

Crystal structure of erbium palladium phosphor (3:20:6), $\text{Er}_3\text{Pd}_{20}\text{P}_6$

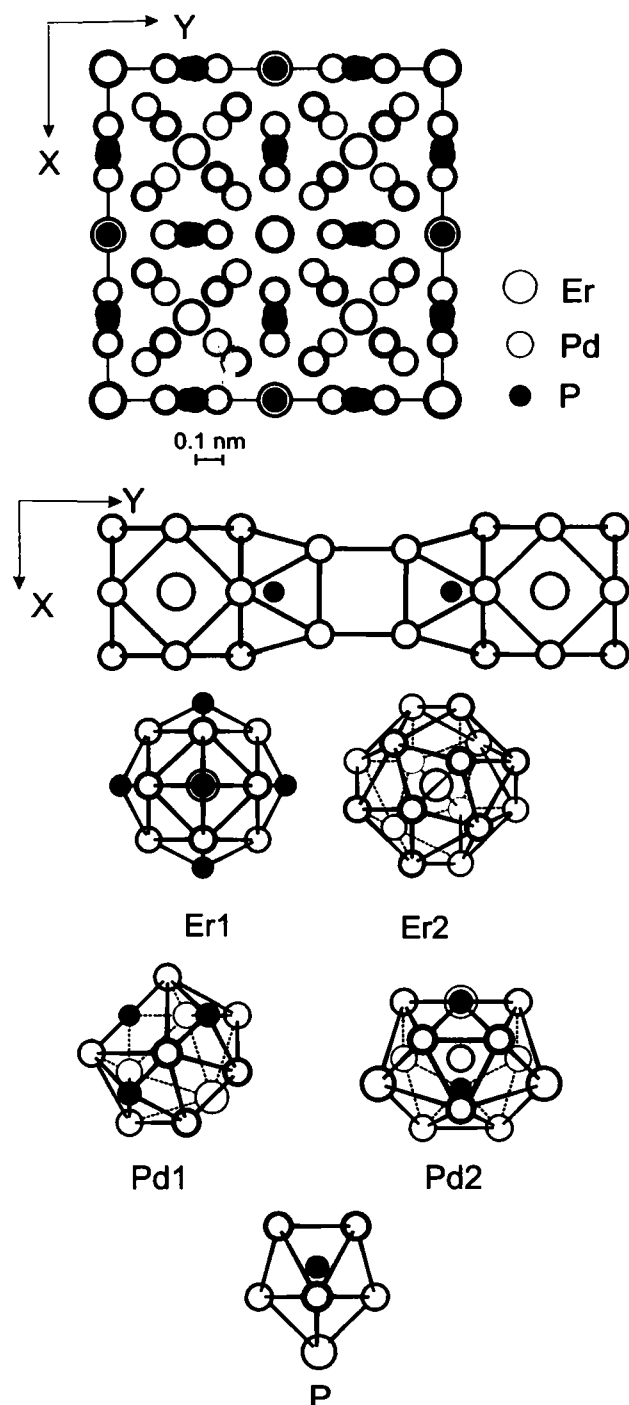
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Abstract

$\text{Er}_3\text{Pd}_{20}\text{P}_6$, cubic, $Fm\bar{3}m$ (no. 225), $a = 12.111(1) \text{ \AA}$, $V = 1776.4 \text{ \AA}^3$, $Z = 4$, $R_{\text{gt}}(F) = 0.030$, $wR_{\text{ref}}(F) = 0.032$, $T = 293 \text{ K}$.

Source of material

Erbium, red phosphorus, and palladium were of 99.99 wt.-% purity. A mixture of the starting components with nominal composition $\text{Er}_3\text{Pd}_{20}\text{P}_6$ was pressed into a small pellet (1 g) and melted in an arc furnace under a Ti/Zr-gettered argon atmosphere at normal pressure. The alloy was heat-treated at 1070 K for 20 d in evacuated fused silica ampules, followed by quenching in cold water. After the sample was crushed, bar-like air-stable single crystals with metallic luster were extracted and were for the structure determination. Energy-dispersive analysis of several crystals in a scanning electron microscope confirmed the presence of erbium, palladium, and phosphorus as the only components in atomic percentages of Er:Pd:P = 11:68:21 (with estimated standard deviations of 2 %).

Discussion

$\text{Er}_3\text{Pd}_{20}\text{P}_6$ adopts the Cr_{23}C_6 -type structure (Pearson symbol cF116) [1], which is built up of fragments of columns of cuboctahedra (figure, top), two tetragonal antiprisms, and cubes (figure, middle). The Er atoms substitute for the transition metals in sites 4a and 8c. The P atoms have tetragonal antiprismatic coordination (figure, bottom). The shortest interatomic distance found ($d(\text{Pd1}-\text{P}) = 235.6(2) \text{ pm}$) is by < 5 % smaller than the sum of atomic radii. Because of the low rare-earth content, the Er atoms remain isolated, and interatomic distances to them are larger than the sum of atomic radii. The greatest contraction (9.4 %) occurs for the $\text{Er2}-\text{Pd2}$ distance of $281.95(5) \text{ pm}$, indicating that the bonding between Er and Pd atoms is probably stronger than that between Pd and P, or Er and P. $\text{Er}_3\text{Pd}_{20}\text{P}_6$ is paramagnetic and its susceptibility follows the Curie-Weiss law.

Table 1. Data collection and handling.

Crystal:	black, bar-like, size $0.05 \times 0.05 \times 0.15 \text{ mm}$
Wavelength:	Mo $K\alpha$ radiation ($0.71073 \text{ \AA}$)
μ :	342.2 cm^{-1}
Diffractometer, scan mode:	Nonius KappaCCD, ω
$2\theta_{\text{max}}$:	79.5°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	327, 327
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 302
$N(\text{param})_{\text{refined}}$:	14
Programs:	WinCSD [2], ORTEP-3 [3]

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Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Er(1)	4 <i>a</i>	0	0	0	0.0074(2)	<i>U</i> ₁₁	<i>U</i> ₁₁	0	0	0
Er(2)	8 <i>c</i>	¼	¼	¼	0.0094(2)	<i>U</i> ₁₁	<i>U</i> ₁₁	0	0	0
Pd(1)	48 <i>h</i>	0.17278(4)	0	<i>x</i>	0.00676(1)	0.0175(3)	<i>U</i> ₁₁	0	−0.0001(2)	0
Pd(2)	32 <i>f</i>	0.11559(4)	<i>x</i>	<i>x</i>	0.00759(1)	<i>U</i> ₁₁	<i>U</i> ₁₁	−0.00073(1)	<i>U</i> ₁₂	<i>U</i> ₁₂
P(1)	24 <i>e</i>	0.2379(3)	0	½	0.009(1)	0.0065(7)	<i>U</i> ₂₂	0	0	0

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References

1. Bowman, A. L.; Arnold, G. P.; Storms, E. K.; Nereson, N. G.: The crystal structure of Cr₂₃C₆. *Acta Crystallogr.* **B28** (1972) 3102–3103.
2. Akselrud, L. G.; Zavalii, P. Y.; Grin, Yu. N.; Pecharsky, V. K.; Baumgartner, B.; Wölfel, E.: Use of the CSD program package for structure determination from powder data. *Mater. Sci. Forum* **133–136** (1993) 335–340.
8. Farrugia, L. J.: ORTEP-3 for Windows – a version of ORTEP-III with a Graphical User Interface (GUI). *J. Appl. Crystallogr.* **30** (1997) 565.