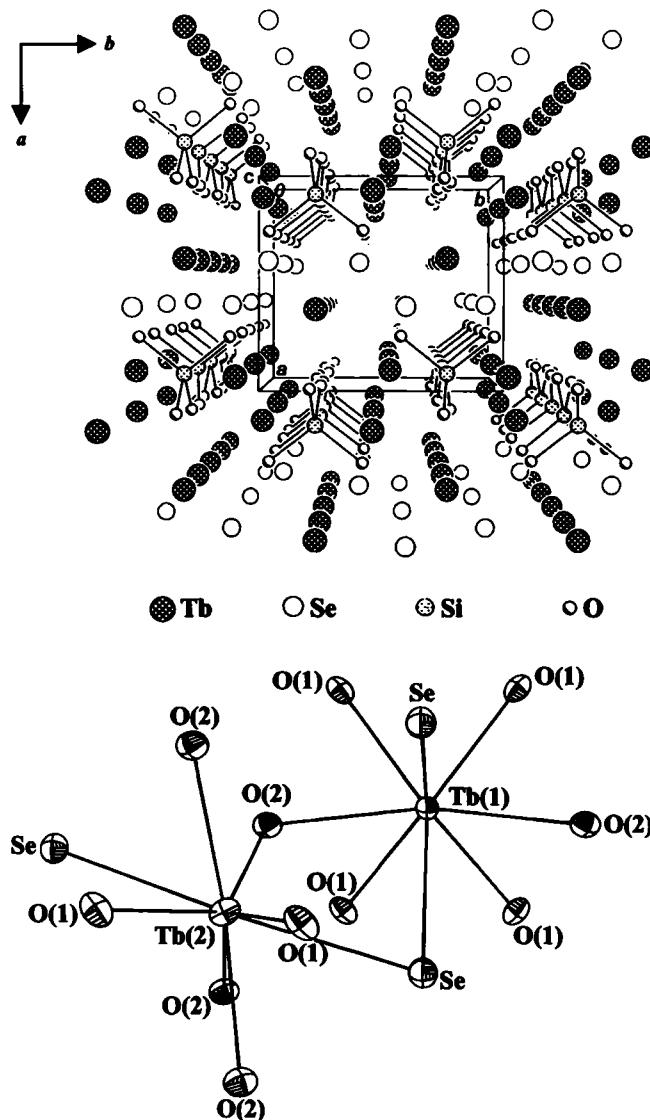


# Crystal structure of diterbium orthosilicate selenide, $\text{Tb}_2(\text{SiO}_4)\text{Se}$

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## Abstract

$\text{O}_4\text{SeSiTb}_2$ , orthorhombic,  $Pbcm$  (No. 57),  $a = 6.0387(4)$  Å,  $b = 6.9855(5)$  Å,  $c = 10.8131(7)$  Å,  $V = 456.1$  Å $^3$ ,  $Z = 4$ ,  $R_{\text{gt}}(F) = 0.030$ ,  $wR_{\text{ref}}(F^2) = 0.073$ ,  $T = 153$  K.

## Source of material

The reactive flux  $\text{Cs}_2\text{Se}_3$  was prepared by the stoichiometric reaction of Cs (Aldrich, 99.5%) and Se (Aldrich, 99.5%) in liquid  $\text{NH}_3$ . Clear needles of  $\text{Tb}_2(\text{SiO}_4)\text{Se}$  were obtained accidentally in the reaction of Tb (0.547 mmol, Alfa Aesar, 99.9%), Cd

(0.543 mmol, Alfa Aesar, 99.9%), Se (1.09 mmol, Aldrich, 99.5%),  $\text{CsCl}$  (0.947 mmol, Aldrich, 99.9%), and  $\text{Cs}_2\text{Se}_3$  (150 mg). The materials were mixed and sealed in an unprotected fused-silica tube that was then evacuated to  $10^{-4}$  Torr. The tube was heated to 1273 K, kept at 1273 K for 50 h, cooled at 4 K/h to 473 K, and then the furnace was turned off. The reaction mixture was washed with water, then  $N,N$ -dimethylformamide, and finally dried with acetone. Semi-quantitative energy dispersive spectroscopy (EDS) analysis confirmed the presence of Tb, Si, and Se in ratio 2:1:1. EDS also indicated the presence of O but provided no evidence for the presence of Cd, Cl, or Cs. The compound resulted from the reaction of the starting components with the fused-silica tube.

## Discussion

$\text{Tb}_2(\text{SiO}_4)\text{Se}$  is a new member of the  $\text{Ln}_2(\text{SiO}_4)\text{Se}$  series ( $\text{Ln} = \text{La}$  [1],  $\text{Nd}$  [2],  $\text{Sm}$  [3],  $\text{Dy}$  [3],  $\text{Ho}$  [3], and  $\text{Er}$  [4]). These isostructural compounds crystallize in space group  $Pbcm$  of the orthorhombic system. In this layered structure, infinite chains of  $\text{SiO}_4^{4-}$  tetrahedra extend along the  $c$  axis, separated by Ln and Se atoms (upper figure). There are two crystallographically distinct Tb atoms, both coordinated by six O atoms and two Se atoms. Atom Tb(1), located within the layer, is in a bicapped trigonal prism; atom Tb(2), located between the layers, is in a dodecahedron (lower figure). The Tb—O distances range from 2.377(5) Å to 2.535(5) Å, which may be compared with the range of 2.177 Å to 2.551 Å in  $\text{Tb}_2\text{O}_3$  [5]. The Tb—Se distances are 2.8958(11) Å, 2.9990(11) Å, and 3.0570(5) Å, compared with 2.876(1) Å to 2.984(1) Å in  $\text{Tb}_2\text{Se}_3$  [6]. Also isostructural with the present compound are the related tellurides  $\text{Ln}_2(\text{SiO}_4)\text{Te}$  ( $\text{Ln} = \text{Nd}$  [7],  $\text{Sm}$  [3,7] and  $\text{Pr}$  [8]). Most of these selenides and tellurides were prepared accidentally.

Table 1. Data collection and handling.

Crystal:	colorless needle, size 0.016 × 0.018 × 0.176 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
$\mu$ :	388.92 $\text{cm}^{-1}$
Diffractometer, scan mode:	Bruker SMART-CCD, $\Delta\omega = 0.3^\circ$
$2\theta_{\text{max}}$ :	57.74°
$N(hkl)$ measured, $N(hkl)$ unique:	3715, 605
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$ , 556
$N(\text{param})_{\text{refined}}$ :	42
Program:	SHELXTL [9]

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**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>11</sub>	<i>U</i> <sub>22</sub>	<i>U</i> <sub>33</sub>	<i>U</i> <sub>12</sub>	<i>U</i> <sub>13</sub>	<i>U</i> <sub>23</sub>
Tb(1)	4d	0.13113(7)	0.03044(7)	1/4	0.0062(3)	0.0049(3)	0.0046(3)	0.0006(2)	0	0
Tb(2)	4c	0.60749(8)	1/4	0	0.0041(3)	0.0057(3)	0.0076(3)	0	0	-0.0013(2)
Se	4d	0.4230(2)	0.3778(1)	1/4	0.0062(5)	0.0073(5)	0.0064(5)	-0.0015(3)	0	0
Si	4c	0.1127(4)	1/4	0	0.005(1)	0.004(1)	0.004(1)	0	0	0.0009(8)
O(1)	8e	0.0542(8)	0.7611(6)	0.1207(5)	0.004(2)	0.007(2)	0.007(2)	0.002(2)	0.001(2)	0.002(2)
O(2)	8e	0.2735(8)	0.0707(7)	0.0310(4)	0.006(2)	0.005(2)	0.008(2)	-0.001(2)	0.001(2)	-0.000(2)

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