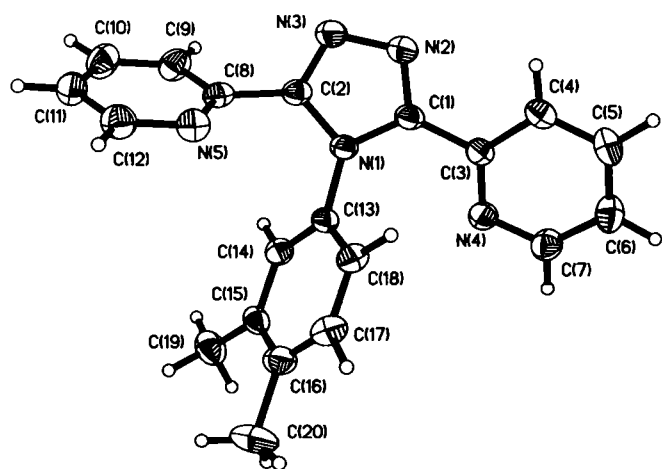


Crystal structure of 4-(3,4-dimethylphenyl)-3,5-bis(2-pyridyl)-1,2,4-triazole, $C_{20}H_{17}N_5$

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which lead to angles of 175.6° and 171.7° for $\angle H19B-C19-C15-C14$ and $\angle H20A-C20-C16-C15$, respectively. There is no obvious weak interaction, such as hydrogen bonds, in the molecular packing in the crystal.

Table 1. Data collection and handling.

Crystal:	colorless rod-like, size 0.20 × 0.26 × 0.32 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	0.80 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
$2\theta_{max}$:	50.06°
$N(hkl)_{measured}$, $N(hkl)_{unique}$:	4482, 2952
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2\sigma(I_{obs})$, 1528
$N(param)_{refined}$:	228
Programs:	SHELXTL [2], SHELXL-97 [3]

Abstract

$C_{20}H_{17}N_5$, triclinic, $P\bar{1}$ (no. 2), $a = 10.006(3)$ Å, $b = 10.318(3)$ Å, $c = 10.371(3)$ Å, $\alpha = 82.630(5)^\circ$, $\beta = 62.925(5)^\circ$, $\gamma = 63.291(5)^\circ$, $V = 847.7$ Å³, $Z = 2$, $R_{gt}(F) = 0.043$, $wR_{ref}(F^2) = 0.102$, $T = 298$ K.

Source of material

All reagents and solvents were used as purchased without further purification. C, H and N elemental analyses were performed on a Perkin-Elmer elemental analyzer.

The compound was prepared by the similar procedure as described in [1]. 3,4-Dimethylphenylamine and 2-pyridinecarboxaldehyde were used instead of 4-methoxyphenylamine and 4-pyridinecarboxaldehyde, respectively (yield 45 %). Elemental analysis: found – C, 73.00 %; H, 5.40 %; N, 21.25 %; calc. for $C_{20}H_{17}N_5$ – C, 73.37 %; H, 5.23 %; N, 21.39 %.

Experimental details

The low ratio of N_{gt}/N_{param} is caused by the poor quality of the crystals indicated by the appearance of micro-blebs.

Discussion

All bond lengths and bond angles are within normal ranges. All four rings are well-defined planes. The dihedral angles between the triazole ring and the pyridine planes are 36.5(2)° (with N4/C3/C4/C5/C6/C7 ring) and 50.1(3)° (with N5/C8/C9/C10/C11/C12 ring), respectively. The dihedral angle between the triazole ring and the substituted phenyl is 57.0(2)°. During refinement the torsions of the methyl groups were set as parameters,

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(4)	2a	0.4745	0.0589	0.1393	0.068
H(5)	2a	0.4608	−0.0060	0.3663	0.080
H(6)	2a	0.2601	0.1599	0.5673	0.082
H(7)	2a	0.0767	0.3798	0.5392	0.077
H(9)	2a	0.1471	0.4888	−0.3005	0.074
H(10)	2a	0.1828	0.6400	−0.4861	0.088
H(11)	2a	0.3496	0.7533	−0.5207	0.082
H(12)	2a	0.4599	0.7226	−0.3630	0.079
H(14)	2a	0.0008	0.6814	0.0625	0.056
H(17)	2a	0.3307	0.7228	0.2352	0.073
H(18)	2a	0.4033	0.4988	0.1373	0.062
H(19A)	2a	−0.2109	0.9890	0.2646	0.111
H(19B)	2a	−0.1806	0.9269	0.1184	0.111
H(19C)	2a	−0.1048	1.0317	0.1159	0.111
H(20A)	2a	0.1580	0.9688	0.3109	0.153
H(20B)	2a	−0.0293	1.0255	0.3433	0.153
H(20C)	2a	0.0988	1.0470	0.1935	0.153

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
N(1)	2a	0.2574(2)	0.4342(2)	0.0183(2)	0.045(1)	0.039(1)	0.046(1)	−0.0124(9)	−0.0221(9)	−0.0021(9)
N(2)	2a	0.3250(2)	0.2050(2)	−0.0288(2)	0.076(1)	0.048(1)	0.061(1)	−0.020(1)	−0.035(1)	−0.001(1)
N(3)	2a	0.3243(2)	0.2764(2)	−0.1508(2)	0.069(1)	0.050(1)	0.059(1)	−0.018(1)	−0.033(1)	−0.004(1)
N(4)	2a	0.1624(2)	0.3565(2)	0.3288(2)	0.050(1)	0.051(1)	0.051(1)	−0.012(1)	−0.022(1)	−0.001(1)
N(5)	2a	0.3796(2)	0.5877(2)	−0.2440(2)	0.051(1)	0.062(1)	0.062(1)	−0.025(1)	−0.027(1)	0.005(1)
C(1)	2a	0.2859(3)	0.3003(2)	0.0704(2)	0.045(1)	0.040(1)	0.052(1)	−0.012(1)	−0.021(1)	−0.004(1)
C(2)	2a	0.2844(2)	0.4128(2)	−0.1205(2)	0.043(1)	0.047(1)	0.048(1)	−0.015(1)	−0.023(1)	−0.003(1)
C(3)	2a	0.2803(3)	0.2604(2)	0.2135(2)	0.047(1)	0.041(1)	0.052(1)	−0.018(1)	−0.023(1)	0.004(1)
C(4)	2a	0.3944(3)	0.1234(2)	0.2222(2)	0.053(2)	0.043(1)	0.064(2)	−0.014(1)	−0.025(1)	0.001(1)
C(5)	2a	0.3862(3)	0.0852(3)	0.3566(3)	0.062(2)	0.054(2)	0.080(2)	−0.021(1)	−0.038(2)	0.022(1)
C(6)	2a	0.2672(3)	0.1827(3)	0.4754(3)	0.064(2)	0.075(2)	0.061(2)	−0.029(2)	−0.030(1)	0.024(1)
C(7)	2a	0.1586(3)	0.3148(3)	0.4570(2)	0.057(2)	0.070(2)	0.051(2)	−0.022(1)	−0.020(1)	0.002(1)
C(8)	2a	0.2879(3)	0.5192(2)	−0.2285(2)	0.043(1)	0.049(1)	0.046(1)	−0.014(1)	−0.019(1)	−0.007(1)
C(9)	2a	0.2119(3)	0.5367(3)	−0.3158(2)	0.066(2)	0.077(2)	0.054(1)	−0.036(1)	−0.033(1)	0.008(1)
C(10)	2a	0.2333(3)	0.6261(3)	−0.4258(3)	0.078(2)	0.087(2)	0.064(2)	−0.033(2)	−0.042(2)	0.009(2)
C(11)	2a	0.3301(3)	0.6943(3)	−0.4453(3)	0.073(2)	0.060(2)	0.056(2)	−0.021(2)	−0.025(1)	0.009(1)
C(12)	2a	0.3973(3)	0.6733(2)	−0.3511(3)	0.066(2)	0.066(2)	0.066(2)	−0.034(1)	−0.025(1)	0.005(1)
C(13)	2a	0.2092(3)	0.5703(2)	0.0882(2)	0.043(1)	0.038(1)	0.039(1)	−0.015(1)	−0.015(1)	−0.001(1)
C(14)	2a	0.0678(3)	0.6905(2)	0.0966(2)	0.042(1)	0.047(1)	0.046(1)	−0.019(1)	−0.016(1)	0.003(1)
C(15)	2a	0.0233(3)	0.8268(2)	0.1562(2)	0.045(1)	0.037(1)	0.045(1)	−0.011(1)	−0.009(1)	0.004(1)
C(16)	2a	0.1267(3)	0.8369(3)	0.2060(2)	0.068(2)	0.056(2)	0.054(1)	−0.033(1)	−0.020(1)	−0.001(1)
C(17)	2a	0.2644(3)	0.7150(3)	0.1992(2)	0.058(2)	0.074(2)	0.054(2)	−0.032(1)	−0.022(1)	−0.004(1)
C(18)	2a	0.3088(3)	0.5805(2)	0.1410(2)	0.048(1)	0.057(2)	0.046(1)	−0.020(1)	−0.019(1)	−0.003(1)
C(19)	2a	−0.1322(3)	0.9551(2)	0.1645(3)	0.064(2)	0.054(2)	0.077(2)	−0.017(1)	−0.020(1)	0.009(1)
C(20)	2a	0.0847(4)	0.9830(3)	0.2693(3)	0.124(3)	0.072(2)	0.114(2)	−0.048(2)	−0.044(2)	−0.021(2)

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