0.015(1)	0.006(1)
C(26)	4e	1.0196(2)	0.2699(1)	0.0651(2)	0.031(1)	0.042(2)	0.036(2)	0.001(1)	0.015(1)	−0.003(1)
C(27)	4e	0.9504(2)	0.2540(1)	0.1101(2)	0.030(1)	0.036(1)	0.038(2)	−0.005(1)	0.013(1)	−0.003(1)
C(28)	4e	1.1640(2)	0.3346(1)	0.0752(2)	0.030(1)	0.038(2)	0.049(2)	−0.004(1)	0.018(1)	0.000(1)
C(29)	4e	1.1707(2)	0.3113(1)	−0.0255(2)	0.025(1)	0.052(2)	0.044(2)	−0.002(1)	0.011(1)	0.004(1)
C(30)	4e	1.2547(2)	0.3101(1)	−0.0593(2)	0.032(1)	0.038(1)	0.038(2)	−0.000(1)	0.018(1)	0.003(1)
C(31)	4e	1.3449(2)	0.3230(2)	0.0230(2)	0.034(1)	0.066(2)	0.036(2)	−0.006(1)	0.018(1)	−0.009(1)
C(32)	4e	1.4224(2)	0.3139(1)	−0.0091(2)	0.029(1)	0.058(2)	0.040(2)	−0.006(1)	0.017(1)	−0.007(1)
C(33)	4e	1.4138(2)	0.2936(1)	−0.1229(2)	0.035(1)	0.031(1)	0.038(2)	−0.000(1)	0.021(1)	0.003(1)
C(34)	4e	1.5006(2)	0.2837(1)	−0.1547(3)	0.043(2)	0.031(1)	0.048(2)	−0.003(1)	0.027(1)	0.001(1)
C(35)	4e	1.3242(2)	0.2825(1)	−0.2060(2)	0.041(2)	0.041(2)	0.034(2)	0.001(1)	0.016(1)	−0.001(1)
C(36)	4e	1.2460(2)	0.2893(1)	−0.1739(2)	0.031(1)	0.046(2)	0.039(2)	−0.005(1)	0.011(1)	−0.001(1)
O(1)	4e	0.5907(1)	−0.08730(9)	0.2007(2)	0.042(1)	0.045(1)	0.037(1)	0.0017(9)	0.0109(9)	0.0004(9)
O(2)	4e	0.6944(1)	−0.05143(9)	0.0534(2)	0.076(1)	0.031(1)	0.060(1)	−0.015(1)	0.041(1)	−0.0112(9)
O(3)	4e	0.8345(1)	0.21103(9)	0.2285(2)	0.048(1)	0.034(1)	0.056(1)	−0.0109(9)	0.035(1)	−0.0067(9)
O(4)	4e	0.8440(1)	0.3124(1)	0.3186(2)	0.068(1)	0.039(1)	0.089(2)	−0.013(1)	0.060(1)	−0.018(1)
O(5)	4e	1.5777(1)	0.2747(1)	−0.0697(2)	0.036(1)	0.090(2)	0.050(1)	0.010(1)	0.025(1)	0.014(1)
O(6)	4e	1.4905(1)	0.2859(1)	−0.2629(2)	0.050(1)	0.061(1)	0.042(1)	−0.003(1)	0.030(1)	−0.003(1)
N(1)	4e	0.6517(1)	0.0883(1)	0.3306(2)	0.037(1)	0.035(1)	0.035(1)	−0.006(1)	0.014(1)	−0.001(1)
N(2)	4e	0.8494(2)	0.1037(1)	0.4240(2)	0.039(1)	0.042(1)	0.036(1)	−0.004(1)	0.011(1)	−0.007(1)
N(3)	4e	0.7822(1)	0.0723(1)	0.1839(2)	0.035(1)	0.029(1)	0.033(1)	−0.0006(9)	0.012(1)	−0.0016(9)
O(1W)	4e	0.7400(1)	0.22794(9)	0.3985(2)	0.037(1)	0.039(1)	0.050(1)	−0.0081(9)	0.0252(9)	−0.0105(9)
O(2W)	4e	0.6270(1)	0.1953(1)	0.1445(2)	0.039(1)	0.047(1)	0.044(1)	0.0049(9)	0.0225(9)	0.0110(9)

Acknowledgments. We thank the Science Foundation for Young Teachers of Northeast Normal University (grant no. 20060304) and the Analysis and Testing Foundation of Northeast Normal University for support.

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