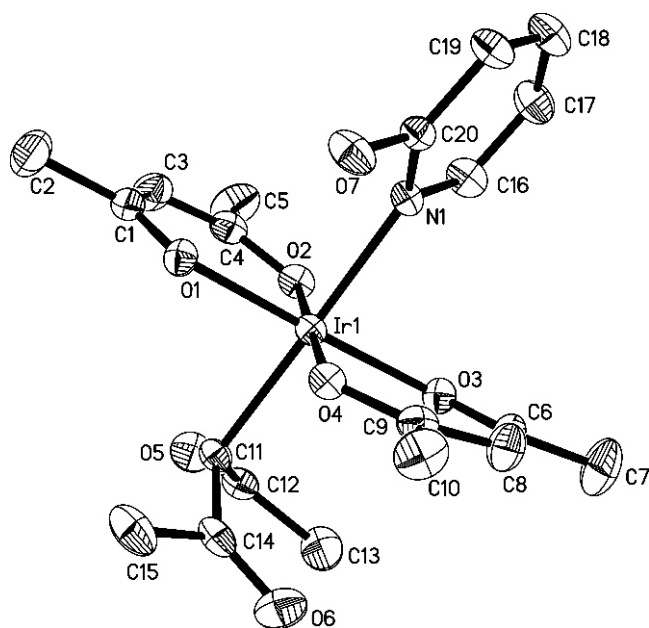


Crystal structure of bis(acetylacetonato- κ^2O,O')-(diacetylmethanido- κC)-(2-hydroxypyridine)iridium(III), $\text{Ir}(\text{C}_5\text{H}_7\text{O}_2)_3(\text{C}_5\text{H}_5\text{NO})$

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Abstract

$\text{C}_{20}\text{H}_{26}\text{IrNO}_7$, orthorhombic, *Pbca* (no. 61), $a = 9.4798(8)$ Å, $b = 14.646(1)$ Å, $c = 31.842(3)$ Å, $V = 4421.0$ Å³, $Z = 8$, $R_{\text{gt}}(F) = 0.036$, $wR_{\text{ref}}(F^2) = 0.074$, $T = 293$ K.

Source of material

The title complex was prepared by mixing the aqueous solution of aquabis(acetylacetonato- κ^2O,O')-(diacetylmethanido- κC)-iridium(III) and equimolar amount of 2-hydroxypyridine. After one week, a yellow crystalline product precipitated. Single crystals were selected from the product for the X-ray diffraction analysis.

Experimental details

All the hydrogen atoms were positioned geometrically with $d(\text{C}—\text{H}) = 0.93 - 0.98$ Å, $d(\text{O}—\text{H}) = 0.82$ Å and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5 U_{\text{eq}}(\text{C})$ and $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{O})$.

Discussion

In the title complex, the Ir atom is six-coordinated and situated in a slightly distorted octahedral environment, formed by four oxygen atoms of two acetylacetonate ligands, one carbon atom of one acetylacetonate ligand and one N atom of 2-hydroxypyridine. The average O—Ir—O chelating angle is $94.1(1)^\circ$. The average Ir—O

bond length of two acetylacetonate ligands is $2.023(3)$ Å, Ir—C bond length is found to be $2.162(5)$ Å, which agree with the literature data of $\text{Ir}(\text{C}_5\text{H}_7\text{O}_2)_3$, $\text{Ir}(\text{C}_5\text{H}_7\text{O}_2)_3(\text{H}_2\text{O})$, $\text{Ir}(\text{C}_5\text{H}_7\text{O}_2)_3(\text{C}_2\text{H}_6\text{SO})$ and $\text{Ir}(\text{C}_5\text{H}_7\text{O}_2)_3(\text{CH}_3\text{CN}) \cdot 1.5\text{H}_2\text{O}$ [1–5]. Ir—N bond length is found to be $2.140(4)$ Å. Changes in the coordination mode of one acetylacetonate ligand of tris(acetylacetonato- O,O')iridium(III) lead to the formation of a direct Ir—C bond. The average Ir—O, Ir—C and Ir—N bond lengths are close to each other. Ir—C bond length is found to be between Ir—C σ -bond ($2.00 - 2.02$ Å) and Ir—C π -bond ($2.27 - 2.31$ Å). In the crystal structure of the title complex there are two different types of coordinating acac ligands: a conventional bidentate (O-bonded acac ligand) and a γ -C bonded acac ligand, as Periana proposed [3]. The crystal packing of the title complex is stabilized by extensive hydrogen bonds formed between 2-hydroxypyridine and carbonyl O atoms of the acetylacetonate ligand with $d(\text{O7}—\text{H7} \cdots \text{O5}) = 2.618(5)$ Å and $\angle \text{O7}—\text{H7} \cdots \text{O5} = 165.9^\circ$.

Table 1. Data collection and handling.

Crystal:	yellow block, size $0.12 \times 0.15 \times 0.19$ mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	60.79 cm^{-1}
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
$2\theta_{\text{max}}$:	56.74°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	28671, 5364
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3751
$N(\text{param})_{\text{refined}}$:	269
Programs:	SHELXS-97, SHELXL-97 [6], DIAMOND [7]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(7)	8c	0.1448	0.4073	0.6475	0.072
H(2A)	8c	0.5749	0.5605	0.7659	0.093
H(2B)	8c	0.4403	0.4994	0.7710	0.093
H(2C)	8c	0.4429	0.5772	0.7373	0.093
H(3)	8c	0.6736	0.3963	0.7603	0.061
H(5A)	8c	0.8133	0.2073	0.7118	0.099
H(5B)	8c	0.7781	0.2484	0.7560	0.099
H(5C)	8c	0.9040	0.2878	0.7298	0.099
H(7A)	8c	0.6912	0.2567	0.5108	0.107
H(7B)	8c	0.5368	0.2539	0.4932	0.107
H(7C)	8c	0.5805	0.1894	0.5302	0.107
H(8)	8c	0.4030	0.3751	0.5125	0.057
H(10A)	8c	0.2026	0.5107	0.5569	0.081
H(10B)	8c	0.2862	0.5127	0.5144	0.081
H(10C)	8c	0.3214	0.5836	0.5498	0.081

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(11)	8c	0.7555	0.5265	0.6508	0.039
H(13A)	8c	1.0150	0.3747	0.5808	0.089
H(13B)	8c	0.9263	0.4491	0.5572	0.089
H(13C)	8c	0.8568	0.3556	0.5694	0.089
H(15A)	8c	0.6990	0.6919	0.6098	0.089
H(15B)	8c	0.5860	0.6315	0.6326	0.089

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(15C)	8c	0.5621	0.6617	0.5860	0.089
H(16)	8c	0.5462	0.1989	0.6398	0.053
H(17)	8c	0.3963	0.0799	0.6474	0.061
H(18)	8c	0.1575	0.1079	0.6580	0.064
H(19)	8c	0.0788	0.2567	0.6581	0.055

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Ir(1)	8c	0.57132(2)	0.39919(1)	0.633268(6)	0.0231(1)	0.0273(1)	0.0325(1)	0.00097(9)	−0.00159(8)	0.00115(9)
N(1)	8c	0.4088(4)	0.3007(3)	0.6443(1)	0.029(2)	0.027(2)	0.037(2)	−0.000(2)	−0.004(2)	0.001(2)
O(1)	8c	0.5125(3)	0.4787(2)	0.6820(1)	0.029(2)	0.037(2)	0.036(2)	0.003(2)	−0.001(2)	−0.005(2)
O(2)	8c	0.6984(3)	0.3185(2)	0.6683(1)	0.029(2)	0.037(2)	0.043(2)	0.004(2)	−0.006(2)	0.007(2)
O(3)	8c	0.6220(3)	0.3180(2)	0.5838(1)	0.033(2)	0.035(2)	0.038(2)	0.005(2)	−0.000(2)	−0.003(2)
O(4)	8c	0.4408(3)	0.4786(2)	0.5991(1)	0.028(2)	0.031(2)	0.038(2)	0.003(2)	−0.003(2)	0.002(2)
O(5)	8c	0.9569(4)	0.4304(3)	0.6506(1)	0.032(2)	0.064(3)	0.066(3)	0.003(2)	−0.002(2)	0.003(2)
O(6)	8c	0.7298(5)	0.5475(3)	0.5541(1)	0.076(3)	0.072(3)	0.049(2)	0.004(3)	0.003(2)	0.016(2)
O(7)	8c	0.2300(4)	0.4036(3)	0.6515(1)	0.030(2)	0.034(2)	0.079(3)	0.002(2)	0.005(2)	0.001(2)
C(1)	8c	0.5563(5)	0.4655(4)	0.7195(2)	0.040(3)	0.043(3)	0.036(3)	−0.010(3)	0.006(3)	−0.004(2)
C(2)	8c	0.4984(7)	0.5316(5)	0.7513(2)	0.081(5)	0.062(5)	0.044(3)	0.006(4)	0.007(3)	−0.010(3)
C(3)	8c	0.6503(6)	0.3968(4)	0.7319(2)	0.057(4)	0.059(4)	0.036(3)	0.000(4)	−0.005(3)	0.007(3)
C(4)	8c	0.7125(5)	0.3302(4)	0.7079(2)	0.032(3)	0.043(4)	0.045(3)	−0.003(3)	−0.005(2)	0.010(3)
C(5)	8c	0.8109(7)	0.2623(5)	0.7282(2)	0.066(4)	0.084(5)	0.050(4)	0.029(4)	−0.006(3)	0.016(4)
C(6)	8c	0.5502(6)	0.3211(4)	0.5492(2)	0.050(4)	0.047(4)	0.033(3)	−0.006(3)	−0.002(3)	−0.006(3)
C(7)	8c	0.5936(7)	0.2487(5)	0.5180(2)	0.096(6)	0.064(5)	0.055(4)	0.027(4)	−0.011(4)	−0.022(4)
C(8)	8c	0.4462(6)	0.3836(4)	0.5385(2)	0.049(3)	0.051(4)	0.042(3)	0.009(3)	−0.012(3)	−0.006(3)
C(9)	8c	0.3996(5)	0.4572(4)	0.5622(2)	0.027(3)	0.044(3)	0.040(3)	0.000(2)	−0.002(2)	0.006(3)
C(10)	8c	0.2928(6)	0.5219(4)	0.5442(2)	0.050(4)	0.061(4)	0.052(3)	0.015(3)	−0.008(3)	0.011(3)
C(11)	8c	0.7463(5)	0.4920(4)	0.6246(1)	0.026(3)	0.032(3)	0.041(3)	−0.004(2)	−0.001(2)	0.000(2)
C(12)	8c	0.8779(6)	0.4412(4)	0.6194(2)	0.033(3)	0.037(3)	0.048(3)	−0.003(3)	0.003(3)	0.005(3)
C(13)	8c	0.9231(6)	0.4016(4)	0.5780(2)	0.041(3)	0.062(4)	0.075(4)	0.003(3)	0.015(3)	−0.013(4)
C(14)	8c	0.7052(6)	0.5595(4)	0.5914(2)	0.038(3)	0.035(3)	0.060(4)	−0.012(3)	−0.001(3)	0.000(3)
C(15)	8c	0.6315(6)	0.6437(4)	0.6063(2)	0.048(4)	0.034(4)	0.095(5)	−0.001(3)	0.001(4)	0.002(4)
C(16)	8c	0.4508(6)	0.2111(4)	0.6437(2)	0.034(3)	0.042(4)	0.056(4)	0.005(3)	−0.004(2)	0.006(3)
C(17)	8c	0.3621(7)	0.1394(4)	0.6483(2)	0.054(4)	0.029(3)	0.068(4)	0.000(3)	−0.004(3)	0.003(3)
C(18)	8c	0.2205(6)	0.1559(4)	0.6543(2)	0.051(4)	0.038(4)	0.070(4)	−0.013(3)	−0.004(3)	0.007(3)
C(19)	8c	0.1745(6)	0.2446(4)	0.6547(2)	0.034(3)	0.045(4)	0.060(4)	−0.008(3)	−0.001(3)	0.005(3)
C(20)	8c	0.2689(5)	0.3170(4)	0.6502(2)	0.031(3)	0.031(3)	0.034(3)	0.000(2)	−0.002(2)	0.004(2)

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