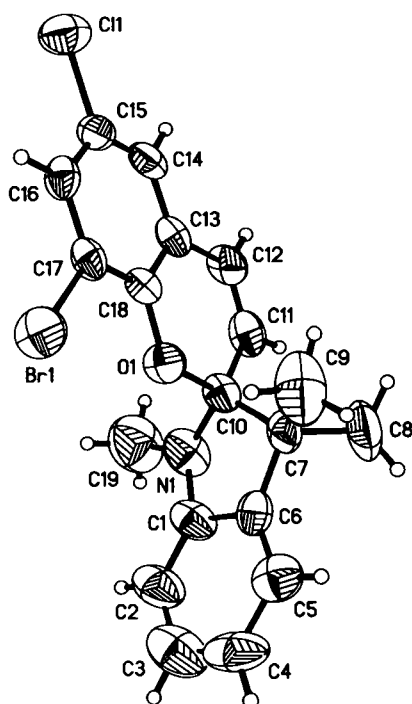


Crystal structure of 8-bromo-6-chloro-1',3',3'-trimethylspiro[2H-1-benzopyran-2,2'-indoline], C₁₉H₁₇BrClNO

A.-Q. Wang^{I,II}, Y. Wu^{III} and H. Guo^{*,I}^I Luoyang Normal University, Department of Chemistry, Luoyang 471022, P. R. China^{II} Zhengzhou Teacher's College, Department of Chemistry, Zhengzhou 450044, P. R. China^{III} Zhengzhou Economic Management Institute, Department of Environment Engineering, Zhengzhou 451191, P. R. China

Received June 2, 2006, accepted and available on-line August 7, 2006; CCDC no. 1267/1809



Abstract

C₁₉H₁₇BrClNO, tetragonal, *P*4₃ (no. 78), *a* = 8.1499(8) Å, *c* = 26.863(5) Å, *V* = 1784.3 Å³, *Z* = 4, *R*_{gt}(*F*) = 0.061, *wR*_{ref}(*F*²) = 0.202, *T* = 294 K.

Source of material

The title compound was synthesized by the method described in [1]. The crude product was purified by silica gel column chromatography using the mixed solvent petroleum ether/ethyl acetate (3:1, *v:v*) as the eluent. Single crystals suitable for X-ray crystallography were obtained from an acetone solution by slow evaporation at room temperature.

Discussion

Spiropyrans and spirooxazines are an important classes of photochromic material [2]. The photochromic compounds continue to attract significant attention in view of their general applicability as optical information storage material or switching devices [3–5]. Many modified spiroyrans and spirooxazines had been prepared in order to develop novel photochromic materials [6–9]. In the title structure, the pyrrolidine ring adopts an envelope conformation, while the pyran ring is in a twist-boat conformation. The bond angles at spiro atom C10 deviate from the normal value

of 109.5° and lie in the range 103.0(1)°–116.8(1)°. The dihedral angle between the C1/C2/C3/C4/C5/C6 plane and the C13/C14/C15/C16/C17/C18 plane is 87.4(4)°. The equation of the C10/C11/C12/C13/C18/O1 plane is 5.62(3)*x* + 5.85(3)*y* – 2.5(1)*z* = 2.2(1) with atom C10 lying only –0.0104 Å from the plane. Although atom C10 has *sp*³ hybridization and the ring is not an aromatic ring, all the six atoms on the ring lie on the same plane. The dihedral angle between the C10/C11/C12/C13/C18/O1 plane and the C13/C14/C15/C16/C17/C18 plane is 1.7(7)°, which shows the two rings are basically on the same plane. The equation of the C1/C6/C7/N1 plane is 4.99(5)*x* – 3.90(5)*y* + 16.9(2)*z* = 14.0(2). The deviation of atom C10 from the plane being 0.3836 Å is relatively large, which shows the five-membered ring composed of C1, C6, C7, N1, C10 is in an envelope conformation. The dihedral angle between this plane and the C1/C2/C3/C4/C5/C6 plane is 4(1)°.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.12 × 0.20 × 0.24 mm
Wavelength:	Mo <i>K</i> _α radiation (0.71073 Å)
<i>μ</i> :	24.57 cm ^{–1}
Diffractometer, scan mode:	Bruker SMART CCD, <i>φ/ω</i>
2 θ _{max} :	50.02°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	8965, 2468
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 1208
<i>N</i> (<i>param</i>) _{refined} :	211
Programs:	SHELXS-97 [10], SHELXL-97 [11], SHELXTL [12]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4a	0.2890	0.9315	0.9540	0.156
H(3)	4a	0.0989	0.9883	1.0110	0.184
H(4)	4a	–0.1001	0.7925	1.0301	0.181
H(5)	4a	–0.1161	0.5505	0.9825	0.134
H(8A)	4a	–0.1395	0.4738	0.8832	0.259
H(8B)	4a	–0.0348	0.3577	0.8492	0.259
H(8C)	4a	–0.0212	0.5485	0.8431	0.259
H(9A)	4a	0.1621	0.2768	0.9565	0.266
H(9B)	4a	0.1170	0.2035	0.9043	0.266
H(9C)	4a	–0.0220	0.2660	0.9399	0.266
H(11)	4a	0.2385	0.4995	0.8069	0.110
H(12)	4a	0.4164	0.3144	0.7821	0.104
H(14)	4a	0.6453	0.1107	0.8045	0.096
H(16)	4a	0.7661	0.0380	0.9446	0.088
H(19A)	4a	0.3943	0.8708	0.8718	0.283
H(19B)	4a	0.4666	0.7150	0.8456	0.283
H(19C)	4a	0.5139	0.7546	0.9009	0.283

* Correspondence author (e-mail: guohui0319@nankai.edu.cn)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Br(1)	4a	0.5605(2)	0.2583(2)	1.00375(6)	0.143(1)	0.172(2)	0.0546(7)	0.046(1)	−0.0243(9)	−0.003(1)
Cl(1)	4a	0.8591(5)	−0.0896(4)	0.8549(2)	0.110(3)	0.097(3)	0.138(4)	0.013(2)	0.026(3)	−0.009(2)
O(1)	4a	0.384(1)	0.408(1)	0.9175(3)	0.107(6)	0.110(6)	0.048(4)	0.026(5)	−0.020(4)	−0.001(4)
N(1)	4a	0.295(2)	0.667(2)	0.8931(5)	0.13(1)	0.098(9)	0.11(1)	−0.045(7)	0.021(8)	0.006(7)
C(1)	4a	0.201(2)	0.716(2)	0.9338(6)	0.14(1)	0.059(9)	0.09(1)	0.016(9)	−0.03(1)	0.001(8)
C(2)	4a	0.206(3)	0.856(2)	0.9601(8)	0.18(2)	0.08(1)	0.13(2)	0.02(1)	−0.03(2)	−0.02(1)
C(3)	4a	0.096(4)	0.889(3)	0.994(1)	0.24(3)	0.09(1)	0.13(2)	0.02(2)	−0.04(2)	0.00(1)
C(4)	4a	−0.025(3)	0.773(3)	1.005(1)	0.19(2)	0.15(2)	0.11(1)	0.11(2)	−0.02(2)	−0.01(2)
C(5)	4a	−0.034(2)	0.627(2)	0.9764(7)	0.12(1)	0.12(1)	0.10(1)	0.02(1)	−0.00(1)	0.01(1)
C(6)	4a	0.078(2)	0.599(2)	0.9410(5)	0.073(8)	0.10(1)	0.081(9)	0.005(8)	−0.023(8)	0.012(8)
C(7)	4a	0.102(2)	0.464(2)	0.9045(5)	0.08(1)	0.086(9)	0.083(9)	0.006(7)	−0.038(7)	−0.003(7)
C(8)	4a	−0.036(2)	0.461(3)	0.8665(9)	0.13(2)	0.17(2)	0.22(2)	−0.03(1)	−0.09(2)	−0.04(2)
C(9)	4a	0.088(2)	0.285(2)	0.9288(9)	0.18(2)	0.11(1)	0.24(3)	−0.04(1)	−0.09(2)	0.08(2)
C(10)	4a	0.272(2)	0.496(1)	0.8837(5)	0.10(1)	0.045(7)	0.082(9)	−0.002(6)	−0.022(7)	0.014(6)
C(11)	4a	0.301(2)	0.448(2)	0.8311(5)	0.11(1)	0.11(1)	0.059(8)	−0.012(9)	−0.028(8)	0.012(7)
C(12)	4a	0.408(2)	0.338(2)	0.8158(4)	0.12(1)	0.11(1)	0.036(6)	0.008(9)	−0.019(7)	−0.007(6)
C(13)	4a	0.511(1)	0.255(1)	0.8510(4)	0.089(8)	0.069(7)	0.053(7)	−0.022(7)	−0.009(6)	0.002(6)
C(14)	4a	0.628(2)	0.136(2)	0.8379(5)	0.090(9)	0.080(9)	0.070(8)	−0.022(7)	0.013(7)	−0.029(7)
C(15)	4a	0.715(1)	0.058(1)	0.8727(5)	0.072(8)	0.067(8)	0.077(9)	−0.003(6)	0.005(7)	−0.010(6)
C(16)	4a	0.701(2)	0.091(2)	0.9212(5)	0.073(8)	0.075(8)	0.072(9)	−0.007(6)	−0.018(6)	0.014(6)
C(17)	4a	0.585(2)	0.210(1)	0.9359(4)	0.088(8)	0.075(8)	0.052(6)	−0.010(7)	−0.020(6)	0.002(6)
C(18)	4a	0.493(1)	0.291(1)	0.9014(4)	0.079(7)	0.060(7)	0.057(7)	−0.010(6)	−0.001(6)	−0.005(6)
C(19)	4a	0.428(3)	0.759(3)	0.877(1)	0.19(1)	0.18(1)	0.19(1)	−0.051(9)	0.020(9)	−0.011(9)

Acknowledgments. This work was financially supported by the National Science Foundation of China (grant nos. 20402009) and the Natural Science Foundation of Henan province (grant no. 0611021000).

References

- Ono, H.; Oasada, T.; Kosuge, K.: Photochromic spiro compounds. US Patent Application 3578602 (1971).
- Winkler, J. D.; Bowen, C. M.; Michelet, V.: Photodynamic Fluorescent Metal Ion Sensors with Parts per Billion Sensitivity. *J. Am. Chem. Soc.* **120** (1998) 3237–3242.
- Duerr, H.: Perspectives in Photochromism: A Novel System Based on the 1,5-Electrocyclization of Heteroanalogous Pentadienyl Anions. *Angew. Chem., Int. Ed.* **28** (1989) 413–431.
- Dueer, H.; Bouas-Laurent, H.: Photochromism: Molecule and Systems. Elsevier, Amsterdam 1990.
- Ichi, M. K.: Photoalignment of Liquid-Crystal Systems. *Chem. Rev.* **100** (2000) 1847–1873.
- Li, X. L.; Wang, Y. M.; Matsuura, T.; Meng, J. B.: Synthesis of new spiro-pyrans and spirooxazines having a heteroaromatic pendant and their photochromic behavior. *Heterocycles* **51** (1999) 2639–2651.
- Zou, W. X.; Chen, P. L.; Gao, Y.; Meng, J. B.: 3-{2-[2-(3,5-Dinitro-2-oxidophenyl)ethyl]-3,3-dimethyl-3H-indol-1-ium-1-yl}propanoic acid bis(dimethylformamide) solvate. *Acta Crystallogr. E59* (2003) 337–339.
- Guo, H.; Gao, Y. B.; Li, Y. X.; Han, J.; Meng, J. B.: 1,3,3-Trimethyl-6'-(4-phenylpiperazin-1-yl)-2,3-dihydrospiro[1H-indole-2,3'-3H-naphth[2,1-b][1,4]oxazine]. *Acta Crystallogr. E61* (2005) o988–o989.
- Guo, H.; Gao, Y. B.; Han, J.; Meng, J. B.: 8-Bromo-6-chloro-1',3',3',5'-tetramethylspiro[2H-1-benzopyran-2,2'-indoline]. *Acta Crystallogr. E61* (2005) o1461–o1462.
- Sheldrick, G. M.: SHELXS-97. Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXL-97. Program for the Refinement of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXTL. Structure Determination Software Suite. Version 5.10. Bruker AXS, Madison, Wisconsin, USA 1998.
- Flack, H. D.: On enantiomorph-polarity estimation. *Acta Crystallogr. A39* (1983) 876–881.