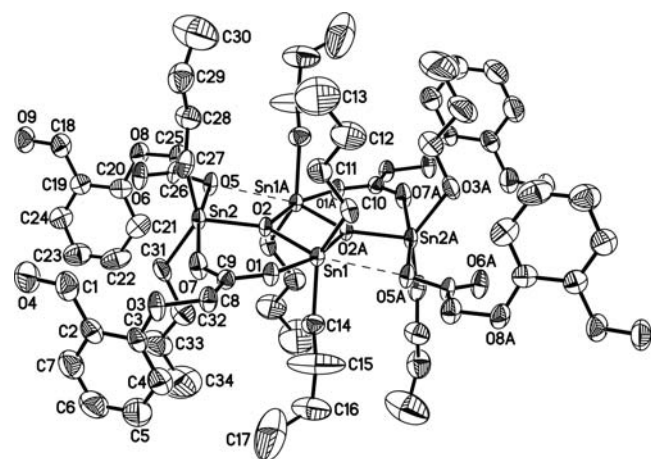


Crystal structure of bis(μ_3 -oxo)-bis[μ_2 -2-(2-formylphenoxy)acetato- O,O']-bis[μ_2 -2-(2-formylphenoxy)acetato- O,O']-octakis(*n*-butyl)tetratin(IV), $\text{Sn}_4\text{O}_2(\text{C}_9\text{H}_7\text{O}_4)_4(\text{C}_4\text{H}_9)_8$

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Abstract

$\text{C}_{68}\text{H}_{100}\text{O}_{18}\text{Sn}_4$, triclinic, $P\bar{1}$ (no. 2), $a = 11.680(7)$ Å, $b = 12.749(8)$ Å, $c = 13.991(9)$ Å, $\alpha = 100.44(1)^\circ$, $\beta = 105.20(1)^\circ$, $\gamma = 98.87(1)^\circ$, $V = 1932.0$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.066$, $wR_{\text{ref}}(F^2) = 0.200$, $T = 273$ K.

Source of material

The chemicals and solvents were procured from commercial sources and used as received without further purification. The title compound was prepared by reaction of di-*n*-butyltin oxide (0.248 g, 1.0 mmol) and 2-(2-formylphenoxy)acetic acid (0.180 g, 1.0 mmol) in absolute benzene (30 ml) under reflux for 7 h, then the solvent was evaporated under vacuum. The residue was recrystallized from dichloromethane/hexane to give fine crystals suitable for X-ray diffraction measurement.

Discussion

The title crystal structure possesses tetranuclear centrosymmetric dimeric structure with a four-membered planar Sn_2O_2 ring where $d(\text{Sn}—\text{O})$ varies from 2.041 to 2.291 Å, which are similar to the corresponding distances in the compounds of di-*n*-butyltin(IV) complexes reported [1]. The four carboxylate ligands are divided into two different types according to their coordinating mode. Two of them are bidentate and connect with each of *exo*- and *endo*-cyclic tin atoms by using both oxygen atoms, whereas the other two are bidentate with one oxygen atom bridging to both of the *exo*- and *endo*-cyclic tin atoms.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.15 × 0.21 × 0.25 mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	13.39 cm ^{−1}
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
$2\theta_{\text{max}}$:	50.1°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	9748, 6687
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 4634
$N(\text{param})_{\text{refined}}$:	410
Programs:	SHELXS-97, SHELXL-97 [2]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	2i	0.6555	0.4489	0.5887	0.139
H(4)	2i	0.6058	0.1641	0.7450	0.147
H(5)	2i	0.6925	0.2461	0.9164	0.178
H(6)	2i	0.7655	0.4261	0.9711	0.181
H(7)	2i	0.7701	0.5308	0.8562	0.161
H(8A)	2i	0.5170	0.1317	0.4763	0.084
H(8B)	2i	0.5385	0.1019	0.5827	0.084
H(10A)	2i	0.7906	−0.1660	0.3152	0.097
H(10B)	2i	0.6736	−0.1517	0.3462	0.097
H(11A)	2i	0.8094	0.0097	0.2824	0.131
H(11B)	2i	0.6890	0.0203	0.3098	0.131
H(12A)	2i	0.7006	−0.1386	0.1421	0.157
H(12B)	2i	0.5801	−0.1315	0.1704	0.157
H(13A)	2i	0.6254	0.0582	0.1475	0.265
H(13B)	2i	0.5582	−0.0378	0.0520	0.265
H(13C)	2i	0.6982	0.0071	0.0783	0.265
H(14A)	2i	0.9333	−0.1132	0.6624	0.126
H(14B)	2i	0.9207	0.0067	0.6933	0.126
H(15A)	2i	0.7397	−0.1744	0.6531	0.222
H(15B)	2i	0.7213	−0.0546	0.6777	0.222
H(16A)	2i	0.7867	−0.1688	0.8143	0.208
H(16B)	2i	0.9068	−0.0851	0.8293	0.208
H(17A)	2i	0.8349	0.0546	0.8841	0.313
H(17B)	2i	0.7732	−0.0351	0.9287	0.313
H(17C)	2i	0.6996	−0.0062	0.8295	0.313
H(18)	2i	0.5402	0.3132	0.4000	0.121
H(21)	2i	0.6310	0.5993	0.2548	0.125
H(22)	2i	0.5824	0.5178	0.0837	0.167
H(23)	2i	0.5266	0.3299	0.0258	0.161
H(24)	2i	0.4854	0.2276	0.1361	0.124
H(25A)	2i	0.6540	0.6328	0.5127	0.106
H(25B)	2i	0.6252	0.6587	0.4045	0.106
H(27A)	2i	1.1623	0.6422	0.5760	0.104
H(27B)	2i	1.1982	0.7666	0.6292	0.104
H(28A)	2i	1.0150	0.7465	0.6790	0.134
H(28B)	2i	0.9897	0.6204	0.6331	0.134
H(29A)	2i	1.1598	0.6034	0.7507	0.154
H(29B)	2i	1.2032	0.7301	0.7893	0.154

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(30A)	2i	1.0908	0.7402	0.8870	0.249
H(30B)	2i	1.1050	0.6196	0.8846	0.249
H(30C)	2i	0.9878	0.6431	0.8143	0.249
H(31A)	2i	1.0203	0.6086	0.2896	0.125
H(31B)	2i	0.8902	0.6287	0.2832	0.125
H(32A)	2i	1.0696	0.7625	0.2370	0.162

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(32B)	2i	0.9463	0.7944	0.2409	0.162
H(33A)	2i	0.9719	0.6181	0.0977	0.179
H(33B)	2i	0.8442	0.6319	0.1074	0.179
H(34A)	2i	0.8713	0.7964	0.0583	0.282
H(34B)	2i	0.8756	0.6991	−0.0262	0.282
H(34C)	2i	0.9964	0.7783	0.0448	0.282

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Sn(1)	2i	0.85879(5)	−0.04856(4)	0.49928(4)	0.0442(4)	0.0377(3)	0.0647(4)	0.0087(2)	0.0169(3)	0.0068(2)
Sn(2)	2i	1.03455(5)	0.74259(4)	0.45621(5)	0.0504(4)	0.0349(3)	0.0888(5)	0.0110(3)	0.0251(3)	0.0031(3)
O(1)	2i	0.7013(5)	0.0372(4)	0.5069(5)	0.057(4)	0.042(3)	0.084(4)	0.015(3)	0.015(3)	0.007(3)
O(2)	2i	0.9737(5)	0.0964(4)	0.5151(4)	0.052(3)	0.031(3)	0.070(4)	0.008(2)	0.019(3)	0.004(2)
O(3)	2i	0.5733(6)	0.2609(5)	0.5922(6)	0.061(4)	0.057(4)	0.114(6)	0.026(3)	0.013(4)	−0.004(4)
O(4)	2i	0.721(1)	0.5818(8)	0.6813(8)	0.19(1)	0.071(6)	0.149(9)	−0.020(7)	0.063(8)	−0.017(6)
O(5)	2i	0.8514(5)	0.7310(4)	0.4735(5)	0.052(3)	0.038(3)	0.111(5)	0.011(3)	0.028(3)	0.010(3)
O(6)	2i	0.8465(6)	0.5543(5)	0.4372(6)	0.063(4)	0.055(4)	0.143(6)	0.026(3)	0.024(4)	0.012(4)
O(7)	2i	1.2203(6)	0.7859(5)	0.4306(6)	0.060(4)	0.056(4)	0.149(7)	−0.002(3)	0.055(4)	−0.011(4)
O(8)	2i	0.6000(6)	0.4995(6)	0.3994(6)	0.063(4)	0.062(4)	0.117(6)	−0.014(3)	0.029(4)	0.009(4)
O(9)	2i	0.508(1)	0.1802(7)	0.3083(8)	0.140(8)	0.072(6)	0.130(7)	−0.002(6)	0.025(6)	−0.003(5)
C(1)	2i	0.684(1)	0.482(1)	0.658(1)	0.11(1)	0.09(1)	0.14(1)	0.011(9)	0.043(9)	0.000(9)
C(2)	2i	0.680(1)	0.412(1)	0.7292(9)	0.074(7)	0.092(9)	0.087(8)	0.017(6)	0.032(6)	−0.005(7)
C(3)	2i	0.632(1)	0.3019(9)	0.6958(9)	0.065(6)	0.075(7)	0.088(7)	0.026(6)	0.032(6)	0.006(6)
C(4)	2i	0.637(2)	0.239(1)	0.766(1)	0.13(1)	0.11(1)	0.13(1)	0.04(1)	0.05(1)	0.02(1)
C(5)	2i	0.689(2)	0.289(2)	0.869(1)	0.19(2)	0.15(2)	0.10(1)	0.07(2)	0.02(1)	0.02(1)
C(6)	2i	0.734(2)	0.395(2)	0.902(1)	0.18(2)	0.14(2)	0.10(1)	0.01(1)	0.05(1)	−0.02(1)
C(7)	2i	0.734(2)	0.457(2)	0.834(1)	0.15(2)	0.12(1)	0.13(1)	0.01(1)	0.06(1)	−0.00(1)
C(8)	2i	0.5706(8)	0.1505(7)	0.5456(8)	0.049(5)	0.052(6)	0.107(8)	0.014(4)	0.028(5)	0.007(5)
C(9)	2i	0.6942(8)	0.1348(7)	0.5446(7)	0.055(5)	0.055(5)	0.072(6)	0.017(5)	0.025(4)	0.015(4)
C(10)	2i	0.752(1)	−0.1129(8)	0.3466(7)	0.077(5)	0.073(4)	0.088(4)	0.019(4)	0.025(4)	0.002(3)
C(11)	2i	0.732(1)	−0.031(1)	0.2817(8)	0.111(6)	0.103(5)	0.114(4)	0.030(4)	0.031(4)	0.024(4)
C(12)	2i	0.657(2)	−0.089(1)	0.1708(9)	0.133(6)	0.138(6)	0.114(4)	0.027(5)	0.025(4)	0.032(4)
C(13)	2i	0.633(2)	−0.008(1)	0.106(1)	0.190(8)	0.179(7)	0.161(6)	0.041(5)	0.033(4)	0.067(5)
C(14)	2i	0.878(1)	−0.064(1)	0.6499(9)	0.103(5)	0.098(5)	0.108(5)	0.035(4)	0.014(4)	0.023(4)
C(15)	2i	0.780(2)	−0.100(2)	0.690(1)	0.178(7)	0.205(8)	0.180(6)	0.024(5)	0.065(5)	0.064(5)
C(16)	2i	0.819(2)	−0.098(2)	0.805(1)	0.175(7)	0.190(7)	0.165(6)	0.031(5)	0.066(5)	0.049(5)
C(17)	2i	0.778(2)	−0.014(2)	0.867(1)	0.212(8)	0.207(8)	0.207(7)	0.052(5)	0.065(5)	0.037(4)
C(18)	2i	0.531(1)	0.280(1)	0.333(1)	0.084(8)	0.076(8)	0.12(1)	−0.012(7)	0.020(7)	−0.001(7)
C(19)	2i	0.5446(9)	0.348(1)	0.263(1)	0.054(6)	0.095(9)	0.099(9)	0.008(6)	0.021(6)	0.023(7)
C(20)	2i	0.5837(9)	0.461(1)	0.2989(9)	0.051(6)	0.105(9)	0.076(7)	0.006(6)	0.012(5)	0.023(7)
C(21)	2i	0.602(1)	0.524(1)	0.232(1)	0.069(8)	0.12(1)	0.12(1)	0.015(7)	0.011(8)	0.039(9)
C(22)	2i	0.575(1)	0.475(2)	0.130(1)	0.11(1)	0.20(2)	0.11(1)	0.01(1)	−0.002(9)	0.08(1)
C(23)	2i	0.538(2)	0.363(2)	0.094(1)	0.12(1)	0.16(2)	0.11(1)	0.01(1)	0.02(1)	0.03(1)
C(24)	2i	0.518(1)	0.302(1)	0.160(1)	0.091(9)	0.12(1)	0.094(9)	0.017(8)	0.016(8)	0.018(8)
C(25)	2i	0.6625(9)	0.6119(9)	0.445(1)	0.060(6)	0.070(7)	0.13(1)	0.001(5)	0.036(7)	0.009(7)
C(26)	2i	0.7968(8)	0.6287(7)	0.4526(7)	0.049(5)	0.035(4)	0.093(7)	0.008(4)	0.014(5)	0.009(4)
C(27)	2i	1.130(1)	0.7063(9)	0.5930(8)	0.079(5)	0.073(5)	0.103(4)	0.028(4)	0.014(4)	0.016(4)
C(28)	2i	1.053(1)	0.685(1)	0.6654(9)	0.106(5)	0.111(6)	0.121(5)	0.035(4)	0.031(4)	0.024(4)
C(29)	2i	1.132(1)	0.671(1)	0.7643(9)	0.133(6)	0.127(6)	0.125(5)	0.032(5)	0.029(4)	0.040(5)
C(30)	2i	1.074(2)	0.668(2)	0.844(1)	0.174(7)	0.179(7)	0.153(5)	0.034(5)	0.057(5)	0.046(5)
C(31)	2i	0.975(1)	0.665(1)	0.3000(9)	0.100(5)	0.096(5)	0.112(4)	0.021(4)	0.040(4)	0.000(4)
C(32)	2i	0.984(2)	0.733(1)	0.227(1)	0.136(6)	0.131(6)	0.140(5)	0.027(5)	0.047(5)	0.027(4)
C(33)	2i	0.925(2)	0.671(1)	0.115(1)	0.153(7)	0.154(6)	0.137(5)	0.022(5)	0.045(5)	0.029(4)
C(34)	2i	0.916(2)	0.743(2)	0.042(1)	0.203(8)	0.191(8)	0.170(6)	0.033(5)	0.050(5)	0.057(5)

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References

1. Ma, C.; Han, Y.; Zhang, R.: Synthesis, characterizations and crystal structures of di-*n*-butyltin(IV) complexes with heteroatomic (N, O or S) acid. J. Organomet. Chem. **689** (2004) 1675–1683.
2. Sheldrick, G. M.: A short history of SHELX. Acta Crystallogr. **A64** (2008) 112–122.