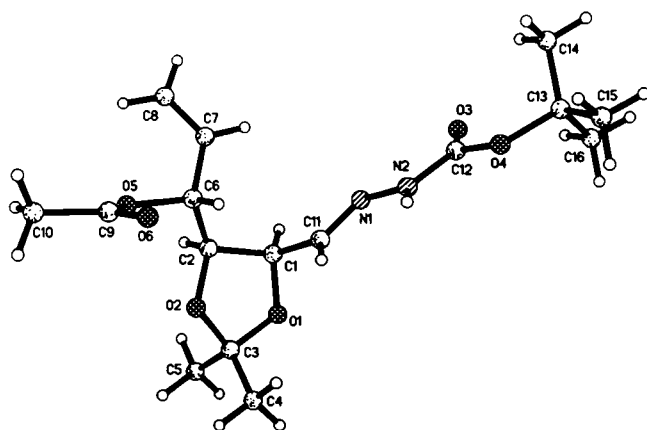


Crystal structure of 4-*O*-acetyl-2,3-*O*-isopropylidene-D-lyxo-5-hexenose-*N*-(*tert*-butoxycarbonyl) hydrazone, C₁₆H₂₆N₂O₆

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Abstract

C₁₆H₂₆N₂O₆, orthorhombic, *P*2₁2₁2₁ (No. 19), *a* = 11.2063(6) Å, *b* = 12.3253(8) Å, *c* = 13.973(2) Å, *V* = 1930.0 Å³, *Z* = 4, *R*_{gt}(*F*) = 0.048, *wR*_{ref}(*F*²) = 0.142, *T* = 293 K.

Source of material

The title compound has been obtained by heating 2,3-*O*-isopropylidene-D-lyxo-5,6-dideoxy-5-hexenose [1] and *N*-(*tert*-butoxycarbonyl) hydrazine in methanol [2–4]. Purification by column chromatography (ethyl acetate/petroleum ether) gave the title compound with a *E*:*Z* ratio > 99:1. Crystallization from *i*-PrOH/ethyl acetate gave colourless crystals (mp 402 K – 403 K, [*α*]₂₀^D = –27, *c* = 0.50, MeOH).

Discussion

The atoms of the terminal moieties of the molecule show large displacement parameters which indicate a strong thermal vibration. We observe a nearly linear inter-molecular hydrogen bond interaction between N2–H2 as donor and O2 of the dioxolan moiety as acceptor. The H2...O2 distance is 2.17 Å and the N2–H2...O2 angle is 175.5°. In the *ab*-view the orientation of the molecules shows a fishbone pattern. We also observe nonpolar channels along the *a*-axis made up by the *tert*-butyl groups.

Table 1. Data collection and handling.

Crystal:	colorless block, size 0.20 × 0.20 × 0.40 mm
Wavelength:	Cu K _α radiation (1.54178 Å)
<i>μ</i> :	7.51 cm ^{–1}
Diffractionmeter, scan mode:	Siemens P4, <i>ω</i>
2 θ _{max} :	134.82°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	2412, 2238
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ (<i>I</i> _{obs}), 1787
<i>N</i> (<i>param</i>) _{refined} :	218
Programs:	SHELXS-97 [5], SHELXL-97 [6]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	4a	1.1019	0.5343	0.4238	0.070
H(2)	4a	0.7346	0.5754	0.4976	0.068
H(2A)	4a	1.2318	0.6496	0.4922	0.068
H(4A)	4a	1.0777	0.9240	0.3294	0.133
H(4B)	4a	0.9712	0.8457	0.3523	0.133
H(4C)	4a	1.0324	0.8423	0.2515	0.133
H(5A)	4a	1.2853	0.8487	0.3185	0.127
H(5B)	4a	1.2470	0.7629	0.2415	0.127
H(5C)	4a	1.3057	0.7248	0.3377	0.127
H(6)	4a	1.0273	0.7010	0.6038	0.075
H(7)	4a	1.0683	0.5142	0.6224	0.107
H(8A)	4a	1.2448	0.5902	0.7371	0.143
H(8B)	4a	1.1907	0.4692	0.7354	0.143
H(10A)	4a	1.2965	0.8731	0.7440	0.168
H(10B)	4a	1.1971	0.8999	0.8194	0.168
H(10C)	4a	1.2154	0.9755	0.7303	0.168
H(11)	4a	0.8982	0.6696	0.4606	0.069
H(14A)	4a	0.5278	0.3147	0.6381	0.194
H(14B)	4a	0.6035	0.2426	0.5684	0.194
H(14C)	4a	0.4660	0.2225	0.5788	0.194
H(15A)	4a	0.3790	0.4669	0.4658	0.224
H(15B)	4a	0.3973	0.4548	0.5765	0.224
H(15C)	4a	0.3254	0.3666	0.5198	0.224
H(16A)	4a	0.4851	0.3541	0.3561	0.170
H(16B)	4a	0.4383	0.2473	0.4041	0.170
H(16C)	4a	0.5758	0.2664	0.3921	0.170

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
N(1)	4a	0.8900(2)	0.5141(2)	0.4620(2)	0.042(1)	0.045(1)	0.082(2)	–0.004(1)	0.005(1)	–0.009(2)
O(1)	4a	1.0865(3)	0.6624(2)	0.3391(2)	0.078(2)	0.066(2)	0.069(2)	–0.027(2)	0.011(1)	–0.019(2)
C(1)	4a	1.0710(3)	0.6085(3)	0.4284(3)	0.053(2)	0.040(2)	0.083(2)	–0.006(2)	0.009(2)	–0.014(2)
N(2)	4a	0.7701(2)	0.5157(2)	0.4834(2)	0.041(1)	0.039(1)	0.090(2)	–0.002(1)	0.008(1)	–0.004(2)
O(2)	4a	1.1391(2)	0.7807(2)	0.4572(2)	0.062(1)	0.047(1)	0.069(1)	–0.010(1)	–0.002(1)	–0.008(1)

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Table 3. Continued.

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
C(2)	4a	1.1489(3)	0.6744(3)	0.4966(3)	0.046(2)	0.044(2)	0.079(2)	-0.003(1)	0.004(2)	-0.008(2)
O(3)	4a	0.7522(2)	0.3337(2)	0.4657(3)	0.054(1)	0.047(1)	0.181(3)	-0.001(1)	0.014(2)	-0.013(2)
C(3)	4a	1.1316(3)	0.7696(3)	0.3545(3)	0.065(2)	0.053(2)	0.066(2)	-0.010(2)	0.003(2)	-0.013(2)
O(4)	4a	0.5957(2)	0.4407(2)	0.5004(2)	0.044(1)	0.054(1)	0.114(2)	-0.007(1)	0.016(1)	-0.015(2)
C(4)	4a	1.0455(5)	0.8528(4)	0.3187(3)	0.099(3)	0.083(3)	0.083(3)	0.009(3)	-0.019(2)	-0.010(3)
O(5)	4a	1.1850(2)	0.7536(2)	0.6495(2)	0.064(1)	0.064(2)	0.078(2)	-0.006(1)	-0.009(1)	-0.009(2)
C(5)	4a	1.2535(4)	0.7772(4)	0.3089(4)	0.076(2)	0.082(3)	0.095(3)	-0.023(2)	0.025(2)	-0.005(3)
C(6)	4a	1.1104(3)	0.6765(3)	0.6003(3)	0.056(2)	0.054(2)	0.079(2)	-0.006(2)	0.003(2)	-0.010(2)
O(6)	4a	1.0268(3)	0.8492(3)	0.6939(3)	0.092(2)	0.096(2)	0.158(3)	0.013(2)	0.008(2)	-0.047(3)
C(7)	4a	1.1198(5)	0.5675(4)	0.6452(4)	0.112(4)	0.070(3)	0.084(3)	-0.010(3)	0.010(3)	0.007(3)
C(8)	4a	1.1917(7)	0.5398(5)	0.7120(5)	0.149(5)	0.092(4)	0.117(4)	0.020(4)	0.034(4)	0.031(4)
C(9)	4a	1.1303(5)	0.8351(4)	0.6971(3)	0.093(3)	0.068(2)	0.069(2)	-0.012(3)	0.006(2)	-0.008(2)
C(10)	4a	1.2177(6)	0.9019(5)	0.7527(4)	0.144(5)	0.111(4)	0.081(3)	-0.042(4)	-0.003(3)	-0.031(3)
C(11)	4a	0.9410(3)	0.6055(3)	0.4526(3)	0.044(2)	0.047(2)	0.082(2)	-0.004(1)	0.005(2)	-0.010(2)
C(12)	4a	0.7093(3)	0.4196(3)	0.4815(3)	0.039(1)	0.045(2)	0.079(2)	-0.000(1)	0.004(2)	-0.000(2)
C(13)	4a	0.5045(3)	0.3543(3)	0.4997(3)	0.046(2)	0.062(2)	0.084(2)	-0.018(2)	0.007(2)	-0.013(2)
C(14)	4a	0.5275(5)	0.2767(5)	0.5781(5)	0.114(4)	0.157(5)	0.117(4)	-0.079(4)	-0.025(3)	0.049(4)
C(15)	4a	0.3914(4)	0.4162(5)	0.5170(6)	0.053(2)	0.112(4)	0.283(9)	-0.015(3)	0.043(4)	-0.067(5)
C(16)	4a	0.5006(4)	0.3008(5)	0.4046(4)	0.089(3)	0.159(5)	0.093(3)	-0.056(4)	0.000(3)	-0.021(4)

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