

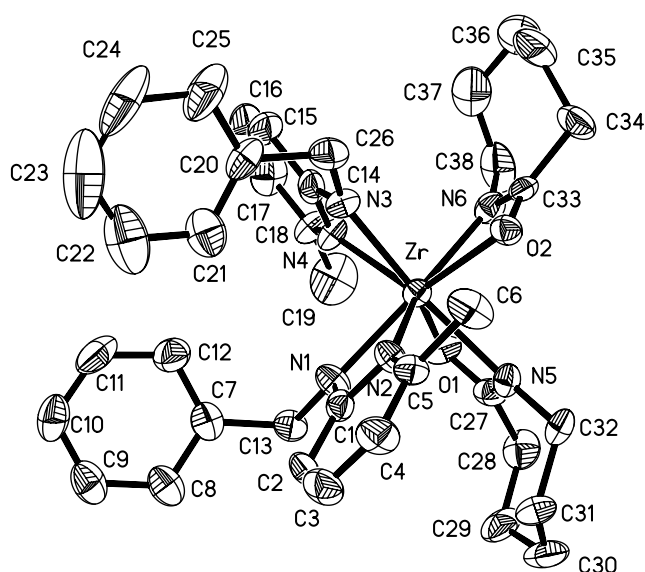
Crystal structure of bis[2-(benzylamino)-6-methylpyridinato]-bis(hexahydro-2-azepinato)zirconium(IV), $C_{38}H_{46}N_6O_2Zr$

R. Kempe^{*,I} and G. Hillebrand^{II}

^I Universität Bayreuth, Lehrstuhl für Anorganische Chemie II, D-95440 Bayreuth, Germany

^{II} Institut für organische Katalyseforschung an der Universität Rostock e.V., Buchbinderstr. 5-6, D-18055 Rostock, Germany

Received July 21, 2003, accepted and available on-line September 2, 2003; CCDC-No. 1267/1125



Abstract

$C_{38}H_{46}N_6O_2Zr$, monoclinic, $P12_1/c1$ (No. 14), $a = 13.575(3)$ Å, $b = 15.903(3)$ Å, $c = 17.404(3)$ Å, $\beta = 106.85(3)^\circ$, $V = 3596.0$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.046$, $wR_{\text{ref}}(F^2) = 0.126$, $T = 293$ K.

Source of material

All operations are done under exclusion of air and moisture using Ar-based Schlenk-techniques. To a solution of 0.15 g (0.27 mmol) bis[2-(benzylamino)-6-methylpyridinato]-chloro-dimethylamido-zirconium(IV) in 10 ml of diethylether 0.03 g (0.3 mmol) hexahydroazepine-2-on is added. The solution is stirred at room temperature for 3 h. The volume of the solution is reduced under vacuum to 5 ml. Crystallization at 243 K affords 0.07 g (35 %) of a crystalline material. NMR data and elemental analysis are consistent with the structure of the title compound.

Discussion

Caprolactamato or hexahydro-2-azepinato transition metal complexes [1–3] have been synthesized with regard to application of such compounds as initiators for ring opening polymerization reactions. In this paper we discuss the solid state structure of the first caprolactamato zirconium compound which is also the first mononuclear biscalprolactamato complex structurally characterized with X-ray diffraction methods.

Due to the strained coordination of the aminopyridinato ligands and the caprolactamato ligands the coordination cannot be described as ideal square antiprismatic or cubic. The two *N*-donor

atoms of each of the two aminopyridinato ligands are forming a plane with the *N,O*-donor set of the “opposite” bound caprolactamato ligand and the zirconium atom. The dihedral angle between these two planes ($N1/N2/N6/O2/Zr$, maximal deviation of 1.4 pm and $N3/N4/N5/O1/Zr$ maximal deviation of 0.9 pm) is 88° . The equivalence of the $Zr-N$ and the $Zr-O$ distances of the caprolactamato ligands indicate a delocalized binding mode. The anionic function is not localized at any of the donor atoms.

Table 1. Data collection and handling.

Crystal:	yellow prism, size $0.1 \times 0.5 \times 0.5$ mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	3.47 cm^{-1}
Diffractometer, scan mode:	Stoe IPDS, φ/ω
$2\theta_{\text{max}}$:	48.64°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	10588, 5450
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3912
$N(\text{param})_{\text{refined}}$:	424
Programs:	SHELXS-97 [4] SHELXL-93 [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4e	0.4514	−0.1114	0.2246	0.042
H(3)	4e	0.5866	−0.1285	0.3400	0.051
H(4)	4e	0.6231	−0.0245	0.4378	0.050
H(6A)	4e	0.4897	0.1711	0.4185	0.065
H(6B)	4e	0.6055	0.1439	0.4496	0.065
H(6C)	4e	0.5243	0.1039	0.4869	0.065
H(8)	4e	0.3238	−0.1790	0.0758	0.055
H(9)	4e	0.2379	−0.2997	0.0957	0.061
H(10)	4e	0.1112	−0.2937	0.1574	0.065
H(11)	4e	0.0711	−0.1661	0.2011	0.072
H(12)	4e	0.1513	−0.0450	0.1842	0.058
H(13A)	4e	0.3664	−0.0369	0.1022	0.044
H(13B)	4e	0.2625	0.0127	0.0728	0.044
H(15)	4e	0.0192	0.1225	0.2987	0.055
H(16)	4e	−0.1187	0.1336	0.1853	0.069
H(17)	4e	−0.0933	0.1469	0.0597	0.067
H(19A)	4e	0.1553	0.1581	0.0281	0.118
H(19B)	4e	0.0514	0.2070	−0.0056	0.118
H(19C)	4e	0.0535	0.1087	−0.0126	0.118
H(21)	4e	0.2584	−0.0265	0.3326	0.064
H(22)	4e	0.2113	−0.1567	0.3665	0.097
H(23)	4e	0.1266	−0.1716	0.4624	0.122
H(24)	4e	0.0902	−0.0542	0.5246	0.110
H(25)	4e	0.1283	0.0791	0.4885	0.088
H(26A)	4e	0.1781	0.1678	0.3921	0.051
H(26B)	4e	0.2933	0.1395	0.4132	0.051

* Correspondence author (e-mail: kempe@uni-bayreuth.de)

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(28A)	4e	0.5035	0.2588	0.0555	0.060
H(28B)	4e	0.4097	0.2123	−0.0036	0.060
H(29A)	4e	0.4978	0.0807	0.0504	0.067
H(29B)	4e	0.5443	0.1306	−0.0086	0.067
H(30A)	4e	0.6796	0.0897	0.0991	0.073
H(30B)	4e	0.6712	0.1877	0.1024	0.073
H(31A)	4e	0.7016	0.1281	0.2301	0.059
H(31B)	4e	0.5990	0.0771	0.2005	0.059
H(32A)	4e	0.5939	0.2028	0.2778	0.045
H(32B)	4e	0.5965	0.2537	0.2015	0.045

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(34A)	4e	0.3354	0.4405	0.3057	0.057
H(34B)	4e	0.3806	0.3917	0.3866	0.057
H(35A)	4e	0.2041	0.3483	0.3740	0.092
H(35B)	4e	0.2314	0.4417	0.3995	0.092
H(36A)	4e	0.1516	0.4863	0.2694	0.105
H(36B)	4e	0.0740	0.4384	0.3049	0.105
H(37A)	4e	0.0847	0.3195	0.2300	0.098
H(37B)	4e	0.0482	0.3968	0.1734	0.098
H(38A)	4e	0.1594	0.3327	0.1208	0.071
H(38B)	4e	0.2231	0.4085	0.1675	0.071

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Zr	4e	0.32185(3)	0.16304(3)	0.22390(2)	0.0234(2)	0.0276(2)	0.0254(2)	0.0017(2)	0.0056(2)	0.0025(2)
N(1)	4e	0.3369(3)	0.0329(2)	0.1881(2)	0.046(2)	0.031(2)	0.025(2)	0.002(2)	0.003(2)	−0.007(2)
N(2)	4e	0.4385(2)	0.0676(2)	0.3087(2)	0.033(2)	0.022(2)	0.026(2)	0.005(2)	0.001(2)	0.002(2)
C(1)	4e	0.4155(3)	0.0065(3)	0.2513(2)	0.032(2)	0.027(2)	0.029(2)	−0.004(2)	0.012(2)	0.001(2)
C(2)	4e	0.4688(3)	−0.0688(3)	0.2628(3)	0.042(3)	0.020(2)	0.042(3)	0.001(2)	0.011(2)	−0.003(2)
C(3)	4e	0.5480(3)	−0.0793(3)	0.3320(3)	0.043(3)	0.029(3)	0.053(3)	0.012(2)	0.010(2)	0.006(2)
C(4)	4e	0.5708(3)	−0.0167(3)	0.3902(3)	0.033(2)	0.041(3)	0.043(3)	0.011(2)	0.001(2)	0.002(2)
C(5)	4e	0.5158(3)	0.0563(3)	0.3768(2)	0.028(2)	0.031(3)	0.029(2)	0.005(2)	0.008(2)	−0.001(2)
C(6)	4e	0.5356(4)	0.1250(3)	0.4385(3)	0.043(3)	0.043(3)	0.033(2)	0.011(2)	−0.005(2)	−0.007(2)
C(7)	4e	0.2466(3)	−0.0990(3)	0.1283(2)	0.037(2)	0.039(3)	0.022(2)	−0.006(2)	−0.006(2)	−0.004(2)
C(8)	4e	0.2725(4)	−0.1758(3)	0.1014(3)	0.055(3)	0.040(3)	0.035(3)	−0.003(2)	0.001(2)	−0.009(2)
C(9)	4e	0.2204(4)	−0.2482(3)	0.1134(3)	0.048(3)	0.039(3)	0.053(3)	−0.005(2)	−0.005(3)	−0.008(3)
C(10)	4e	0.1458(4)	−0.2453(4)	0.1499(3)	0.039(3)	0.055(4)	0.062(3)	−0.019(3)	0.004(3)	0.003(3)
C(11)	4e	0.1224(4)	−0.1686(4)	0.1756(3)	0.039(3)	0.082(5)	0.067(4)	−0.012(3)	0.030(3)	−0.005(3)
C(12)	4e	0.1699(4)	−0.0958(3)	0.1659(3)	0.047(3)	0.055(4)	0.049(3)	−0.008(3)	0.023(3)	−0.013(3)
C(13)	4e	0.3056(3)	−0.0200(3)	0.1169(2)	0.045(3)	0.040(3)	0.027(2)	−0.003(2)	0.014(2)	−0.009(2)
N(3)	4e	0.2203(3)	0.1331(2)	0.2968(2)	0.028(2)	0.039(2)	0.033(2)	0.003(2)	0.011(2)	0.001(2)
N(4)	4e	0.1416(2)	0.1433(2)	0.1661(2)	0.023(2)	0.043(3)	0.034(2)	−0.004(2)	0.010(2)	0.006(2)
C(14)	4e	0.1292(3)	0.1342(3)	0.2402(3)	0.033(2)	0.023(2)	0.050(3)	−0.001(2)	0.011(2)	0.005(2)
C(15)	4e	0.0294(3)	0.1295(3)	0.2485(3)	0.029(2)	0.044(3)	0.066(3)	0.004(2)	0.019(3)	0.003(3)
C(16)	4e	−0.0520(4)	0.1354(3)	0.1809(4)	0.026(3)	0.044(3)	0.093(4)	0.003(2)	0.004(3)	0.006(3)
C(17)	4e	−0.0373(4)	0.1441(3)	0.1056(3)	0.032(3)	0.062(4)	0.062(3)	0.000(2)	−0.005(3)	0.009(3)
C(18)	4e	0.0618(3)	0.1483(3)	0.1001(3)	0.027(2)	0.056(3)	0.042(3)	0.000(2)	0.013(2)	0.007(2)
C(19)	4e	0.0823(4)	0.1562(4)	0.0203(3)	0.065(4)	0.109(6)	0.043(3)	−0.005(4)	−0.014(3)	0.011(4)
C(20)	4e	0.1948(3)	0.0404(3)	0.4053(2)	0.028(2)	0.062(3)	0.029(2)	−0.009(2)	0.005(2)	0.008(2)
C(21)	4e	0.2221(4)	−0.0314(3)	0.3703(3)	0.065(3)	0.045(3)	0.046(3)	−0.002(3)	0.007(3)	0.007(3)
C(22)	4e	0.1950(5)	−0.1092(4)	0.3915(4)	0.088(5)	0.052(4)	0.075(4)	−0.005(3)	−0.019(4)	0.023(3)
C(23)	4e	0.1448(6)	−0.1186(5)	0.4485(4)	0.110(6)	0.091(6)	0.066(5)	−0.046(5)	−0.032(4)	0.053(5)
C(24)	4e	0.1224(5)	−0.0484(6)	0.4844(4)	0.077(4)	0.144(8)	0.058(4)	−0.045(5)	0.026(4)	0.035(5)
C(25)	4e	0.1457(4)	0.0320(4)	0.4635(3)	0.074(4)	0.106(5)	0.045(3)	−0.038(4)	0.027(3)	0.005(3)
C(26)	4e	0.2241(4)	0.1261(3)	0.3806(3)	0.049(3)	0.046(3)	0.036(3)	0.001(2)	0.020(2)	−0.006(2)
O(1)	4e	0.3154(2)	0.1883(2)	0.0958(2)	0.029(2)	0.053(2)	0.032(2)	0.008(1)	0.012(1)	0.005(1)
N(5)	4e	0.4642(3)	0.1961(2)	0.1898(2)	0.029(2)	0.029(2)	0.034(2)	−0.003(2)	0.007(2)	0.005(2)
C(27)	4e	0.4144(3)	0.1976(3)	0.1134(3)	0.044(3)	0.029(3)	0.036(3)	0.008(2)	0.020(2)	0.006(2)
C(28)	4e	0.4631(4)	0.2075(3)	0.0473(3)	0.061(3)	0.056(3)	0.036(3)	−0.006(3)	0.019(2)	0.008(2)
C(29)	4e	0.5329(4)	0.1322(4)	0.0439(3)	0.047(3)	0.080(4)	0.053(3)	−0.011(3)	0.035(3)	−0.013(3)
C(30)	4e	0.6373(4)	0.1355(4)	0.1084(3)	0.044(3)	0.072(4)	0.077(4)	0.008(3)	0.034(3)	−0.020(3)
C(31)	4e	0.6320(3)	0.1296(3)	0.1942(3)	0.027(2)	0.052(3)	0.072(4)	0.010(2)	0.018(3)	−0.003(3)
C(32)	4e	0.5745(3)	0.2012(3)	0.2196(3)	0.028(2)	0.042(3)	0.044(3)	−0.009(2)	0.013(2)	0.005(2)
O(2)	4e	0.3896(2)	0.2535(2)	0.3243(2)	0.029(2)	0.027(2)	0.035(2)	0.005(1)	0.008(1)	0.000(1)
N(6)	4e	0.2698(3)	0.2976(2)	0.2186(2)	0.029(2)	0.035(2)	0.039(2)	0.002(2)	0.012(2)	0.010(2)
C(33)	4e	0.3287(3)	0.3136(3)	0.2909(3)	0.031(2)	0.027(3)	0.047(3)	0.001(2)	0.026(2)	0.002(2)
C(34)	4e	0.3251(4)	0.3925(3)	0.3368(3)	0.061(3)	0.026(3)	0.057(3)	0.008(2)	0.021(3)	−0.006(2)
C(35)	4e	0.2224(5)	0.4021(4)	0.3557(4)	0.104(5)	0.032(3)	0.111(5)	0.004(3)	0.059(5)	−0.015(3)
C(36)	4e	0.1344(5)	0.4315(4)	0.2863(5)	0.067(4)	0.045(4)	0.165(7)	0.016(3)	0.057(5)	−0.010(4)
C(37)	4e	0.1067(4)	0.3729(4)	0.2135(4)	0.037(3)	0.045(4)	0.154(7)	0.016(3)	0.014(4)	0.012(4)
C(38)	4e	0.1907(4)	0.3556(3)	0.1740(3)	0.052(3)	0.030(3)	0.078(4)	0.008(2)	−0.009(3)	0.024(3)

Acknowledgments. We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for financial support and Anke Spannenberg for assistance in the X-ray experiment.

References

1. Arndt, P.; Lefebvre, C.; Kempe, R.; Tillack, A.; Rosenthal, U.: Reactions of lactams with titanocene- and zirconocene-alkyne complexes as elemental steps in catalytic anionic ring-opening polymerization of lactams. *Chem. Ber.* (1996) 1281-1285.
2. Oberthur, M.; Hillebrand, G.; Arndt, P.; Kempe, R.: Ligand-substrate communication in Group 4 aminopyridinato complexes. *Chem. Ber./Recueil* (1997) 789-794.
3. Evans, W. J.; Fujimoto, C. H.; Ziller, J. W.: Organolanthanide-Based Coordination and Insertion Reactivity of the Anion Formed by Deprotonation of ϵ -Caprolactam. *Organometallics* **20** (2001) 4529-4536.
4. Sheldrick, G. M.: Phase Annealing in SHELX-90: Direct Methods for large structures. *Acta Crystallogr. A* **46** (1990) 467-473.
5. Sheldrick, G. M.: SHELXL-93, A program for refining crystal structures, University of Göttingen, Germany 1993.