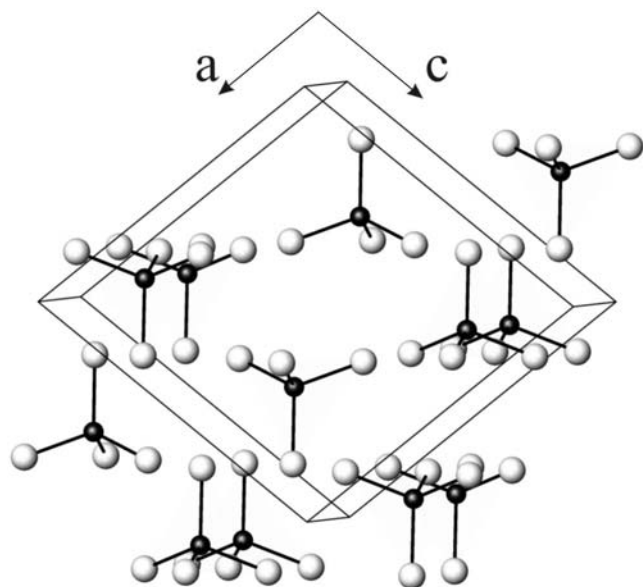


# Refinement of the crystal structure of $\beta$ -silicon tetrabromide, $\beta$ -SiBr<sub>4</sub>, at 271 K

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## Abstract

Br<sub>4</sub>Si, monoclinic,  $P12_1/c1$  (no. 14),  $a = 10.285(2)$  Å,  $b = 6.763(1)$  Å,  $c = 10.377(2)$  Å,  $\beta = 103.15(3)^\circ$ ,  $V = 702.8$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{\text{gt}}(F) = 0.037$ ,  $wR_{\text{ref}}(F^2) = 0.084$ ,  $T = 271$  K.

## Source of material

Liquid SiBr<sub>4</sub> (Aldrich, 99.995 %) was filled in capillaries with 0.2 mm diameter under argon. In the capillary colorless transparent single crystals of SiBr<sub>4</sub> were grown by rapidly freezing one end of the liquid and solidifying the remaining part slowly at 271 K on a STOE IPDS II diffractometer with a temperature control equipment.

## Discussion

SiBr<sub>4</sub> (m.p. 278 K) is known since more than 100 years [1,2]. In 1931 Pohland described that it exists in two modifications by polarization microscopy analysis [2]. That there might exist some more modifications of SiBr<sub>4</sub> could recently be shown by global minimization of the lattice energy using the program CRYSCA were found [3]. Our studies show that the cubic high-temperature  $\alpha$ -phase of SiBr<sub>4</sub> transforms into a monoclinic low-temperature  $\beta$ -phase at 168 K. In capillaries of small diameters crystals of  $\beta$ -SiBr<sub>4</sub> can form at even higher temperatures when cooling liquid SiBr<sub>4</sub>, which is possibly due to the preferred nucleation of  $\beta$ -SiBr<sub>4</sub> versus  $\alpha$ -SiBr<sub>4</sub> under such conditions.

The cubic  $\alpha$ -SiBr<sub>4</sub> [4] is isotypic to  $\alpha$ -GeBr<sub>4</sub> [5] and adopts the SnI<sub>4</sub> type structure. Monoclinic  $\beta$ -SiBr<sub>4</sub> crystallizes like  $\beta$ -GeBr<sub>4</sub> [6] in the SnBr<sub>4</sub> type structure [3]. Monoclinic  $\beta$ -SiBr<sub>4</sub> can be described in terms of a distorted hexagonal close packing of Br atoms, in which 1/8 of the tetrahedral holes are occupied by Si atoms. Each Si atom is surrounded by four crystallographically different Br atoms with Si—Br distances ranging from 2.165(2) to 2.169(2) Å and the Br—Si—Br angles lie between 109.18(9)° and 109.80(9)°. Each Br atom is surrounded by other 12 Br atoms as *hcp*. The three Br—Br distances at the edges of a SiBr<sub>4</sub> tetrahedron range from 3.53 to 3.55 Å, and the remaining nine Br—Br distances are slightly longer (3.93 to 4.55 Å).

**Table 1.** Data collection and handling.

|                                                         |                                                  |
|---------------------------------------------------------|--------------------------------------------------|
| Crystal:                                                | colorless cube, size 0.2 × 0.2 × 0.2 mm          |
| Wavelength:                                             | Ag K $\alpha$ radiation (0.56086 Å)              |
| $\mu$ :                                                 | 122.70 cm <sup>−1</sup>                          |
| Diffractometer, scan mode:                              | STOE IPDS II image plate, $\omega$               |
| $2\theta_{\text{max}}$ :                                | 39.98°                                           |
| $N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ : | 5931, 128                                        |
| Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ : | $I_{\text{obs}} > 2\sigma(I_{\text{obs}})$ , 825 |
| $N(\text{param})_{\text{refined}}$ :                    | 47                                               |
| Programs:                                               | SHELXS-97, SHELXL-97 [7]                         |

**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

| Atom  | Site | <i>x</i>   | <i>y</i>  | <i>z</i>   | <i>U</i> <sub>11</sub> | <i>U</i> <sub>22</sub> | <i>U</i> <sub>33</sub> | <i>U</i> <sub>12</sub> | <i>U</i> <sub>13</sub> | <i>U</i> <sub>23</sub> |
|-------|------|------------|-----------|------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|
| Br(1) | 4e   | 0.6938(1)  | 0.1265(1) | 0.0924(1)  | 0.0806(7)              | 0.0447(4)              | 0.0841(7)              | −0.0106(4)             | 0.0148(5)              | −0.0103(4)             |
| Br(2) | 4e   | 0.58636(9) | 0.5803(1) | 0.2009(1)  | 0.0545(6)              | 0.0812(7)              | 0.0866(7)              | 0.0134(4)              | 0.0159(5)              | −0.0190(5)             |
| Br(3) | 4e   | 0.7995(1)  | 0.5781(2) | −0.0234(1) | 0.0868(7)              | 0.0813(7)              | 0.0588(6)              | −0.0148(5)             | 0.0164(5)              | 0.0156(5)              |
| Br(4) | 4e   | 0.92379(8) | 0.4347(1) | 0.3092(1)  | 0.0539(5)              | 0.0776(6)              | 0.0640(6)              | 0.0004(4)              | −0.0084(4)             | 0.0031(5)              |
| Si(1) | 4e   | 0.7509(2)  | 0.4294(3) | 0.1443(2)  | 0.043(1)               | 0.040(1)               | 0.046(1)               | −0.0007(8)             | 0.0072(9)              | −0.0002(9)             |

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