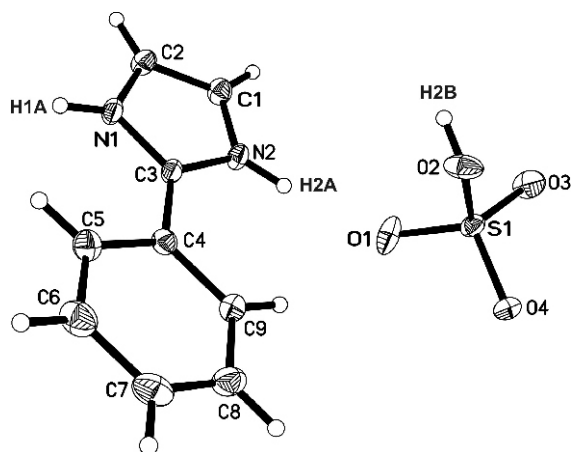


Crystal structure of 2-phenylimidazolium hydrogen sulfate, $[\text{C}_9\text{H}_9\text{N}_2][\text{HSO}_4]$

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Abstract

$\text{C}_9\text{H}_{10}\text{N}_2\text{O}_4\text{S}$, triclinic, $P\bar{1}$ (no. 2), $a = 4.481(5)$ Å, $b = 9.300(5)$ Å, $c = 13.014(5)$ Å, $\alpha = 93.829(5)^\circ$, $\beta = 99.633(5)^\circ$, $\gamma = 102.739(5)^\circ$, $V = 518.5$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.048$, $wR_{\text{ref}}(F^2) = 0.132$, $T = 293$ K.

Source of material

A mixture of 2-phenylimidazole (0.5 mmol) and H_2SO_4 (0.2 mL) was dissolved in water (10 mL). Colorless crystals were obtained at room temperature after two weeks.

Experimental details

All H atoms bound to carbon atoms were positioned geometrically and treated as riding with $d(\text{C}—\text{H}) = 0.93$ Å and $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$. H atoms bonded to N and O were located in difference Fourier maps and refined isotropically with distance restraints of $d(\text{O}—\text{H}) = 0.82(1)$ Å and $d(\text{N}—\text{H}) = 0.86(1)$ Å.

Discussion

2-Phenylimidazole is an important N-containing ligand with excellent coordinating abilities and fruitful aromatic systems. The current interest in 2-phenylimidazole or its derivative is rapidly

expanding for their potential applications in medical chemistry, host-guest chemistry, ion exchange, and nonlinear optics [1].

There are one 2-phenylimidazolium cation and one hydrogen sulfate anion in the asymmetric unit of the title compound, $[(\text{C}_9\text{H}_9\text{N}_2)(\text{HSO}_4)]$. The bond lengths $d(\text{C1}—\text{C2})$, $d(\text{C1}—\text{N2})$, $d(\text{C2}—\text{N1})$, $d(\text{C3}—\text{N1})$ and $d(\text{C3}—\text{N2})$ are 1.339(3), 1.360(3), 1.367(3), 1.331(3) and 1.332(3) Å, respectively. The C1/C2/N1/C3/N2 ring is an imidazole group.

In the crystal, molecules are linked into layer by H-bonding interactions: $d(\text{N1}—\text{H1A} \cdots \text{O4}) = 2.963(3)$ Å, $\angle \text{N1}—\text{H1A} \cdots \text{O4} = 178(3)^\circ$ and $d(\text{N2}—\text{H2A} \cdots \text{O1}) = 2.736(3)$ Å, $\angle \text{N2}—\text{H2A} \cdots \text{O1} = 175(3)^\circ$, $d(\text{O2}—\text{H2B} \cdots \text{O4}) = 2.637(3)$ Å, $\angle \text{O2}—\text{H2B} \cdots \text{O4} = 174(3)^\circ$.

Table 1. Data collection and handling.

Crystal:	colorless sheet, size $0.12 \times 0.14 \times 0.21$ mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	3.13 cm^{-1}
Diffractionmeter, scan mode:	Oxford Diffraction Gemini R Ultra, ω
$2\theta_{\text{max}}$:	58.38°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	3733, 2348
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1714
$N(\text{param})_{\text{refined}}$:	158
Programs:	SHELXS-97, SHELXL-97 [2]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(1)	2i	0.4846	0.3285	0.0090	0.050
H(2)	2i	0.2593	0.0753	0.0425	0.051
H(5)	2i	0.9312	0.0495	0.3471	0.059
H(6)	2i	1.3042	0.0795	0.4962	0.073
H(7)	2i	1.7082	0.2895	0.5326	0.073
H(8)	2i	1.7354	0.4699	0.4193	0.065
H(9)	2i	1.3577	0.4447	0.2724	0.051
H(2A)	2i	0.936(5)	0.419(2)	0.140(2)	0.055(7)
H(1A)	2i	0.577(7)	0.028(2)	0.195(2)	0.066(8)
H(2B)	2i	0.777(4)	0.741(3)	0.274(2)	0.061(9)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
C(1)	2i	0.5613(6)	0.2783(2)	0.0631(2)	0.047(1)	0.040(1)	0.038(1)	0.009(1)	0.007(1)	0.0094(9)
C(2)	2i	0.4376(6)	0.1397(2)	0.0812(2)	0.039(1)	0.038(1)	0.043(1)	−0.0002(9)	0.004(1)	0.002(1)
C(3)	2i	0.8588(5)	0.2298(2)	0.2029(2)	0.034(1)	0.0263(9)	0.038(1)	0.0043(8)	0.0114(9)	0.0026(8)
C(4)	2i	1.1028(5)	0.2445(2)	0.2944(2)	0.036(1)	0.035(1)	0.037(1)	0.0107(9)	0.0086(9)	0.0034(9)

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Table 3. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
C(5)	2 <i>i</i>	1.0912(6)	0.1350(3)	0.3620(2)	0.051(2)	0.045(1)	0.055(1)	0.013(1)	0.011(1)	0.014(1)
C(6)	2 <i>i</i>	1.3148(7)	0.1525(3)	0.4506(2)	0.064(2)	0.072(2)	0.053(2)	0.027(2)	0.008(1)	0.023(1)
C(7)	2 <i>i</i>	1.5565(7)	0.2781(4)	0.4724(2)	0.055(2)	0.083(2)	0.047(1)	0.029(2)	0.000(1)	0.001(1)
C(8)	2 <i>i</i>	1.5716(6)	0.3860(3)	0.4051(2)	0.044(1)	0.056(2)	0.056(2)	0.007(1)	0.001(1)	−0.011(1)
C(9)	2 <i>i</i>	1.3471(5)	0.3707(2)	0.3172(2)	0.042(1)	0.037(1)	0.046(1)	0.007(1)	0.006(1)	0.003(1)
O(1)	2 <i>i</i>	1.1560(5)	0.6169(2)	0.1418(1)	0.089(2)	0.0317(8)	0.065(1)	−0.0065(9)	0.024(1)	0.0035(8)
O(2)	2 <i>i</i>	0.9321(4)	0.7087(2)	0.2786(2)	0.037(1)	0.084(1)	0.070(1)	0.0259(9)	0.0206(9)	0.048(1)
O(3)	2 <i>i</i>	1.0100(4)	0.8519(2)	0.1328(1)	0.049(1)	0.058(1)	0.065(1)	0.0061(8)	0.0024(8)	0.0345(9)
O(4)	2 <i>i</i>	1.4461(3)	0.8280(2)	0.2581(1)	0.0291(8)	0.0429(9)	0.065(1)	0.0020(7)	0.0008(8)	0.0030(8)
S(1)	2 <i>i</i>	1.1431(1)	0.75412(5)	0.19663(4)	0.0294(3)	0.0326(3)	0.0450(4)	−0.0004(2)	0.0037(2)	0.0129(2)
N(1)	2 <i>i</i>	0.6257(4)	0.1118(2)	0.1680(2)	0.041(1)	0.0262(9)	0.046(1)	0.0018(8)	0.0062(8)	0.0092(8)
N(2)	2 <i>i</i>	0.8195(4)	0.3322(2)	0.1384(1)	0.041(1)	0.0275(9)	0.0399(9)	0.0012(8)	0.0075(8)	0.0079(8)

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