

Crystal structure of diaqua-cinnamate- κO -(1,12,15-triaza-3,4:9,10-dibenzo-5,8-dioxacycloheptadecane- $\kappa^3 N, N', N''$)nickel(II) cinnamate hydrate, $[\text{Ni}(\text{H}_2\text{O})_2(\text{C}_9\text{H}_7\text{O}_2)(\text{C}_{20}\text{H}_{27}\text{N}_3\text{O}_2)](\text{C}_9\text{H}_7\text{O}_2) \cdot \text{H}_2\text{O}$

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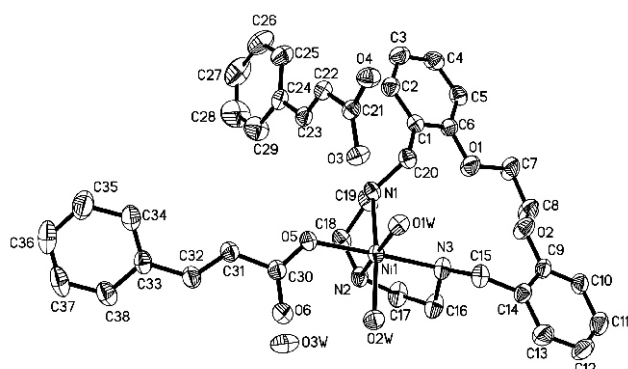


Table 1. Data collection and handling.

Crystal:	green block, size 0.150 × 0.304 × 0.314 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	5.65 cm ⁻¹
Diffractionmeter, scan mode:	Rigaku RAXIS-RAPID, ω
$2\theta_{\text{max}}$:	54.96°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	34507, 8729
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 5443
$N(\text{param})_{\text{refined}}$:	487
Programs:	SHELXS-97, SHELXL-97 [1]

Abstract

$\text{C}_{38}\text{H}_{47}\text{N}_3\text{NiO}_9$, monoclinic, $P2_1/n$ (no. 14), $a = 10.267(2)$ Å, $b = 12.421(4)$ Å, $c = 30.146(9)$ Å, $\beta = 97.08(1)^\circ$, $V = 3815.1$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.055$, $wR_{\text{ref}}(F^2) = 0.143$, $T = 293$ K.

Source of material

A mixture of NiCO_3 (11.9 mg, 0.10 mmol) and cinnamic acid (29.6 mg, 0.20 mmol) in water was stirred for 10 min at 60 °C, then 1,12,15-triaza-3,4:9,10-dibenzo-5,8-dioxacycloheptadecane (L, 34.1 mg, 0.10 mmol) was added. After stirring for 30 min, a necessary amount of ammonia (14 mL) was added to get a clear solution. Green single crystals of the title compound were obtained by slow evaporation of the solution at ambient temperature (yield 64 %, based on NiCO_3).

Experimental details

All H atoms on C atoms were generated geometrically and refined as riding atoms with $d(\text{C}—\text{H}) = 0.97$ (CH₂) or 0.93 Å (CH), and $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$. The H atoms bonded to nitrogen and water were located from difference Fourier map and refined with $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{O/N})$.

Discussion

In the asymmetric unit of the crystal structure of the title compound, there are one Ni(II) cation, one L ligand, two cinnamate anions, and three water molecules. The Ni(II) cation is six-coordinated by three nitrogen atoms from the L ligand, one oxygen atom from one cinnamate anion and two water molecules, forming a distorted octahedral coordination. Adjacent molecules are further linked by hydrogen bonds to form a dimer.

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4e	0.5500	−0.0525	0.3424	0.074
H(3)	4e	0.3555	−0.1184	0.3066	0.086
H(4)	4e	0.1619	−0.0841	0.3360	0.082
H(5)	4e	0.1626	0.0221	0.3981	0.072
H(7A)	4e	0.2547	0.0568	0.4806	0.096
H(7B)	4e	0.1857	0.1493	0.4511	0.096
H(8A)	4e	0.2845	0.2764	0.4943	0.096
H(8B)	4e	0.2376	0.1915	0.5275	0.096
H(10)	4e	0.2976	0.2765	0.5875	0.079
H(11)	4e	0.3863	0.3630	0.6517	0.096
H(12)	4e	0.6070	0.3852	0.6673	0.104
H(13)	4e	0.7446	0.3326	0.6155	0.091
H(15A)	4e	0.7833	0.2453	0.5493	0.067
H(15B)	4e	0.6862	0.1506	0.5355	0.067
H(16A)	4e	0.7333	0.4252	0.5057	0.087
H(16B)	4e	0.5822	0.4142	0.5090	0.087
H(17A)	4e	0.5327	0.4126	0.4331	0.095
H(17B)	4e	0.6307	0.5089	0.4436	0.095
H(18A)	4e	0.7077	0.3361	0.3508	0.078
H(18B)	4e	0.5823	0.4010	0.3598	0.078
H(19A)	4e	0.5024	0.2541	0.3954	0.077
H(19B)	4e	0.5346	0.2158	0.3483	0.077
H(20A)	4e	0.5895	0.0797	0.4472	0.063
H(20B)	4e	0.6666	0.0060	0.4175	0.063
H(22)	4e	0.8625	−0.2041	0.3569	0.060
H(23)	4e	0.9020	0.0108	0.3697	0.059
H(25)	4e	0.7217	−0.1751	0.2967	0.084
H(26)	4e	0.6212	−0.1422	0.2254	0.113
H(27)	4e	0.6427	0.0232	0.1937	0.131
H(28)	4e	0.7621	0.1542	0.2330	0.149
H(29)	4e	0.8590	0.1234	0.3044	0.103
H(31)	4e	1.0760	0.1925	0.3382	0.066
H(32)	4e	1.2011	0.3844	0.3517	0.075
H(34)	4e	1.2142	0.1396	0.2912	0.094
H(35)	4e	1.3508	0.105	0.2374	0.118
H(36)	4e	1.4844	0.2375	0.2155	0.118
H(37)	4e	1.4853	0.4032	0.248	0.109
H(38)	4e	1.3495	0.4405	0.3010	0.089
H(1C)	4e	0.907(3)	0.057(2)	0.460(1)	0.077

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1D)	4e	0.911(3)	0.105(3)	0.5020(7)	0.077
H(2C)	4e	0.999(3)	0.349(3)	0.469(1)	0.091
H(2D)	4e	0.979(4)	0.303(3)	0.5092(9)	0.091
H(3D)	4e	0.917(5)	0.613(3)	0.409(2)	0.155

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(3C)	4e	0.993(4)	0.572(4)	0.376(2)	0.155
H(1A)	4e	0.719(3)	0.142(3)	0.381(1)	0.071
H(2A)	4e	0.770(3)	0.428(3)	0.411(1)	0.082
H(3A)	4e	0.590(2)	0.248(3)	0.481(1)	0.082

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Ni(1)	4e	0.80211(3)	0.24425(3)	0.44555(1)	0.0396(2)	0.0457(2)	0.0440(2)	0.0055(2)	0.0074(1)	0.0043(2)
C(1)	4e	0.4759(3)	0.0270(2)	0.3913(1)	0.044(2)	0.047(2)	0.053(2)	0.002(1)	0.007(1)	0.006(1)
C(2)	4e	0.4720(3)	−0.0365(3)	0.3537(1)	0.056(2)	0.065(2)	0.064(2)	0.000(2)	0.014(2)	−0.008(2)
C(3)	4e	0.3559(4)	−0.0770(3)	0.3323(1)	0.077(2)	0.073(3)	0.064(2)	−0.006(2)	0.004(2)	−0.015(2)
C(4)	4e	0.2404(3)	−0.0553(3)	0.3496(1)	0.058(2)	0.072(2)	0.073(2)	−0.014(2)	−0.003(2)	−0.003(2)
C(5)	4e	0.2406(3)	0.0081(3)	0.3865(1)	0.040(2)	0.072(2)	0.068(2)	−0.006(2)	0.005(2)	0.005(2)
C(6)	4e	0.3568(3)	0.0515(3)	0.4065(1)	0.046(2)	0.055(2)	0.055(2)	−0.002(1)	0.009(1)	−0.004(2)
C(7)	4e	0.2672(3)	0.1280(4)	0.4686(1)	0.050(2)	0.120(4)	0.073(2)	−0.005(2)	0.022(2)	−0.019(2)
C(8)	4e	0.2975(3)	0.2034(4)	0.5055(1)	0.056(2)	0.110(3)	0.076(3)	0.018(2)	0.020(2)	−0.018(2)
C(9)	4e	0.4692(3)	0.2501(3)	0.5636(1)	0.066(2)	0.052(2)	0.048(2)	0.006(2)	0.020(1)	0.002(2)
C(10)	4e	0.3878(4)	0.2866(3)	0.5933(1)	0.082(2)	0.057(2)	0.066(2)	0.002(2)	0.035(2)	−0.001(2)
C(11)	4e	0.4412(5)	0.3380(3)	0.6316(1)	0.118(4)	0.061(2)	0.067(3)	0.012(2)	0.038(2)	0.001(2)
C(12)	4e	0.5721(5)	0.3528(3)	0.6406(1)	0.135(4)	0.072(3)	0.054(2)	−0.002(3)	0.014(2)	−0.016(2)
C(13)	4e	0.6549(4)	0.3195(3)	0.6098(1)	0.092(3)	0.073(3)	0.061(2)	−0.004(2)	0.005(2)	−0.002(2)
C(14)	4e	0.6048(3)	0.2672(3)	0.5707(1)	0.069(2)	0.050(2)	0.045(2)	0.008(2)	0.010(1)	0.004(1)
C(15)	4e	0.6933(3)	0.2283(3)	0.5378(1)	0.050(2)	0.069(2)	0.050(2)	0.014(2)	0.010(1)	0.012(2)
C(16)	4e	0.6523(4)	0.3928(3)	0.4920(1)	0.096(3)	0.061(2)	0.064(2)	0.036(2)	0.029(2)	0.015(2)
C(17)	4e	0.6225(4)	0.4312(3)	0.4444(1)	0.103(3)	0.062(2)	0.078(3)	0.037(2)	0.033(2)	0.024(2)
C(18)	4e	0.6443(3)	0.3463(3)	0.3718(1)	0.061(2)	0.073(2)	0.058(2)	−0.003(2)	−0.003(2)	0.023(2)
C(19)	4e	0.5728(3)	0.2421(3)	0.3773(1)	0.052(2)	0.070(2)	0.067(2)	0.001(2)	−0.006(2)	0.022(2)
C(20)	4e	0.6042(3)	0.0649(3)	0.4167(1)	0.042(2)	0.061(2)	0.055(2)	−0.000(1)	0.004(1)	0.007(2)
C(21)	4e	0.9534(3)	−0.1584(3)	0.4200(1)	0.038(1)	0.051(2)	0.045(2)	0.006(1)	0.007(1)	−0.003(1)
C(22)	4e	0.8903(3)	−0.1429(3)	0.3732(1)	0.052(2)	0.053(2)	0.046(2)	−0.001(1)	0.011(1)	−0.004(1)
C(23)	4e	0.8710(3)	−0.0495(3)	0.3534(1)	0.043(2)	0.056(2)	0.050(2)	0.001(1)	0.013(1)	−0.004(1)
C(24)	4e	0.8051(3)	−0.0305(3)	0.3079(1)	0.049(2)	0.063(2)	0.043(2)	0.009(2)	0.011(1)	−0.000(2)
C(25)	4e	0.7310(4)	−0.1079(3)	0.2838(1)	0.093(3)	0.061(2)	0.055(2)	0.010(2)	0.004(2)	−0.003(2)
C(26)	4e	0.6702(5)	−0.0885(4)	0.2412(1)	0.109(3)	0.108(4)	0.061(3)	0.023(3)	−0.011(2)	−0.019(3)
C(27)	4e	0.6827(6)	0.0094(5)	0.2225(2)	0.143(5)	0.127(5)	0.051(3)	0.046(4)	−0.007(3)	0.013(3)
C(28)	4e	0.7535(6)	0.0870(5)	0.2459(2)	0.183(6)	0.094(4)	0.089(4)	0.000(4)	−0.009(4)	0.042(3)
C(29)	4e	0.8129(4)	0.0683(4)	0.2885(1)	0.104(3)	0.076(3)	0.074(3)	−0.003(2)	−0.003(2)	0.023(2)
C(30)	4e	1.0229(3)	0.2867(3)	0.3903(1)	0.052(2)	0.057(2)	0.058(2)	−0.001(2)	0.011(2)	0.002(2)
C(31)	4e	1.0957(3)	0.2579(3)	0.3525(1)	0.049(2)	0.058(2)	0.061(2)	−0.001(2)	0.011(1)	0.002(2)
C(32)	4e	1.1859(3)	0.3180(3)	0.3377(1)	0.063(2)	0.065(2)	0.062(2)	−0.005(2)	0.017(2)	0.006(2)
C(33)	4e	1.2657(3)	0.2939(3)	0.3020(1)	0.049(2)	0.075(2)	0.050(2)	0.002(2)	0.009(1)	0.014(2)
C(34)	4e	1.2681(4)	0.1939(4)	0.2826(1)	0.077(3)	0.086(3)	0.079(3)	−0.006(2)	0.032(2)	0.004(2)
C(35)	4e	1.3504(5)	0.1730(4)	0.2503(2)	0.101(3)	0.108(4)	0.093(3)	0.010(3)	0.044(3)	−0.005(3)
C(36)	4e	1.4301(4)	0.2515(5)	0.2374(2)	0.077(3)	0.150(5)	0.075(3)	0.012(3)	0.035(2)	0.021(3)
C(37)	4e	1.4298(4)	0.3499(5)	0.2565(2)	0.067(2)	0.130(4)	0.079(3)	−0.015(3)	0.022(2)	0.035(3)
C(38)	4e	1.3487(4)	0.3721(3)	0.2884(1)	0.068(2)	0.085(3)	0.070(2)	−0.013(2)	0.012(2)	0.017(2)
O(1)	4e	0.3644(2)	0.1239(2)	0.44143(9)	0.057(1)	0.108(2)	0.080(2)	−0.015(1)	0.024(1)	−0.032(2)
O(2)	4e	0.4255(2)	0.1922(2)	0.52574(8)	0.054(1)	0.103(2)	0.066(2)	0.008(1)	0.011(1)	−0.025(2)
O(3)	4e	0.9887(2)	−0.0774(2)	0.44345(7)	0.066(1)	0.050(1)	0.050(1)	0.009(1)	−0.003(1)	−0.006(1)
O(4)	4e	0.9661(2)	−0.2538(2)	0.43318(7)	0.069(1)	0.048(1)	0.057(1)	0.002(1)	−0.002(1)	−0.001(1)
O(6)	4e	1.0640(3)	0.3651(2)	0.41485(9)	0.092(2)	0.073(2)	0.090(2)	−0.023(2)	0.036(2)	−0.016(2)
O(5)	4e	0.9296(2)	0.2254(2)	0.39708(7)	0.044(1)	0.061(1)	0.061(1)	−0.006(1)	0.017(1)	−0.004(1)
O(1W)	4e	0.8720(2)	0.1021(2)	0.47560(7)	0.053(1)	0.052(1)	0.049(1)	0.015(1)	−0.001(1)	−0.000(1)
O(2W)	4e	0.9394(2)	0.3346(2)	0.48611(8)	0.061(1)	0.064(2)	0.056(1)	−0.000(1)	0.003(1)	0.000(1)
O(3W)	4e	0.9140(4)	0.5505(3)	0.3893(1)	0.125(3)	0.080(2)	0.099(2)	−0.011(2)	−0.009(2)	−0.025(2)
N(1)	4e	0.6658(2)	0.1613(2)	0.39909(8)	0.038(1)	0.052(2)	0.053(2)	−0.001(1)	0.007(1)	0.010(1)
N(2)	4e	0.7127(3)	0.3822(2)	0.41532(9)	0.064(2)	0.049(2)	0.054(2)	0.006(1)	0.016(1)	0.012(1)
N(3)	4e	0.6650(3)	0.2746(2)	0.49250(9)	0.058(2)	0.056(2)	0.052(2)	0.020(1)	0.016(1)	0.012(1)

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References

- Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A64** (2008) 112–122.