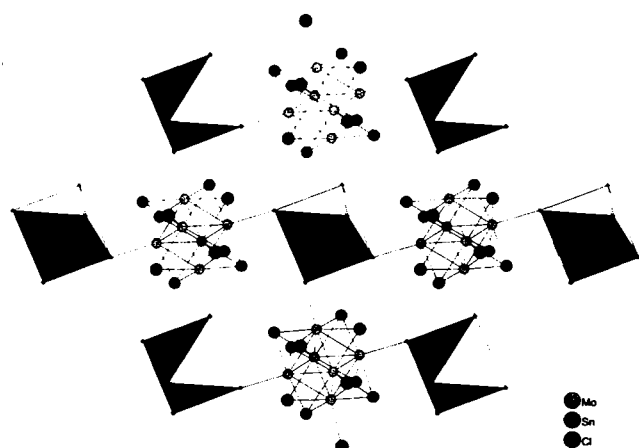


Crystal structure of tin octa- μ -chloro-hexachloro-octahedro-molybdate(II), $\text{Sn}[(\text{Mo}_6\text{Cl}_8)\text{Cl}_6]$.

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Abstract

$\text{Cl}_{14}\text{Mo}_6\text{Sn}$, cubic, $Pn\bar{3}$ (No. 201), $a = 12.952 \text{ \AA}$, $V = 2172.8 \text{ \AA}^3$, $Z = 4$, $R_{\text{gt}}(F) = 0.048$, $wR_{\text{ref}}(F) = 0.063$, $T = 293 \text{ K}$.

Source of material

Single crystals of $\text{Sn}[(\text{Mo}_6\text{Cl}_8)\text{Cl}_6]$ were grown in evacuated quartz tubes from a well ground mixture of MoCl_2 [1] and commercial SnCl_2 in the molar ratio 2:3 at 1073 K for 19 h. The crystals were yellow, slightly transparent needles.

Experimental details

Due to the sensitivity to moisture all preparative work had to be done in a glove box under Ar-atmosphere. The crystal chosen for the X-ray experiment was mounted on a quartz fibre of 0.1 mm diameter and then encapsulated in a quartz capillary of 0.3 mm diameter.

Discussion

Tin octa- μ -chloro-hexachloro-octahedro-molybdate(II), $\text{Sn}[(\text{Mo}_6\text{Cl}_8)\text{Cl}_6]$, has been investigated as part of a project aimed at synthesizing extended molecules with Mo_6Cl_8 as a base. The title compound $\text{Sn}[(\text{Mo}_6\text{Cl}_8)\text{Cl}_6]$ is isostructural with $\text{Pb}[(\text{Mo}_6\text{Cl}_8)\text{Cl}_6]$ [2], with the Sn atom situated in a chloride

octahedron formed by the bridging chlorine atoms. The Mo atoms are, as in all other reported structures of Mo(II)halides, arranged in an octahedron encapsulated in a chlorine cube [3] as shown in the figure. The structure may be described as an double skutterudite. Skutterudite CoAs_3 [4] is derived from the aristotype, 'ideal', cubic perovskite, by the concerted rotation of the 'ideal' perovskite octahedron around the three fold axes, ([111] direction). The rotation angle φ in skutterudite is $\varphi \approx 35^\circ$ [5]. Due to the differences in size between the octahedra in the title compound the rotation angle φ will differ. The φ value for the smaller chloride octahedron surrounding tin is 23.8° , and for the bigger chloride octahedron surrounding the Mo_6Cl_8 unit, $\varphi = 36.5^\circ$.

Table 1. Data collection and handling.

Crystal:	yellow needle, size $0.1 \times 0.2 \times 0.3 \text{ mm}$
Wavelength:	Mo K_α radiation (0.71073 \AA)
μ :	4.7 cm^{-1}
Diffractometer, scan mode:	STOE IPDS, φ
$2\theta_{\text{max}}$:	56.44°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	18628, 900
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 3 \sigma(I_{\text{obs}})$, 253
$N(\text{param})_{\text{refined}}$:	33
Programs:	SHELXS 97 [6], Jana 2000 [7]

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Table 2. Atomic coordinates and displacement parameters (in \AA^2).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Mo	24h	0.8649(2)	0.0282(2)	0.963(2)	0.020(8)	0.0266(7)	0.0239(7)	-0.0007(7)	-0.0016(8)	0.0014(7)
Sn	4c	1/2	1/2	1/2	0.0292(8)	U_{11}	U_{11}	0.0010(8)	U_{12}	U_{12}
Cl(1)	8e	0.1363(4)	x	x	0.038(2)	U_{11}	U_{11}	-0.008(2)	U_{12}	U_{12}
Cl(2)	24h	0.3776(4)	0.6883(4)	0.9337(4)	0.030(3)	0.031(3)	0.030(3)	0.009(2)	-0.003(2)	0.002(2)
Cl(3)	24h	0.5891(5)	0.8133(4)	0.0698(5)	0.044(4)	0.020(2)	0.068(5)	-0.004(2)	-0.010(3)	-0.006(3)

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