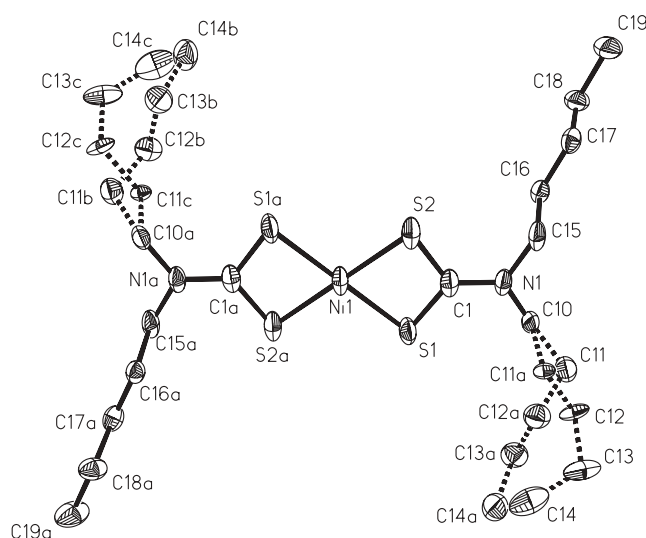


Crystal structure of bis(dipentyldithiocarbamato)nickel(II), $\text{Ni}(\text{C}_{11}\text{H}_{22}\text{NS}_2)_2$

J. Kameníček^{*I}, R. Pastorek^I, B. Cvek^I and J. Taraba^{II}^I Palacký University, Department of Inorganic and Physical Chemistry, Křížkovského 10, 771 47 Olomouc, Czech Republic^{II} Masaryk University, Department of Inorganic Chemistry, Kotlářská 10, 611 37 Brno, Czech Republic

Received January 24, 2003, accepted in final form and available on-line April 15, 2003; CCDC-No. 201016



Abstract

$\text{C}_{22}\text{H}_{44}\text{N}_2\text{NiS}_4$, monoclinic, $P12_1/n1$ (No. 14), $a = 10.402(2) \text{ \AA}$, $b = 13.261(3) \text{ \AA}$, $c = 10.701(2) \text{ \AA}$, $\beta = 114.79(3)^\circ$, $V = 1340.1 \text{ \AA}^3$, $Z = 2$, $R_{\text{gt}}(F) = 0.049$, $wR_{\text{ref}}(F^2) = 0.118$, $T = 120 \text{ K}$.

Source of material

The title compound, a known species [1], was isolated as a green blocks by the slow evaporation of an *n*-hexane solution.

Discussion

From the figure, it is apparent that the central atom is four-coordinated. The NiS_4 chromophore is essentially planar and deviation of carbon atom from dithiocarbamate group plane is about $0.07 \text{ \AA}$ [2]. The Ni—S distances are $2.190(1) \text{ \AA}$ and $2.179(1) \text{ \AA}$; the bond lengths of S1—C1 and S2—C1 are $1.700(4) \text{ \AA}$ and $1.705(4) \text{ \AA}$, respectively. The N1—C1 distance is very short ($1.315 \text{ \AA}$) which is well known for the dithiocarbamate complexes [3]. The best convergence was reached for two positions for C11 to C14 atoms with occupying factors 0.5.

Table 1. Data collection and handling.

Crystal:	dark green prism, size $0.35 \times 0.35 \times 0.25 \text{ mm}$
Wavelength:	Mo K_α radiation ($0.71073 \text{ \AA}$)
μ :	10.47 cm^{-1}
Diffractionmeter, scan mode:	KUMA KM-4, ω
$2\theta_{\text{max}}$:	57.02°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	10565, 3120
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2199
$N(\text{param})_{\text{refined}}$:	169
Programs:	PARST 95 [2], SHELXS-97 [4], SHELXL-97 [5]

Table 2. Atomic coordinates and displacement parameters (in $\text{\AA}^2$).

Atom	Site	Occ.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(10B)	4e		0.6094	0.1738	0.4050	0.050
H(10A)	4e		0.7829	0.2365	0.4811	0.050
H(11C)	4e	0.50	0.6703	0.3770	0.3287	0.050
H(11B)	4e	0.50	0.7167	0.3650	0.3861	0.050
H(11A)	4e	0.50	0.5175	0.3065	0.2384	0.050
H(11D)	4e	0.50	0.6242	0.3668	0.2799	0.050
H(12D)	4e	0.50	0.6478	0.3632	0.5155	0.050
H(12A)	4e	0.50	0.5665	0.3223	0.4991	0.050
H(12C)	4e	0.50	0.5141	0.4403	0.2858	0.050
H(12B)	4e	0.50	0.4719	0.2861	0.3113	0.050
H(13D)	4e	0.50	0.2979	0.3851	0.3261	0.050
H(13C)	4e	0.50	0.3741	0.2443	0.3331	0.050
H(13A)	4e	0.50	0.4960	0.4937	0.3491	0.050
H(13B)	4e	0.50	0.4422	0.3343	0.4513	0.050
H(14A)	4e	0.50	0.4170	0.4094	0.3281	0.050
H(14B)	4e	0.50	0.2623	0.4124	0.1988	0.050
H(14E)	4e	0.50	0.1987	0.3593	0.1328	0.050
H(14C)	4e	0.50	0.2559	0.4710	0.2732	0.050
H(14D)	4e	0.50	0.3393	0.3666	0.2359	0.050
H(14F)	4e	0.50	0.1449	0.3475	0.2825	0.050
H(15B)	4e		0.9060	0.2762	0.3570	0.046
H(15A)	4e		0.8764	0.2098	0.2257	0.044
H(16B)	4e		0.9874	0.1374	0.5052	0.040
H(16A)	4e		0.9492	0.0684	0.3614	0.049
H(17B)	4e		1.1616	0.2272	0.4583	0.047
H(17A)	4e		1.1241	0.1512	0.3230	0.060
H(18B)	4e		1.2428	0.0926	0.6104	0.046
H(18A)	4e		1.1964	0.0032	0.4640	0.090
H(19C)	4e		1.4249	0.1640	0.5707	0.080
H(19B)	4e		1.3832	0.0966	0.4182	0.090
H(19A)	4e		1.4595	0.0454	0.5808	0.130

* Correspondence author (e-mail: kamen@risc.upol.cz)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Ni(1)	2b		1/2	0	0	0.0531(4)	0.0207(3)	0.0302(3)	−0.0019(2)	−0.0101(2)	−0.0002(2)
S(1)	4e		0.48692(9)	0.09220(6)	0.16405(9)	0.0483(5)	0.0439(5)	0.0435(5)	0.0003(4)	−0.0115(4)	−0.0137(4)
S(2)	4e		0.7045(1)	0.07592(6)	0.07825(8)	0.0608(5)	0.0303(4)	0.0334(4)	−0.0071(3)	−0.0032(4)	−0.0021(3)
N(1)	4e		0.7343(3)	0.1876(2)	0.2979(3)	0.048(2)	0.024(1)	0.039(1)	0.006(1)	−0.009(1)	−0.006(1)
C(1)	4e		0.6544(3)	0.1286(2)	0.1962(3)	0.051(2)	0.021(1)	0.038(2)	0.003(1)	−0.009(1)	0.002(1)
C(10)	4e		0.6883(3)	0.2242(3)	0.4006(4)	0.034(2)	0.048(2)	0.064(2)	0.010(2)	−0.013(2)	−0.028(2)
C(11)	4e	0.50	0.6396(8)	0.3263(5)	0.4172(9)	0.045(4)	0.044(4)	0.045(4)	−0.004(3)	0.010(4)	−0.001(4)
C(11A)	4e	0.50	0.6066(6)	0.3240(4)	0.3317(7)	0.028(3)	0.028(3)	0.026(3)	0.008(2)	0.017(3)	0.008(2)
C(12)	4e	0.50	0.5597(7)	0.3669(5)	0.4406(6)	0.055(4)	0.045(3)	0.039(3)	0.027(3)	0.035(3)	0.010(3)
C(12A)	4e	0.50	0.5016(7)	0.3562(6)	0.3078(7)	0.047(4)	0.049(4)	0.062(5)	0.001(3)	0.025(4)	0.005(3)
C(13)	4e	0.50	0.3772(8)	0.3265(6)	0.3434(7)	0.054(4)	0.053(4)	0.049(4)	0.007(3)	0.016(3)	0.007(3)
C(13A)	4e	0.50	0.4517(9)	0.4486(6)	0.3915(9)	0.069(5)	0.047(4)	0.103(7)	0.028(4)	0.062(5)	0.030(4)
C(14)	4e	0.50	0.245(1)	0.3735(7)	0.248(1)	0.064(6)	0.061(6)	0.049(5)	0.013(4)	0.005(4)	−0.006(4)
C(14A)	4e	0.50	0.310(1)	0.4165(8)	0.289(1)	0.11(1)	0.053(6)	0.095(9)	0.027(6)	0.061(8)	0.022(6)
C(15)	4e		0.8785(3)	0.2106(2)	0.3182(3)	0.054(2)	0.022(1)	0.039(2)	−0.005(1)	−0.003(1)	−0.003(1)
C(16)	4e		0.9830(3)	0.1346(2)	0.4104(3)	0.041(2)	0.027(1)	0.030(1)	0.000(1)	0.009(1)	−0.001(1)
C(17)	4e		1.1298(3)	0.1534(2)	0.4214(3)	0.053(2)	0.030(2)	0.042(2)	−0.010(1)	0.024(2)	−0.006(1)
C(18)	4e		1.2381(3)	0.0829(2)	0.5199(4)	0.041(2)	0.042(2)	0.071(2)	−0.001(1)	0.031(2)	−0.001(2)
C(19)	4e		1.3848(4)	0.1014(3)	0.5291(5)	0.056(2)	0.062(2)	0.127(4)	−0.010(2)	0.060(3)	−0.020(3)

References

1. Pastorek, R.; Kameníček, J.; Cvek, B.; Pavlíček, M.; Šindelář, Z.: Nickel(II) di(pentyl)dithiocarbamates with P-ligands in the coordination sphere. *J. Coord. Chem.*, in press.
2. Nardelli, M.: PARST 95. Program for calculating molecular parameters from crystal structure parameters. *J. Appl. Crystallogr.* **28** (1995) 659-689.
3. Maxfield, P. L.: Dialkylthiocarbamatonickel complexes containing substituted phosphines. *Inorg. Nucl. Chem. Lett.* **6** (1970) 693-696.
4. Sheldrick, G. M.: Phase annealing in SHELX-90 – direct methods for larger structures. *Acta Crystallogr. A* **46** (1990) 467-473.
5. Sheldrick, G. M.: SHELXL-97. Program for crystal structure refinement. University of Göttingen, Germany 1997.