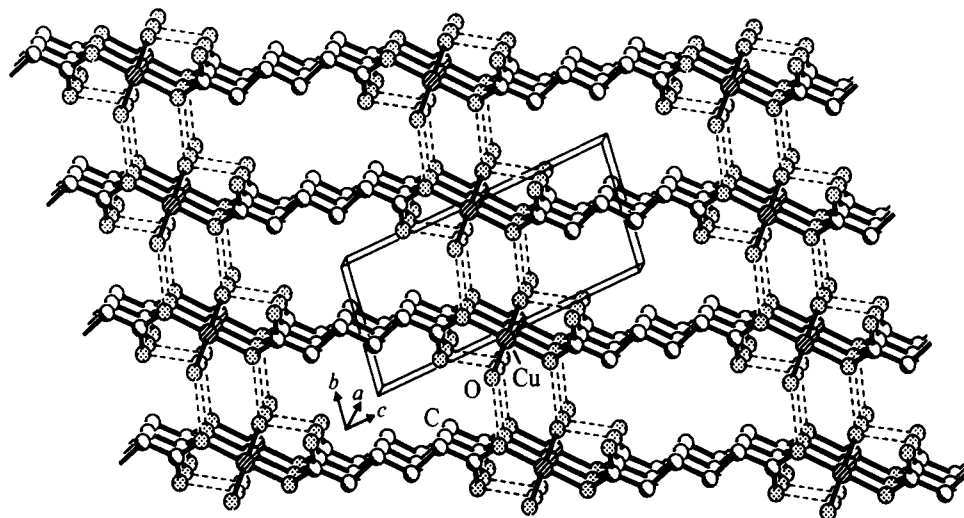


Crystal structure of *catena*-adipato-*O,O'*-diaquacopper(II), $\text{Cu}(\text{C}_6\text{H}_8\text{O}_4)(\text{H}_2\text{O})_2$

Y.-Q. Zheng*, A.-Y. Pan and J.-L. Lin

Ningbo University, Institute for Solid State Chemistry, Ningbo, Zhejiang, 315211 P. R. China

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Abstract

$\text{C}_6\text{H}_{12}\text{CuO}_6$, triclinic, $P\bar{1}$ (No. 2), $a = 3.861(1) \text{ \AA}$, $b = 4.928(1) \text{ \AA}$, $c = 11.326(2) \text{ \AA}$, $\alpha = 84.34(2)^\circ$, $\beta = 87.08(1)^\circ$, $\gamma = 83.45(2)^\circ$, $V = 212.9 \text{ \AA}^3$, $Z = 1$, $R_{\text{gt}}(F) = 0.040$, $wR_{\text{ref}}(F^2) = 0.104$, $T = 293 \text{ K}$.

Source of material

The title compound was synthesized by the reaction of 0.50 g (3.42 mmol) adipic acid with an excess of freshly prepared CuCO_3 in 50 ml doubly-distilled water, followed by removing the unreacted CuCO_3 . Blue crystals grew from the filtrate at room temperature by slow evaporation for several weeks.

Discussion

In the crystal structure, each Cu atom is square-planarly coordinated by four O atoms from two bis-monodentate adipate ligands and two water molecules to form a *trans* configuration with $d(\text{Cu—O}) = 1.930 \text{ \AA}$ (2 \times), 1.969 \AA (2 \times). The Cu atom is exactly located in the least-squares plane defined by the four coordinating O atoms. Through the carboxyl O atoms on both ends, the adipate ($\text{C}_6\text{H}_8\text{O}_4$)²⁻ ligands bridge the Cu atoms to form 1D chains $[\text{Cu}(\text{C}_6\text{H}_8\text{O}_4)_2(\text{H}_2\text{O})_2]$, which propagate parallel to $[0\bar{1}1]$ and are held together by interchain hydrogen bonds between water and carboxyl oxygen atoms with $d(\text{O}\cdots\text{O}) = 2.630 \text{ \AA}$ and 2.688 \AA and $\angle\text{O—H}\cdots\text{O} = 154^\circ$ and 169° . The adipate ($\text{C}_6\text{H}_8\text{O}_4$)²⁻ ligands are centered at 1e positions and their bridging fashion is similar to that observed in the Co adipate reported previously [1]. All backbone C atoms are nearly coplanar and the dihedral angle between the carboxyl plane and the C backbone plane is 19.9° . The terminal C—C bonds of 1.510 \AA are significantly shorter than the rests

averaged at 1.523 \AA and the terminal $\angle\text{C—C—C}$ angles of 113.9° are slightly larger than the rests of 111.8° . The C—O bond length of 1.297 \AA to the coordinating O atom is substantially larger than that of 1.235 \AA to the non-coordinating one. The closest Cu \cdots Cu distance of 3.861 \AA does not indicate interaction between the Cu atoms.

Table 1. Data collection and handling.

Crystal:	blue block, size $0.089 \times 0.178 \times 0.222 \text{ mm}$
Wavelength:	Mo K_α radiation (0.71073 \AA)
μ :	25.61 cm^{-1}
Diffractometer, scan mode:	Bruker P4, $\theta/2\theta$
$2\theta_{\text{max}}$:	55°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	1413, 982
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 966
$N(\text{param})_{\text{refined}}$:	62
Programs:	SHELXS-97 [2], SHELXL-97 [3]

Table 2. Atomic coordinates and displacement parameters (in \AA^2).

Atom	Site	x	y	z	U_{iso}
H(2A)	2i	0.3925	−0.6305	0.8100	0.028
H(2B)	2i	0.7504	−0.5296	0.8100	0.037
H(3A)	2i	0.5704	−0.2185	0.9632	0.028
H(3B)	2i	0.198	−0.3234	0.9689	0.027
HA(O3)	2i	0.2387	−0.3626	0.4156	0.050
HB(O3)	2i	0.1217	−0.1382	0.3597	0.050

* Correspondence author (e-mail: zhengcm@nbu.edu.cn)

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Cu	1 <i>f</i>	1/2	0	1/2	0.0431(3)	0.0153(3)	0.0119(3)	0.0032(2)	0.0005(2)	0.0016(2)
C(1)	2 <i>i</i>	0.4047(6)	−0.2584(5)	0.7300(2)	0.029(1)	0.020(1)	0.017(1)	0.0006(8)	−0.0035(8)	0.0018(9)
C(2)	2 <i>i</i>	0.5169(6)	−0.4766(5)	0.8279(2)	0.028(1)	0.022(1)	0.016(1)	0.0055(8)	0.0001(8)	0.0025(9)
C(3)	2 <i>i</i>	0.4477(6)	−0.3847(5)	0.9523(2)	0.027(1)	0.024(1)	0.014(1)	0.0036(9)	−0.0001(8)	0.0023(9)
O(1)	2 <i>i</i>	0.5481(6)	−0.2942(4)	0.6258(2)	0.049(1)	0.0182(8)	0.0152(9)	0.0032(7)	0.0008(7)	0.0021(7)
O(2)	2 <i>i</i>	0.1904(5)	−0.0608(4)	0.7502(2)	0.039(1)	0.028(1)	0.024(1)	0.0117(8)	−0.0049(8)	0.0016(8)
O(3)	2 <i>i</i>	0.1752(5)	−0.2022(4)	0.4238(2)	0.044(1)	0.0182(8)	0.0197(9)	0.0058(7)	−0.0029(7)	−0.0009(7)

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