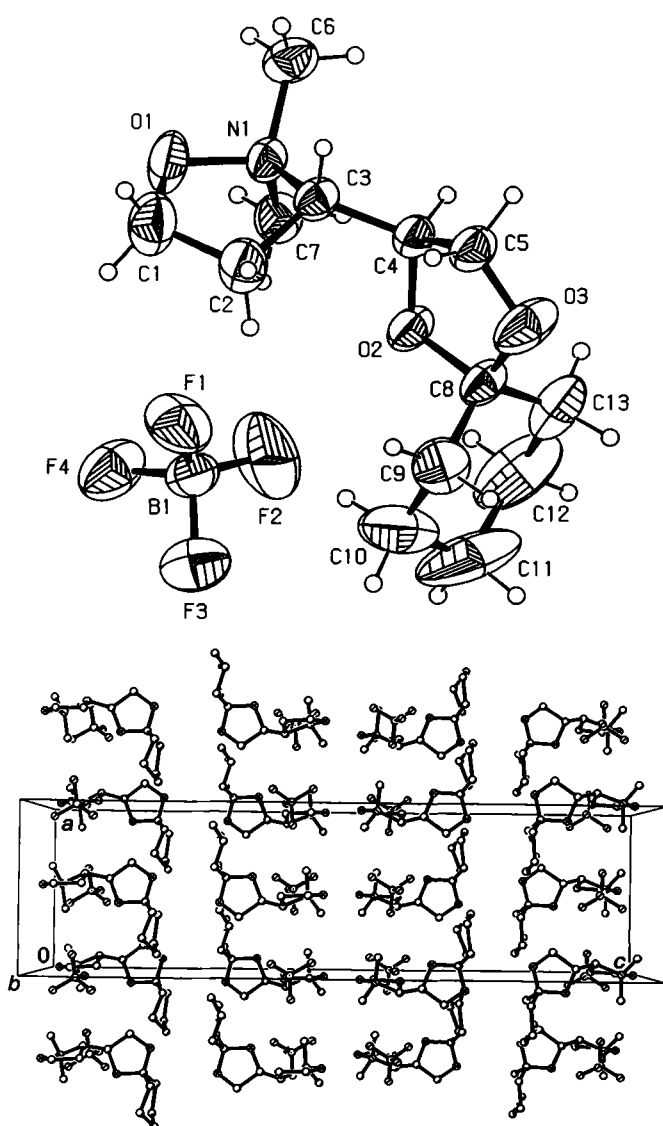


Crystal structure of (3*S*,1'*S*)-2,2-dimethyl-3-[1,2-cyclohexylidenedioxyethyl]-tetrahydro-1,2-oxazolium tetrafluoroborate, (C₁₃H₂₄NO₃)[BF₄]

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glyceraldehyde [1–3]. This was followed by *N*-methylation of the intermediate isoxazoline with trimethyloxonium tetrafluoroborate and reduction on treatment with sodium borohydride in ethanol [2]. Further methylation of the corresponding major isoxazolidine isomer (separated by MPLC, ratio of diastereoisomers = 84:16) with trimethyloxonium tetrafluoroborate [3–5] and recrystallization from ethanol afforded the isoxazolidinium salt in the form of colorless crystals (m.p. 443–444 K, $\alpha_D^{20} = -29.4^\circ$, $c = 0.81$, CH₂Cl₂).

Discussion

The isoxazolidine ring system shows an *envelope* conformation (figure, top). Three atoms (C10, C11, C12) of the cyclohexylidene moiety exhibit very large displacement parameters. A resolution of eventual disorder via split positions was not performed, the distances of the C–C bonds deviate less than 5 % from normal values. The fluorine atoms of the tetrafluoroborate anion have also large displacement parameters, which indicates considerable rotation of the anion.

The packing diagram reveals a bilayer-type stacking of the molecules with alternating polarity along the *c* axis (figure, bottom). The non-polar bilayer is built up by a face-to-face orientation of the cyclohexylidene moieties; the polar bilayer is built by the hetero atoms of the isoxazolidines and the tetrafluoroborate anions.

Table 1. Data collection and handling.

Crystal:	colorless lath, size 0.10 × 0.25 × 1.2 mm
Wavelength:	Cu K α radiation (1.54178 Å)
μ :	10.24 cm ⁻¹
Diffractometer, scan mode:	Siemens P4, ω
$2\theta_{\max}$:	109.94°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	1613, 1501
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1174
$N(\text{param})_{\text{refined}}$:	199
Programs:	SHELXS-97 [6], SHELXL-97 [7]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1A)	8c	0.0618	0.5178	0.4513	0.146
H(1B)	8c	-0.1065	0.5198	0.4543	0.146
H(2A)	8c	0.0584	0.4363	0.3923	0.120
H(2B)	8c	-0.1006	0.4873	0.3909	0.120
H(3)	8c	-0.1956	0.3044	0.4095	0.082
H(4)	8c	-0.0728	0.1290	0.3796	0.083
H(5A)	8c	-0.2387	0.1880	0.3358	0.102
H(5B)	8c	-0.1904	0.3296	0.3366	0.102

Abstract

C₁₃H₂₄BF₄NO₃, orthorhombic, *C*222₁ (no. 20), $a = 9.320(2)$ Å, $b = 10.6761(8)$ Å, $c = 33.786(2)$ Å, $V = 3361.6$ Å³, $Z = 8$, $R_{\text{gt}}(F) = 0.064$, $wR_{\text{ref}}(F^2) = 0.185$, $T = 293$ K.

Source of material

The title compound has been obtained by 1,3-dipolar cycloaddition of ethylene and the nitrile oxide obtained *in situ* from hydroximoyl chloride, derived from 2,3-*O*-cyclohexylidene-D-

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Table 2. Continued.

Atom	Site	x	y	z	U _{iso}
H(6A)	8c	-0.2233	0.2026	0.4674	0.139
H(6B)	8c	-0.0951	0.1492	0.4923	0.139
H(6C)	8c	-0.1353	0.0880	0.4516	0.139
H(7A)	8c	0.1786	0.2594	0.4292	0.126
H(7B)	8c	0.1156	0.1234	0.4278	0.126
H(7C)	8c	0.1547	0.1845	0.4685	0.126
H(9A)	8c	0.0829	0.4215	0.3099	0.135
H(9B)	8c	0.1214	0.3536	0.2702	0.135

Table 2. Continued.

Atom	Site	x	y	z	U _{iso}
H(10A)	8c	0.3019	0.3713	0.3357	0.206
H(10B)	8c	0.3223	0.4437	0.2957	0.206
H(11A)	8c	0.4682	0.2682	0.2992	0.278
H(11B)	8c	0.3641	0.2628	0.2627	0.278
H(12A)	8c	0.3568	0.0628	0.2933	0.230
H(12B)	8c	0.3242	0.1229	0.3348	0.230
H(13A)	8c	0.1327	0.1176	0.2698	0.157
H(13B)	8c	0.1093	0.0479	0.3102	0.157

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
F(1)	8c	0.3766(5)	0.3963(5)	0.4213(2)	0.080(3)	0.142(4)	0.220(6)	-0.006(3)	0.018(3)	-0.028(4)
F(2)	8c	0.516(1)	0.2274(4)	0.4090(2)	0.288(9)	0.068(2)	0.266(7)	-0.005(5)	-0.029(8)	-0.011(3)
F(3)	8c	0.5787(6)	0.4134(5)	0.3900(2)	0.135(4)	0.143(4)	0.128(3)	0.007(3)	0.025(3)	0.032(3)
F(4)	8c	0.5860(5)	0.3683(6)	0.4524(1)	0.113(3)	0.222(6)	0.092(3)	0.017(4)	-0.004(2)	-0.004(3)
B(1)	8c	0.512(1)	0.3437(8)	0.4183(2)	0.092(6)	0.094(6)	0.086(5)	0.000(5)	0.019(5)	0.004(5)
O(1)	8c	-0.0208(7)	0.3607(4)	0.4703(1)	0.184(5)	0.081(3)	0.064(2)	-0.003(4)	-0.010(3)	-0.019(2)
O(2)	8c	0.0799(4)	0.2356(5)	0.35652(9)	0.068(2)	0.122(3)	0.052(2)	-0.001(3)	0.003(2)	-0.012(2)
O(3)	8c	-0.0721(5)	0.2239(8)	0.3040(1)	0.088(3)	0.257(8)	0.057(2)	0.005(5)	-0.005(2)	-0.008(4)
N(1)	8c	-0.0300(5)	0.2522(4)	0.4455(1)	0.084(3)	0.072(3)	0.050(2)	-0.004(3)	-0.003(2)	-0.004(2)
C(1)	8c	-0.025(1)	0.4689(6)	0.4470(2)	0.20(1)	0.079(4)	0.090(5)	-0.007(6)	-0.002(6)	-0.008(4)
C(2)	8c	-0.035(1)	0.4329(5)	0.4051(2)	0.154(7)	0.066(3)	0.081(4)	-0.006(5)	0.002(5)	0.005(3)
C(3)	8c	-0.0912(6)	0.2996(5)	0.4065(1)	0.072(3)	0.079(3)	0.053(3)	0.002(3)	-0.001(2)	0.003(3)
C(4)	8c	-0.0614(6)	0.2168(6)	0.3717(1)	0.069(3)	0.085(4)	0.052(3)	-0.006(3)	0.001(3)	-0.008(3)
C(5)	8c	-0.1564(7)	0.2438(8)	0.3361(2)	0.077(4)	0.123(5)	0.055(3)	-0.009(4)	0.000(3)	-0.007(4)
C(6)	8c	-0.1297(8)	0.1653(7)	0.4660(2)	0.108(5)	0.097(5)	0.071(4)	-0.002(4)	0.013(4)	0.018(3)
C(7)	8c	0.1178(7)	0.2003(7)	0.4425(2)	0.077(4)	0.098(4)	0.077(4)	0.001(4)	-0.012(3)	-0.003(3)
C(8)	8c	0.0728(7)	0.2347(7)	0.3140(1)	0.081(4)	0.107(5)	0.046(3)	0.008(4)	0.001(3)	-0.002(3)
C(9)	8c	0.135(1)	0.3512(8)	0.2987(2)	0.138(7)	0.108(6)	0.092(5)	-0.003(6)	-0.012(5)	0.020(5)
C(10)	8c	0.288(1)	0.366(2)	0.3073(3)	0.127(9)	0.27(2)	0.120(8)	-0.08(1)	-0.012(7)	0.07(1)
C(11)	8c	0.368(1)	0.261(3)	0.2914(3)	0.080(6)	0.49(3)	0.128(8)	0.04(2)	0.036(5)	0.08(2)
C(12)	8c	0.308(2)	0.131(2)	0.3065(4)	0.18(1)	0.26(2)	0.129(8)	0.15(1)	0.032(9)	0.03(1)
C(13)	8c	0.148(1)	0.1230(9)	0.2981(2)	0.18(1)	0.135(8)	0.074(4)	0.037(8)	0.019(6)	-0.012(5)

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