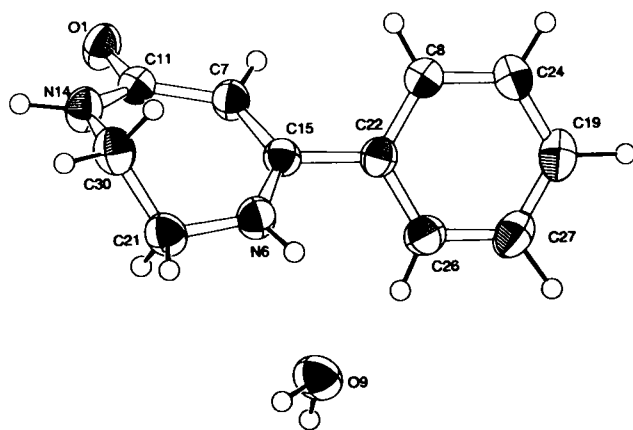


Crystal structure of 7-phenyl-1,4-diazepin-5-one monohydrate, $C_{11}H_{12}N_2O \cdot H_2O$

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Abstract

$C_{11}H_{12}N_2O_2$, triclinic, $P\bar{1}$ (No. 2), $a = 7.1014(4)$ Å, $b = 8.6896(7)$ Å, $c = 10.1265(8)$ Å, $\alpha = 70.175(3)^\circ$, $\beta = 88.517(5)^\circ$, $\gamma = 66.283(4)^\circ$, $V = 533.7$ Å³, $Z = 2$, $R_{gt}(F) = 0.044$, $wR_{ref}(F^2) = 0.064$, $T = 298$ K.

Source of material

A solution of (0.1 mol) of ethyl benzoylacetate and 10 ml of xylene was added, drop wise, over 40 min. to a refluxing solution of 0.1 mol ethylene diamine and 100 ml of xylene. During this time an oily layer separated. On cooling, the oil solidified to a hard mass. The xylene layer was decanted and discarded. The solid was suspended in about 100 ml of chloroform and the mixture was filtered. The solid was recrystallized from ethanol to give 50 % of product with mp 483 K – 485 K. The melting point was taken in a capillary tube. Chemical analysis for $C_{11}H_{12}N_2O$: C, 69.70; H, 6.43; N, 15.09 (calculated: C, 70.18; H, 6.43 ; N, 14.88).

Discussion

The interest carried to derivatives of the 1,4-diazepine is due to their remarkable pharmacological activity. Let's note for instance their antitumoral [1], and anticonvulsant [2,3] actions. In the course of our research in the 1,4-diazepine series [4,5], we were interested in the synthesis of the 7-phenyl-1,4-diazepin-5-one. In the literature [6] the synthesis of the compound is performed by condensation of ethylenediamine with ethyl benzoylacetate in xylene. The obtained diazepine can theoretically adopt one of the two tautomeric enamine or imine forms. In a previous work [5], we have shown by a study based on the ¹H-NMR spectral data and

on the electronic impact fragmentations, that the compound should adopt, preferentially, the enaminic form.

The structure was determined to identify the tautomeric form present in the crystal and the enamine conformation was observed. The seven-membered ring is composed of a planar fragment $C_{21}N_6C_{15}C_{7}C_{11}$ (at $136.2(2)^\circ$ of the phenyl ring) with N14 at $0.4329(5)$ Å and C30 at $0.9566(6)$ Å. The organic molecule crystallizes with one structural water molecule (O9) which shows three intermolecular hydrogen bonds, one as acceptor ($N6-O9 = 2.9506(6)$ Å) and two as donor ($O1-O9 = 2.7990(6)$ Å and $O1-O9 = 2.8975(7)$ Å).

Table 1. Data collection and handling.

Crystal:	colorless cube, size $0.30 \times 0.35 \times 0.4$ mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	0.9 cm^{-1}
Diffractometer, scan mode:	Kappa CCD, ϕ -scan
$2\theta_{\max}$:	50.24°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	1857, 1857
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 3 \sigma(I_{\text{obs}})$, 1656
$N(\text{param})_{\text{refined}}$:	144
Programs:	DENZO [7], SCALEPACK [7], SIR92 [8], maXus [9], ORTEP-II [10]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(7)	2i	0.23393	-0.180201	0.251538	0.0564
H(8)	2i	0.192488	0.047041	0.02255	0.0658
H(19)	2i	0.402887	0.440645	-0.035683	0.0692
H(21A)	2i	-0.411958	0.153415	0.382936	0.0614
H(21B)	2i	-0.213158	0.003515	0.481737	0.0614
H(24)	2i	0.364517	0.221478	-0.106589	0.0725
H(26)	2i	0.084883	0.319495	0.293148	0.0621
H(27)	2i	0.264958	0.489384	0.16537	0.0701
H(30A)	2i	-0.435058	-0.097007	0.376588	0.0627
H(30B)	2i	-0.333358	-0.037807	0.240288	0.0627
H(6)	2i	-0.215(1)	0.2565(9)	0.2747(7)	0.0500
H(14)	2i	-0.165957	-0.354906	0.45561	0.0614
H(9A)	2i	-0.547(2)	0.518(1)	0.3509(8)	0.084(3)
H(9B)	2i	-0.348(2)	0.480(1)	0.408(1)	0.102(3)

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
O(1)	2i	0.19420(6)	-0.43687(5)	0.41247(4)	0.0411(2)	0.0304(2)	0.0569(3)	-0.0141(2)	0.0098(2)	-0.0111(2)
N(6)	2i	-0.16927(8)	0.14163(6)	0.29777(5)	0.0428(3)	0.0317(2)	0.0501(3)	-0.0116(2)	0.0137(2)	-0.0133(2)
C(7)	2i	0.10653(9)	-0.13080(7)	0.28554(6)	0.0361(3)	0.0327(3)	0.0416(3)	-0.0134(2)	0.0100(2)	-0.0117(2)
C(8)	2i	0.1995(1)	0.13784(7)	0.05225(6)	0.0549(4)	0.0390(3)	0.0444(3)	-0.0236(3)	0.0132(3)	-0.0170(2)
O(9)	2i	-0.41198(8)	0.49325(6)	0.32548(5)	0.0484(3)	0.0568(3)	0.0570(3)	-0.0173(2)	0.0158(2)	-0.0256(2)
C(11)	2i	0.05250(9)	-0.27685(7)	0.36560(5)	0.0369(3)	0.0334(3)	0.0361(3)	-0.0155(2)	0.0063(2)	-0.0128(2)
N(14)	2i	-0.14046(7)	-0.25171(6)	0.39211(5)	0.0377(3)	0.0376(2)	0.0493(3)	-0.0186(2)	0.0101(2)	-0.0113(2)
C(15)	2i	0.00821(8)	0.05132(6)	0.25527(5)	0.0355(3)	0.0335(3)	0.0321(3)	-0.0134(2)	0.0034(2)	-0.0102(2)
C(19)	2i	0.3147(1)	0.37084(8)	0.01792(6)	0.0520(4)	0.0484(3)	0.0483(4)	-0.0299(3)	0.0032(3)	-0.0051(3)
C(21)	2i	-0.29176(9)	0.06041(7)	0.38654(6)	0.0392(3)	0.0409(3)	0.0459(3)	-0.0114(2)	0.0117(3)	-0.0145(2)
C(22)	2i	0.11277(8)	0.16489(6)	0.17173(5)	0.0357(3)	0.0287(2)	0.0353(3)	-0.0111(2)	0.0013(2)	-0.0071(2)
C(24)	2i	0.2990(1)	0.24108(8)	-0.02399(6)	0.0610(4)	0.0508(3)	0.0449(3)	-0.0290(3)	0.0172(3)	-0.0157(3)
C(26)	2i	0.1279(1)	0.29750(7)	0.21225(6)	0.0507(4)	0.0380(3)	0.0385(3)	-0.0181(2)	0.0020(3)	-0.0132(2)
C(27)	2i	0.2295(1)	0.39938(8)	0.13587(6)	0.0584(4)	0.0424(3)	0.0504(3)	-0.0275(3)	-0.0043(3)	-0.0117(3)
C(30)	2i	-0.31846(9)	-0.07941(8)	0.34179(6)	0.0320(3)	0.0483(3)	0.0493(3)	-0.0161(2)	0.0063(3)	-0.0131(3)

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