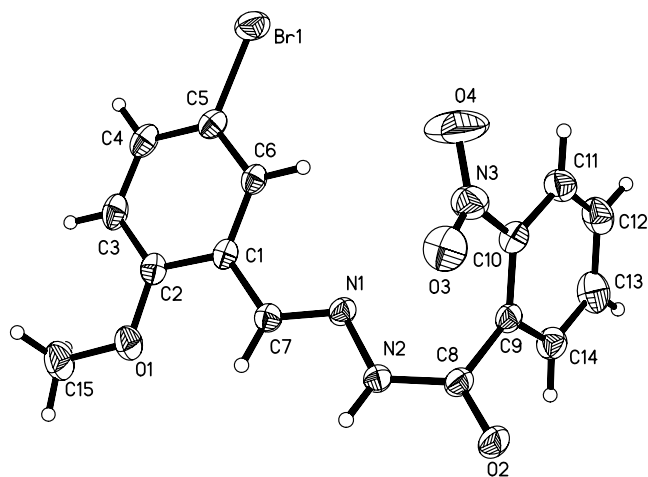


Crystal structure of 2-nitro-*N'*-(5-bromo-2-methoxybenzylidene)-benzohydrazide, C₁₅H₁₂BrN₃O₄

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Abstract

C₁₅H₁₂BrN₃O₄, monoclinic, *P*12₁/*c*1 (no. 14), *a* = 11.3077(3) Å, *b* = 7.4907(2) Å, *c* = 18.3286(5) Å, β = 90.124(1)°, *V* = 1552.5 Å³, *Z* = 4, *R*_g(*F*) = 0.037, *wR*_{ref}(*F*²) = 0.102, *T* = 298 K.

Source of material

Both starting materials of AR grade were purchased from Aldrich and were used as obtained. 5-Bromo-2-methoxybenzaldehyde (215.1 mg, 1.0 mmol) and 2-nitrobenzohydrazide (181.2 mg, 1.0 mmol) were dissolved in a methanol solution (50 ml). The mixture was stirred for 30 min at room temperature. The resulting solution was left in air for a few days, yielding colorless block-shaped crystals.

Experimental details

H2 atom was located in a difference map and refined with N—H distance restrained to 0.90(1) Å. The remaining H atoms were positioned geometrically [*d*(C—H) = 0.93–0.96 Å] and refined using a riding model, with *U*_{iso}(H) = 1.2 *U*_{eq}(C) and 1.5 *U*_{eq}(O). A rotating group model was used for the methyl group.

Discussion

Schiff bases are a kind of versatile compounds, which possess excellent biological properties [1–3]. Recently, a large number of Schiff bases derived from the reaction of aldehydes with benzohydrazides have been reported [4–6].

In the crystal structure of the title compound, all the bond lengths are comparable with those observed in other similar compounds [4–8]. The dihedral angle between two substituted benzene rings

is 73.0(2)°. The dihedral angle between the O3/N3/O4 plane and the C9–C14 benzene ring is 20.2(2)°. The C15 atom lies in the least-squares plane of C1–C6, with mean deviation from the plane of 0.001(2) Å. The torsion angles of C7–N1–N2–C8 and N1–N2–C8–C9 are 1.9(2) and 11.5(2)°, respectively. The crystal structure is stabilized by intermolecular N—H⋯O hydrogen bonds.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.28 × 0.28 × 0.30 mm
Wavelength:	Mo <i>K</i> _α radiation (0.71073 Å)
μ :	26.72 cm ^{−1}
Diffractometer, scan mode:	Bruker SMART 1000 CCD, ω
2 θ _{max} :	54°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	9004, 3371
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 2326
<i>N</i> (<i>param</i>) _{refined} :	212
Programs:	SHELXS-97 [9], SHELXL-97 [10], SHELXTL [11]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(3)	4e	−0.1658	0.2121	−0.1173	0.075
H(4)	4e	−0.2507	0.3247	−0.0124	0.074
H(6)	4e	0.0612	0.3034	0.0893	0.056
H(7)	4e	0.2217	0.0929	−0.0533	0.056
H(11)	4e	0.2996	0.5027	0.2658	0.076
H(12)	4e	0.3969	0.7608	0.2381	0.087
H(13)	4e	0.5218	0.7750	0.1392	0.094
H(14)	4e	0.5526	0.5265	0.0682	0.071
H(15A)	4e	−0.0712	0.0072	−0.1914	0.134
H(15B)	4e	0.0437	0.0405	−0.2372	0.134
H(15C)	4e	−0.0396	0.2006	−0.2175	0.134
H(2)	4e	0.395(3)	0.066(3)	−0.005(1)	0.080

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Br(1)	4e	−0.17214(3)	0.41529(5)	0.13373(2)	0.0509(2)	0.0949(3)	0.0716(2)	0.0093(2)	0.0081(1)	0.0009(2)
O(1)	4e	0.0572(2)	0.1179(3)	−0.1353(1)	0.068(1)	0.083(2)	0.051(1)	−0.008(1)	−0.013(1)	−0.008(1)
O(2)	4e	0.5630(2)	0.1377(3)	0.0703(1)	0.037(1)	0.077(1)	0.091(2)	0.013(1)	−0.007(1)	−0.031(1)
O(3)	4e	0.3406(3)	0.0567(3)	0.1784(2)	0.099(2)	0.054(1)	0.097(2)	−0.008(1)	0.003(2)	0.010(1)
O(4)	4e	0.2187(4)	0.2088(5)	0.2390(2)	0.161(3)	0.133(3)	0.192(4)	−0.059(3)	0.125(3)	−0.047(3)
N(1)	4e	0.2653(2)	0.2213(3)	0.0326(1)	0.036(1)	0.048(1)	0.054(1)	0.005(1)	−0.0030(9)	−0.004(1)
N(2)	4e	0.3791(2)	0.1547(3)	0.0263(1)	0.037(1)	0.057(1)	0.057(1)	0.007(1)	−0.002(1)	−0.016(1)
N(3)	4e	0.3055(3)	0.1969(4)	0.2006(2)	0.074(2)	0.068(2)	0.068(2)	−0.012(2)	0.014(1)	−0.004(1)
C(1)	4e	0.0693(2)	0.2147(3)	−0.0146(1)	0.041(1)	0.044(1)	0.051(2)	−0.005(1)	−0.007(1)	0.005(1)
C(2)	4e	−0.0009(2)	0.1870(4)	−0.0768(2)	0.053(2)	0.049(2)	0.051(2)	−0.008(1)	−0.008(1)	0.005(1)
C(3)	4e	−0.1198(3)	0.2292(4)	−0.0758(2)	0.053(2)	0.072(2)	0.062(2)	−0.010(2)	−0.024(1)	0.008(2)
C(4)	4e	−0.1706(3)	0.2969(4)	−0.0131(2)	0.040(2)	0.068(2)	0.077(2)	−0.003(1)	−0.012(1)	0.010(2)
C(5)	4e	−0.1024(2)	0.3230(4)	0.0481(2)	0.043(1)	0.053(2)	0.059(2)	−0.000(1)	−0.002(1)	0.008(1)
C(6)	4e	0.0160(2)	0.2837(4)	0.0476(1)	0.039(1)	0.051(2)	0.051(2)	−0.002(1)	−0.009(1)	0.006(1)
C(7)	4e	0.1938(2)	0.1659(4)	−0.0160(1)	0.045(1)	0.049(2)	0.047(1)	0.000(1)	−0.002(1)	−0.002(1)
C(8)	4e	0.4646(2)	0.2049(4)	0.0723(2)	0.037(1)	0.051(2)	0.060(2)	0.002(1)	0.000(1)	−0.008(1)
C(9)	4e	0.4407(2)	0.3612(4)	0.1220(1)	0.036(1)	0.051(2)	0.052(2)	0.007(1)	−0.007(1)	−0.003(1)
C(10)	4e	0.3664(2)	0.3598(4)	0.1808(2)	0.046(2)	0.052(2)	0.059(2)	−0.001(1)	−0.004(1)	0.002(1)
C(11)	4e	0.3496(3)	0.5084(5)	0.2256(2)	0.062(2)	0.064(2)	0.063(2)	0.016(2)	−0.003(2)	−0.013(2)
C(12)	4e	0.4077(3)	0.6602(5)	0.2092(2)	0.083(2)	0.059(2)	0.076(2)	0.013(2)	−0.014(2)	−0.020(2)
C(13)	4e	0.4830(4)	0.6689(5)	0.1501(2)	0.098(3)	0.052(2)	0.084(2)	−0.014(2)	−0.015(2)	−0.001(2)
C(14)	4e	0.5005(3)	0.5208(4)	0.1073(2)	0.057(2)	0.060(2)	0.061(2)	−0.009(2)	−0.000(1)	0.003(2)
C(15)	4e	−0.0077(4)	0.0892(6)	−0.2007(2)	0.094(3)	0.116(3)	0.058(2)	−0.019(2)	−0.018(2)	−0.014(2)

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