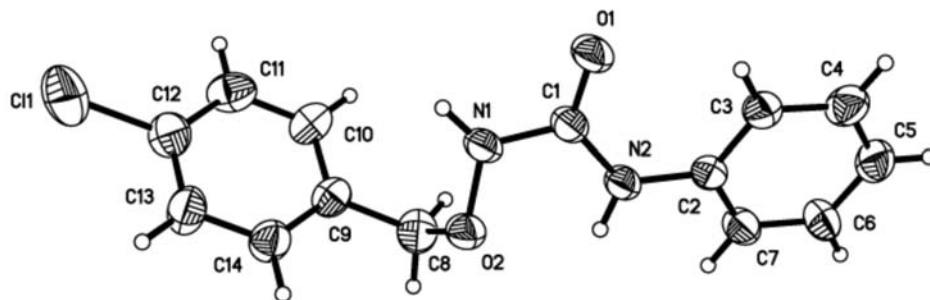


Crystal structure of 1-(4-chlorobenzoyloxy)-3-phenylurea, C₁₄H₁₃ClN₂O₂

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Abstract

C₁₄H₁₃ClN₂O₂, monoclinic, *P*12₁/*c*1 (no. 14), *a* = 5.3549(4) Å, *b* = 15.555(1) Å, *c* = 16.561(2) Å, β = 95.759(1)°, *V* = 1372.5 Å³, *Z* = 4, *R*_{gt}(*F*) = 0.049, *wR*_{ref}(*F*²) = 0.149, *T* = 298 K.

Source of material

Synthesis of the title compound was accomplished by reaction of aniline (6 mmol) with 4-nitrophenyl-4-chlorobenzoyloxy-carbamate (6.5 mmol) in anhydrous dichloromethane (40 mL) in the presence of anhydrous triethylamine (0.5 mL). The reaction mixture was stirred for 12 h, then washed with 1 M NaOH (30 mL for three times), 1 M HCl (30 mL) and water (30 mL) [1]. The solution was dried with MgSO₄, filtered, and distilled under reduced pressure at 25 °C to gain the desired compound as a yellow solid. The substance was soaked in ether at 0–5 °C for 4 h, filtered, then recrystallized from acetic ether/light petroleum, filtered and dried. The structure was confirmed by IR, MS and NMR. Colorless needle-shaped single crystals of the title compound suitable for X-ray diffraction analysis were obtained by slow evaporation of the mixed solvent acetone/cyclohexane (1:2, *v/v*) at room temperature for one week (m.p. 130–132 °C).

Experimental details

All hydrogen atoms were placed in idealized positions and treated as riding with *d*(C—H) = 0.93 Å for benzene ring, *d*(C—H) = 0.97 Å for methylene, *d*(N—H) = 0.86 Å, and *U*_{iso}(H) = 1.2 *U*_{eq}(C,N).

Discussion

In the title crystal structure, the bond length of the carbonyl bond (C=O) is 1.225 Å which is in the normal range of 1.19–1.23 Å, similar to that observed in our previous papers [2–4]. There are three significant atomic planes in the structure, two benzene rings and the urea plane (O1, N1, N2, C1). The dihedral angle between the two benzene rings is 68.08(7)°, while the urea group forms di-

hedral angles of 14.07(2)° and 58.05(7)° with benzene and 4-chlorophenyl rings, respectively. In the crystal structure, molecules are linked into parallel chains along [001] by intermolecular N—H⋯O hydrogen bonding.

Table 1. Data collection and handling.

Crystal:	colorless needle, size 0.08 × 0.12 × 0.40 mm
Wavelength:	Mo <i>K</i> _α radiation (0.71073 Å)
μ:	2.77 cm ^{−1}
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
2θ _{max} :	50.02°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	7122, 2423
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ(<i>I</i> _{obs}), 1544
<i>N</i> (<i>param</i>) _{refined} :	172
Programs:	SHELXS-97, SHELXL-97 [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	4e	0.1094	0.4721	0.4168	0.077
H(2)	4e	0.4811	0.6359	0.4145	0.072
H(3)	4e	0.0175	0.7523	0.4985	0.080
H(4)	4e	0.0053	0.9007	0.5043	0.098
H(5)	4e	0.2897	0.9826	0.4433	0.101
H(6)	4e	0.5979	0.9158	0.3789	0.097
H(7)	4e	0.6163	0.7683	0.3730	0.083
H(8A)	4e	0.6602	0.4614	0.3160	0.096
H(8B)	4e	0.3972	0.5019	0.2885	0.096
H(10)	4e	0.0672	0.4066	0.2453	0.093
H(11)	4e	−0.1012	0.2699	0.2314	0.103
H(13)	4e	0.4762	0.1781	0.3767	0.105
H(14)	4e	0.6436	0.3147	0.3884	0.094

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Cl(1)	4e	0.0507(2)	0.10849(6)	0.29104(7)	0.162(1)	0.0943(7)	0.159(1)	−0.0310(6)	0.0019(8)	−0.0437(6)
N(1)	4e	0.2208(3)	0.5121(1)	0.4212(1)	0.047(1)	0.057(1)	0.091(1)	−0.0091(8)	0.017(1)	−0.010(1)
N(2)	4e	0.3404(3)	0.6549(1)	0.4282(1)	0.051(1)	0.053(1)	0.079(1)	−0.0048(8)	0.0172(9)	−0.0042(9)
O(1)	4e	−0.0374(3)	0.6106(1)	0.4663(1)	0.059(1)	0.064(1)	0.123(2)	−0.0066(8)	0.038(1)	−0.006(1)
O(2)	4e	0.4739(3)	0.4972(1)	0.4083(1)	0.054(1)	0.065(1)	0.090(1)	−0.0018(7)	0.0169(9)	−0.0069(8)
C(1)	4e	0.1653(4)	0.5951(1)	0.4411(2)	0.055(1)	0.056(1)	0.071(2)	−0.008(1)	0.014(1)	−0.004(1)
C(2)	4e	0.3181(4)	0.7446(1)	0.4346(1)	0.052(1)	0.055(1)	0.060(1)	−0.004(1)	0.001(1)	−0.002(1)
C(3)	4e	0.1350(4)	0.7847(2)	0.4742(2)	0.058(2)	0.063(2)	0.081(2)	0.003(1)	0.010(1)	−0.002(1)
C(4)	4e	0.1280(5)	0.8738(2)	0.4773(2)	0.071(2)	0.067(2)	0.107(2)	0.012(1)	0.003(2)	−0.013(2)
C(5)	4e	0.2980(5)	0.9229(2)	0.4415(2)	0.082(2)	0.055(2)	0.111(2)	−0.000(1)	−0.011(2)	−0.005(2)
C(6)	4e	0.4804(5)	0.8830(2)	0.4029(2)	0.082(2)	0.063(2)	0.097(2)	−0.018(1)	0.006(2)	0.004(1)
C(7)	4e	0.4913(5)	0.7947(2)	0.3993(2)	0.064(2)	0.063(2)	0.080(2)	−0.010(1)	0.013(1)	−0.004(1)
C(8)	4e	0.4862(5)	0.4635(2)	0.3275(2)	0.086(2)	0.076(2)	0.083(2)	0.006(1)	0.035(2)	0.004(1)
C(9)	4e	0.3739(5)	0.3750(2)	0.3180(1)	0.070(2)	0.070(2)	0.058(1)	0.012(1)	0.015(1)	−0.001(1)
C(10)	4e	0.1515(5)	0.3607(2)	0.2716(2)	0.078(2)	0.086(2)	0.068(2)	0.020(2)	0.006(1)	0.001(1)
C(11)	4e	0.0499(5)	0.2788(2)	0.2631(2)	0.072(2)	0.112(2)	0.070(2)	0.005(2)	−0.007(1)	−0.020(2)
C(12)	4e	0.1736(6)	0.2115(2)	0.3018(2)	0.087(2)	0.078(2)	0.078(2)	−0.000(2)	0.002(2)	−0.023(2)
C(13)	4e	0.3941(6)	0.2240(2)	0.3495(2)	0.103(2)	0.065(2)	0.091(2)	0.012(2)	−0.011(2)	−0.001(1)
C(14)	4e	0.4925(5)	0.3060(2)	0.3566(2)	0.079(2)	0.074(2)	0.077(2)	0.006(1)	−0.011(1)	−0.004(1)

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