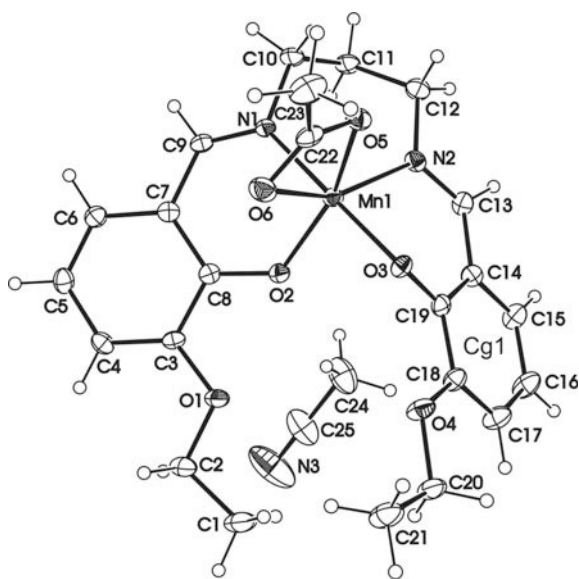


Crystal structure of acetato[*N,N'*-bis(3-ethoxysalicylidene)-1,3-diiminato]manganese(III) — acetonitrile (1:1), $\text{Mn}(\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4)(\text{CH}_3\text{COO}) \cdot \text{CH}_3\text{CN}$

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Abstract

$\text{C}_{25}\text{H}_{30}\text{MnN}_3\text{O}_6$, monoclinic, $P2_1/n$ (no. 14), $a = 9.8702(6)$ Å, $b = 12.0550(7)$ Å, $c = 22.083(1)$ Å, $\beta = 102.628(1)^\circ$, $V = 2564.0$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.050$, $wR_{\text{ref}}(F^2) = 0.140$, $T = 200$ K.

Source of material

$\text{Mn}(\text{CH}_3\text{COO})_3 \cdot 2\text{H}_2\text{O}$ (0.50 g, 1.86 mmol), NaCl (0.11 g, 1.88 mmol) and *N,N'*-bis(3-ethoxysalicylidene)-1,3-diiminopropane (0.70 g, 1.89 mmol) in EtOH (70 ml) and H_2O (5 ml) were stirred for 3 h at room temperature and then filtered. The solvent was evaporated in vacuo, the residue washed with acetone/ H_2O and dried, to give a dark brown powder (0.93 g). Crystals suitable for X-ray diffraction analysis were obtained by slow evaporation from a CH_3CN solution.

Experimental details

Hydrogen atoms were positioned geometrically and allowed to ride on their respective carrier atoms [$d(\text{C}—\text{H}) = 0.95$ Å (CH), 0.99 Å (CH_2) or 0.98 Å (CH_3) and $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{CH}, \text{CH}_2)$ or $1.5 U_{\text{eq}}(\text{CH}_3)$].

Discussion

The crystal structure of the title compound consists of a neutral complex $[\text{Mn}(\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4)(\text{CH}_3\text{COO})]$ and an acetonitrile solvent molecule. Mn(III) ion is six-coordinated by two N and two O

atoms from the *N,N'*-bis(3-ethoxysalicylidene)-1,3-diiminopropane dianionic ligand and two O atoms of the acetate anion in a considerably distorted octahedral environment. In the octahedral structures of metal complexes of salen or salen-like ligands (where salen is dianion of *N,N'*-bis(salicylidene)ethylenediamine), the equatorial planes are generally defined by the two N and two O atoms of the dianion [1,2]. However, in this complex, the four atoms (N1, N2, O2 and O3) are not coplanar, with the torsion angle $\angle\text{N1}—\text{N2}—\text{O3}—\text{O2} = 40.80(8)^\circ$. The N2, O2, O5 and O6 atoms lie on the equatorial plane ($\angle\text{O5}—\text{O6}—\text{O2}—\text{N2} = 0.70(9)^\circ$). Within the plane, the $\angle\text{O5}—\text{Mn1}—\text{O6}$ chelate angle is $58.21(8)^\circ$, which results in the distortion of the octahedral structure. The apical bond angle $\angle\text{N1}—\text{Mn1}—\text{O3}$ is $176.09(9)^\circ$. While the Mn—N and Mn—O_{phenoxo} bond lengths are nearly equivalent, respectively ($d(\text{Mn}—\text{N}) = 1.998(2)$ Å and $2.160(2)$ Å; $d(\text{Mn}—\text{O}) = 1.891(2)$ Å and $1.889(2)$ Å), the two Mn—O_{acetato} bond lengths are somewhat different ($2.043(2)$ Å and $2.402(2)$ Å). The compound displays the intermolecular C—H $\cdots$ O hydrogen bonds with $d(\text{C}\cdots\text{O}) = 3.266(4)$ Å - $3.411(3)$ Å. Moreover, there are weak intermolecular π - π interactions between the adjacent benzene rings. For Cg1 (the centroid of 6-membered ring C14–C19) and Cg1ⁱ (symmetry code i: $-x, 1-y, -z$), the centroid-centroid distance is $5.254(2)$ Å and the ring planes are parallel and shifted for 3.902 Å.

Table 1. Data collection and handling.

Crystal:	brown rod, size $0.17 \times 0.25 \times 0.26$ mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	5.59 cm^{-1}
Diffractometer, scan mode:	Bruker SMART 1000 CCD, ϕ/ω
$2\theta_{\text{max}}$:	56.64°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	18767, 6349
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3746
$N(\text{param})_{\text{refined}}$:	320
Programs:	SHELXS-97 [3], SHELXL-97 [4], ORTEP [5], PLATON [6]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(1A)	4e	0.5884	0.3449	0.0580	0.075
H(1B)	4e	0.7152	0.4237	0.0889	0.075
H(1C)	4e	0.5649	0.4437	0.1030	0.075
H(2A)	4e	0.7290	0.2511	0.1418	0.043
H(2B)	4e	0.7130	0.3519	0.1874	0.043
H(4)	4e	0.7531	0.1934	0.2434	0.034

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(5)	4e	0.7478	0.0741	0.3269	0.038
H(6)	4e	0.5401	0.0023	0.3409	0.035
H(9)	4e	0.2969	−0.0285	0.3040	0.029
H(10A)	4e	−0.0206	0.0313	0.2709	0.038
H(10B)	4e	0.0857	−0.0701	0.2884	0.038
H(11A)	4e	0.0558	−0.1037	0.1774	0.039
H(11B)	4e	−0.0753	−0.1338	0.2058	0.039
H(12A)	4e	−0.1535	−0.0416	0.1127	0.042
H(12B)	4e	−0.1821	0.0353	0.1676	0.042
H(13)	4e	−0.1166	0.0670	0.0432	0.035
H(15)	4e	−0.0540	0.1380	−0.0466	0.042
H(16)	4e	0.0388	0.2758	−0.0980	0.051

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(17)	4e	0.1535	0.4255	−0.0430	0.047
H(20A)	4e	0.1785	0.5857	0.0135	0.046
H(20B)	4e	0.3209	0.5166	0.0246	0.046
H(21A)	4e	0.2510	0.6638	0.1139	0.076
H(21B)	4e	0.3571	0.6942	0.0710	0.076
H(21C)	4e	0.3951	0.5986	0.1222	0.076
H(23A)	4e	0.0681	0.3868	0.3622	0.074
H(23B)	4e	−0.0715	0.3161	0.3412	0.074
H(23C)	4e	−0.0537	0.4308	0.3073	0.074
H(24A)	4e	0.2973	0.1612	−0.0001	0.098
H(24B)	4e	0.3582	0.1779	0.0727	0.098
H(24C)	4e	0.3300	0.0556	0.0446	0.098

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Mn(1)	4e	0.12211(5)	0.19081(4)	0.19214(2)	0.0235(3)	0.0214(3)	0.0264(3)	−0.0012(2)	0.0072(2)	0.0028(2)
O(1)	4e	0.5380(2)	0.2640(2)	0.15879(9)	0.031(1)	0.034(1)	0.035(1)	−0.005(1)	0.014(1)	0.006(1)
O(2)	4e	0.3018(2)	0.1817(2)	0.17550(9)	0.025(1)	0.035(1)	0.027(1)	0.0016(9)	0.0078(9)	0.007(1)
O(3)	4e	0.0804(2)	0.3091(2)	0.13505(9)	0.033(1)	0.024(1)	0.020(1)	0.0003(9)	0.0039(9)	0.0002(9)
O(4)	4e	0.1913(2)	0.4654(2)	0.07715(9)	0.053(2)	0.026(1)	0.027(1)	−0.010(1)	0.011(1)	0.002(1)
O(5)	4e	−0.0199(2)	0.2416(2)	0.24112(9)	0.028(1)	0.032(1)	0.036(1)	0.003(1)	0.014(1)	−0.001(1)
O(6)	4e	0.1845(2)	0.3118(2)	0.2803(1)	0.035(1)	0.032(1)	0.046(1)	−0.007(1)	0.009(1)	0.002(1)
N(1)	4e	0.1651(2)	0.0586(2)	0.2476(1)	0.024(1)	0.021(1)	0.025(1)	−0.002(1)	0.006(1)	0.004(1)
N(2)	4e	−0.0291(2)	0.0939(2)	0.1282(1)	0.025(1)	0.021(1)	0.034(2)	−0.003(1)	0.005(1)	0.000(1)
C(1)	4e	0.6305(4)	0.3874(3)	0.0952(2)	0.065(3)	0.048(2)	0.040(2)	−0.024(2)	0.019(2)	0.004(2)
C(2)	4e	0.6661(3)	0.3106(3)	0.1499(2)	0.036(2)	0.038(2)	0.038(2)	−0.008(2)	0.017(2)	0.001(2)
C(3)	4e	0.5463(3)	0.1937(2)	0.2085(1)	0.031(2)	0.013(1)	0.026(2)	−0.002(1)	0.011(1)	−0.001(1)
C(4)	4e	0.6670(3)	0.1650(2)	0.2490(1)	0.022(2)	0.028(2)	0.037(2)	0.001(1)	0.009(1)	−0.004(1)
C(5)	4e	0.6637(3)	0.0936(2)	0.2989(1)	0.030(2)	0.028(2)	0.035(2)	0.006(1)	0.000(1)	0.000(2)
C(6)	4e	0.5409(3)	0.0519(2)	0.3075(1)	0.032(2)	0.026(2)	0.029(2)	0.001(1)	0.007(1)	0.002(1)
C(7)	4e	0.4153(3)	0.0821(2)	0.2671(1)	0.028(2)	0.022(2)	0.027(2)	0.002(1)	0.010(1)	−0.003(1)
C(8)	4e	0.4156(3)	0.1535(2)	0.2163(1)	0.027(2)	0.021(2)	0.026(2)	0.004(1)	0.009(1)	−0.001(1)
C(9)	4e	0.2882(3)	0.0311(2)	0.2753(1)	0.032(2)	0.020(2)	0.024(2)	0.001(1)	0.012(1)	−0.000(1)
C(10)	4e	0.0497(3)	−0.0139(2)	0.2563(1)	0.031(2)	0.027(2)	0.037(2)	−0.007(1)	0.010(1)	0.008(1)
C(11)	4e	−0.0179(3)	−0.0717(2)	0.1962(1)	0.037(2)	0.023(2)	0.040(2)	−0.009(1)	0.011(2)	0.005(1)
C(12)	4e	−0.1082(3)	0.0032(2)	0.1491(2)	0.034(2)	0.031(2)	0.040(2)	−0.009(1)	0.007(2)	0.001(2)
C(13)	4e	−0.0532(3)	0.1145(2)	0.0698(1)	0.033(2)	0.024(2)	0.028(2)	−0.001(1)	0.002(1)	−0.004(1)
C(14)	4e	0.0073(3)	0.2034(2)	0.0408(1)	0.029(2)	0.023(2)	0.031(2)	0.001(1)	0.006(1)	−0.001(1)
C(15)	4e	−0.0065(3)	0.1987(3)	−0.0239(1)	0.048(2)	0.026(2)	0.028(2)	−0.002(2)	0.003(2)	−0.004(2)
C(16)	4e	0.0475(4)	0.2802(3)	−0.0544(2)	0.068(3)	0.037(2)	0.021(2)	−0.004(2)	0.008(2)	−0.003(2)
C(17)	4e	0.1150(4)	0.3695(3)	−0.0216(1)	0.058(2)	0.036(2)	0.026(2)	−0.006(2)	0.013(2)	0.001(2)
C(18)	4e	0.1272(3)	0.3786(2)	0.0416(1)	0.039(2)	0.022(2)	0.026(2)	0.001(1)	0.006(1)	0.001(1)
C(19)	4e	0.0732(3)	0.2953(2)	0.0748(1)	0.024(2)	0.023(2)	0.028(2)	0.006(1)	0.005(1)	0.003(1)
C(20)	4e	0.2515(4)	0.5498(3)	0.0454(2)	0.050(2)	0.032(2)	0.035(2)	−0.010(2)	0.011(2)	0.009(2)
C(21)	4e	0.3196(4)	0.6338(3)	0.0922(2)	0.075(3)	0.036(2)	0.039(2)	−0.022(2)	0.009(2)	0.003(2)
C(22)	4e	0.0579(4)	0.3035(2)	0.2803(2)	0.042(2)	0.020(2)	0.034(2)	0.005(2)	0.014(2)	0.010(2)
C(23)	4e	−0.0053(4)	0.3647(3)	0.3268(2)	0.073(3)	0.041(2)	0.042(2)	0.010(2)	0.029(2)	−0.002(2)
N(3)	4e	0.6087(5)	0.1237(4)	0.0183(3)	0.061(3)	0.084(3)	0.202(5)	−0.015(2)	0.049(3)	−0.062(3)
C(24)	4e	0.3601(4)	0.1307(4)	0.0367(2)	0.045(3)	0.078(3)	0.070(3)	0.003(2)	0.006(2)	−0.030(2)
C(25)	4e	0.4995(5)	0.1267(3)	0.0265(2)	0.046(3)	0.052(3)	0.094(4)	−0.001(2)	0.011(3)	−0.034(2)

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