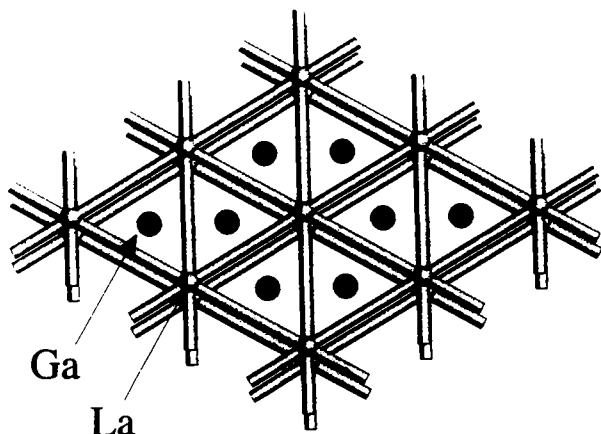


# Refinement of the crystal structure of lanthanum digallide, LaGa<sub>2</sub>

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## Abstract

Ga<sub>2</sub>La, hexagonal, *P6/mmm* (No. 191),  $a = 4.3092(6)$  Å,  $c = 4.4251(9)$  Å,  $V = 71.2$  Å<sup>3</sup>,  $Z = 1$ ,  $R_{\text{gt}}(F) = 0.019$ ,  $wR_{\text{ref}}(F^2) = 0.044$ ,  $T = 293$  K.

## Source of material

Single crystals of the title compound was grown from a mixture of elemental La and Ga in LaBr<sub>3</sub> flux under Ar atmosphere in sealed Ta ampoule at 1173 K for 12 days.

## Experimental details

Absorption correction was applied using the  $\psi$ -scan method [1].

## Discussion

La and Ga form a number of binary compounds: La<sub>3</sub>Ga, La<sub>5</sub>Ga<sub>3</sub>, LaGa, LaGa<sub>2</sub> and LaGa<sub>6</sub> [2]. Of these, LaGa<sub>2</sub> is interesting because its electronic structure is known [3] and has been reported to possess a homogeneity range of 66.7 at. % – 80.0 at. % of Ga [4]. Although its structure has been determined from powder data [2], the single crystal structure has not been reported. We present here the single crystal structure of LaGa<sub>2</sub>.

Our structure refinement did not show excess Ga. The La—Ga distance in the title compound is 3.3294(4) Å, much larger than the sum of the covalent radii of La (1.69 Å) and Ga (1.25 Å). This indicates that the La—Ga interaction is significantly ionic. The Ga—Ga distance is 2.4879(3) Å, close to the sum of covalent radii of Ga. It is also comparable to the Ga—Ga distance of 2.465 Å in

$\alpha$ -Ga, which is the thermodynamically stable allotrope at room temperature [5]. However, it is smaller than the Ga—Ga distance in many Ga-containing intermetallic compounds that are characterized by multi-center, delocalized bonding. For example, the Ga—Ga distance in the superconducting V<sub>2</sub>Ga<sub>5</sub> solid is 2.683(2) Å [6]. This fact implies that the Ga—Ga bond is close to the localized two-center, two-electron type. It should be also noted that the thermal ellipsoid of Ga is elongated along the *c*-axis, indicating the larger degree of freedom for motion along this direction which is not blocked by the La atoms.

Table 1. Data collection and handling.

Crystal:	metallic dark grey plate, size 0.04 × 0.12 × 0.12 mm
Wavelength:	Ag K $\alpha$ radiation (0.56085 Å)
$\mu$ :	174.39 cm <sup>-1</sup>
Diffractometer, scan mode:	CAD4, $\omega$
$2\theta_{\text{max}}$ :	53.9°
$N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ :	368, 86
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 78
$N(\text{param})_{\text{refined}}$ :	6
Programs:	SIR97 [7], SHELXTL [8]

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## References

1. North, A. C. T.; Phillips, D. C.; Mathews, F. S.: A Semi-empirical Method of Absorption Correction. *Acta Crystallogr.* **A24** (1968) 351–359.
2. Villars, P.; Calvert, L. D.: *Pearson's Handbook of Crystallographic Data for Intermetallic Phases*. ASM-International, Materials Park 1991.
3. Harima, H.; Yanase, A.: Electronic Structure and Fermi Surface of LaGa<sub>2</sub>. *J. Phys. Soc. Jpn.* **60** (1991) 2718–2723.
4. Massalski, T. B.: *Binary Alloy Phase Diagrams*. American Society For Metals, Metal Park 1986.
5. Donohue, J.: *The structures of the elements*. Krieger, Malabar, Florida 1982.
6. Kitchingman, W. J.; Norman, P. L.: X-ray diffraction studies of the ternary alloys (V,Mn)<sub>2</sub>Ga<sub>5</sub>. *Acta Crystallogr.* **B28** (1972) 1311–1312.
7. Altomare, A.; Burla, M. C.; Camalli, M.; Cascarano, G. L.; Giacovazzo, C.; Guagliardi, A.; Moliterni, A. G. G.; Polidori, G.; Spagna, R.: SIR97: A New Tool for Crystal Structure Determination and Refinement. *J. Appl. Crystallogr.* **32** (1999) 115–119.
8. Sheldrick, G. M.: SHELXTL. Version 5.1. Bruker Analytical X-Ray Systems, Madison, Wisconsin, USA 1997.

Table 2. Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>11</sub>	<i>U</i> <sub>22</sub>	<i>U</i> <sub>33</sub>	<i>U</i> <sub>12</sub>	<i>U</i> <sub>13</sub>	<i>U</i> <sub>23</sub>
La	1a	0	0	0	0.0056(2)	<i>U</i> <sub>11</sub>	0.0069(3)	<i>U</i> <sub>11</sub> /2	0	0
Ga	2d	2/3	1/3	1/2	0.0051(3)	<i>U</i> <sub>11</sub>	0.0184(5)	<i>U</i> <sub>11</sub> /2	0	0

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