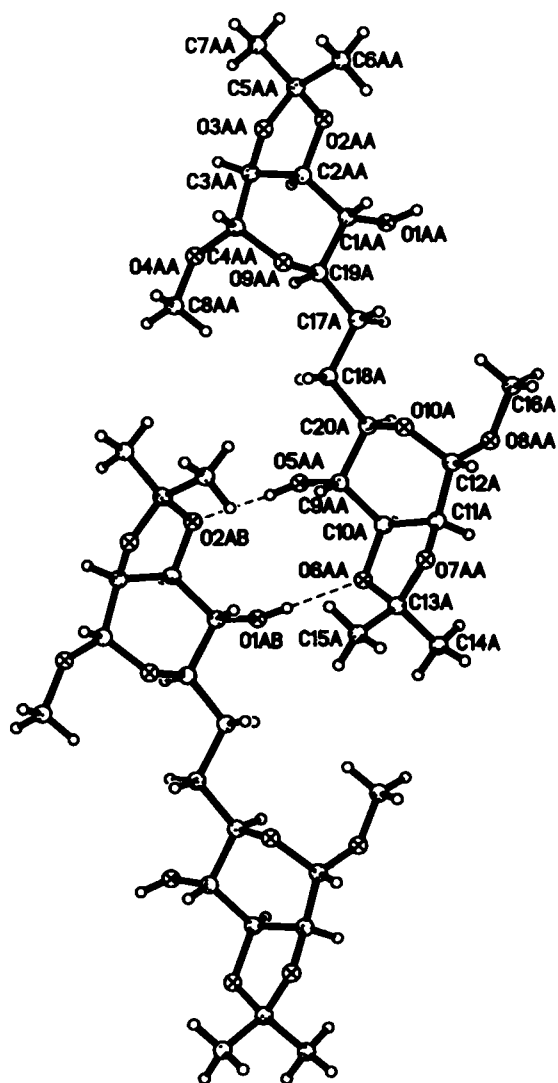


Crystal structure of 6,6-bis(methyl2,3-*O*-isopropylidene-6-deoxy- α -D-mannopyranoside), $C_{20}H_{34}O_{10}$

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title compound was obtained after separation from the main product (2*R*/3*S*,4*S*,5*R*)-2-[-*N*-(*tert*-butoxycarbonyl)-hydrazino]-3-4-isopropylidendioxy-5-vinyltetrahydrofuran [ratio 3:1] by flash chromatographie using petroleum ether / ethyl acetate in the form of colourless crystals: mp 333 K – 335 K; $[\alpha]_{20}^D = +27$ ($c = 0.50$, in MeOH).

Discussion

The compound crystallizes with two independent molecules in the asymmetric unit. We observe a strong intermolecular hydrogen bond interaction between the hydroxy groups (O1, O5) and the oxygen atoms (O2, O6) of the dioxolan moieties. The H1A...O6A, H5A...O2A, H1B...O6B and H5B...O2B distances are 2.08 Å, 2.07 Å, 2.11 Å and 2.06 Å, respectively. The bond angles of these hydrogen bonds are nearly linear ($\sim 180^\circ$).

Table 1. Data collection and handling.

Crystal:	colourless block, size 0.15 × 0.2 × 0.5 mm
Wavelength:	Cu $K\alpha$ radiation (1.54178 Å)
μ :	8.56 cm ⁻¹
Diffractometer, scan mode:	Siemens P4, ω
$2\theta_{\max}$:	134.98°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	4876, 4195
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3069
$N(\text{param})_{\text{refined}}$:	546
Programs:	SHELXS-97 [4], SHELXL-97 [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1AA)	2a	0.6842	0.0150	1.0846	0.089
H(1A)	2a	0.5409	0.0767	1.0389	0.058
H(2A)	2a	0.7868	0.1442	1.0634	0.059
H(3A)	2a	0.6792	0.2294	1.0401	0.066
H(4A)	2a	0.4682	0.2196	0.8895	0.059
H(5A)	2a	0.7080	0.0677	0.3319	0.100
H(6A1)	2a	0.4754	0.1377	1.3956	0.157
H(6A2)	2a	0.4076	0.1203	1.2591	0.157
H(6A3)	2a	0.3692	0.1796	1.3241	0.157
H(7A1)	2a	0.6648	0.2156	1.4037	0.169
H(7A2)	2a	0.5648	0.2609	1.3352	0.169
H(7A3)	2a	0.7052	0.2453	1.2722	0.169
H(8A1)	2a	0.6579	0.2483	0.6237	0.120
H(8A2)	2a	0.5086	0.2528	0.6770	0.120
H(8A3)	2a	0.5694	0.1920	0.6408	0.120
H(9A)	2a	0.5224	0.0303	0.3662	0.061
H(10A)	2a	0.7002	-0.0663	0.3607	0.058
H(11A)	2a	0.5360	-0.1321	0.3768	0.061

Abstract

$C_{20}H_{34}O_{10}$, monoclinic, $P12_11$ (No. 4), $a = 9.6895$ Å, $b = 23.0104$ Å, $c = 10.1973$ Å, $\beta = 92.239^\circ$, $V = 2271.8$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.062$, $wR_{\text{ref}}(F^2) = 0.196$, $T = 293$ K.

Source of material

The title compound has been obtained by heating of methyl 6-deoxy-6-iodo-2,3-*O*-isopropylidene- α -D-mannopyranoside [1], Zn, NH₄Cl, vitamin B 12, and NH₂NHBoc in methanol [2,3]. The

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Table 2. Continued.

Atom	Site	x	y	z	<i>U</i> _{iso}
H(12A)	2a	0.3453	-0.0960	0.5148	0.071
H(14A)	2a	0.5524	-0.1285	0.0238	0.125
H(14B)	2a	0.4255	-0.1587	0.0860	0.125
H(14C)	2a	0.5673	-0.1553	0.1650	0.125
H(15A)	2a	0.4209	-0.0353	-0.0098	0.134
H(15B)	2a	0.3604	-0.0033	0.1109	0.134
H(15C)	2a	0.2879	-0.0593	0.0542	0.134
H(16A)	2a	0.5314	-0.1477	0.7849	0.156
H(16B)	2a	0.3804	-0.1322	0.7359	0.156
H(16C)	2a	0.4866	-0.0825	0.7694	0.156
H(17A)	2a	0.4909	0.0329	0.8013	0.071
H(17B)	2a	0.6459	0.0136	0.8025	0.071
H(18A)	2a	0.5070	0.0797	0.6051	0.076
H(18B)	2a	0.6687	0.0774	0.6129	0.076
H(19A)	2a	0.7099	0.1125	0.8318	0.057
H(20A)	2a	0.6536	-0.0259	0.5861	0.063
H(1BB)	2a	0.2024	0.2297	0.5032	0.096
H(1B)	2a	0.0277	0.2730	0.4419	0.057
H(2B)	2a	0.2364	0.3590	0.4673	0.057
H(3B)	2a	0.0843	0.4333	0.4531	0.066
H(4B)	2a	-0.0869	0.4144	0.2742	0.069
H(5B)	2a	0.1949	0.2782	-0.2674	0.093
H(6B1)	2a	-0.0788	0.3162	0.7810	0.113
H(6B2)	2a	-0.1185	0.2927	0.6403	0.113
H(6B3)	2a	-0.2046	0.3431	0.7008	0.113

Table 2. Continued.

Atom	Site	x	y	z	<i>U</i> _{iso}
H(7B1)	2a	0.0330	0.4131	0.8078	0.116
H(7B2)	2a	-0.0900	0.4431	0.7296	0.116
H(7B3)	2a	0.0612	0.4500	0.6824	0.116
H(8B1)	2a	0.1430	0.4545	0.0477	0.167
H(8B2)	2a	-0.0148	0.4553	0.0766	0.167
H(8B3)	2a	0.0581	0.3965	0.0441	0.167
H(9B)	2a	0.0427	0.2205	-0.2235	0.059
H(10B)	2a	0.2840	0.1499	-0.2251	0.058
H(11B)	2a	0.1641	0.0670	-0.2097	0.066
H(12B)	2a	-0.0347	0.0796	-0.0676	0.063
H(14D)	2a	0.1401	0.0760	-0.5721	0.155
H(14E)	2a	0.0328	0.0355	-0.5059	0.155
H(14F)	2a	0.1784	0.0460	-0.4375	0.155
H(15D)	2a	-0.0283	0.1576	-0.5815	0.131
H(15E)	2a	-0.0872	0.1809	-0.4505	0.131
H(15F)	2a	-0.1417	0.1221	-0.5107	0.131
H(16D)	2a	0.1714	0.0529	0.2026	0.121
H(16E)	2a	0.0197	0.0460	0.1457	0.121
H(16F)	2a	0.0770	0.1078	0.1829	0.121
H(17C)	2a	-0.0070	0.2286	0.2039	0.077
H(17D)	2a	0.1525	0.2174	0.2114	0.077
H(18C)	2a	-0.0020	0.2689	0.0047	0.077
H(18D)	2a	0.1548	0.2849	0.0210	0.077
H(19B)	2a	0.1858	0.3199	0.2367	0.063
H(20B)	2a	0.2109	0.1855	-0.0025	0.059

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
O(1A)	2a	0.7240(4)	0.0388(2)	1.0396(4)	0.066(2)	0.048(2)	0.064(3)	0.013(2)	0.006(2)	0.006(2)
C(1A)	2a	0.6328(6)	0.0857(2)	1.0076(5)	0.056(3)	0.043(3)	0.045(3)	0.001(2)	0.006(2)	0.001(2)
O(2A)	2a	0.6630(4)	0.1390(2)	1.2137(4)	0.075(2)	0.055(2)	0.042(2)	0.020(2)	-0.002(2)	-0.005(2)
C(2A)	2a	0.6874(6)	0.1408(2)	1.0760(5)	0.055(3)	0.047(3)	0.046(3)	0.008(2)	0.002(2)	0.000(2)
O(3A)	2a	0.5070(5)	0.2018(2)	1.1181(4)	0.087(3)	0.078(3)	0.045(2)	0.037(3)	0.003(2)	0.000(2)
C(3A)	2a	0.6157(6)	0.1964(3)	1.0293(6)	0.059(3)	0.055(3)	0.050(3)	0.008(3)	-0.001(2)	-0.005(3)
O(4A)	2a	0.6467(4)	0.2234(2)	0.8085(4)	0.071(3)	0.060(2)	0.051(2)	-0.009(2)	0.002(2)	0.002(2)
C(4A)	2a	0.5522(6)	0.1958(2)	0.8900(5)	0.060(3)	0.040(3)	0.048(3)	0.001(2)	-0.001(2)	0.001(2)
O(5A)	2a	0.7255(4)	0.0406(2)	0.3821(4)	0.070(3)	0.065(3)	0.065(3)	-0.027(2)	-0.003(2)	0.002(2)
C(5A)	2a	0.5591(8)	0.1815(3)	1.2431(6)	0.092(4)	0.066(4)	0.043(3)	0.037(4)	0.000(3)	-0.004(3)
O(6A)	2a	0.5870(4)	-0.0455(2)	0.2006(4)	0.066(2)	0.063(2)	0.044(2)	-0.015(2)	0.002(2)	0.001(2)
C(6A)	2a	0.442(1)	0.1521(5)	1.3118(9)	0.106(6)	0.120(8)	0.091(6)	0.056(6)	0.038(5)	0.030(5)
O(7A)	2a	0.3945(4)	-0.0849(2)	0.2824(4)	0.064(2)	0.073(3)	0.047(2)	-0.014(2)	0.000(2)	-0.002(2)
C(7A)	2a	0.630(1)	0.2303(4)	1.3207(8)	0.21(1)	0.076(5)	0.054(4)	0.045(7)	-0.035(6)	-0.021(4)
O(8A)	2a	0.5183(6)	-0.1213(2)	0.6007(4)	0.109(4)	0.057(2)	0.043(2)	-0.007(3)	0.010(2)	0.005(2)
C(8A)	2a	0.5910(9)	0.2296(4)	0.6765(6)	0.107(6)	0.093(5)	0.040(3)	-0.029(5)	0.005(3)	0.008(3)
C(9A)	2a	0.6016(6)	0.0089(3)	0.4045(5)	0.052(3)	0.050(3)	0.051(3)	-0.007(2)	0.001(2)	0.004(2)
O(9A)	2a	0.5155(4)	0.1404(2)	0.8428(4)	0.053(2)	0.052(2)	0.053(2)	-0.006(2)	-0.005(2)	-0.002(2)
C(10A)	2a	0.6087(6)	-0.0495(2)	0.3412(5)	0.049(3)	0.049(3)	0.047(3)	-0.005(2)	-0.001(2)	-0.002(2)
O(10A)	2a	0.4487(4)	-0.0260(2)	0.5622(4)	0.060(2)	0.058(2)	0.059(2)	-0.009(2)	0.010(2)	-0.010(2)
C(11A)	2a	0.4990(6)	-0.0926(3)	0.3823(5)	0.059(3)	0.055(3)	0.038(3)	-0.006(3)	0.002(2)	0.000(2)
C(12A)	2a	0.4421(7)	-0.0835(3)	0.5174(6)	0.066(3)	0.058(3)	0.053(3)	-0.014(3)	0.004(3)	-0.008(3)
C(13A)	2a	0.4635(6)	-0.0774(3)	0.1625(6)	0.062(3)	0.064(4)	0.044(3)	-0.014(3)	0.003(2)	-0.001(3)
C(14A)	2a	0.506(1)	-0.1354(4)	0.1039(8)	0.119(6)	0.068(4)	0.064(4)	-0.008(4)	0.005(4)	-0.020(4)
C(15A)	2a	0.3752(8)	-0.0405(5)	0.0712(8)	0.070(4)	0.124(7)	0.073(5)	0.004(5)	-0.011(4)	0.025(5)
C(16A)	2a	0.476(1)	-0.1209(4)	0.7337(7)	0.19(1)	0.071(5)	0.048(4)	-0.023(6)	0.026(5)	0.003(4)
C(17A)	2a	0.5831(7)	0.0452(3)	0.7797(6)	0.082(4)	0.050(3)	0.045(3)	-0.007(3)	-0.001(3)	-0.004(3)
C(18A)	2a	0.5858(8)	0.0558(3)	0.6317(6)	0.088(4)	0.051(3)	0.050(3)	-0.018(3)	0.003(3)	-0.001(3)
C(19A)	2a	0.6221(6)	0.0968(2)	0.8607(5)	0.052(3)	0.041(3)	0.050(3)	-0.001(2)	0.002(2)	-0.001(2)
C(20A)	2a	0.5825(6)	0.0007(3)	0.5509(6)	0.060(3)	0.048(3)	0.048(3)	-0.005(3)	-0.004(2)	-0.005(2)
O(1B)	2a	0.2238(5)	0.2510(2)	0.4429(4)	0.069(2)	0.061(3)	0.064(3)	0.025(2)	0.011(2)	0.005(2)
C(1B)	2a	0.1143(6)	0.2896(2)	0.4123(5)	0.059(3)	0.037(3)	0.047(3)	0.004(2)	0.006(2)	0.002(2)
O(2B)	2a	0.1143(4)	0.3444(2)	0.6178(4)	0.053(2)	0.062(2)	0.042(2)	0.014(2)	0.002(2)	-0.003(2)
C(2B)	2a	0.1405(6)	0.3471(2)	0.4790(5)	0.049(3)	0.049(3)	0.044(3)	0.007(2)	0.004(2)	-0.004(2)
O(3B)	2a	-0.0737(4)	0.3854(2)	0.5090(4)	0.065(2)	0.068(3)	0.053(2)	0.020(2)	0.004(2)	-0.001(2)
C(3B)	2a	0.0436(6)	0.3954(3)	0.4310(6)	0.071(4)	0.043(3)	0.051(3)	0.004(3)	-0.003(3)	0.002(2)
O(4B)	2a	0.1023(6)	0.4247(2)	0.2233(4)	0.110(4)	0.058(2)	0.049(2)	-0.015(3)	0.002(2)	0.008(2)

Table 3. Continued.

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
C(4B)	2a	0.0011(7)	0.3937(2)	0.2867(6)	0.084(4)	0.040(3)	0.047(3)	0.007(3)	−0.001(3)	−0.001(2)
O(5B)	2a	0.2294(4)	0.2563(2)	−0.2118(4)	0.065(2)	0.058(2)	0.063(3)	−0.015(2)	−0.009(2)	0.009(2)
C(5B)	2a	−0.0211(6)	0.3700(3)	0.6373(6)	0.054(3)	0.059(3)	0.052(3)	0.014(3)	0.006(2)	−0.002(3)
O(6B)	2a	0.1607(4)	0.1547(2)	−0.3862(4)	0.064(2)	0.064(2)	0.038(2)	−0.015(2)	0.001(2)	−0.004(2)
C(6B)	2a	−0.1139(7)	0.3267(3)	0.6949(8)	0.071(4)	0.063(4)	0.094(5)	0.006(3)	0.024(4)	0.011(4)
O(7B)	2a	−0.0045(4)	0.0986(2)	−0.2964(4)	0.070(2)	0.058(2)	0.055(2)	−0.015(2)	−0.005(2)	0.004(2)
C(7B)	2a	−0.0025(9)	0.4240(4)	0.7221(7)	0.095(5)	0.072(4)	0.064(4)	0.011(4)	0.011(4)	−0.006(4)
O(8B)	2a	0.1472(4)	0.0772(2)	0.0193(4)	0.073(3)	0.061(2)	0.051(2)	0.012(2)	0.002(2)	0.008(2)
C(8B)	2a	0.069(1)	0.4335(4)	0.0867(7)	0.19(1)	0.085(6)	0.054(4)	−0.036(7)	−0.003(6)	0.026(4)
C(9B)	2a	0.1340(6)	0.2107(3)	−0.1850(5)	0.055(3)	0.050(3)	0.043(3)	−0.002(2)	−0.002(2)	0.002(2)
O(9B)	2a	−0.0158(4)	0.3365(2)	0.2378(4)	0.073(3)	0.048(2)	0.052(2)	−0.001(2)	−0.007(2)	−0.003(2)
C(10B)	2a	0.1852(5)	0.1548(3)	−0.2462(5)	0.050(3)	0.050(3)	0.045(3)	−0.001(2)	−0.007(2)	−0.001(2)
O(10B)	2a	0.0146(4)	0.1594(2)	−0.0245(4)	0.059(2)	0.049(2)	0.051(2)	0.003(2)	0.010(2)	−0.002(2)
C(11B)	2a	0.1058(7)	0.1015(3)	−0.2019(5)	0.075(4)	0.050(3)	0.038(3)	−0.003(3)	0.000(2)	−0.002(2)
C(12B)	2a	0.0492(6)	0.1033(3)	−0.0666(5)	0.061(3)	0.049(3)	0.046(3)	0.003(3)	0.002(2)	0.007(2)
C(13B)	2a	0.0499(7)	0.1139(3)	−0.4199(6)	0.080(4)	0.064(4)	0.044(3)	−0.023(3)	−0.003(3)	−0.004(3)
C(14B)	2a	0.105(1)	0.0635(4)	−0.4900(8)	0.170(9)	0.077(5)	0.065(5)	−0.029(6)	0.023(5)	−0.021(4)
C(15B)	2a	−0.0621(8)	0.1466(4)	−0.4978(8)	0.095(5)	0.095(6)	0.069(4)	−0.033(5)	−0.036(4)	0.017(4)
C(16B)	2a	0.100(1)	0.0704(4)	0.1480(6)	0.125(6)	0.070(4)	0.048(4)	0.012(5)	0.013(4)	0.013(3)
C(17B)	2a	0.0819(8)	0.2453(3)	0.1847(6)	0.100(5)	0.044(3)	0.047(3)	0.004(3)	0.003(3)	−0.004(3)
C(18B)	2a	0.0873(8)	0.2549(3)	0.0372(6)	0.099(5)	0.047(3)	0.047(3)	0.006(3)	0.005(3)	−0.005(3)
C(19B)	2a	0.1026(7)	0.2999(2)	0.2646(6)	0.069(3)	0.042(3)	0.047(3)	0.007(3)	0.005(3)	−0.001(2)
C(20B)	2a	0.1238(6)	0.2015(2)	−0.0385(5)	0.062(3)	0.041(3)	0.045(3)	0.000(2)	0.000(2)	−0.003(2)

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References

- Kleban, M.; Kautz, U.; Greul, J.; Hilgers, P.; Kugler, R.; Dong, H.-Q.; Jäger, V.: Vitamin B12 Catalysis of Zinc-Mediated 6-Deoxy-6-iodopyranoside Fragmentation: A Mild and Convenient Preparation of ω -Unsaturated Hexose Derivatives (5-Hexenoses). *Synthesis* (2000) 1027-1033.
- Jäger, V.; Bierer, L.; Dong, H.-Q.; Palmer, A. M.: New Heterocyclic Structures from Unsaturated Aldehyde Derivatives. Inhibition Of α -L-Fucosidases. *J. Heterocycl. Chem.* **37** (2000) 455-465.
- Jäger, V.; Gültekin, Z.: Unpublished results, Stuttgart (2001).
- Sheldrick, G. M.: SHELXS-97, Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXL-97, Program for refining crystal structures. University of Göttingen, Germany 1997.