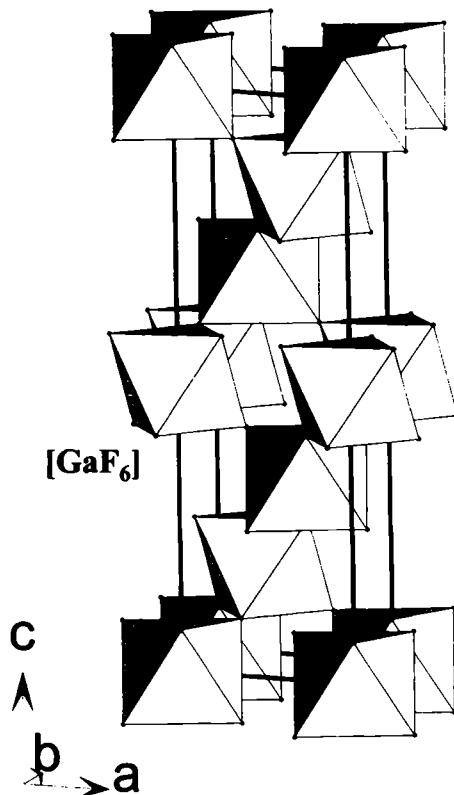


Refinement of the crystal structure of gallium trifluoride, GaF_3

M. Roos and G. Meyer*

Universität zu Köln, Institut für Anorganische Chemie, Greinstraße 6, D-50939 Köln, Germany

Received July 14, 2000, CSD-No. 409507



Abstract

F_3Ga , trigonal, $R\bar{3}c$ (No. 167), $a = 5.012(4)$ Å, $c = 12.99(1)$ Å, $V = 282.6$ Å³, $Z = 6$, $R_{\text{gt}}(F) = 0.022$, $wR_{\text{ref}}(F^2) = 0.053$, $T = 293$ K.

Source of material

GaF_3 may be obtained through the reaction of gallium metal (Ventron 4 N) with NH_4F (Merck, 99.8 %, recrystallized from CH_3OH to remove traces of water) in the molar ratio of 1:3 in a sealed Monel container. The reaction temperature is 748 K with a crystallization time of six weeks. Colourless square shaped nearly cubic crystals are obtained [1].

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Ga	6b	0	0	0	0.0033(4)	U_{11}	0.0030(5)	0.0016(2)	0	0
F	18e	0.0591(3)	$x-1/3$	1/12	0.0090(7)	U_{11}	0.0083(7)	0.0057(7)	-0.0020(3)	$-U_{13}$

Discussion

GaF_3 crystallizes in the VF_3 type of structure with the hexagonal space group $R\bar{3}c$ [1]. A previous description of the crystal structure [2] uses the rhombohedral setting of the space group. The cell parameters obtained by powder diffraction refinement are $a = 4.995(1)$ Å and $c = 13.063(4)$ Å. GaF_3 is built of a three-dimensional network of $[\text{GaF}_6]$ octahedra sharing common corners with a Ga–F–Ga angle of 147.79(8)°.

Table 1. Data collection and handling.

Crystal:	colourless, square-formed, nearly cubic, size 0.16 × 0.20 × 0.20 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	143.37 cm ⁻¹
Diffractometer, scan mode:	STOE IPDS, ϕ
$2\theta_{\text{max}}$:	55.7°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	726, 77
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 74
$N(\text{param})_{\text{refined}}$:	8
Programs:	SHELXS-97 [3], SHELXL-97 [4], DIAMOND [5]

Acknowledgment. We are grateful to the Deutsche Forschungsgemeinschaft, Bonn, for financial support.

References

1. Roos, M.: Aufbau und Abbau komplexer Ammoniumfluorometallate der Elemente Aluminium, Gallium und Indium unter Ammoniakatmosphäre. Dissertation, Universität zu Köln, Germany 1999.
2. Brewer, F. M.; Garton, G.; Goodgame, D. M. L.: The preparation and crystal structure of gallium trifluoride. *J. Inorg. Nucl. Chem.* **9** (1959) 56.
3. Sheldrick, G. M.: SHELXS-97-2, Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
4. Sheldrick, G. M.: SHELXL-97-2, Program for Crystal Structure Refinement. University of Göttingen, Germany 1997.
5. Brandenburg, K.: Diamond (Version 2.1a). Crystal Impact GbR, Germany 1996-1999.

* Correspondence author (e-mail: gerd.meyer@uni-koeln.de)