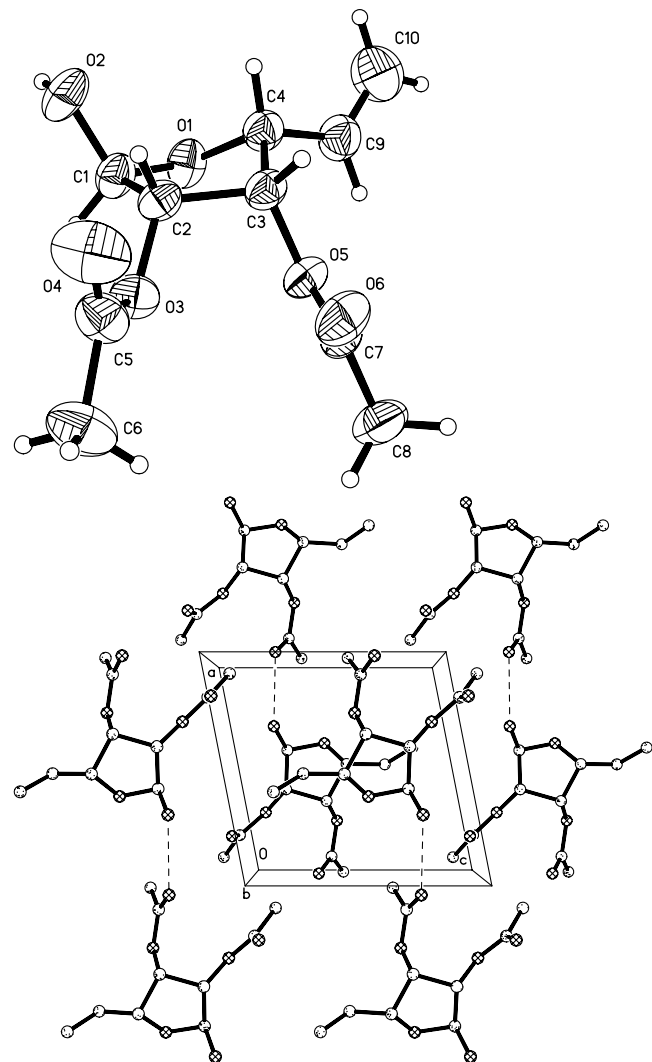


Crystal structure of 2,3-di-*O*-acetyl-5,6-dideoxy- α -*D*-lyxo-5-hexenofuranose, C₁₀H₁₄O₆

Wolfgang Frey, Robert Sardzik and Volker Jäger*

Universität Stuttgart, Institut für Organische Chemie, Pfaffenwaldring 55, 70569 Stuttgart, Germany

Received January 21, 2008, accepted and available on-line July 7, 2008; CCDC no. 1267/2274



Abstract

C₁₀H₁₄O₆, monoclinic, *P*2₁ (no. 4), *a* = 8.195(2) Å, *b* = 8.672(2) Å, *c* = 8.479(1) Å, β = 101.02(1)°, *V* = 591.5 Å³, *Z* = 2, *R*_{gt}(*F*) = 0.051, *wR*_{ref}(*F*²) = 0.131, *T* = 293 K.

Source of material

The title compound has been obtained by vitamin B₁₂-catalysed fragmentation of methyl 2,3-di-*O*-acetyl-6-deoxy-6-iodo- α -*D*-mannopyranoside in methanol [1,2]. Isolation by chromatography (petroleum ether/ethyl acetate) gave a (67 : 33) mixture of α - and β -anomers as colourless oil. Crystallization from ethyl ace-

tate/hexane gave the title compound as a pure α -anomer in the form of colourless crystals [1] (m.p. 352–353 K); $[\alpha]_D^{20}$ = +70.6 (*c* = 1.00, mixture of anomers in equilibrium).

Discussion

The furanose ring system of the molecule has an envelope conformation. C4 deviates by 0.58 Å from the plane given by C1/C2/C3/O1 (rmsd = 0.026 Å; figure, top; 40 % probability ellipsoids). The hydrogens of the methyl groups C6 and C8 of the acetyl moieties shows a rotational disorder with population parameters of 0.5 each. The double bond is clearly identified by the distance of 1.286(4) Å between C9 and C10. The crystal packing shows along [010] an antiparallel orientation of the molecules in the (010) plane. This effect is forced by the influence of the intermolecular hydrogen bond between the hydroxy group O2–H2A and the carbonyl oxygen O6 of the acetyl moiety (figure, bottom). The H2A...O6 distance is 1.94(3) Å and the angle O2–H2A...O6 is 167(3)°.

Table 1. Data collection and handling.

Crystal:	colourless block, size 0.3 × 0.5 × 0.7 mm
Wavelength:	Mo <i>K</i> α radiation (0.71073 Å)
μ :	1.08 cm ⁻¹
Diffractometer, scan mode:	Nicolet P3, Wyckoff scan
2 θ _{max} :	64.96°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	8521, 4266
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 3354
<i>N</i> (<i>param</i>) _{refined} :	178
Programs:	SHELXS-97 [3], SHELXL-97 [4], SHELXTL [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	2a		0.397(3)	−0.006(3)	0.790(3)	0.049(6)
H(2A)	2a		0.212(4)	0.172(4)	0.751(3)	0.072(8)
H(2)	2a		0.591(3)	0.260(3)	0.813(3)	0.055(6)
H(3)	2a		0.707(3)	0.251(2)	0.599(2)	0.039(5)
H(4)	2a		0.434(2)	0.255(3)	0.487(3)	0.042(5)
H(6A)	2a	0.50	0.8822	−0.0898	1.0615	0.121
H(6B)	2a	0.50	0.8857	0.0118	1.2151	0.121
H(6C)	2a	0.50	1.0216	0.0347	1.1097	0.121
H(6D)	2a	0.50	0.9774	0.0609	1.1960	0.121
H(6E)	2a	0.50	0.9739	−0.0407	1.0425	0.121
H(6F)	2a	0.50	0.8381	−0.0636	1.1478	0.121
H(8A)	2a	0.50	0.8985	−0.1860	0.5446	0.103
H(8B)	2a	0.50	1.0377	−0.1631	0.6976	0.103
H(8C)	2a	0.50	1.0613	−0.0958	0.5318	0.103
H(8D)	2a	0.50	1.0998	−0.1106	0.6381	0.103
H(8E)	2a	0.50	0.9606	−0.1335	0.4851	0.103

* Correspondence author (e-mail: jaeger.ioc@oc.uni-stuttgart.de)

Table 2. Continued.

Atom	Site	Occ.	x	y	z	<i>U</i> _{iso}
H(8F)	2a	0.50	0.9371	-0.2008	0.6509	0.103
H(9)	2a		0.545(4)	0.000(5)	0.320(4)	0.11(1)

Table 2. Continued.

Atom	Site	Occ.	x	y	z	<i>U</i> _{iso}
H(10A)	2a		0.351(5)	0.262(5)	0.206(5)	0.12(1)
H(10B)	2a		0.414(5)	0.107(5)	0.090(5)	0.11(1)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
O(1)	2a	0.3833(1)	0.0493(2)	0.5703(1)	0.0348(5)	0.0612(8)	0.0432(6)	-0.0107(5)	0.0053(4)	-0.0103(6)
C(1)	2a	0.4158(2)	0.0891(2)	0.7353(2)	0.0306(6)	0.056(1)	0.0414(8)	-0.0043(6)	0.0063(5)	-0.0049(7)
O(2)	2a	0.3096(2)	0.2062(2)	0.7681(2)	0.0321(6)	0.087(1)	0.0605(8)	0.0043(6)	0.0090(5)	-0.0215(7)
C(2)	2a	0.5927(2)	0.1560(2)	0.7716(2)	0.0291(5)	0.0441(7)	0.0394(7)	-0.0003(6)	0.0021(5)	-0.0028(7)
O(3)	2a	0.6993(1)	0.0589(1)	0.8826(2)	0.0401(6)	0.0469(7)	0.0446(6)	-0.0010(5)	-0.0026(5)	0.0003(5)
C(3)	2a	0.6449(2)	0.1628(2)	0.6072(2)	0.0301(6)	0.0384(7)	0.0463(8)	-0.0015(6)	0.0074(5)	-0.0012(6)
C(4)	2a	0.4775(2)	0.1540(2)	0.4913(2)	0.0364(6)	0.0465(8)	0.0400(7)	0.0047(7)	0.0037(5)	-0.0013(7)
O(4)	2a	0.7990(3)	0.2627(2)	1.0328(2)	0.100(1)	0.065(1)	0.072(1)	-0.009(1)	-0.029(1)	-0.0133(8)
O(5)	2a	0.7375(1)	0.0264(1)	0.5812(2)	0.0325(5)	0.0446(6)	0.0517(6)	0.0034(5)	0.0080(4)	-0.0046(5)
C(5)	2a	0.7992(2)	0.1279(3)	1.0084(2)	0.0447(8)	0.065(1)	0.0438(8)	-0.0093(8)	-0.0046(7)	0.0048(8)
O(6)	2a	0.9730(2)	0.1400(2)	0.7027(2)	0.0317(5)	0.0668(9)	0.081(1)	-0.0040(6)	0.0099(6)	-0.0061(8)
C(6)	2a	0.9068(3)	0.0107(4)	1.1075(3)	0.073(2)	0.082(2)	0.072(2)	0.001(1)	-0.024(1)	0.018(1)
C(7)	2a	0.9021(2)	0.0306(2)	0.6353(2)	0.0338(7)	0.060(1)	0.0460(8)	0.0050(7)	0.0110(6)	0.0039(8)
C(8)	2a	0.9821(3)	-0.1167(3)	0.5991(3)	0.054(1)	0.081(2)	0.069(1)	0.026(1)	0.007(1)	-0.007(1)
C(9)	2a	0.4803(3)	0.0944(3)	0.3263(3)	0.055(1)	0.069(1)	0.0437(9)	0.0046(9)	0.0064(7)	-0.0045(9)
C(10)	2a	0.4048(5)	0.1606(5)	0.1971(3)	0.105(2)	0.102(2)	0.046(1)	0.013(2)	0.003(1)	0.005(1)

Acknowledgments. We are grateful to Fonds der Chemischen Industrie and to Landesgraduierten-förderung Baden-Württemberg (Ph. D. scholarship to R. Sardzik) for financial support.

References

- Kleban, M.; Kautz, U.; Greul, J.; Hilgers, P.; Kugler, R.; Dong, H.-Q.; Jäger, V.: Vitamin B12 Catalysis of Zink-Mediated 6-Deoxy-6-iodopyranoside Fragmentation: A Mild and Convenient Preparation of *o*-Unsaturated Hexose Derivatives (5-Hexenoses). *Synthesis* (2000) 1027-1033.
- Sardzik, R.: -Cyclopropanozucker- durch Vitamin B12-katalysierte Cyclisierung von Diiodiden; Carbonyldiimidazol als Reagenz für die Cyclisierung ungesättigter Aminoalkohole zu 3-Pyrrolinen. Dissertation, Universität Stuttgart (2007).
- Sheldrick, G. M.: SHELXS-97. Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXL-97. Program for the Refinement of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXTL. Structure Determination Software Suite. Version 6.14. Bruker AXS, Madison, Wisconsin, USA 2000.