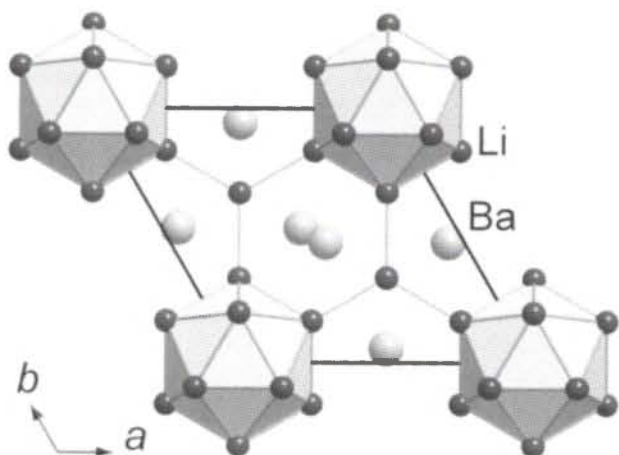


Refinement of the crystal structure of barium tetralithium, BaLi₄

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more than 95 at. % BaLi₄, according to powder X-ray diffraction analysis. All further handlings were carried out under argon atmosphere in a glove box or through the Schlenk technique.

Discussion

The crystal structure of BaLi₄ is known, but established only from Weissenberg photographs [1]. In the present work it was confirmed and refined on the basis of single-crystal diffractometer data.

The structure is characterized by rows of face sharing Li₁₂ icosahedra, each centered by a Li atom ($d(\text{Li—Li})$: 2.91 Å – 3.32 Å). The rows are interconnected via bridging Li atoms which in turn are surrounded by six Ba atoms. Each Ba atom takes a position between two rows. Ba—Ba distances range from 4.5453(4) Å to 4.5966(8) Å, Ba—Li distances – from 3.842(6) Å to 4.085(8) Å. Discrete or condensed Li₁₃ icosahedra occur frequently in ternary Li-rich phases (Li₁₃Na₂₉Ba₁₉ [2], Li₈₀Ba₃₉N₉ [3]).

Abstract

BaLi₄, hexagonal, $P6_3/mmc$ (no. 194), $a = 10.936(1)$ Å, $c = 8.943(2)$ Å, $V = 926.3$ Å³, $Z = 6$, $R_{\text{gt}}(F) = 0.029$, $wR_{\text{ref}}(F^2) = 0.068$, $T = 293$ K.

Source of material

The compound was synthesized from stoichiometric amounts of barium metal (Merck, 99 %; distilled twice with intermediate heating at 1270 K under vacuum in a closed Ta container to remove traces of H) and Li metal (Merck, 99 %) placed in a Ta ampoule under an Ar atmosphere. The ampoule was closed by arc welding and heated at 620 K for 10 days followed by cooling to room temperature at 1 K/h. The resulting product consisted of

Table 1. Data collection and handling.

Crystal:	grey metallic, xenomorphic fragment, size 0.087 × 0.116 × 0.164 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	93.85 cm ⁻¹
Diffractometer, scan mode:	Stoe IPDS II, ω
$2\theta_{\text{max}}$:	63.88°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	10729, 641
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 592
$N(\text{param})_{\text{refined}}$:	21
Programs:	SHELXS-97 [4], SHELXL-97 [5], DIAMOND [6]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Ba(1)	6h	0.47188(2)	2x	¼	0.0335(2)	0.0334(2)	0.0378(2)	½ U_{22}	0	0
Li(1)	2a	0	0	0	0.051(8)	U_{11}	0.05(1)	½ U_{11}	0	0
Li(2)	4f	½	¾	0.585(2)	0.050(6)	U_{11}	0.06(1)	½ U_{11}	0	0
Li(3)	6h	0.0986(8)	2x	¼	0.052(6)	0.042(7)	0.053(8)	½ U_{22}	0	0
Li(4)	12k	0.1645(6)	2x	0.566(1)	0.046(3)	0.050(5)	0.049(5)	½ U_{22}	0.001(2)	0.002(4)

References

- Wang, F.; Kanda, F.; Miskell, C.; King, A.: The crystal structures of Sr₆Mg₂₃, SrMg₄, Ba₆Mg₂₃ and BaLi₄. *Acta Crystallogr.* **18** (1965) 24–31.
- Smetana, V.; Babizhetskyy, V.; Vajenine, G.; Simon, A.: Li₂₆ Clusters in the Compound Li₁₃Na₂₉Ba₁₉. *Angew. Chem.* **45** (2006) 6051–6053; *Angew. Chem., Int. Ed. Engl.* **118** (2006) 6197–6200.
- Smetana, V.; Babizhetskyy, V.; Vajenine, G.; Simon, A.: Li₈₀Ba₃₉N₉: the first Li/Ba subnitride. *Inorg. Chem.* **45** (2006) 10786–10789.
- Sheldrick, G. M.: SHELXS-97. Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXL-97. Program for the Refinement of Crystal Structures. University of Göttingen, Germany 1997.
- Brandenburg, K.: DIAMOND. Visual Crystal Structure Information System. Version 3.1b. Crystal Impact, Bonn, Germany 2006.

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