

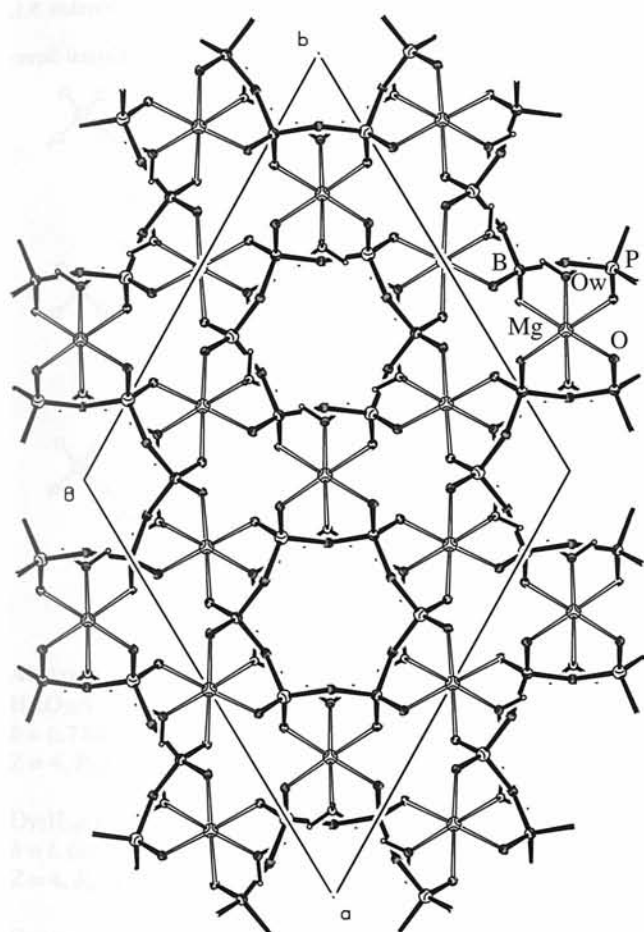
Crystal structure of diaquamagnesium (dihydrogenmonoborate-monophosphate), $\text{Mg}[\text{BPO}_4(\text{OH})_2](\text{H}_2\text{O})_2$, containing isolated six-membered rings of tetrahedra

H.-Z. Shi^I, Y.-K. Shan^{*,II}, M.-Y. He^{II} and Y.-Y. Liu^I

^I East China Normal University, Laboratory for Quantum Optics of National Education Ministry, Department of Chemistry, Shanghai 200062, P. R. China

^{II} East China Normal University, Department of Chemistry, Shanghai 200062, P. R. China

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Abstract

$\text{BH}_6\text{MgO}_8\text{P}$, trigonal, $R\bar{3}c$ (No. 167), $a = 14.9518(9)$ Å, $c = 13.809(2)$ Å, $V = 2673.6$ Å³, $Z = 18$, $R_{\text{gt}}(F) = 0.030$, $wR_{\text{ref}}(F^2) = 0.086$, $T = 293$ K.

Source of material

A mixture of $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$, H_3PO_4 (85 wt%), H_3BO_3 , $\text{C}_5\text{H}_5\text{N}$, $\text{NH}_2\text{NHCSNH}_2$ and H_2O in molar ratio of 1 : 1.5 : 8 : 2.5 : 1 : 22, was loaded in a sealed thick wall Pyrex tube and then placed in an oven at 383 K for 6 days. Colorless cubic crystals were obtained after washing with deionized water. The TG curve showed a two step mass loss of 27.3% (calc. 27 % for a total loss of nine water

molecules per formula unit) accompanied by two endothermic DTA peaks (onset: 390 K, peak maximum: 419 K; onset: 478 K, peak maximum: 514 K, respectively).

Discussion

The structure of $\text{Mg}[\text{BPO}_4(\text{OH})_2](\text{H}_2\text{O})_2$, consists of B and P tetrahedra and Mg octahedra. The fundamental unit can be regarded as $\text{MgO}_2(\text{OH})_2(\text{H}_2\text{O})_2$ octahedra linking unbranched six-membered rings $[\text{B}_3\text{P}_3\text{O}_{12}(\text{OH})_6]^{6-}$ of alternating corner sharing $\text{BO}_2(\text{OH})_2$ and PO_4 tetrahedra (see figure). These units are repeated along c axis with H bonds spanning the space between units. The coordination octahedra $\text{MgO}_2[(\text{OH})_2(\text{H}_2\text{O})_2]$ are formed by two protonated B–O3 groups, two P–O1 groups from two adjacent six-membered rings and two H_2O ligands. The Mg octahedron has bonds ranging from $d(\text{Mg}—\text{O}) = 2.024(1)$ Å to $2.099(1)$ Å with an average of 2.050 Å. The mean distances $\bar{d}(\text{P}—\text{O}) = 1.530$ Å and $\bar{d}(\text{B}—\text{O}) = 1.470$ Å are consistent with those for seamanite ($\bar{d}(\text{P}—\text{O}) = 1.534$ Å, $\bar{d}(\text{B}—\text{O}) = 1.466$ Å) [1].

The title structure is very similar to the structure of mineral diopside $\text{Cu}_6[\text{Si}_6\text{O}_{18}] \cdot 6\text{H}_2\text{O}$ [2]. The main difference is the number of cations (6 vs. 12) per ring unit ($\text{B}_3\text{P}_3\text{O}_{12}(\text{OH})_6]^{6-}$ vs. $[\text{Si}_6\text{O}_{18}]^{12-}$). Also, the crystallographic c -axis length of the title compound is approximately doubled compared to that of diopside. The present synthesis of the nickel-free magnesium borophosphate hydrate validates the deduction made in [3].

Table 1. Data collection and handling.

Crystal:	colorless fragment, size $0.23 \times 0.25 \times 0.25$ mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	5.68 cm ⁻¹
Diffractometer, scan mode:	Brucker SMART CCD, φ/ω
$2\theta_{\text{max}}$:	54.1°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	4855, 665
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 649
$N(\text{param})_{\text{refined}}$:	52
Programs:	SHELXTL [4], SHELXL-97 [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(1)	36f	0.008(2)	0.292(2)	0.168(2)	0.025
H(2W)	36f	0.172(2)	0.051(2)	0.009(2)	0.025
H(3W)	36f	0.223(2)	0.116(2)	0.062(2)	0.025

* Correspondence author (e-mail: ykshan@chem.ecnu.edu.cn)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Mg	18 <i>e</i>	0.49740(5)	0	1/4	0.0102(3)	0.0115(4)	0.0099(4)	<i>U</i> ₂₂ /2	0.0011(1)	2 <i>U</i> ₁₃
B	18 <i>e</i>	0.1893(2)	0	1/4	0.0106(8)	0.009(1)	0.010(1)	<i>U</i> ₂₂ /2	−0.0004(4)	2 <i>U</i> ₁₃
P	18 <i>e</i>	0.81148(4)	0	1/4	0.0093(3)	0.0076(3)	0.0081(3)	<i>U</i> ₂₂ /2	−0.00034(9)	2 <i>U</i> ₁₃
O(1)	36 <i>f</i>	0.03790(9)	0.26338(9)	0.17022(9)	0.0129(6)	0.0146(6)	0.0136(6)	0.0093(5)	0.0033(4)	0.0046(4)
O(2)	36 <i>f</i>	0.22693(9)	0.26132(9)	0.16481(9)	0.0117(6)	0.0179(6)	0.0142(6)	0.0084(5)	0.0028(4)	0.0066(4)
O(3)	36 <i>f</i>	0.2065(1)	0.1027(1)	0.0112(1)	0.0203(7)	0.0152(6)	0.0139(6)	0.0078(5)	0.0010(5)	−0.0003(5)
O(4)	36 <i>f</i>	0.1733(1)	0.08390(9)	0.21350(9)	0.0297(7)	0.0143(6)	0.0165(7)	0.0153(5)	−0.0105(5)	−0.0056(5)

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