

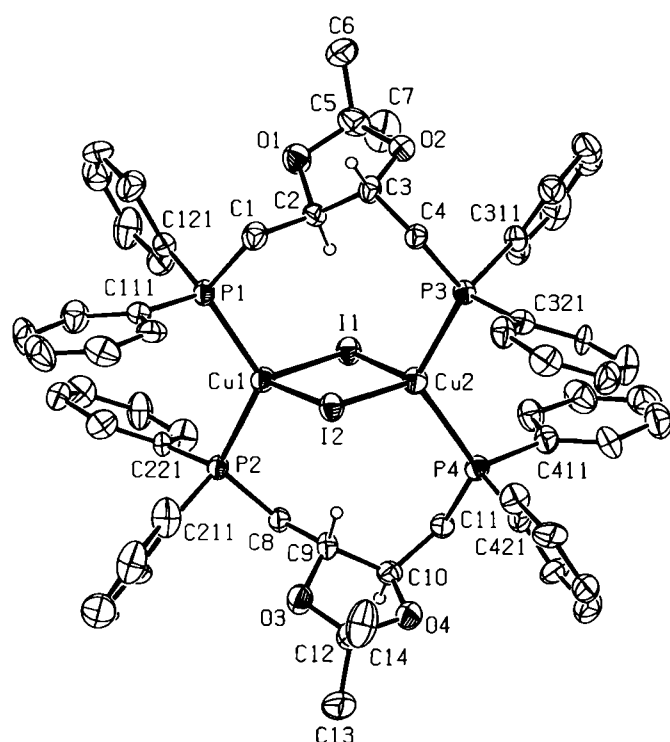
Crystal structure of diiodo-bis[(4*R*,5*R*)-*trans*-4,5-bis[(diphenylphosphino)methyl]-2,2-dimethyl-1,3-dioxalane]dicopper(I), Cu₂I₂(C₃₁H₃₂O₂P₂)₂

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Abstract

C₆₂H₆₄Cu₂I₂O₄P₄, triclinic, *P*1 (no. 1), *a* = 10.0989(2) Å, *b* = 12.8084(1) Å, *c* = 12.8423(2) Å, α = 65.074(1)°, β = 84.31°, γ = 87.854(1)°, *V* = 1499.0 Å³, *Z* = 1, *R*_{gt}(*F*) = 0.036, *wR*_{ref}(*F*²) = 0.098, *T* = 293 K.

Source of material

The title compound was synthesized following a published procedure for the silver(I) analogs [1]. To a solution of optically pure DIOP (50.0 mg, 0.10 mmol); DIOP = (4*R*,5*R*)-*trans*-4,5-bis[(diphenylphosphino)methyl]-2,2-dimethyl-1,3-dioxalane) in CH₂Cl₂ (5 mL) was added an excess of CuI (40.0 mg, 0.21 mmol). After refluxing the mixture for 0.5 h, a precipitate was removed by filtration and the filtrate treated with dried diethyl ether (10 mL) to give a white powder (yield 76 %). Colorless crystals suitable for X-ray studies were obtained by slow evaporation of its solution in CH₂Cl₂/*n*-heptane (1:1).

Elemental analysis – found: C, 53.58 %; H, 4.82% (VarioEL); calc. for C₆₂H₆₄Cu₂I₂O₄P₄: C, 54.04 %; H, 4.68%.

Experimental details

The absolute configuration of the chiral complex was confirmed by a value of the Flack parameter of 0.02(3) [2].

Discussion

The chiral diphosphine DIOP is an important bidentate ligand for transition metal complexes that can be used for asymmetric catalytic reactions [3–5]. In the search for such complexes we have obtained a dimeric Cu(I) compound, [Cu(I-*R,R*-DIOP)]₂, which features a novel Cu₂ core quadruply bridged by all the four ligands. Each copper atom resides at a slightly distorted tetrahedral environment composed of two iodine atoms and two phosphorous atoms each from a DIOP ligand. The four bridging ligands and the two copper atoms form a rhombic four-membered Cu₂I₂ ring (I–Cu–I, 109.11(6) and 110.42(6)°; Cu–I–Cu, 70.22(5)° and 70.24(5)°, respectively) and an outer 14-membered ring, Cu–DIOP–Cu–DIOP. Both of the Cu1–I1–Cu2–I2 and Cu1–P1–P3–Cu2–P4–P2 cores are coplanar (largest deviation 0.0069 Å and 0.0317 Å, respectively), intersecting at Cu1...Cu2 with a dihedral angle of 97.4(6)°. The coordination parameters around the two Cu(I) centers are slightly different. The Cu1–P1 and Cu1–P2 distances (2.252(4) Å and 2.254(4) Å) are shorter than the Cu2–P3 and Cu2–P4 distances (2.305(4) Å and 2.320(4) Å), while the P1–Cu1–P2 angle (122.5(2)°) is slightly wider than P3–Cu2–P4 (120.2(2)°). The bond angles about the two bridging iodine atoms are nearly identical ($\angle$ Cu2–I1–Cu1 = 70.22(5)°; $\angle$ Cu2–I2–Cu1 = 70.24(5)°). The Cu...Cu separation of 3.141 Å indicates no significant interaction of the metals (the sum of the covalent radii is 2.76 Å). Similar coordination mode has been reported for some Ag(I)-DIOP compounds [1] and ((4*R*,5*R*)-*trans*-4,5-bis(((3-methylphenyl)phosphino)-methyl)-2,2-dimethyl-1,3-dioxalane)copper(I) chloride [5].

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.25 × 0.30 × 0.33 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	18.91 cm ^{−1}
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
2 θ _{max} :	49.94°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	7502, 6122
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 5213
<i>N</i> (<i>param</i>) _{refined} :	667
Programs:	SHELXS-97 [6], SHELXL-97 [7]

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Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U _{iso}
H(1A)	1a	0.3841	0.1906	0.1975	0.066
H(1B)	1a	0.3117	0.2000	0.3070	0.066
H(2A)	1a	0.1583	0.0512	0.3038	0.052
H(3A)	1a	0.4121	-0.0057	0.3871	0.055
H(4A)	1a	0.2709	0.0386	0.5128	0.050
H(4B)	1a	0.3152	-0.0895	0.5684	0.050
H(6A)	1a	0.4605	-0.1313	0.1524	0.148
H(6B)	1a	0.4664	-0.2249	0.2791	0.148
H(6C)	1a	0.5209	-0.1005	0.2438	0.148
H(7A)	1a	0.1223	-0.1501	0.2795	0.144
H(7B)	1a	0.2169	-0.2556	0.3008	0.144
H(7C)	1a	0.2042	-0.1622	0.1750	0.144
H(8A)	1a	-0.4245	0.3747	0.2353	0.049
H(8B)	1a	-0.3682	0.2501	0.2661	0.049
H(9A)	1a	-0.2658	0.2752	0.4232	0.051
H(10A)	1a	-0.5429	0.2407	0.4416	0.052
H(11F)	1a	-0.4350	0.0788	0.4531	0.051
H(11G)	1a	-0.5142	0.0574	0.5705	0.051
H(13A)	1a	-0.6191	0.4369	0.4728	0.157
H(13B)	1a	-0.5323	0.5288	0.4861	0.157
H(13C)	1a	-0.5974	0.4268	0.5960	0.157
H(14A)	1a	-0.2577	0.3347	0.5952	0.157
H(14B)	1a	-0.3693	0.3672	0.6702	0.157
H(14C)	1a	-0.2966	0.4640	0.5602	0.157
H(11A)	1a	0.2214	0.4925	-0.0542	0.081
H(11B)	1a	0.2915	0.6771	-0.1030	0.104
H(11C)	1a	0.3761	0.7190	0.0375	0.109
H(11D)	1a	0.3756	0.5822	0.2241	0.091
H(11E)	1a	0.2951	0.3988	0.2715	0.074
H(12A)	1a	0.0040	0.2451	0.0022	0.080
H(12B)	1a	0.0243	0.2458	-0.1709	0.100
H(12C)	1a	0.2300	0.2475	-0.2632	0.098

Table 2. Continued.

Atom	Site	x	y	z	U _{iso}
H(12D)	1a	0.4174	0.2669	-0.1850	0.101
H(12E)	1a	0.4016	0.2786	-0.0110	0.094
H(21A)	1a	-0.3523	0.5900	0.1005	0.075
H(21B)	1a	-0.3489	0.7686	0.0917	0.122
H(21C)	1a	-0.1989	0.8232	0.1826	0.117
H(21D)	1a	-0.0390	0.6925	0.2729	0.101
H(21E)	1a	-0.0263	0.5149	0.2704	0.083
H(22A)	1a	-0.3447	0.2860	0.0666	0.078
H(22B)	1a	-0.3362	0.3170	-0.1207	0.081
H(22C)	1a	-0.2020	0.4466	-0.2589	0.080
H(22D)	1a	-0.0577	0.5577	-0.2157	0.080
H(22E)	1a	-0.0705	0.5368	-0.0216	0.073
H(31A)	1a	0.2375	-0.2917	0.6678	0.076
H(31B)	1a	0.2385	-0.4716	0.6650	0.095
H(31C)	1a	0.0701	-0.5150	0.5803	0.113
H(31D)	1a	-0.0976	-0.3861	0.4999	0.117
H(31E)	1a	-0.0821	-0.1987	0.4908	0.070
H(32A)	1a	-0.0447	-0.2349	0.7813	0.056
H(32B)	1a	-0.0491	-0.2652	0.9683	0.086
H(32C)	1a	0.0619	-0.1445	1.0208	0.083
H(32D)	1a	0.2054	-0.0056	0.8836	0.079
H(32E)	1a	0.2175	0.0181	0.6917	0.067
H(41A)	1a	-0.3697	-0.1086	0.4772	0.073
H(41B)	1a	-0.4488	-0.2913	0.5287	0.095
H(41C)	1a	-0.4743	-0.4217	0.7190	0.098
H(41D)	1a	-0.4078	-0.3756	0.8551	0.100
H(41E)	1a	-0.3310	-0.1893	0.8138	0.080
H(42A)	1a	-0.1166	0.0668	0.7308	0.070
H(42B)	1a	-0.1363	0.0768	0.9134	0.081
H(42C)	1a	-0.3414	0.0541	1.0165	0.089
H(42D)	1a	-0.5296	0.0250	0.9415	0.090
H(42E)	1a	-0.5068	0.0110	0.7676	0.073

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
I(1)	1a	-0.14501(5)	0.06170(5)	0.28491(5)	0.0491(6)	0.0502(5)	0.0463(5)	-0.0026(4)	-0.0089(4)	-0.0146(4)
I(2)	1a	0.04143(5)	0.24304(5)	0.46430(5)	0.0494(6)	0.0503(5)	0.0460(5)	-0.0058(4)	-0.0085(4)	-0.0133(4)
Cu(1)	1a	-0.0213(2)	0.2610(2)	0.2550(1)	0.035(1)	0.053(1)	0.046(1)	0.0020(8)	-0.0018(8)	-0.0098(9)
Cu(2)	1a	-0.0839(2)	0.0473(1)	0.4916(1)	0.039(1)	0.050(1)	0.045(1)	-0.0016(8)	-0.0083(8)	-0.0129(9)
P(1)	1a	0.1781(4)	0.2823(3)	0.1527(3)	0.036(2)	0.045(2)	0.037(2)	-0.001(2)	-0.004(2)	-0.003(2)
P(2)	1a	-0.1901(4)	0.3851(3)	0.1907(3)	0.035(2)	0.042(2)	0.033(2)	-0.002(2)	-0.003(2)	-0.007(2)
P(3)	1a	0.0856(4)	-0.0808(3)	0.5649(3)	0.035(2)	0.038(2)	0.039(2)	-0.001(2)	-0.008(2)	-0.013(2)
P(4)	1a	-0.2901(4)	0.0221(3)	0.5965(4)	0.033(2)	0.046(2)	0.048(2)	-0.004(2)	-0.002(2)	-0.013(2)
O(1)	1a	0.3015(8)	0.0124(8)	0.2055(7)	0.078(6)	0.062(5)	0.051(5)	0.006(4)	-0.010(4)	-0.021(4)
O(2)	1a	0.3095(8)	-0.1323(7)	0.3873(7)	0.061(4)	0.049(4)	0.052(5)	0.005(3)	0.000(3)	-0.018(4)
O(3)	1a	-0.3826(9)	0.4061(6)	0.4070(7)	0.089(6)	0.043(4)	0.052(5)	-0.012(4)	0.011(4)	-0.024(4)
O(4)	1a	-0.4730(7)	0.2579(6)	0.5688(6)	0.066(5)	0.055(4)	0.048(4)	0.000(3)	0.000(3)	-0.020(3)
C(1)	1a	0.299(1)	0.181(1)	0.243(1)	0.044(7)	0.058(8)	0.052(7)	-0.005(5)	-0.003(5)	-0.013(6)
C(2)	1a	0.2555(9)	0.0566(7)	0.2899(9)	0.038(5)	0.038(5)	0.047(6)	0.005(4)	-0.004(4)	-0.012(4)
C(3)	1a	0.3178(9)	-0.0261(7)	0.3960(7)	0.041(5)	0.052(5)	0.034(5)	-0.003(4)	-0.007(4)	-0.007(4)
C(4)	1a	0.260(1)	-0.036(1)	0.512(1)	0.037(5)	0.043(6)	0.037(6)	-0.001(4)	-0.004(4)	-0.009(5)
C(5)	1a	0.317(2)	-0.108(1)	0.265(1)	0.084(9)	0.059(8)	0.068(9)	0.010(6)	-0.016(7)	-0.040(8)
C(6)	1a	0.454(2)	-0.145(2)	0.232(2)	0.10(2)	0.11(2)	0.06(1)	0.04(1)	0.02(1)	-0.01(1)
C(7)	1a	0.205(1)	-0.175(1)	0.254(1)	0.12(1)	0.11(1)	0.079(9)	-0.025(8)	-0.016(8)	-0.050(8)
C(8)	1a	-0.353(1)	0.323(1)	0.2703(9)	0.039(5)	0.039(5)	0.037(6)	0.000(4)	-0.003(4)	-0.009(5)
C(9)	1a	-0.3534(9)	0.3028(8)	0.3964(9)	0.042(5)	0.039(5)	0.036(5)	-0.006(4)	-0.007(4)	-0.005(4)
C(10)	1a	-0.4597(9)	0.2227(7)	0.4789(7)	0.038(5)	0.049(5)	0.042(5)	0.006(4)	-0.008(4)	-0.018(4)
C(11)	1a	-0.436(1)	0.096(1)	0.520(1)	0.033(5)	0.046(6)	0.044(6)	0.002(4)	-0.006(4)	-0.012(5)
C(12)	1a	-0.433(2)	0.376(1)	0.521(1)	0.075(9)	0.07(1)	0.069(9)	-0.007(8)	0.010(7)	-0.033(8)
C(13)	1a	-0.557(2)	0.449(2)	0.519(2)	0.13(2)	0.07(1)	0.11(2)	0.01(1)	0.04(1)	-0.04(1)
C(14)	1a	-0.330(2)	0.386(1)	0.593(1)	0.13(1)	0.12(1)	0.055(8)	-0.046(9)	-0.010(7)	-0.027(8)
C(111)	1a	0.252(1)	0.425(1)	0.115(1)	0.037(7)	0.043(7)	0.045(7)	0.006(5)	-0.003(5)	-0.015(6)
C(112)	1a	0.252(2)	0.511(2)	0.002(2)	0.06(1)	0.07(1)	0.06(1)	-0.009(9)	-0.008(8)	-0.017(9)
C(113)	1a	0.295(2)	0.621(2)	-0.028(2)	0.09(2)	0.04(1)	0.10(2)	-0.020(9)	-0.01(1)	0.00(1)
C(114)	1a	0.343(2)	0.646(2)	0.057(2)	0.08(1)	0.04(1)	0.13(2)	-0.012(9)	-0.01(1)	-0.02(1)
C(115)	1a	0.344(2)	0.564(2)	0.168(2)	0.071(9)	0.07(1)	0.10(1)	0.008(8)	-0.022(8)	-0.05(1)

Table 3. Continued.

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
C(116)	1a	0.297(2)	0.454(1)	0.196(2)	0.048(7)	0.055(8)	0.08(1)	0.009(6)	0.006(6)	−0.030(7)
C(121)	1a	0.198(2)	0.267(1)	0.021(1)	0.048(9)	0.037(7)	0.053(9)	0.002(6)	0.004(7)	−0.012(7)
C(122)	1a	0.089(2)	0.253(2)	−0.036(2)	0.042(9)	0.06(1)	0.08(1)	−0.008(7)	0.007(8)	−0.015(9)
C(123)	1a	0.100(2)	0.250(2)	−0.137(2)	0.09(1)	0.09(1)	0.05(1)	−0.03(1)	−0.017(9)	−0.013(9)
C(124)	1a	0.222(2)	0.253(2)	−0.193(2)	0.12(2)	0.06(1)	0.06(1)	−0.01(1)	0.00(1)	−0.03(1)
C(125)	1a	0.334(2)	0.264(2)	−0.146(2)	0.09(1)	0.08(1)	0.07(1)	0.03(1)	0.03(1)	−0.02(1)
C(126)	1a	0.325(2)	0.271(2)	−0.041(2)	0.05(1)	0.10(1)	0.06(1)	0.007(9)	0.000(8)	−0.01(1)
C(211)	1a	−0.190(2)	0.526(1)	0.185(1)	0.051(9)	0.063(9)	0.036(7)	−0.001(7)	0.009(6)	−0.009(6)
C(212)	1a	−0.285(2)	0.609(1)	0.134(2)	0.06(1)	0.049(9)	0.09(1)	0.001(7)	0.010(7)	−0.039(8)
C(213)	1a	−0.285(2)	0.716(2)	0.131(2)	0.12(2)	0.06(1)	0.13(2)	0.02(1)	0.04(1)	−0.05(1)
C(214)	1a	−0.196(3)	0.750(2)	0.183(2)	0.14(2)	0.07(1)	0.09(1)	−0.01(1)	0.02(1)	−0.05(1)
C(215)	1a	−0.101(2)	0.672(2)	0.236(2)	0.12(1)	0.08(1)	0.060(9)	−0.05(1)	0.006(9)	−0.040(8)
C(216)	1a	−0.095(2)	0.564(2)	0.236(1)	0.09(1)	0.08(1)	0.043(7)	−0.021(9)	−0.029(7)	−0.023(7)
C(221)	1a	−0.210(1)	0.412(1)	0.040(1)	0.045(7)	0.031(6)	0.032(6)	0.000(5)	0.002(5)	−0.003(5)
C(222)	1a	−0.289(2)	0.343(2)	0.011(2)	0.07(1)	0.066(9)	0.064(9)	−0.020(7)	−0.011(7)	−0.034(8)
C(223)	1a	−0.282(2)	0.361(2)	−0.100(1)	0.08(1)	0.063(9)	0.051(9)	−0.027(7)	−0.014(7)	−0.011(7)
C(224)	1a	−0.201(2)	0.438(2)	−0.183(1)	0.08(1)	0.08(1)	0.047(8)	0.003(8)	−0.021(7)	−0.034(8)
C(225)	1a	−0.117(2)	0.506(1)	−0.159(1)	0.08(1)	0.06(1)	0.05(1)	−0.001(8)	0.017(8)	−0.012(8)
C(226)	1a	−0.124(2)	0.492(2)	−0.042(1)	0.06(1)	0.07(1)	0.058(9)	−0.004(8)	−0.007(7)	−0.037(9)
C(311)	1a	0.079(1)	−0.230(1)	0.574(1)	0.044(8)	0.030(6)	0.051(7)	−0.011(5)	−0.002(6)	−0.018(6)
C(312)	1a	0.174(2)	−0.311(1)	0.631(2)	0.052(9)	0.06(1)	0.062(8)	0.003(7)	−0.005(7)	−0.011(7)
C(313)	1a	0.173(2)	−0.418(2)	0.632(2)	0.07(1)	0.049(9)	0.10(1)	0.000(8)	0.003(9)	−0.017(9)
C(314)	1a	0.072(3)	−0.443(2)	0.581(2)	0.14(2)	0.05(1)	0.09(1)	−0.02(1)	0.04(1)	−0.04(1)
C(315)	1a	−0.028(3)	−0.365(2)	0.530(2)	0.14(2)	0.08(1)	0.07(1)	−0.04(1)	0.03(1)	−0.04(1)
C(316)	1a	−0.019(2)	−0.254(1)	0.526(1)	0.08(1)	0.046(8)	0.052(8)	−0.016(7)	0.006(7)	−0.020(7)
C(321)	1a	0.087(1)	−0.107(1)	0.714(1)	0.039(7)	0.050(8)	0.044(8)	0.005(6)	−0.016(5)	−0.022(6)
C(322)	1a	0.008(2)	−0.189(1)	0.800(1)	0.058(9)	0.040(7)	0.026(6)	−0.018(6)	0.007(6)	−0.001(5)
C(323)	1a	0.002(2)	−0.205(2)	0.913(1)	0.10(1)	0.07(1)	0.044(9)	−0.033(9)	−0.007(8)	−0.014(8)
C(324)	1a	0.071(2)	−0.136(2)	0.945(1)	0.09(1)	0.08(1)	0.037(7)	−0.003(8)	0.000(6)	−0.016(7)
C(325)	1a	0.155(2)	−0.052(2)	0.863(1)	0.08(1)	0.08(1)	0.056(9)	0.006(8)	−0.021(7)	−0.041(8)
C(326)	1a	0.162(2)	−0.038(1)	0.747(1)	0.045(7)	0.058(8)	0.050(8)	−0.005(6)	−0.018(5)	−0.006(6)
C(411)	1a	−0.345(2)	−0.129(1)	0.643(2)	0.031(6)	0.049(8)	0.060(9)	−0.005(5)	−0.004(6)	−0.004(7)
C(412)	1a	−0.379(2)	−0.161(1)	0.554(2)	0.061(9)	0.054(8)	0.058(8)	−0.007(6)	−0.017(6)	−0.012(7)
C(413)	1a	−0.426(2)	−0.270(1)	0.586(2)	0.08(1)	0.053(9)	0.11(1)	−0.010(7)	−0.014(9)	−0.032(9)
C(414)	1a	−0.439(2)	−0.349(2)	0.699(2)	0.07(1)	0.06(1)	0.11(2)	−0.002(9)	−0.00(1)	−0.04(1)
C(415)	1a	−0.402(2)	−0.320(2)	0.779(2)	0.09(2)	0.06(1)	0.07(1)	0.00(1)	−0.01(1)	−0.00(1)
C(416)	1a	−0.354(2)	−0.208(1)	0.755(2)	0.08(1)	0.038(9)	0.06(1)	0.007(8)	−0.015(8)	−0.005(8)
C(421)	1a	−0.311(2)	0.037(1)	0.735(1)	0.05(1)	0.053(9)	0.042(8)	0.003(7)	−0.005(7)	−0.005(7)
C(422)	1a	−0.199(2)	0.057(1)	0.774(1)	0.08(1)	0.07(1)	0.046(8)	−0.001(8)	−0.015(7)	−0.031(8)
C(423)	1a	−0.211(2)	0.064(1)	0.884(1)	0.08(1)	0.07(1)	0.07(1)	0.011(8)	−0.003(9)	−0.056(9)
C(424)	1a	−0.334(2)	0.051(2)	0.945(2)	0.11(2)	0.06(1)	0.05(1)	0.01(1)	0.01(1)	−0.018(9)
C(425)	1a	−0.446(2)	0.032(2)	0.901(2)	0.07(1)	0.09(1)	0.046(9)	0.005(9)	0.002(8)	−0.012(9)
C(426)	1a	−0.432(2)	0.025(2)	0.797(1)	0.047(9)	0.09(1)	0.043(8)	0.012(8)	0.003(7)	−0.023(8)

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