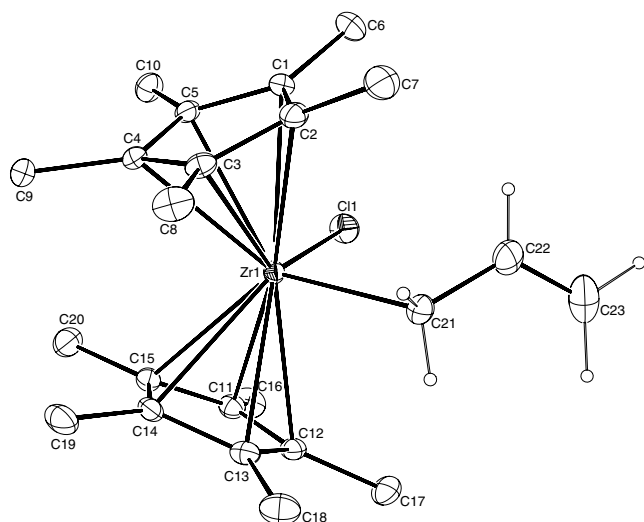


Crystal structure of chlorobis(η^5 -pentamethylcyclopentadienyl)(η^1 -2-propenyl)zirconium, $C_{23}H_{35}ClZr$

R. D. Ernst*, B. G. Harvey and A. M. Arif

University of Utah, Department of Chemistry, 315 S. 1400 E., Rm. 2020, Salt Lake City, Utah 84112-0850, USA

Received October 19, 2002, accepted and available on-line December 12, 2002; CCDC-No. 1267/954



Abstract

$C_{23}H_{35}ClZr$, monoclinic, $P12_1/n1$ (No. 14), $a = 8.6485(1)$ Å, $b = 26.9339(5)$ Å, $c = 9.9577(2)$ Å, $\beta = 112.3270(9)^\circ$, $V = 2145.7$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.031$, $wR_{\text{ref}}(F^2) = 0.067$, $T = 150$ K.

Source of material

The compound was synthesized as described previously [1] and was crystallized by evaporation of a saturated hexane solution.

Discussion

The structural result reveals that the compound is a 16 electron σ allyl, rather than an 18 electron π allyl, complex. The average Zr—C distance for the pentamethylcyclopentadienyl ligands is 2.559(7) Å, although the distances for the carbon atoms close to the Cl or allyl ligands are longer than the others. The Zr—C21 and Zr—Cl distances are shorter, at 2.325(2) Å and 2.4362(5) Å, respectively. As is normal, the methyl substituents tilt out of the cyclopentadienyl planes by an average of 7.6°, away from the zirconium center.

Table 1. Data collection and handling.

Crystal:	yellow plate, size 0.08 × 0.13 × 0.23 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	6.40 cm ⁻¹
Diffractometer, scan mode:	Nonius Kappa CCD, ϕ/ω
$2\theta_{\text{max}}$:	54.96°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	7788, 4753
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3934
$N(\text{param})_{\text{refined}}$:	366
Programs:	SIR97 [2], SHELXL-97 [3]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(6A)	4e	0.064(4)	0.045(1)	0.038(3)	0.059(9)
H(6B)	4e	0.030(3)	0.094(1)	-0.065(3)	0.056(8)
H(6C)	4e	0.068(4)	0.040(1)	-0.109(3)	0.060(9)
H(7A)	4e	0.442(3)	-0.032(1)	0.128(3)	0.059(9)
H(7B)	4e	0.279(5)	-0.016(1)	0.155(4)	0.08(1)
H(7C)	4e	0.458(3)	-0.014(1)	0.275(3)	0.052(8)
H(8A)	4e	0.739(3)	0.011(1)	0.185(3)	0.047(8)
H(8B)	4e	0.758(3)	0.035(1)	0.322(3)	0.046(8)
H(8C)	4e	0.823(3)	0.061(1)	0.215(3)	0.044(8)
H(9A)	4e	0.658(3)	0.1215(9)	-0.082(3)	0.036(7)
H(9B)	4e	0.763(4)	0.134(1)	0.077(3)	0.055(9)
H(9C)	4e	0.631(3)	0.170(1)	-0.016(3)	0.053(8)
H(10A)	4e	0.223(4)	0.139(1)	-0.216(3)	0.059(9)
H(10B)	4e	0.167(4)	0.164(1)	-0.106(3)	0.053(8)
H(10C)	4e	0.338(3)	0.180(1)	-0.123(3)	0.046(7)
H(16A)	4e	0.241(3)	0.243(1)	0.361(3)	0.048(8)
H(16B)	4e	0.284(3)	0.263(1)	0.232(3)	0.050(8)
H(16C)	4e	0.374(4)	0.284(1)	0.387(3)	0.058(9)
H(17A)	4e	0.486(3)	0.154(1)	0.663(3)	0.047(8)
H(17B)	4e	0.326(3)	0.1719(9)	0.551(3)	0.037(7)
H(17C)	4e	0.452(3)	0.211(1)	0.646(3)	0.039(7)
H(18A)	4e	0.721(4)	0.114(1)	0.661(4)	0.08(1)
H(18B)	4e	0.802(4)	0.095(1)	0.573(3)	0.08(1)
H(18C)	4e	0.877(5)	0.140(1)	0.664(4)	0.10(1)
H(19A)	4e	0.890(4)	0.135(1)	0.349(3)	0.06(1)
H(19B)	4e	0.956(4)	0.180(1)	0.433(3)	0.067(9)
H(19C)	4e	0.869(4)	0.186(1)	0.267(4)	0.08(1)
H(20A)	4e	0.666(4)	0.237(1)	0.146(3)	0.055(9)
H(20B)	4e	0.475(4)	0.235(1)	0.078(3)	0.062(9)
H(20C)	4e	0.568(3)	0.277(1)	0.187(3)	0.054(8)
H(21A)	4e	0.507(3)	0.045(1)	0.424(3)	0.044(7)
H(21B)	4e	0.470(3)	0.0848(9)	0.521(3)	0.036(7)
H(22)	4e	0.208(3)	0.032(1)	0.325(3)	0.051(8)
H(23A)	4e	0.059(4)	0.043(1)	0.477(3)	0.063(9)
H(23B)	4e	0.236(4)	0.074(1)	0.595(3)	0.07(1)

* Correspondence author (e-mail: ernst@chemistry.chem.utah.edu)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Zr(1)	4e	0.42750(2)	0.127924(7)	0.25744(2)	0.0154(1)	0.0179(1)	0.0173(1)	−0.00074(8)	0.00710(8)	0.00058(8)
Cl(1)	4e	0.14086(6)	0.15763(2)	0.18191(6)	0.0180(3)	0.0404(3)	0.0351(3)	0.0027(2)	0.0106(2)	−0.0039(2)
C(1)	4e	0.2810(3)	0.07364(8)	0.0322(2)	0.022(1)	0.023(1)	0.020(1)	−0.0054(8)	0.0078(9)	−0.0050(8)
C(2)	4e	0.4104(3)	0.04351(8)	0.1267(2)	0.032(1)	0.019(1)	0.022(1)	−0.0017(9)	0.011(1)	−0.0021(9)
C(3)	4e	0.5658(3)	0.06580(8)	0.1444(2)	0.024(1)	0.023(1)	0.019(1)	0.0053(9)	0.0072(9)	−0.0026(8)
C(4)	4e	0.5318(2)	0.10865(7)	0.0543(2)	0.022(1)	0.020(1)	0.017(1)	−0.0005(8)	0.0092(9)	−0.0027(8)
C(5)	4e	0.3558(2)	0.11415(8)	−0.0115(2)	0.023(1)	0.022(1)	0.015(1)	−0.0004(8)	0.0067(9)	−0.0015(8)
C(6)	4e	0.0970(3)	0.0621(1)	−0.0274(3)	0.026(1)	0.037(2)	0.034(2)	−0.011(1)	0.008(1)	−0.005(1)
C(7)	4e	0.3932(4)	−0.00809(9)	0.1756(3)	0.051(2)	0.021(1)	0.035(2)	−0.002(1)	0.017(1)	0.001(1)
C(8)	4e	0.7332(3)	0.0418(1)	0.2230(3)	0.029(1)	0.033(1)	0.030(1)	0.012(1)	0.007(1)	−0.001(1)
C(9)	4e	0.6554(3)	0.13514(9)	0.0066(3)	0.027(1)	0.033(1)	0.025(1)	−0.006(1)	0.015(1)	−0.005(1)
C(10)	4e	0.2634(3)	0.15283(9)	−0.1211(3)	0.029(1)	0.031(1)	0.023(1)	0.003(1)	0.009(1)	0.006(1)
C(11)	4e	0.4654(3)	0.21617(8)	0.3643(2)	0.023(1)	0.021(1)	0.025(1)	−0.0003(8)	0.0090(9)	−0.0042(9)
C(12)	4e	0.5177(3)	0.18349(8)	0.4842(2)	0.020(1)	0.026(1)	0.020(1)	−0.0033(8)	0.0079(9)	−0.0070(9)
C(13)	4e	0.6642(3)	0.15850(8)	0.4868(2)	0.019(1)	0.025(1)	0.022(1)	0.0005(8)	0.0043(9)	−0.0052(9)
C(14)	4e	0.7009(3)	0.17585(8)	0.3676(2)	0.019(1)	0.027(1)	0.027(1)	−0.0076(9)	0.0114(9)	−0.0108(9)
C(15)	4e	0.5739(3)	0.21066(8)	0.2893(2)	0.030(1)	0.022(1)	0.025(1)	−0.0083(9)	0.014(1)	−0.0069(9)
C(16)	4e	0.3313(3)	0.2547(1)	0.3337(3)	0.038(2)	0.026(1)	0.044(2)	0.008(1)	0.012(1)	−0.003(1)
C(17)	4e	0.4388(3)	0.1804(1)	0.5942(3)	0.034(1)	0.042(2)	0.028(1)	−0.001(1)	0.017(1)	−0.007(1)
C(18)	4e	0.7764(3)	0.1251(1)	0.6040(3)	0.027(1)	0.039(2)	0.031(1)	0.005(1)	0.000(1)	−0.001(1)
C(19)	4e	0.8645(3)	0.1678(1)	0.3523(3)	0.023(1)	0.054(2)	0.045(2)	−0.011(1)	0.017(1)	−0.023(2)
C(20)	4e	0.5672(4)	0.2428(1)	0.1634(3)	0.061(2)	0.026(1)	0.033(2)	−0.014(1)	0.025(2)	−0.005(1)
C(21)	4e	0.4230(3)	0.07010(9)	0.4292(3)	0.032(1)	0.027(1)	0.023(1)	−0.002(1)	0.011(1)	0.003(1)
C(22)	4e	0.2581(3)	0.0501(1)	0.4076(3)	0.036(1)	0.043(2)	0.027(1)	−0.002(1)	0.007(1)	0.012(1)
C(23)	4e	0.1761(4)	0.0555(1)	0.4955(4)	0.036(2)	0.055(2)	0.060(2)	0.004(1)	0.022(2)	0.025(2)

Acknowledgments. We thank the University of Utah and the National Science Foundation for partial support of this work.

References

1. Tjaden, E. B.; Stryker, J. M.: Zirconacyclobutanes from Thermal Rearrangement of Permethylzirconocene Bis(allyl) and Related Complexes. An Unprecedented Synthesis of B-Substituted Metallacyclobutanes. *J. Amer. Chem. Soc.* **115** (1993) 2083-2085.
2. Altomare, A.; Burla, M. C.; Cascarano, G. L.; Giacovazzo, C.; Guagliardi, A.; Moliterni, A. G. G.; Polidori, G.; Spagna, R.: SIR97. *J. Appl. Crystallogr.* **32** (1999) 115-119.
3. Sheldrick, G. M.: SHELXL-97. Program for Crystal Structure Solution, University of Göttingen, Germany 1997.