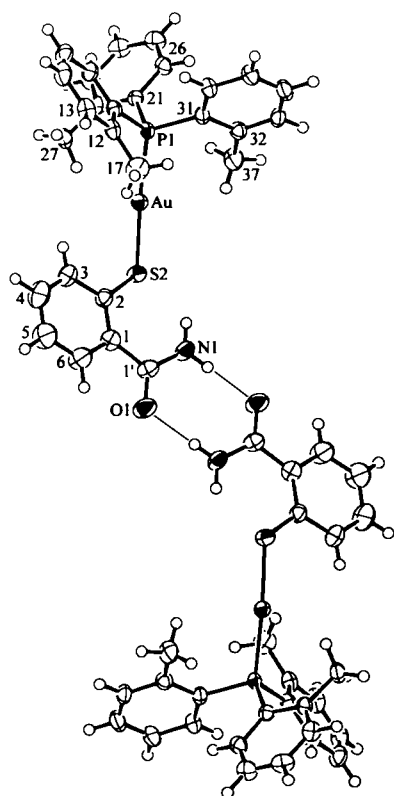


Crystal structure of benzamide-2-thiolato(tri-*o*-tolylphosphine)gold(I), $C_{28}H_{27}AuNOPS$

D. R. Smyth and E. R. T. Tiekink*¹

The University of Adelaide, Department of Chemistry, Australia 5005

Received June 7, 2002, accepted and available on-line August 10, 2002; CCDC-No. 1267/873



Abstract

$C_{28}H_{27}AuNOPS$, triclinic, $P\bar{1}$ (No. 2), $a = 11.731(2)$ Å, $b = 13.237(4)$ Å, $c = 9.245(3)$ Å, $\alpha = 105.06(2)^\circ$, $\beta = 91.32(2)^\circ$, $\gamma = 75.65(2)^\circ$, $V = 1341.7$ Å³, $Z = 2$, $R_{gt}(F) = 0.034$, $wR_{ref}(F^2) = 0.088$, $T = 173$ K.

Source of material

The title compound was prepared in 95% yield from the reaction between *o*-tol₃PAuCl and 2-mercaptobenzamide [1]. Colourless crystals were obtained from the slow evaporation of an acetonitrile solution of the compound (mp 437 K – 439 K).

Experimental details

The H atoms were placed in their geometrically calculated positions and included in the final refinement in the riding model approximation.

Discussion

The gold atom exists in the expected linear environment defined by the sulfur, $d(Au—S) = 2.295(1)$ Å, and phosphorus, $d(Au—P)$

$= 2.271(1)$ Å, atoms which define an angle of $174.97(4)^\circ$ at gold. Centrosymmetric molecules associate via hydrogen bonding interactions as shown in the figure. The parameters associated with these interactions are $d(H1a \cdots O1^i) = 1.97$ Å, $d(N1 \cdots O1^i) = 2.840(7)$ Å with an angle of 176° subtended at H1a for symmetry operation $i: -1-x, 1-y, 1-z$. The second amide hydrogen atom forms an intramolecular S \cdots H interaction so that $d(H1b \cdots S)$ is 2.33 Å and the angle at H1b is 131° .

Table 1. Data collection and handling.

Crystal:	colourless plate, size $0.04 \times 0.25 \times 0.25$ mm
Wavelength:	Mo $K\alpha$ radiation (0.7107 Å)
μ :	56.58 cm ⁻¹
Diffractometer, scan mode:	Rigaku AFC7R, $\omega/2\theta$
$2\theta_{max}$:	55°
$N(hkl)_{measured}$, $N(hkl)_{unique}$:	6440, 6141
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 4823
$N(param)_{refined}$:	301
Programs:	teXsan [2], DIRDIF92 [3], SHELXL-97 [4], DIFABS [5], ORTEPII [6]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(1a)	2i	-0.4080	0.4439	0.5327	0.046
H(1b)	2i	-0.2836	0.3794	0.4751	0.046
H(3)	2i	-0.0086	0.2643	0.0891	0.056
H(4)	2i	-0.0227	0.3836	-0.0570	0.085
H(5)	2i	-0.1870	0.5279	-0.0333	0.086
H(6)	2i	-0.3384	0.5541	0.1430	0.064
H(13)	2i	0.4006	0.3729	0.2863	0.048
H(14)	2i	0.5675	0.2473	0.1627	0.050
H(15)	2i	0.5901	0.0630	0.1388	0.049
H(16)	2i	0.4453	0.0063	0.2431	0.038
H(17a)	2i	0.2257	0.4228	0.4265	0.071
H(17b)	2i	0.2046	0.3422	0.5196	0.071
H(17c)	2i	0.1444	0.3418	0.3625	0.071
H(23)	2i	0.2788	-0.2491	0.0642	0.042
H(24)	2i	0.3713	-0.3421	0.2335	0.046
H(25)	2i	0.4064	-0.2470	0.4730	0.045
H(26)	2i	0.3417	-0.0592	0.5458	0.037
H(27a)	2i	0.2061	-0.0958	-0.0144	0.053
H(27b)	2i	0.2385	0.0132	0.0764	0.053
H(27c)	2i	0.1152	-0.0089	0.1111	0.053
H(33)	2i	0.1397	0.1417	0.9317	0.044
H(34)	2i	0.3002	0.2067	1.0190	0.049
H(35)	2i	0.4301	0.2322	0.8515	0.047
H(36)	2i	0.3971	0.1899	0.5942	0.039
H(37a)	2i	0.0349	0.0593	0.7446	0.068
H(37b)	2i	0.0895	0.0160	0.5763	0.068
H(37c)	2i	0.0031	0.1343	0.6313	0.068

* Correspondence author (e-mail: chmtert@nus.edu.sg)

¹ Current address: National University of Singapore, Department of Chemistry, Singapore 117543

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Au	2i	0.03792(2)	0.17399(2)	0.34937(2)	0.0269(1)	0.0271(1)	0.0283(1)	-0.00609(7)	-0.00153(7)	0.00679(7)
S(2)	2i	-0.1571(1)	0.2340(1)	0.3040(2)	0.0272(6)	0.0329(7)	0.0428(7)	-0.0046(5)	0.0006(5)	0.0120(5)
P(1)	2i	0.2258(1)	0.1067(1)	0.4075(1)	0.0273(6)	0.0270(6)	0.0210(5)	-0.0088(5)	-0.0011(4)	0.0083(5)
O(1)	2i	-0.4598(4)	0.5192(4)	0.3185(6)	0.050(3)	0.057(3)	0.073(3)	0.014(2)	0.021(2)	0.040(3)
N(1)	2i	-0.3510(4)	0.4231(4)	0.4630(5)	0.039(3)	0.030(2)	0.041(2)	-0.001(2)	0.011(2)	0.007(2)
C(1')	2i	-0.3666(5)	0.4578(4)	0.3397(7)	0.035(3)	0.025(3)	0.049(3)	-0.006(2)	0.003(2)	0.010(2)
C(1)	2i	-0.2638(5)	0.4264(4)	0.2266(6)	0.043(3)	0.029(3)	0.040(3)	-0.011(2)	0.002(2)	0.010(2)
C(2)	2i	-0.1663(5)	0.3356(5)	0.2084(6)	0.028(3)	0.043(3)	0.029(2)	-0.011(2)	0.001(2)	0.007(2)
C(3)	2i	-0.0763(6)	0.3232(5)	0.1011(7)	0.041(3)	0.057(4)	0.042(3)	-0.011(3)	0.008(3)	0.015(3)
C(4)	2i	-0.0846(7)	0.3940(7)	0.0140(9)	0.059(5)	0.099(6)	0.068(5)	-0.023(4)	0.020(4)	0.042(5)
C(5)	2i	-0.1813(7)	0.4797(7)	0.0283(9)	0.072(5)	0.074(5)	0.083(5)	-0.005(4)	0.018(4)	0.056(5)
C(6)	2i	-0.2710(6)	0.4953(5)	0.1340(8)	0.052(4)	0.045(4)	0.068(4)	-0.002(3)	0.014(3)	0.030(3)
C(11)	2i	0.3358(4)	0.1558(4)	0.3265(5)	0.028(2)	0.028(2)	0.023(2)	-0.012(2)	-0.004(2)	0.007(2)
C(12)	2i	0.3198(5)	0.2673(4)	0.3397(5)	0.040(3)	0.036(3)	0.024(2)	-0.019(2)	-0.004(2)	0.011(2)
C(13)	2i	0.4086(6)	0.2984(5)	0.2777(6)	0.057(4)	0.041(3)	0.032(3)	-0.030(3)	-0.006(2)	0.011(2)
C(14)	2i	0.5086(5)	0.2236(5)	0.2038(6)	0.046(3)	0.059(4)	0.033(3)	-0.032(3)	-0.002(2)	0.015(3)
C(15)	2i	0.5223(5)	0.1144(5)	0.1900(6)	0.034(3)	0.059(4)	0.034(3)	-0.019(3)	0.004(2)	0.011(3)
C(16)	2i	0.4360(5)	0.0811(4)	0.2518(5)	0.033(3)	0.034(3)	0.031(2)	-0.011(2)	0.002(2)	0.009(2)
C(17)	2i	0.2147(6)	0.3507(5)	0.4189(7)	0.059(4)	0.027(3)	0.058(4)	-0.012(3)	0.003(3)	0.011(3)
C(21)	2i	0.2735(4)	-0.0401(4)	0.3468(5)	0.024(2)	0.026(2)	0.028(2)	-0.008(2)	0.000(2)	0.009(2)
C(22)	2i	0.2546(4)	-0.0976(4)	0.1999(5)	0.031(3)	0.030(2)	0.026(2)	-0.009(2)	0.001(2)	0.011(2)
C(23)	2i	0.2917(5)	-0.2095(4)	0.1620(6)	0.040(3)	0.033(3)	0.034(3)	-0.013(2)	-0.001(2)	0.007(2)
C(24)	2i	0.3472(5)	-0.2654(4)	0.2625(7)	0.038(3)	0.025(3)	0.052(3)	-0.009(2)	-0.002(2)	0.010(2)
C(25)	2i	0.3670(5)	-0.2093(4)	0.4044(6)	0.042(3)	0.029(3)	0.046(3)	-0.010(2)	-0.002(2)	0.019(2)
C(26)	2i	0.3294(5)	-0.0975(4)	0.4469(6)	0.033(3)	0.034(3)	0.032(2)	-0.011(2)	-0.002(2)	0.016(2)
C(27)	2i	0.1987(5)	-0.0425(4)	0.0831(5)	0.046(3)	0.035(3)	0.023(2)	-0.007(2)	-0.002(2)	0.007(2)
C(31)	2i	0.2474(4)	0.1411(4)	0.6089(5)	0.034(3)	0.027(2)	0.021(2)	-0.006(2)	0.001(2)	0.007(2)
C(32)	2i	0.1688(5)	0.1258(4)	0.7105(5)	0.031(3)	0.033(3)	0.029(2)	-0.010(2)	-0.002(2)	0.011(2)
C(33)	2i	0.1918(5)	0.1514(4)	0.8622(6)	0.045(3)	0.038(3)	0.026(2)	-0.001(2)	0.008(2)	0.013(2)
C(34)	2i	0.2870(5)	0.1902(5)	0.9147(6)	0.054(4)	0.041(3)	0.023(2)	-0.006(3)	-0.007(2)	0.008(2)
C(35)	2i	0.3641(5)	0.2051(5)	0.8156(6)	0.043(3)	0.044(3)	0.031(3)	-0.015(3)	-0.008(2)	0.010(2)
C(36)	2i	0.3441(5)	0.1801(4)	0.6625(5)	0.040(3)	0.036(3)	0.026(2)	-0.014(2)	-0.003(2)	0.012(2)
C(37)	2i	0.0651(5)	0.0799(6)	0.6614(7)	0.039(3)	0.069(4)	0.042(3)	-0.028(3)	-0.001(2)	0.023(3)

Acknowledgments. The Australian Research Council and The University of Adelaide are thanked for support.

References

- Smyth, D. R.: Structural studies of phosphinegold(I) thiolates with potential medicinal applications. Ph. D. Thesis. The University of Adelaide, Australia 2001.
- teXsan: Single Crystal Structure Analysis Software. Version 1.05. Molecular Structure Corporation. The Woodlands, TX, USA 1997.
- Beurskens, P. T.; Admiraal, G.; Beurskens, G.; Bosman, W. P.; García-Granda, S.; Smits, J. M. M.; Smykalla, C.: The DIRDIF program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands 1992.
- Sheldrick, G. M.: SHELXL-97. Program for crystal structure refinement. University of Göttingen, Germany 1997.
- Walker, N.; Stuart, D.: An empirical method for correcting diffractometer data for absorption effects. *Acta Crystallogr.* **A39** (1983) 158-166.
- Johnson, C. K.: ORTEPII. Report ORNL-5138, Oak Ridge National Laboratory, Tennessee, USA 1976.