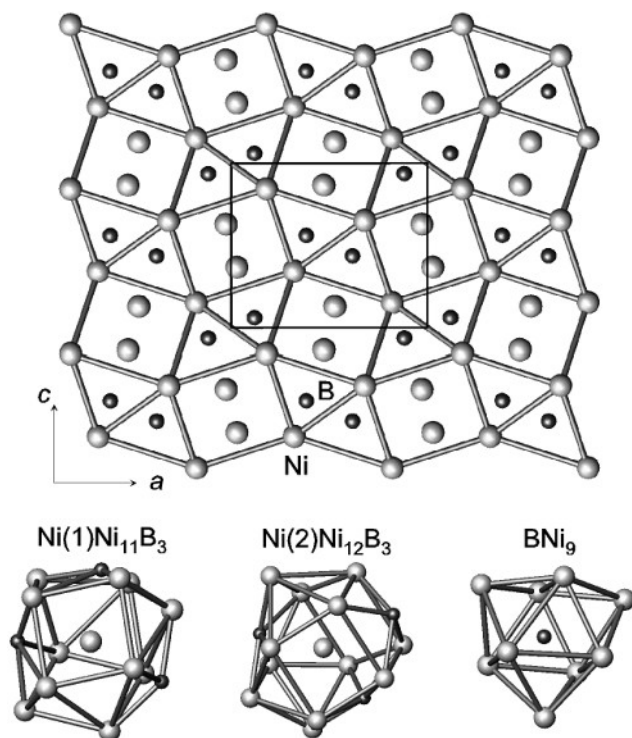


Refinement of the crystal structures of trinickel boron, Ni₃B, and tripalladium boron, Pd₃B

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Abstract

BNi₃, orthorhombic, *Pnma* (no. 62), $a = 5.2219(2)$ Å, $b = 6.6171(2)$ Å, $c = 4.3918(1)$ Å, $V = 151.8$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.022$, $wR_{\text{ref}}(F^2) = 0.036$, $T = 293$ K.

BPd₃, orthorhombic, *Pnma* (no. 62), $a = 5.4644(2)$ Å, $b = 7.5632(3)$ Å, $c = 4.8506(1)$ Å, $V = 200.5$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.025$, $wR_{\text{ref}}(F^2) = 0.049$, $T = 293$ K.

Source of material

Samples were obtained by arc-melting of mixtures of elemental components: Ni foil (99.99 %, Alfa Aesar), Pd foil (99.95 %, Chempur) and crystalline B powder (99.995 %, Chempur). The samples were sealed in tantalum tubes and annealed in evacuated silica tubes at 920 °C for 7 days. Needle-like single crystals were mechanically extracted from crushed samples. All manipulations were carried out in a argon-filled glove box ($p(\text{O}_2/\text{H}_2\text{O}) \leq 1$ ppm).

Experimental details

Lattice parameters of both compounds were refined from 58 reflections by least-squares fitting of powder diffraction data (Huber G670 imaging plate Guinier-camera, $\text{CuK}\alpha_1$ radiation, $\lambda = 1.54056$ Å) with LaB₆ as internal standard ($a = 4.15692$ Å).

In order to get sufficiently large data sets scattering intensities for Ni₃B and Pd₃B were measured up to high $\sin\theta/\lambda$ values, 1.414 and 1.293, respectively. This ensured an equal amount of independent intensities for all three crystals. During the crystal structure refinement special attention was paid to correlations between displacement and occupational parameters. After initial refinement of atomic coordinates, an extinction coefficient, displacement parameters and weighting scheme the occupational parameters for Ni1 (Pd1) were refined while occupancies of the other atoms were kept fixed. No deviation from full occupancy was observed. The same procedure was applied for the other atoms with similar results. After these steps displacement parameters for all atoms were refined anisotropically while fixing the occupation to 100 %. This resulted in a significant reduction of the reliability factors. In the further course of the refinement the occupational parameters for Ni (Pd) and B atoms were again treated separately for each species and finally the occupational parameters were refined freely for all atoms.

Discussion

The crystal structures of the isotopic compounds Ni₃B and Pd₃B have been subject of several investigations [1–4]. Nevertheless, in all previous studies the occupancies of the metal and boron positions were not refined. With respect to a discussion and evaluation of chemical composition, a detailed refinement of high resolution single-crystal XRD data was desirable. For this purpose we studied two Ni₃B and one Pd₃B single crystals of high quality which were isolated from the bulk samples with the nominal composition of 3:1.

For all refinement cycles full occupation at all atomic sites (within standard deviations of < 2 %) was observed without any significant variation of the displacement parameters. Moreover refined crystallographic parameters for the two Ni₃B single crystals are identical within one e.s.d. Derived compositions of Ni_{75.0(9)}B_{25.1(4)} and Pd₇₅₍₁₎B_{25.1(6)} are in very good agreement with results from WDXS (Ni₇₂₍₃₎B_{27.6(8)} and Pd₇₅₍₂₎B₂₄₍₂₎) and chemical analysis (Ni_{75.1(6)}B₂₄₍₁₎ and Pd_{74.0(1)}B_{26.0(1)}).

Ni₃B and Pd₃B crystallize with the cementite (Fe₃C) type structure, which is closely related to the U₃Si₂ and CuAl₂ structure types, as discussed in detail in [5] and [6]. Ni (Pd) atoms form puckered 3².4.3.4 nets, which are stacked along the *c* axis (figure, top). One more peculiarity of the Fe₃C structure type is the presence of empty octahedra, formed by *d* elements. In Ni₃B and Pd₃B structures Ni and Pd, respectively, are situated in the centers of 15- and 14-vertices polyhedra (figure, bottom), while B atoms are located in 3 capped trigonal prisms formed by Ni (Pd). The shortest interatomic distances Ni1—B at 2.0337(7) Å, Ni1—Ni1 at 2.4538(1) Å, Pd1—B at 2.158(2) Å and Pd1—Pd1 at 2.7163(2) Å are in good agreement with the sum of atomic radii of the elements ($r(\text{Ni}) = 1.25$ Å, $r(\text{Pd}) = 1.38$ Å and $r(\text{B}) = 0.83$ Å [7]; shortening is less than 3 %.

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1. Trinickel boron, Ni₃B

Table 1. Data collection and handling.

Crystal:	metallic needle, size 0.017 × 0.024 × 0.054 mm
Wavelength:	Ag K α radiation (0.5608 Å)
μ :	47.08 cm ⁻¹
Diffractometer, scan mode:	Rigaku R-AXIS SPIDER, φ/ω
$2\theta_{\max}$:	104.96°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	9768, 1622
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1404
$N(\text{param})_{\text{refined}}$:	26
Programs:	SHELXL-97 [8], WinCSD [9], ATOMS [10]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Ni(1)	8d	1.00(1)	0.17986(1)	0.06150(1)	0.15443(2)	0.00598(2)	0.00438(4)	0.00514(2)	0.00008(2)	-0.00045(2)	-0.00035(2)
Ni(2)	4c	1.00(1)	0.02798(2)	¼	0.63106(2)	0.00545(3)	0.00510(5)	0.00611(3)	0	0.00091(2)	0
B	4c	1.01(2)	0.3822(2)	¼	0.4380(2)	0.0061(3)	0.0078(4)	0.0085(3)	0	-0.0007(2)	0

2. Tripalladium boron, Pd₃B

Table 3. Data collection and handling.

Crystal:	metallic platelet, size 0.007 × 0.022 × 0.043 mm
Wavelength:	Ag K α radiation (0.5608 Å)
μ :	137.77 cm ⁻¹
Diffractometer, scan mode:	Rigaku R-AXIS SPIDER, φ/ω
$2\theta_{\max}$:	93°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	7184, 1754
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1381
$N(\text{param})_{\text{refined}}$:	26
Programs:	SHELXL-97 [8], WinCSD [9], ATOMS [10]

Table 4. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Pd(1)	8d	1.00(2)	0.18010(2)	0.07043(2)	0.17244(3)	0.00570(3)	0.00622(4)	0.00681(5)	-0.00003(3)	-0.00034(3)	-0.00017(3)
Pd(2)	4c	1.00(2)	0.03773(3)	¼	0.65475(3)	0.00519(4)	0.00652(5)	0.00657(6)	0	0.00017(4)	0
B(1)	4c	1.01(2)	0.3890(5)	¼	0.4370(5)	0.0082(7)	0.0145(9)	0.0100(9)	0	-0.0011(7)	0

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