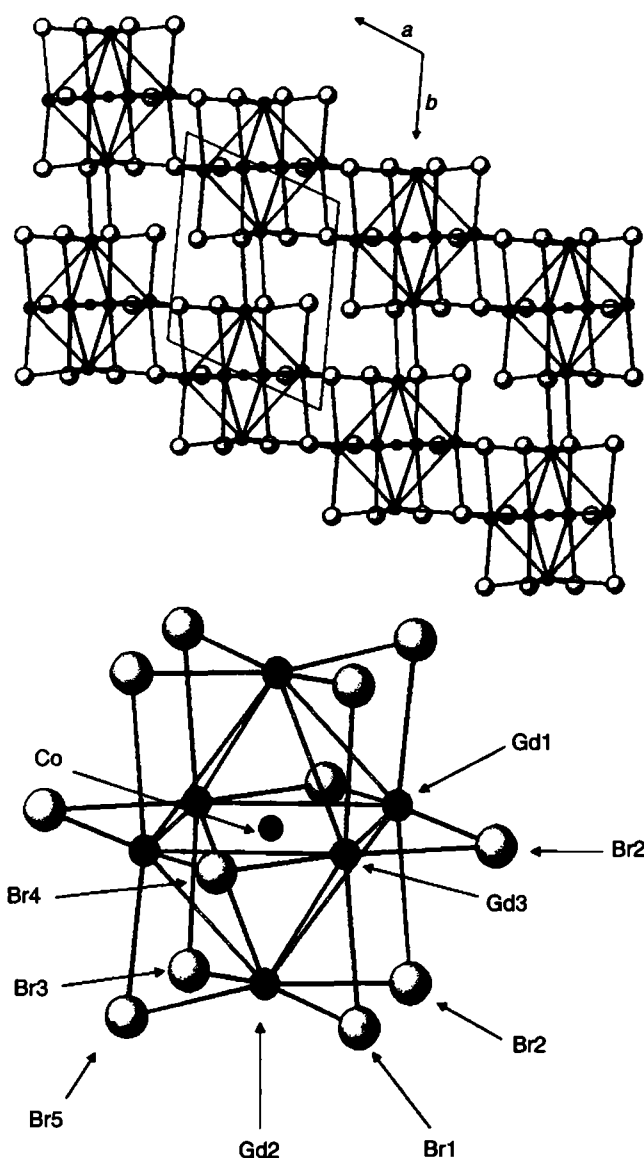


Crystal structure of hexagadolinium cobalt decabromide, $\text{Gd}_6\text{CoBr}_{10}$, a $\text{Y}_6\text{RuI}_{10}$ -type structure

C. Lefevre*, C. Hoch, R. Eger and A. Simon

Max-Planck-Institut für Festkörperforschung, Heisenbergstraße 1, 70569 Stuttgart, Germany

Received December 14, 2004, accepted and available on-line January 10, 2005; CSD no. 409809



Abstract

$\text{Br}_{10}\text{CoGd}_6$, triclinic, $P\bar{1}$ (no. 2), $a = 7.3051(8) \text{ \AA}$, $b = 9.040(1) \text{ \AA}$, $c = 9.090(1) \text{ \AA}$, $\alpha = 108.868(1)^\circ$, $\beta = 97.260(1)^\circ$, $\gamma = 106.299(1)^\circ$, $V = 529.4 \text{ \AA}^3$, $Z = 1$, $R_{\text{gt}}(F) = 0.031$, $wR_{\text{ref}}(F^2) = 0.096$, $T = 293 \text{ K}$.

Source of material

$\text{Gd}_6\text{CoBr}_{10}$ formed in a reaction of GdBr_3 , Gd and Co in a molar ratio 2:2:1. The mixture was sealed in a tantalum capsule under argon and heated at 1173 K for 10 days. Three different phases were obtained, which were analyzed through SEM as $\text{Gd}_2\text{Co}_2\text{Br}$, GdCo_3 and $\text{Gd}_6\text{CoBr}_{10}$. The new compound $\text{Gd}_6\text{CoBr}_{10}$ forms thin dark silvery platelet-like crystals.

Discussion

$\text{Gd}_6\text{CoBr}_{10}$ (figure, top) is isostructural to $\text{Y}_6\text{RuI}_{10}$ [1]. This structure has been observed for several compounds, Y_6MI_{10} ($M = \text{Co, Ni, Rh, Os, Ir, Pt}$) and $\text{Pr}_6\text{MI}_{10}$ ($M = \text{Co, Ru, Os}$) [2,3]. It can be described as a chain of distorted cuboctahedral clusters $\text{Gd}_6\text{Br}_{12}$ (figure, bottom) in which the cobalt atom occupies the center of the distorted Gd_6 octahedron ($d_{\text{Co-Gd1}} = 2.6833 \text{ \AA}$, $d_{\text{Co-Gd2}} = 2.6142 \text{ \AA}$ and $d_{\text{Co-Gd3}} = 2.6313 \text{ \AA}$). Each cuboctahedron is connected to others via halogen atoms Br2 and Br4 resulting in a cluster chain $[\text{Gd}_6\text{CoBr}_8\text{Br}_{4/2}]$ oriented along the a -axis (figure, top). The distance between two chains is equal to $2.572 \text{ \AA}$.

Table 1. Data collection and handling.

Crystal:	dark silvery platelet, size $0.06 \times 0.12 \times 0.18 \text{ mm}$
Wavelength:	Ag K_α radiation ($0.56086 \text{ \AA}$)
μ :	203.80 cm^{-1}
Diffractometer, scan mode:	IPDS 1, φ
$2\theta_{\text{max}}$:	42°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	8780, 2215
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2041
$N(\text{param})_{\text{refined}}$:	80
Programs:	SHELX-97 [4], ATOMS [5]

Table 2. Atomic coordinates and displacement parameters (in $\text{\AA}^2$).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Gd(1)	2i	$-0.13224(5)$	$0.14171(5)$	$0.68380(5)$	$0.0166(2)$	$0.0216(2)$	$0.0182(2)$	$0.0070(2)$	$0.0047(2)$	$0.0084(2)$
Gd(2)	2i	$0.47492(5)$	$0.29765(5)$	$0.59422(5)$	$0.0194(2)$	$0.0205(2)$	$0.0211(3)$	$0.0082(2)$	$0.0061(2)$	$0.0098(2)$
Gd(3)	2i	$0.37701(5)$	$-0.04370(4)$	$0.74777(5)$	$0.0183(2)$	$0.0201(2)$	$0.0183(2)$	$0.0077(2)$	$0.0049(2)$	$0.0086(2)$

* Correspondence author (e-mail: c.lefevre@fkf.mpg.de)

Table 2. Continued.

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Br(1)	2i	0.2990(1)	0.2685(1)	0.8595(1)	0.0234(4)	0.0231(4)	0.0203(4)	0.0092(3)	0.0067(3)	0.0065(3)
Br(2)	2i	-0.0505(1)	-0.1816(1)	0.5867(1)	0.0185(4)	0.0236(4)	0.0231(5)	0.0073(3)	0.0049(3)	0.0102(3)
Br(3)	2i	0.5890(2)	0.3709(1)	0.3225(1)	0.0499(6)	0.0245(4)	0.0311(5)	0.0197(4)	0.0199(4)	0.0159(4)
Br(4)	2i	-0.2362(1)	0.0877(1)	0.9640(1)	0.0239(4)	0.0479(6)	0.0204(5)	0.0044(4)	0.0038(3)	0.0169(4)
Br(5)	2i	-0.1334(1)	0.4693(1)	0.7789(1)	0.0281(5)	0.0211(4)	0.0408(6)	0.0062(3)	-0.0035(4)	0.0037(4)
Co	1f	½	0	½	0.0151(7)	0.0158(6)	0.0143(7)	0.0064(5)	0.0040(5)	0.0071(5)

References

1. Hughbanks, T.; Corbett, J. D.: Encapsulation of Heavy Transition Metals in Iodide Clusters. Synthesis, Structure and Bonding of the Unusual Cluster Phase Y₆I₁₀Ru. *Inorg. Chem.* **28** (1989) 631-635.
2. Payne, M. W.; Corbett, J. D.: Encapsulation of the Platinum Metals within Cluster Iodides of Rare Earth Elements. *Inorg. Chem.* **29** (1990) 2246-2251.
3. Llusar, R.; Corbett, J. D.: Reduced Praseodymium Cluster Bromides Stabilized by Transition Metals. *Inorg. Chem.* **33** (1994) 849-853.
4. Sheldrick, G. M.: SHELXL-97. Program for the Refinement of Crystal Structures. University of Göttingen, Germany 1997.
5. Dowty, E.: ATOMS for Windows and Macintosh. Version 5.0. Shape Software, Kingsport, Tennessee, USA 1999.