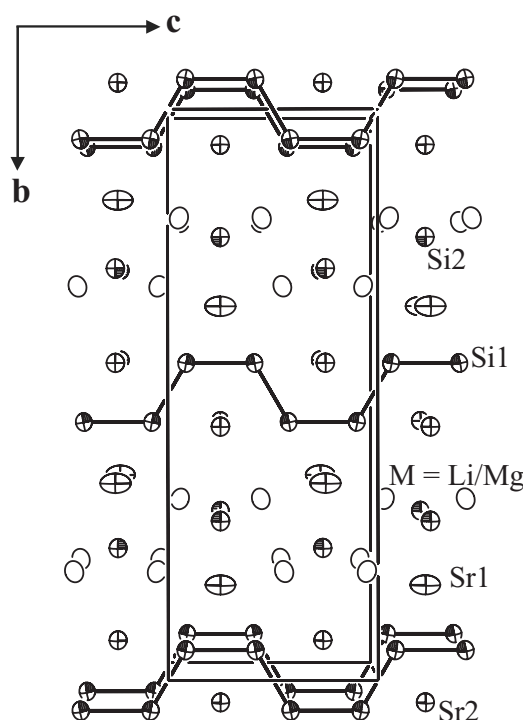


Crystal structure of distrontium lithium magnesium trisilicide, $\text{Sr}_2(\text{Li}_x\text{Mg}_{2-x})\text{Si}_3$ ($x = 1$)

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Abstract

$\text{LiMgSi}_3\text{Sr}_2$, orthorhombic, *Cmcm* (No. 63), $a = 4.6179(5)$ Å, $b = 19.488(2)$ Å, $c = 7.1618(8)$ Å, $V = 644.5$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.029$, $wR_{\text{ref}}(F^2) = 0.063$, $T = 293$ K.

Source of material

$\text{Sr}_2\text{LiMgSi}_3$ was synthesized from a mixture of pure elements at stoichiometric amounts via direct heating in argon atmosphere up to 1123 K for 2h. After cooling down by a rate of 50 K/h, the product was obtained as dark grey crystals with metallic lustre.

Discussion

$\text{Sr}_2\text{LiMgSi}_3$ crystallizes in a layered structure similar to EALiX_2 (EA = Sr, Ba; X = Si, Ge) [1]. It contains one-dimensional Zintl anion chains in *cis-trans* conformation (Si—Si distance is 2.381(4) Å; the bond angle is 120.17(9)°). Every silicon atom on the chains is coordinated by a trigonal Sr_6 prism. In addition, there are also isolated Si^{4-} anions. It can be formulated as $(\text{Sr}^{2+})_2(\text{Li}^+)(\text{Mg}^{2+})(\text{Si}^{1.5-})_2(\text{Si}^{4-})$ according to the Zintl-Klemm concept [2–4]. Si^{4-} is also coordinated in a very typical way, situated at the centre of a trigonal Sr_5 bi-pyramid with all faces capped by M (M = Li, Mg) atoms. The structure relates also to the LaSi [5] structure type if a M_2Si unit is omitted. It can be considered as being isostructural to $\text{Sr}_2\text{Mg}_{(2-x)}\text{Si}_3$ [6]. However, it should be pointed out that the Sr1 site in the latter case is split into three positions (0 0.3495(4) 1/4; 0 0.3490(6) 0.2881(7); 0.065(1) 0.3371(3) 0.3236(8)), while it is well refined by one position here yet with a quite large displacement along *c* direction. This observation can be traced back to the irregular coordination sphere of Sr1. It is difficult to say if electronic factors might also play an important role. The M position is coordinated by a Si_4 tetrahedron, it is situated at the centre of a Sr_5 tetragonal pyramid.

Table 1. Data collection and handling.

Crystal:	dark grey plate, size 0.12 × 0.18 × 0.18 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	170.65 cm ^{−1}
Diffractometer, scan mode:	Bruker SMART CCD, ω
$2\theta_{\text{max}}$:	67.66°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	5018, 732
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 515
$N(\text{param})_{\text{refined}}$:	26
Programs:	SHELXS-97 [7], SHELXL-97 [8], ORTEP-3 [9]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Sr(1)	4 <i>c</i>	0	0.34517(3)	1/4	0.0140(3)	0.0144(3)	0.0350(4)	0	0	0
Sr(2)	4 <i>c</i>	0	0.55585(3)	1/4	0.0127(2)	0.0128(3)	0.0118(2)	0	0	0
Si(1)	8 <i>f</i>	0	0.05282(5)	0.5834(2)	0.0155(5)	0.0126(5)	0.0101(5)	0	0	0.0005(4)
Si(2)	4 <i>c</i>	0	0.72110(8)	1/4	0.0120(7)	0.0134(7)	0.0117(7)	0	0	0
Mg/Li(1) ^a	8 <i>f</i>	0	0.1906(1)	0.0529(3)	0.016(1)	0.017(1)	0.013(1)	0	0	−0.0017(9)

a: $\text{Mg/Li}(1) = 0.502(8)\text{Mg} + 0.498\text{Li}$

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References

1. Xie, Q.; Wörle, M.; Nesper, R.: Conformational Changes of Unbranched Tetrelide Polymer Chains in Zintl Phases. *Helv. Chim. Acta.*, in preparation.
2. Zintl, E.: Intermetallische Verbindungen. *Angew. Chem.* **52** (1939) 1-6.
3. Klemm, W.: Metalloids and their Compounds with the Alkali Metals. *Proc. Chem. Soc. London* (1958) 329-341.
4. Nesper, R.: Structure and chemical bonding in Zintl-phases containing lithium. *Prog. Solid St. Chem.* **20** (1990) 1-45.
5. Mattausch, H.; Oeckler, O.; Simon, A.: Eine neue Modifikation von Lanthanmonosilicide - IT-LaSi, *Z. Anorg. Allg. Chem.* **625** (1999) 1151-1154.
6. Currao, A.: Synthese, Struktur und Eigenschaften von Zintlphasen der Erdalkali- und Alkali-Erdalkalimetalle mit Silicium. Dissertation, ETH-Nr. 11747, Zurich 1996.
7. Sheldrick, G. M.: SHELXS-97. Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
8. Sheldrick, G. M.: SHELXL-97. Program for the Refinement of Crystal Structures. University of Göttingen, Germany 1997.
9. Farrugia, L. J.: Ortep-3 for Windows (version 1.075). *J. Appl. Crystallogr.* **30** (1997) 565.