

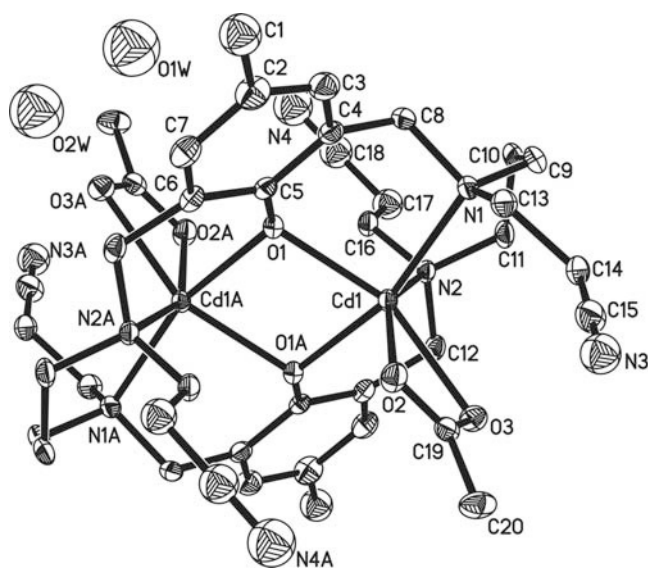
Crystal structure of [μ -11,23-dimethyl-3,7,15,19-tetraazatricyclo-[19.3.1.1^{9,13}]-3,7,15,19-tetracyanomethyltetracos-1(24),-9(25),10,12,21,23-hexaene-25,26-diolato- $\kappa^8 N^3, N^7, O^{25}, O^{26}; N^{15}, N^{19}, O^{25}, O^{26}$]-bis[(acetato- $\kappa^2 O, O'$)cadmium(II)] hydrate, $Cd_2(C_{36}H_{46}N_8O_2)(CH_3COO)_2 \cdot 0.8H_2O$

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Abstract

$C_{40}H_{53.6}Cd_2N_8O_{6.80}$, trigonal, $R\bar{3}$ (no. 148), $a = 31.46(1)$ Å, $c = 12.33(3)$ Å, $V = 10568$ Å³, $Z = 9$, $R_{gt}(F) = 0.045$, $wR_{ref}(F^2) = 0.148$, $T = 293$ K.

Source of material

Tetraaminodiphenol macrocyclic ligand was refluxed in acrylonitrile to give the H_2L [1]. A mixture of H_2L (0.0632 g, 0.1 mmol) and $Cd(CH_3COO)_2$ (0.05 g, 0.2 mmol) in methanol (10 ml) and $CHCl_3$ (10 ml) was stirred for 10 min and white precipitate was obtained. The precipitate was then dissolved in a minimum amount of ammonia solution. The resulting solution was filtered. Colorless single crystals of the title compound were obtained by slow evaporation of the filtrate at room temperature (yield 56 %). Chemical analysis — found: C, 49.11 %; H, 5.56 %; N, 11.51 %; calculated for $C_{40}H_{53.6}Cd_2N_8O_{6.8}$: C, 49.02 %; H, 5.51 %; N, 11.43 %.

Experimental details

All H atoms on C atoms were generated geometrically and refined as riding atoms with $d(C-H) = 0.93$ Å and $U_{iso}(H) = 1.2 U_{eq}(C)$. The water H atoms were not located from a difference Fourier map.

The terminal group atoms C1, C17, C18, N3 and N4 were refined isotropically.

Discussion

Crown ether compounds have attracted much interest as a result of their significance in biological transport systems [2]. To further widen the scope of applications of crown ethers, a variety of armed crown ether derivatives were prepared.

The crystal structure of the title compound contains a dinuclear cadmium complex. Each of the two Cd(II) atoms has a distorted octahedral environment, defined by two N atoms and two O atoms from the macrocyclic ligand L^{2-} and two O atoms from a chelating acetate anion. Atoms Cd1, O1, O1A, Cd1A form a distorted square plane in which the two metal centres are separated by 3.51(1) Å.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.17 × 0.20 × 0.25 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	9.57 cm ⁻¹
Diffractometer, scan mode:	Oxford Diffraction Gemini R Ultra, ω
$2\theta_{max}$:	58.76°
$N(hkl)_{measured}$, $N(hkl)_{unique}$:	27271, 5850
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 4178
$N(param)_{refined}$:	236
Programs:	SHELXS-97 [3], SHELXL-97 [4]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
C(1)	18f	0.5918(3)	0.2517(3)	0.5365(6)	0.112(2)
H(1A)	18f	0.5775	0.2568	0.6005	0.168
H(1B)	18f	0.5844	0.2656	0.4749	0.168
H(1C)	18f	0.6267	0.2672	0.5455	0.168
H(3)	18f	0.6064	0.1996	0.3855	0.081
H(8A)	18f	0.5922	0.1371	0.2668	0.065
H(8B)	18f	0.5463	0.0851	0.2838	0.065
H(9A)	18f	0.6503	0.0722	0.2423	0.078
H(9B)	18f	0.6340	0.1059	0.1847	0.078
H(10A)	18f	0.5531	0.0356	0.1694	0.073
H(10B)	18f	0.5929	0.0337	0.0969	0.073
H(11A)	18f	0.5579	-0.0401	0.1606	0.068
H(11B)	18f	0.5988	-0.0141	0.2481	0.068
H(12A)	18f	0.5263	-0.1015	0.2868	0.063
H(12B)	18f	0.5588	-0.0722	0.3855	0.063
H(13A)	18f	0.6449	0.1340	0.4616	0.071
H(13B)	18f	0.6682	0.1565	0.3486	0.071

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(01A)	18 <i>f</i>	0.6681	0.0611	0.7281	0.116
H(01B)	18 <i>f</i>	0.6947	0.0446	0.6448	0.116
H(01C)	18 <i>f</i>	0.6526	0.0055	0.7183	0.116
H(14A)	18 <i>f</i>	0.6758	0.0759	0.4218	0.073
H(14B)	18 <i>f</i>	0.7117	0.1176	0.3428	0.073
H(16A)	18 <i>f</i>	0.4874	−0.0122	0.2460	0.061
H(16B)	18 <i>f</i>	0.4606	−0.0563	0.3262	0.061

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
C(17)	18 <i>f</i>	0.4624(2)	−0.0787(2)	0.1708(5)	0.089(2)
H(17A)	18 <i>f</i>	0.4869	−0.0688	0.1140	0.107
H(17B)	18 <i>f</i>	0.4534	−0.1116	0.1937	0.107
C(18)	18 <i>f</i>	0.4199(3)	−0.0783(3)	0.1298(6)	0.106(2)
N(3)	18 <i>f</i>	0.7497(2)	0.1567(2)	0.5681(5)	0.107(2)
N(4)	18 <i>f</i>	0.3873(3)	−0.0725(3)	0.1003(7)	0.154(3)
H(7)	18 <i>f</i>	0.5264	0.1754	0.6495	0.079

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
C(2)	18 <i>f</i>		0.5705(2)	0.1966(2)	0.5185(5)	0.072(3)	0.048(3)	0.100(4)	0.027(2)	−0.005(3)	−0.005(3)
C(3)	18 <i>f</i>		0.5829(2)	0.1779(2)	0.4339(5)	0.060(3)	0.056(3)	0.085(4)	0.027(2)	0.004(2)	0.015(2)
C(4)	18 <i>f</i>		0.5625(1)	0.1282(2)	0.4154(3)	0.042(2)	0.051(2)	0.058(2)	0.026(2)	−0.002(2)	0.008(2)
C(5)	18 <i>f</i>		0.5266(1)	0.0944(1)	0.4871(3)	0.037(2)	0.048(2)	0.045(2)	0.025(2)	−0.006(2)	0.001(2)
C(8)	18 <i>f</i>		0.5762(2)	0.1101(2)	0.3178(3)	0.049(2)	0.068(3)	0.048(2)	0.032(2)	0.007(2)	0.021(2)
C(9)	18 <i>f</i>		0.6228(2)	0.0774(2)	0.2314(4)	0.062(3)	0.082(3)	0.056(3)	0.040(3)	0.026(2)	0.023(2)
C(10)	18 <i>f</i>		0.5824(2)	0.0328(2)	0.1713(3)	0.060(3)	0.095(3)	0.035(2)	0.045(3)	0.016(2)	0.009(2)
C(11)	18 <i>f</i>		0.5692(2)	−0.0163(2)	0.2190(3)	0.052(2)	0.086(3)	0.042(2)	0.041(2)	0.009(2)	−0.006(2)
C(19)	18 <i>f</i>		0.6271(1)	0.0288(2)	0.5945(3)	0.035(2)	0.060(3)	0.044(2)	0.015(2)	−0.002(2)	0.009(2)
C(12)	18 <i>f</i>		0.5282(2)	−0.0814(2)	0.3484(3)	0.056(2)	0.067(3)	0.048(2)	0.041(2)	−0.001(2)	−0.015(2)
C(13)	18 <i>f</i>		0.6539(2)	0.1267(2)	0.3916(4)	0.038(2)	0.058(3)	0.070(3)	0.015(2)	0.003(2)	0.014(2)
C(20)	18 <i>f</i>		0.6639(2)	0.0356(2)	0.6789(4)	0.064(3)	0.076(3)	0.075(3)	0.022(3)	−0.028(3)	0.008(3)
C(14)	18 <i>f</i>		0.6924(2)	0.1110(2)	0.4085(4)	0.038(2)	0.068(3)	0.071(3)	0.022(2)	0.012(2)	0.016(2)
C(15)	18 <i>f</i>		0.7245(2)	0.1362(2)	0.4976(4)	0.048(3)	0.091(4)	0.071(3)	0.021(3)	−0.001(2)	0.012(3)
C(16)	18 <i>f</i>		0.4838(1)	−0.0435(2)	0.2667(3)	0.042(2)	0.082(3)	0.036(2)	0.036(2)	−0.003(2)	−0.007(2)
N(1)	18 <i>f</i>		0.6093(1)	0.0891(1)	0.3372(3)	0.037(2)	0.060(2)	0.045(2)	0.023(2)	0.006(1)	0.012(2)
N(2)	18 <i>f</i>		0.5315(1)	−0.0354(1)	0.3053(2)	0.039(2)	0.066(2)	0.032(2)	0.033(2)	0.001(1)	−0.004(1)
O(1)	18 <i>f</i>		0.50646(9)	0.04667(9)	0.4673(2)	0.037(1)	0.044(1)	0.041(1)	0.021(1)	0.003(1)	−0.002(1)
O(2)	18 <i>f</i>		0.6083(1)	0.0562(1)	0.5978(2)	0.053(2)	0.081(2)	0.052(2)	0.037(2)	−0.008(1)	−0.012(2)
O(3)	18 <i>f</i>		0.6162(1)	−0.0022(1)	0.5225(2)	0.056(2)	0.061(2)	0.054(2)	0.030(2)	−0.010(1)	−0.005(1)
Cd(1)	18 <i>f</i>		0.558896(9)	0.01850(1)	0.44930(2)	0.0313(2)	0.0494(2)	0.0326(2)	0.0204(1)	0.00082(9)	−0.0002(1)
C(6)	18 <i>f</i>		0.5128(2)	0.1126(2)	0.5765(3)	0.047(2)	0.054(2)	0.053(2)	0.032(2)	−0.006(2)	−0.007(2)
C(7)	18 <i>f</i>		0.5352(2)	0.1633(2)	0.5901(4)	0.069(3)	0.061(3)	0.077(3)	0.040(3)	−0.011(3)	−0.015(2)
O(1W)	18 <i>f</i>	0.25	0.512(1)	0.341(1)	0.699(3)	0.18(1)	0.18(1)	0.18(1)	0.090(5)	−0.000(1)	−0.000(1)
O(2W)	18 <i>f</i>	0.15	0.571(2)	0.346(2)	0.815(4)	0.15(1)	0.15(1)	0.15(1)	0.075(6)	0.000(1)	0.000(1)

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