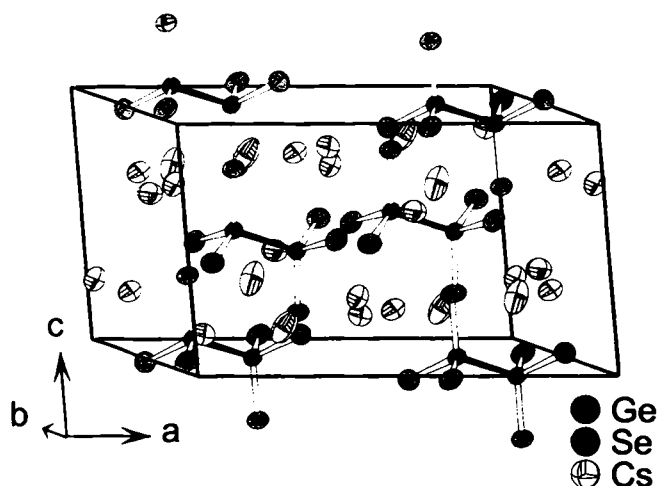


Crystal structure of hexacesium hexaselenidodigermanate(III), $\text{Cs}_6\text{Ge}_2\text{Se}_6$

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Received July 25, 2000, CSD-No. 409512



Abstract

$\text{Cs}_6\text{Ge}_2\text{Se}_6$, monoclinic, $C12/c1$ (No. 15), $a = 16.280(3) \text{ \AA}$, $b = 13.528(3) \text{ \AA}$, $c = 9.621(2) \text{ \AA}$, $\beta = 96.70(3)^\circ$, $V = 2104.4 \text{ \AA}^3$, $Z = 4$, $R_{\text{gt}}(F) = 0.027$, $wR_{\text{ref}}(F^2) = 0.060$, $T = 293 \text{ K}$.

Source of material

The title compound was prepared starting from an inhomogeneous sample obtained by thermal decomposition of CsN_3 mixed with GeSe_2 (molar ratio 6 : 2). The reaction was carried out at 630 K in a quartz glass ampule. Together with additional Se (molar composition 6 Cs : 2 Ge : 6 Se), the raw product was heated to 1273 K. After quenching in icewater, homogenization and subsequent annealing at 953 K for two weeks the reaction yielded air sensitive, transparent, orange single crystals of $\text{Cs}_6\text{Ge}_2\text{Se}_6$.

Discussion

With $\text{Cs}_6\text{Ge}_2\text{Se}_6$ another crystalline solid could be prepared using the "azide-method" [1]. $\text{Cs}_6\text{Ge}_2\text{Se}_6$ crystallizes isotypically to $\text{K}_6\text{Ge}_2\text{Te}_6$ [2]. The hitherto known compounds with the composition $\text{M}_6\text{M}'_2\text{X}_6$, $\text{M} = \text{Na}$ ($\text{M}' = \text{Si}$, $\text{X} = \text{Te}$; $\text{M}' = \text{Ge}$, $\text{X} = \text{Se}$, Te), K ($\text{M}' = \text{Ge}$, $\text{X} = \text{S}$, Se , Te ; $\text{M}' = \text{Sn}$, $\text{X} = \text{Se}$, Te), $\text{Rb}_6\text{Sn}_2\text{Te}_6$ and $\text{M} = \text{Cs}$ ($\text{M}' = \text{Ge}$, $\text{X} = \text{Te}$; $\text{M}' = \text{Sn}$, $\text{X} = \text{Se}$, Te) [2, 3 and cited literature] crystallize in three different structure types characterized by isolated $\text{M}'_2\text{X}_6$ units with covalent $\text{M}'\text{—M}'$ -bondings. In $\text{Cs}_6\text{Ge}_2\text{Se}_6$, the alkali metals are distorted octahedral and distorted trigonal prismatic coordinated by selenium ($d(\text{Cs—Se}) = 3.49 \text{ \AA} - 3.96 \text{ \AA}$, see fig.). The ethane analogous Ge_2Se_6 unit contains two trigonal GeSe_3 -pyramids ($d(\text{Ge—Se}) = 2.36 \text{ \AA} - 2.38 \text{ \AA}$) in staggered conformation connected by a covalent Ge—Ge bond ($d(\text{Ge—Ge}) = 2.54 \text{ \AA}$).

Table 1. Data collection and handling.

Crystal:	orange, transparent, polygonal isometric, size $0.16 \times 0.16 \times 0.18 \text{ mm}$
Wavelength:	$\text{Mo K}\alpha$ radiation ($0.71069 \text{ \AA}$)
μ :	234.36 cm^{-1}
Diffractometer, scan mode:	STOE IPDS, ϕ
$2\theta_{\text{max}}$:	55.94°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	11351, 2525
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2102
$N(\text{param})_{\text{refined}}$:	66
Programs:	SHELXS-97 [4], SHELXL-97 [5], DIAMOND [6]

Table 2. Atomic coordinates and displacement parameters (in $\text{\AA}^2$).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Cs(1)	8f	0.77802(2)	−0.02458(3)	−0.19190(5)	0.0392(2)	0.0330(2)	0.0640(3)	−0.0084(2)	0.0241(2)	−0.0184(2)
Cs(2)	8f	0.59492(2)	−0.20100(3)	0.12209(4)	0.0317(2)	0.0512(2)	0.0267(2)	0.0045(2)	0.0027(1)	0.0023(2)
Cs(3)	4e	1/2	0.09239(4)	1/4	0.0315(2)	0.0387(3)	0.0256(2)	0	0.0086(2)	0
Cs(4)	4e	1/2	0.57846(4)	3/4	0.0266(2)	0.0286(2)	0.0331(2)	0	0.0078(2)	0
Se(1)	8f	0.68376(3)	0.26385(4)	0.28565(5)	0.0264(2)	0.0272(3)	0.0175(2)	0.0014(2)	0.0044(2)	0.0000(2)
Se(2)	8f	0.62156(3)	0.08581(4)	−0.02693(6)	0.0289(2)	0.0236(3)	0.0263(3)	−0.0052(2)	0.0089(2)	−0.0055(2)
Se(3)	8f	0.59077(3)	0.36589(4)	−0.07126(6)	0.0273(2)	0.0275(3)	0.0236(2)	0.0059(2)	0.0028(2)	0.0046(2)
Ge	8f	0.67842(3)	0.24344(4)	0.03914(5)	0.0177(2)	0.0191(2)	0.0163(2)	−0.0003(2)	0.0041(2)	−0.0003(2)

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Acknowledgment. This work was supported by the Deutsche Forschungsgemeinschaft.

References

1. Reiner, C.: Gemischte Alkalimetallthioindate und Chalkogenide mit heteronuklearen Clustern der Tetrele und der Triefe. Dissertation, Universität Siegen, Germany 1999.
2. Dittmar, G.: Die Kristallstrukturen von $K_6[Ge_2Te_6]$ und $K_6[Sn_2Te_6]$ und ihre kristall-chemische Beziehung zum $K_6[Si_2Te_6]$ -Typ. Z. Anorg. Allg. Chem. **453** (1978) 68-78.
3. Friede, B.; Jansen, M.: Darstellung und Kristallstruktur von Hexacesium-hexatellurido-digermanat(III) ($Cs_6Ge_2Te_6$). Z. Naturforsch. **54b** (1999) 1095-1097.
4. Sheldrick, G. M.: SHELXS-97, Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
5. Sheldrick, G. M.: SHELXL-97, Program for Crystal Structure Refinement. University of Göttingen, Germany 1997.
6. Brandenburg, K.: Diamond (Version 2.1a). Crystal Impact GbR, Germany 1996-1999.