

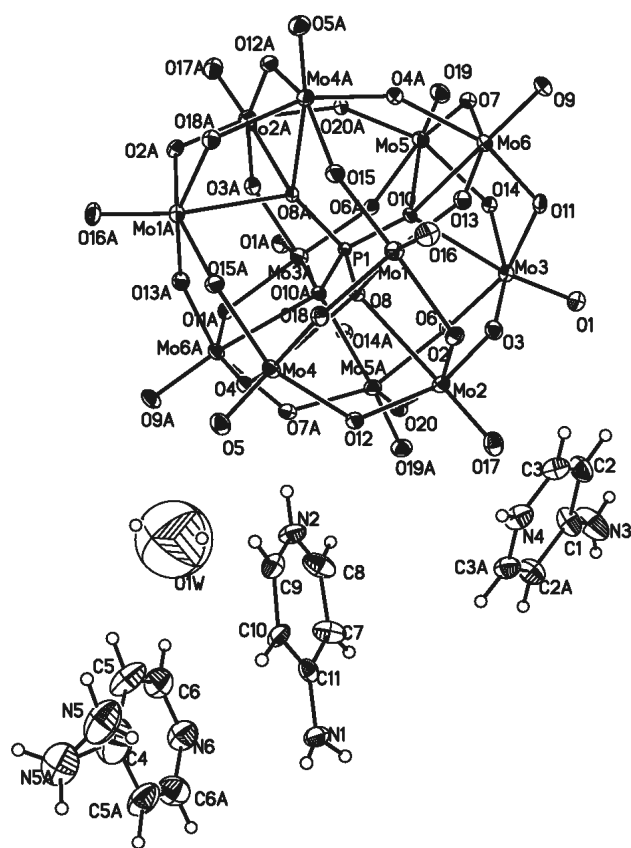
Crystal structure of tris(4-aminopyridinium) phosphatopolyoxododecamolybdate — 4-aminopyridine — water (1:1:1), $[\text{C}_5\text{H}_7\text{N}_2]_3[\text{PMo}_{12}\text{O}_{40}] \cdot \text{C}_5\text{H}_6\text{N}_2 \cdot \text{H}_2\text{O}$

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Abstract

$\text{C}_{20}\text{H}_{29}\text{Mo}_{12}\text{N}_8\text{O}_{41}\text{P}$, monoclinic, $C12/c1$ (no. 15), $a = 11.7094(4)$ Å, $b = 25.7946(9)$ Å, $c = 16.6748(6)$ Å, $\beta = 97.205(1)^\circ$, $V = 4996.7$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.044$, $wR_{\text{ref}}(F^2) = 0.142$, $T = 293$ K.

Source of material

A mixture of $\text{Na}_3\text{PO}_4 \cdot 12\text{H}_2\text{O}$ (0.1520 g, 0.40 mmol), $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ (0.2900 g, 1.20 mmol), 4-aminopyridine (0.0940 g, 0.0319 mmol), H_2SO_4 (0.4 ml, 1:1 v/v), H_2O (15.00 ml) was sealed in a 25 ml Teflon-lined bomb at 453 K for 120 h, then slowly cooled to room temperature. Red block-like crystals were collected by filtration, washed with distilled water, and air-dried giving a yield of 30 % based on the initial $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$.

Experimental details

In the 4-aminopyridine molecule, the N5 atom which links on C4 atom is disordered, the occupancy factor was set to 50 %.

Discussion

The title compound consists of a Keggin $(\text{PMo}_{12}\text{O}_{40})^{3-}$ polyoxoanion, three 4-aminopyridine $(\text{C}_5\text{H}_7\text{N}_2)^+$ cation, one 4-aminopyridine molecule and one water molecule. The Keggin $(\text{PMo}_{12}\text{O}_{40})^{3-}$ polyoxoanion contains a central PO_4 tetrahedron surrounded by four groups Mo_3O_{10} of three edge-shared octahedra MoO_6 . These Mo_3O_{10} triplets are linked to each other, and linked to the PO_4 tetrahedron by sharing corners. The PO_4 tetrahedron in which the P—O bond lengths are in the normal range of 1.531(5) Å - 1.535(5) Å and the O—P—O bond angles vary in the range of 108.9(2)° - 110.6(2)°, is a almost regular tetrahedron. Each of MoO_6 octahedrons is distorted. There are one terminal oxygen atom (O_t), four doubly bridging oxygen atoms (O_b), one triply bridging oxygen atom (O_a). The bond angles of O—Mo—O range from 71.2(2)° to 172.6(2)°. The Mo— O_t , Mo— O_b and Mo— O_a distances are in ranges 1.553(4) Å to 1.683(6) Å, 1.796(5) Å to 2.032(4) Å and 2.414(5) to 2.430(5) Å, respectively. In the 4-aminopyridine cations, the bond lengths C—C are in range from 1.22(2) to 1.48(2) Å and C—N are in range from 1.32(2) to 1.33(2) Å, exhibiting normal C—C and C—N bond distances. A lot of intermolecular N—H⋯O hydrogen bonds ($d(\text{N—H} \cdots \text{O}) = 2.940 - 3.377$ Å) were found in the title crystal structure. All of the bond lengths and bond angles are within the normal ranges and are consistent with those described in the literature [1]. There are profuse hydrogen bondings in the target crystal, which are involved with Keggin type heteropolyanion, 4-aminopyridine and water molecular result in an intricate three-dimensional supramolecular network.

Table 1. Data collection and handling.

Crystal:	red block, size 0.25 × 0.26 × 0.32 mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	30.54 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
$2\theta_{\text{max}}$:	50°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	15366, 4276
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3903
$N(\text{param})_{\text{refined}}$:	383
Programs:	SHELXS-97, SHELXL-97 [2]

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Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(5A)	8 <i>f</i>	0.50	0.9217	0.0052	0.2981	0.116
H(5B)	8 <i>f</i>	0.50	1.0501	0.0052	0.3251	0.116
H(5)	8 <i>f</i>		0.8375	0.0791	0.2857	0.092
H(6A)	8 <i>f</i>		0.8482	0.1606	0.2812	0.082
H(1A)	8 <i>f</i>		1.2094	0.0945	0.4530	0.049
H(1B)	8 <i>f</i>		1.2091	0.1525	0.4602	0.049
H(2A)	8 <i>f</i>		0.7860	0.1159	0.5527	0.058
H(7)	8 <i>f</i>		1.0351	0.1993	0.4860	0.051
H(8)	8 <i>f</i>		0.8543	0.1930	0.5261	0.063

Table 2. Continued.

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(9)	8 <i>f</i>		0.8773	0.0424	0.5416	0.055
H(10)	8 <i>f</i>		1.0492	0.0432	0.4903	0.043
H(3A)	8 <i>f</i>		0.9635	0.4076	0.7744	0.094
H(4A)	4 <i>e</i>		1.0000	0.1914	⅓	0.052
H(2C)	8 <i>f</i>		0.9138	0.3201	0.8482	0.055
H(3)	8 <i>f</i>		0.9197	0.2318	0.845	0.051
H(1)	8 <i>f</i>	0.50	0.7506	0.1303	0.4111	0.672
H(2)	8 <i>f</i>	0.50	0.6642	0.1197	0.3410	0.672

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Mo(1)	8 <i>f</i>		0.64597(5)	0.01185(3)	0.86443(4)	0.0194(4)	0.0234(4)	0.0178(4)	0.0023(3)	0.0011(2)	0.0043(3)
Mo(2)	8 <i>f</i>		0.80290(5)	0.10861(3)	0.79475(4)	0.0152(4)	0.0264(4)	0.0207(4)	−0.0007(2)	0.0013(3)	0.0002(3)
Mo(3)	8 <i>f</i>		0.63374(6)	0.21400(3)	0.87148(4)	0.0203(4)	0.0242(4)	0.0187(4)	−0.0046(3)	0.0016(3)	−0.0033(3)
Mo(4)	8 <i>f</i>		0.67051(5)	0.01999(3)	0.66296(4)	0.0197(4)	0.0218(4)	0.0167(4)	0.0042(3)	0.0026(2)	−0.0014(2)
Mo(5)	8 <i>f</i>		0.34045(5)	0.20634(3)	0.84653(4)	0.0191(4)	0.0245(4)	0.0202(4)	0.0020(3)	0.0040(3)	−0.0029(3)
Mo(6)	8 <i>f</i>		0.50072(6)	0.11762(3)	0.96295(4)	0.0197(4)	0.0254(4)	0.0127(3)	−0.0025(3)	0.0022(2)	−0.0004(2)
P(1)	4 <i>e</i>		½	0.1132(1)	⅓	0.011(1)	0.016(1)	0.010(1)	0	0.0006(9)	0
O(1)	8 <i>f</i>		0.7190(5)	0.2598(2)	0.9161(3)	0.026(3)	0.028(3)	0.026(3)	−0.004(2)	0.002(2)	−0.004(2)
O(2)	8 <i>f</i>		0.7892(4)	0.0562(2)	0.8638(3)	0.014(2)	0.030(3)	0.021(3)	0.003(2)	−0.002(2)	0.002(2)
O(3)	8 <i>f</i>		0.7455(4)	0.1579(2)	0.8560(3)	0.023(3)	0.027(3)	0.018(3)	0.002(2)	0.001(2)	0.001(2)
O(4)	8 <i>f</i>		0.6225(4)	0.0704(2)	0.5894(3)	0.020(2)	0.023(3)	0.016(2)	0.004(2)	0.003(2)	0.000(2)
O(5)	8 <i>f</i>		0.7297(5)	−0.0245(2)	0.6083(3)	0.027(3)	0.026(3)	0.026(3)	0.007(2)	0.003(2)	−0.006(2)
O(6)	8 <i>f</i>		0.6370(4)	0.2294(2)	0.7631(3)	0.017(2)	0.025(3)	0.016(3)	0.004(2)	0.000(2)	−0.002(2)
O(7)	8 <i>f</i>		0.3729(4)	0.1695(2)	0.9416(3)	0.019(3)	0.024(3)	0.019(3)	−0.003(2)	0.005(2)	−0.003(2)
O(8)	8 <i>f</i>		0.6073(4)	0.0785(2)	0.7624(3)	0.016(2)	0.016(3)	0.013(2)	0.003(2)	0.001(2)	0.002(2)
O(9)	8 <i>f</i>		0.4960(5)	0.1058(2)	1.0612(3)	0.031(3)	0.036(4)	0.014(3)	−0.005(2)	0.004(2)	−0.001(2)
O(10)	8 <i>f</i>		0.4949(4)	0.1471(2)	0.8250(3)	0.014(2)	0.028(3)	0.009(2)	0.000(2)	0.001(2)	−0.002(2)
O(11)	8 <i>f</i>		0.6018(4)	0.1729(2)	0.9672(3)	0.019(3)	0.021(3)	0.017(2)	0.000(2)	−0.003(2)	−0.005(2)
O(12)	8 <i>f</i>		0.8013(4)	0.0570(2)	0.7087(3)	0.016(2)	0.025(3)	0.019(3)	0.002(2)	0.002(2)	−0.001(2)
O(13)	8 <i>f</i>		0.6004(4)	0.0679(2)	0.9341(3)	0.020(3)	0.032(3)	0.017(3)	0.000(2)	0.000(2)	0.000(2)
O(14)	8 <i>f</i>		0.4881(4)	0.2431(2)	0.8744(3)	0.019(3)	0.022(3)	0.017(2)	−0.004(2)	0.002(2)	−0.002(2)
O(15)	8 <i>f</i>		0.4899(4)	−0.0023(2)	0.8455(3)	0.018(3)	0.028(3)	0.020(3)	0.001(2)	0.003(2)	−0.001(2)
O(16)	8 <i>f</i>		0.6961(5)	−0.0325(2)	0.9322(3)	0.027(3)	0.026(3)	0.028(3)	0.005(2)	−0.001(2)	0.011(2)
O(17)	8 <i>f</i>		0.9330(4)	0.1205(2)	0.8174(3)	0.032(3)	0.026(3)	0.026(3)	0.006(2)	−0.001(2)	−0.003(2)
O(18)	8 <i>f</i>		0.6808(4)	−0.0168(2)	0.7679(3)	0.022(3)	0.017(3)	0.020(3)	0.001(2)	0.002(2)	−0.001(2)
O(19)	8 <i>f</i>		0.2498(5)	0.2524(2)	0.8721(3)	0.028(3)	0.032(3)	0.027(3)	0.005(2)	0.007(2)	−0.007(2)
O(20)	8 <i>f</i>		0.7568(4)	0.1571(2)	0.7044(3)	0.017(2)	0.024(3)	0.016(2)	0.003(2)	0.002(2)	0.004(2)
N(5)	8 <i>f</i>	0.50	0.992(3)	0.015(1)	0.288(2)	0.09(2)	0.06(2)	0.14(3)	−0.02(1)	−0.01(2)	0.01(1)
N(6)	4 <i>e</i>		0	0.1684(7)	¼	0.058(9)	0.06(1)	0.08(1)	0	0.003(8)	0
C(4)	4 <i>e</i>		0	0.063(1)	¼	0.12(1)	0.10(1)	0.12(1)	0	−0.016(9)	0
C(5)	8 <i>f</i>		0.903(1)	0.0947(6)	0.271(1)	0.041(7)	0.06(1)	0.12(1)	−0.007(6)	−0.010(7)	0.018(9)
C(6)	8 <i>f</i>		0.911(1)	0.1417(6)	0.2686(8)	0.066(8)	0.07(1)	0.069(8)	−0.013(7)	0.010(6)	−0.004(7)
N(1)	8 <i>f</i>		1.1671(7)	0.1232(3)	0.4571(5)	0.027(4)	0.044(5)	0.053(5)	−0.003(3)	0.010(4)	−0.008(4)
N(2)	8 <i>f</i>		0.8528(7)	0.1171(4)	0.5365(6)	0.023(4)	0.071(7)	0.051(6)	−0.001(4)	0.009(4)	−0.001(4)
C(7)	8 <i>f</i>		1.0037(9)	0.1670(4)	0.4950(6)	0.052(6)	0.031(5)	0.048(6)	0.016(4)	0.022(5)	0.025(4)
C(8)	8 <i>f</i>		0.897(1)	0.1631(5)	0.5202(6)	0.058(7)	0.058(8)	0.045(6)	0.030(6)	0.019(5)	0.012(5)
C(9)	8 <i>f</i>		0.9088(8)	0.0738(5)	0.5279(6)	0.035(5)	0.051(7)	0.051(6)	−0.010(5)	0.005(4)	−0.015(5)
C(10)	8 <i>f</i>		1.0128(7)	0.0744(4)	0.4992(6)	0.022(4)	0.036(5)	0.048(5)	−0.005(4)	0.000(4)	−0.006(4)
C(11)	8 <i>f</i>		1.0651(8)	0.1210(4)	0.4829(5)	0.037(5)	0.034(5)	0.021(4)	0.008(4)	0.000(3)	−0.001(3)
N(3)	4 <i>e</i>		0	0.3825(6)	¾	0.14(2)	0.05(1)	0.06(1)	0	0.04(1)	0
N(4)	4 <i>e</i>		0	0.2247(5)	¾	0.038(6)	0.022(6)	0.070(8)	0	0.009(6)	0
C(1)	4 <i>e</i>		0	0.3310(5)	¾	0.08(1)	0.016(7)	0.038(7)	0	0.011(7)	0
C(2)	8 <i>f</i>		0.949(1)	0.3030(4)	0.8087(6)	0.062(7)	0.040(6)	0.038(5)	−0.003(5)	0.013(5)	−0.012(4)
C(3)	8 <i>f</i>		0.9522(8)	0.2505(4)	0.8060(7)	0.034(5)	0.040(6)	0.054(6)	0.000(4)	0.011(4)	0.004(5)
O(1W)	8 <i>f</i>	0.50	0.713(6)	0.140(4)	0.367(5)	0.45(6)	0.45(6)	0.45(6)	0.00(1)	0.06(1)	0.00(1)

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