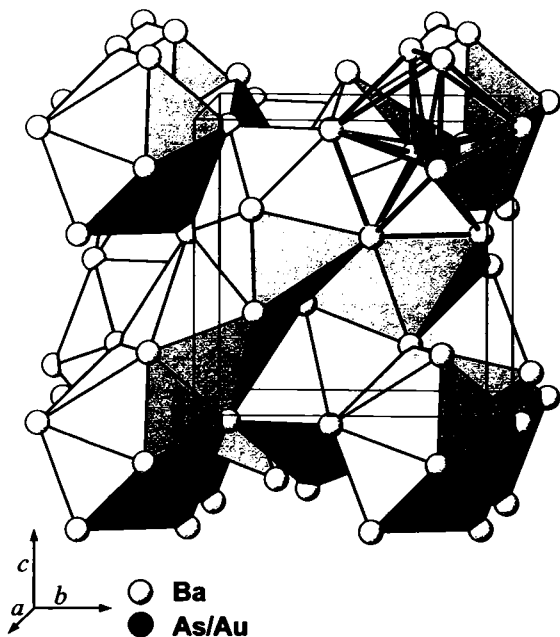


# Crystal structure of barium arsenide auride, Ba<sub>8</sub>As<sub>5</sub>Au

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## Abstract

As<sub>5</sub>AuBa<sub>8</sub>, cubic,  $\bar{I}43d$  (No. 220),  $a = 9.9607(7)$  Å,  $V = 988.3$  Å<sup>3</sup>,  $Z = 2$ ,  $R_{\text{gt}}(F) = 0.033$ ,  $wR_{\text{ref}}(F^2) = 0.053$ ,  $T = 293$  K.

## Source of material

Ba<sub>8</sub>As<sub>5</sub>Au was synthesized from BaAs, Ba and Au (ratio 5:3:1) in sealed tantalum ampoules at 1175 K. Single crystals were isolated from slowly cooled samples (cooling rate: 25 K/h).

## Experimental details

The absolute configuration in respect to the polarity of the space group was proven by changing the sign of the  $hkl$  indices. The  $wR(F^2)$  values for the possible orientations were 0.053 and 0.109, respectively. The first one represents the absolute configuration of the crystal studied.

## Discussion

Ba<sub>8</sub>As<sub>5</sub>Au crystallizes in the anti-Th<sub>3</sub>P<sub>4</sub> type of structure where arsenic and gold share the thorium position (12a). The site occupation factors were refined with the restraint  $\text{SOF}(12a) =$

$\text{SOF}(\text{As}) + \text{SOF}(\text{Au}) = 1$ , converging to  $\text{SOF}(\text{As}) \cong 5/6$  and  $\text{SOF}(\text{Au}) \cong 1/6$ , within standard deviations. Therefore  $\text{SOF}(\text{As}) = 5/6$  and  $\text{SOF}(\text{Au}) = 1/6$  were fixed, during the final refinements. This is compatible with the charge balance ( $8 \times \text{Ba}^{2+}$ ;  $5 \times \text{As}^{3-}$  and  $1 \times \text{Au}^-$ ). Similar occupancies have been reported for several other compounds as  $\text{Eu}_4\text{X}_2\text{Y}$  ( $\text{X} = \text{P, As, Sb, Bi}$ ;  $\text{Y} = \text{S, Se, Te}$ ) [1] or  $\text{Rb}_4\text{Cl}_2\text{O}$  [2] with  $\text{SOF}(\text{X}):\text{SOF}(\text{Y}) = 2:1$ . The existence of isotopic barium phosphide chloride  $\text{Ba}_{16}(\text{P}_{10.05}\text{Cl}_{1.86}\square_{0.09})$  [3] supports the assumption of an oxidation state of  $-1$  for gold [4]. Barium is surrounded by six anions, and the coordination polyhedron can be described either by a distorted octahedron or a twisted trigonal antiprism of As/Au. The Ba atoms form a 3D framework of condensed Ba<sub>8</sub> dodecahedra, which are centered by As or Au (see figure) ( $d(\text{As/Au}-\text{Ba}) = 3.302(1)$  Å ( $4\times$ ) and  $3.604(1)$  Å ( $4\times$ )). The Ba<sub>8</sub> dodecahedra with  $x_{\text{Ba}} = 0.06930(3)$  deviate from the regular one with  $x = 0.0833 = 1/12$ .

Table 1. Data collection and handling.

Crystal:	block with dark metallic lustre size $0.15 \times 0.20 \times 0.25$ mm
Wavelength:	Mo $K\alpha$ radiation ( $0.71073$ Å)
$\mu$ :	$313.28 \text{ cm}^{-1}$
Diffractionmeter, scan mode:	SMART APEX Bruker AXS, $\omega$
$2\theta_{\text{max}}$ :	$69.92^\circ$
$N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ :	6972, 363
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 349
$N(\text{param})_{\text{refined}}$ :	7
Programs:	SHELXTL-97 [5], ATOMS [6]

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Table 2. Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	Occ.	x	y	z	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>12</sub>	U <sub>13</sub>	U <sub>23</sub>
Ba(1)	16c		0.06930(3)	x	x	0.0187(2)	U <sub>11</sub>	U <sub>11</sub>	0.0011(1)	U <sub>12</sub>	U <sub>12</sub>
As(2)	12a	0.83	3/8	0	1/4	0.0214(4)	0.0149(3)	U <sub>22</sub>	0	0	0
Au(2)	12a	0.17	3/8	0	1/4	U <sub>11</sub> (As2)	U <sub>22</sub> (As2)	U <sub>22</sub> (As2)	0	0	0

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