Inhibitory Effect of Hydroxyindoles and their Analogues on Human Melanoma Tyrosinase

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A recent study showed that N-acylserotonin derivatives have strong inhibitory activity against tyrosinase. To clarify the role of the 5-hydroxy group in the indole ring, 2-, 4-, 5-, 6-, and 7-hydroxyindole and 11 related compounds such as 5-hydroxyindan and 6-hydroxyquinoline were tested for their inhibition of catecholase activity of tyrosinase from human HMV-II melanoma cells. 6-Hydroxyindole (5) and 7-hydroxyindole (6) were potent inhibitors, while 5-hydroxyindole (4) was a weaker inhibitor than the above-mentioned compounds (IC_{50} = 20, 79, 366, and 342 μ M for **5**, **6**, **4**, and kojic acid, respectively). 2-Hydroxycarbazole was also active (IC₅₀ = 190 μ M), 5-hydroxyindan, 4-aminophenol, and harmalol were slightly active, and other compounds were inactive as an inhibitor. A similar pattern of inhibition was found with these compounds against mouse B16 melanoma tyrosinase, but with some differences from that for HMV-II tyrosinase. Kinetic analysis with HMV-II tyrosinase showed that the inhibition by hydroxyindoles 4, 5, and 6 was competitive with respect to the substrate L-DOPA. Melanin formation in HMV-II cells was suppressed by 14% with 10 μm 5 without cytotoxicity, but 30 or 100 μ M 5 decreased the cell viability. The present results suggest that 6-hydroxyindole is a potential and useful pharmacophore of antimelanogenic agents and that the position of a phenolic hydroxy group in a specific heterocyclic ring such as in indole is possibly optimized to yield more active inhibitors for tyrosinase.

Key words: Tyrosinase, Inhibitor, Hydroxyindole

Introduction

Tyrosinase (EC 1.14.18.1) inhibitors are widely studied for developing efficient skin-lightening agents (Chang, 2009; Khan, 2007). Recently, we found that N-acylserotonin derivatives with phenolic substituents, such as N-caffeoylserotonin, strongly inhibited tyrosinase from mouse and human melanoma cells (Yamazaki et al., 2009). The inhibitory activity of these compounds is closely connected with the presence of the 5-hydroxy group in the indole ring. Almost a half century ago, 5-hydroxyindole (4) was reported as a tyrosinase inhibitor (Seiji, 1961), and later its activity was compared with those of some tryptophan derivatives and metal chelators (Pomerantz, 1963), but no detailed structure-activity relationship (SAR) study concerning the 5-hydroxyindole moiety has been reported. Phenolic compounds are the main class of tyrosinase inhibitors as alternative substrates. While many studies were carried out on the effect of the number and position of the hydroxy groups in stilbene (Kim et al., 2002), chalcone (Nishida et al., 2007), flavone (Kubo et al., 2000), and phenylalkylamide skeletons (Okombi et al., 2006; Cho et al., 2006) on tyrosinase inhibitory activity, such an analysis has been rarely reported for N-heterocyclic phenols. On the other hand, previous screening of tyrosinase inhibitors has been done mostly with mushroom tyrosinase (Cho et al., 2006; Tamura et al., 2007). Some studies used tyrosinase from mouse B16 melanoma cells (Kim et al., 2002), and fewer studies used human tyrosinase (Okombi et al., 2006). Mushroom tyrosinase showed, with some compounds, a considerably different behaviour from those of the mammalian enzymes, and the B16 tyrosinase also does not necessarily behave in parallel with a human melanoma tyrosinase (Yamazaki et al., 2009). Screening with tyrosinase from human melanocytes is beneficial in developing skin lighteners, but because of some difficulty in cultivating normal melanocytes the next best way would be the use of human melanoma tyrosinase. The nucleotide and amino acid sequences of a human melanoma tyrosinase were reported to be nearly identical to those of the human melanocyte tyrosinase (Bouchard *et al.*, 1989), suggesting that their enzymatic properties are, if not the same, very similar to each other.

Here, we report several hydroxyindoles as potent inhibitors of HMV-II human melanoma tyrosinase in L-3,4-dihydroxyphenylalanine (L-DOPA) oxidation, with SAR focused on the position of the hydroxy group in indole and several related ring structures.

Material and Methods

Reagents

Indole (1), 2- (2), 4- (3), and 6-hydroxyindole (5), harmalol (14), and harmaline (15) were purchased from Wako Pure Chemicals Inc. (Osaka, Japan), 5-hydroxyindole (4), 5-hydroxyindan (7), 3- (11) and 4-aminophenol (10), 4-hydroxycarbazole (12), 3,4-dihydro-7-hydroxy-2(1*H*)-quinolinone (16), and kojic acid (17) were from Tokyo Chemical Industries Ltd. (Tokyo, Japan), and 7-hydroxyindole (6) from Kanto Chemical Company Inc. (Tokyo, Japan). 2-Hydroxycarbazole (13), 6-(8) and 8-hydroxyquinoline (9), and mushroom tyrosinase were obtained from Sigma-Aldrich Co. (St. Louis, MO, USA).

Tyrosinase assay

Human HMV-II melanoma cells (Dai-Nippon-Sumitomo Pharmaceuticals Inc., Osaka, Japan) were grown in RPMI-1640 medium containing 15% fetal bovine serum (FBS). Tyrosinase was extracted from the cells according to the published method (Yamazaki et al., 2009). Assay solutions were prepared by mixing 0.94 ml 50 mm phosphate buffer (pH 6.8) containing 2.6 mm L-DOPA and $10 \,\mu l$ ethanol containing $0-30 \,\mathrm{mm}$ test compound, unless otherwise stated. The reaction was started by adding the enzyme solution (50 μ l) to the substrate solutions prewarmed to 37 °C, and the dopachrome formation during the incubation for 5 min was determined by light absorption at 475 nm [$\varepsilon = 3700 \,\mathrm{M}^{-1} \,\mathrm{cm}^{-1}$ (Kubo *et al.*, 2003)]. Usually, 2.5–3 mU enzyme (Cabanes et al., 1984) were applied per assay. Experiments with tyrosinase extracted from B16 mouse melanoma cells (Riken Cell Bank, Tsukuba, Japan) and mushroom tyrosinase were done in the same way as described above. IC₅₀ values were obtained from

Fig. 1. Chemical structures of the compounds.

plots of concentration vs. relative activity (% of control).

Antimelanogenic assay

HMV-II cells were cultured in 96-well plates $(1 \cdot 10^4 \text{ cells in } 0.2 \text{ ml})$ medium per well) for 1 d. The medium of each well was changed with fresh medium (0.2 ml) containing $500 \, \mu\text{M}$ theophylline, and then $2 \, \mu$ l ethanol solutions containing $0-10 \, \text{mM}$ test compounds were added to the wells. After incubation for 3 d, the medium was aspirated off, and cell viability was assayed with a tetrazolium reagent (CellTiter 96° , Promega Corp., Madison, WI, USA). The intracellular melanin content was measured by the published method (Liu *et al.*, 2007), but with some modifications. Cells were cultured in 12-well plates $(2 \cdot 10^5 \text{ cells in } 2 \, \text{ml})$ medium per well) for 1 d, and 40 mM

theophylline solution in water (25 μ l each) and ethanol solutions of test compounds (2 μ l each) were added to the wells. After 3 d of incubation, cells of each well were collected by a cell scraper, washed with phosphate-buffered saline, and then homogenized in 0.2 ml 1 m NaOH by sonication. The absorbance at 415 nm of the homogenates was measured to evaluate the intracellular melanin content.

Results and Discussion

Among the tested compounds (Fig. 1), 5- (4), 6- (5), and 7-hydroxyindole (6), and 2-hydroxycarbazole (13) clearly inhibited the L-DOPA oxidation by HMV-II melanoma tyrosinase. The typical result for the dose-dependency of inhibition is shown in Fig. 2, and IC₅₀ values and percentage of inhibition at a concentration of $100 \,\mu\text{M}$ are summarized in Table I. The order of the activity strength was 5 > 6 > 13 > kojic acid (17, the usual antimelanogenic agent) ≥ 4 , as shown by the IC₅₀ values. 5-Hydroxyindan (7), 4-aminophenol (10), and harmalol (14) indicated marginal

inhibitory activities, while 2- (2) and 4-hydroxyindole (3), 6- (8) and 8-hydroxyquinoline (9), 3-aminophenol (11), 4-hydroxycarbazole (12), harmaline (15), 3,4-dihydro-7-hydroxy-2(1H)quinolinone (16), and indole (1) were inactive in inhibiting this tyrosinase. A similar pattern of inhibitory activity with these compounds was found with B16 mouse melanoma tyrosinase, but 4-hydroxyindole (3) weakly inhibited B16 tyrosinase in contrast to the inactivity against HMV-II enzyme (Table I). Importantly, 6-hydroxyindole (5) and 7-hydroxyindole (6) were stronger inhibitors than 5-hydroxyindole (4) and kojic acid (17) against B16 tyrosinase as well as HMV-II tyrosinase. In addition, 4- (3), 5- (4) and 6-hydroxyindole (5), and kojic acid (17) inhibited mushroom tyrosinase $[IC_{50} = (31 \pm 7), (77 \pm 4), (51 \pm 8), and$ $(59 \pm 6) \mu \text{M}$ (mean \pm sd, n = 3), respectively, under the same assay conditions as above], while 2-hydroxyindole (2), 7-hydroxyindole (6), and 6-hydroxyquinoline (8) promoted the oxidation of L-DOPA by the mushroom enzyme [initial rates = (144 ± 7) , (136 ± 9) , and $(176 \pm 12)\%$ of the control (mean \pm sd, n = 3) in the presence

Table I. Tyrosinase inhibitory activity of hydroxyindoles and the related compounds^a.

Compound	Cell line			
	HMV-II		B16	
-	IC ₅₀ [μM]	Inhibition at 100 μ M ^b (%)	IC ₅₀ [μM]	Inhibition at 100 μm ^b (%)
Indole (1)	_	−13 ± 8	_	-8 ± 3
2-Hydroxyindole (2)	_	-7 ± 6	_	1 ± 5
4-Hydroxyindole (3)	_	1 ± 5	1055 ± 69	9 ± 4
5-Hydroxyindole (4)	366 ± 41	27 ± 3	45 ± 9	75 ± 1
6-Hydroxyindole (5)	20 ± 3	83 ± 4	18 ± 10	89 ± 9
7-Hydroxyindole (6)	79 ± 13	54 ± 2	19 ± 3	81 ± 3
5-Hydroxyindan (7)	> 1000	9 ± 4	> 1000	7 ± 2
6-Hydroxyquinoline (8)	_	-1 ± 4	_	-5 ± 4
8-Hydroxyquinoline (9)	_	1 ± 5	_	0 ± 2
4-Aminophenol (10)	_	13 ± 3	173 ± 59	40 ± 6
3-Aminophenol (11)	_	-4 ± 4	_	-1 ± 2
4-Hydroxycarbazole (12)	_	1 ± 3	_	-3 ± 1
2-Hydroxycarbazole (13)	190 ± 18	39 ± 1	_	12 ± 6
Harmalol (14)	_	16 ± 3	95 ± 14	52 ± 3
Harmaline (15)	_	-2 ± 5	_	0 ± 2
3,4-Dihydro-7-hydroxy-2(1 <i>H</i>)-quinolinone (16)	_	-2 ± 3	_	-4 ± 5
Kojic acid (17)	342 ± 35	31 ± 10	48 ± 3	68 ± 1

^a Values are means \pm sd (standard deviation), n = 3-4.

^b Calculated by subtracting the relative activity (% of control) from 100. A negative value means that the enzyme reaction is accelerated by the compound.

^{-,} too large to be determined.

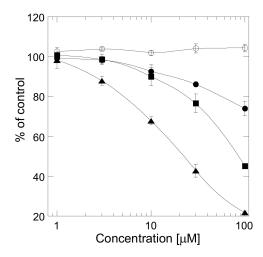


Fig. 2. Dose-dependency of the HMV-II tyrosinase inhibition by hydroxyindoles $\mathbf{4}$ ($\mathbf{\bullet}$), $\mathbf{5}$ (\mathbf{A}), and $\mathbf{6}$ ($\mathbf{\bullet}$). 6-Hydroxyquinoline ($\mathbf{8}$, $\mathbf{\circ}$) was inactive. Values (% of control) are the means \pm sd (n=3) in a representative experiment that was repeated with comparable results.

of $100 \,\mu\text{M}$ **2**, **6**, and **8**, respectively]. Similar phenomena with the mushroom enzyme have been reported for serotonin derivatives (Yamazaki *et al.*, 2009).

The present result shows that hydroxyindoles are a unique class of mammalian tyrosinase inhibitors and that the position of the hydroxy group in the indole ring strongly affects the inhibitory activity. The best position is 6, not 5 found previously with serotonin derivatives (Yamazaki et

al., 2009). The 7-hydroxy group is also superior to the 5-hydroxy group for tyrosinase inhibition. On the other hand, 4- and 2-hydroxy groups are substantially ineffective on the inhibitory activity. A similar relationship is seen from the comparison of 2- (13) and 4-hydroxycarbazoles (12). The activity of an indole alkaloid, harmalol (14), and the inactivity of its O-methyl derivative, harmaline (15), also suggest the effectiveness of the 6-hydroxy group (7-OH in β -carboline ring) in inhibiting tyrosinase. The inhibitory function of these hydroxy groups needs the indole structure, since the present analogues, 5-hydroxyindan (7), 6- (8) and 8-hydroxyquinoline (9), 4- (10) and 3-aminophenol (11), and 3,4-dihydro-7-hydroxy-2(1H)-quinolinone (16), are all weak or inactive inhibitors for the melanoma tyrosinase. The weak activity of 7 as compared to 5- (4) and 6-hydroxyindole (5) implies an essential role of the nitrogen atom that is located at the γ or δ position to the phenolic hydroxy group. However, the contrasts of activities for 4 vs. 8 and 10, 5 vs. 11 and 16, and 6 vs. 9 suggest that the spatial arrangement of a phenolic hydroxy group and a nitrogen atom (or an NH group) is not a critical determinant for inhibition of tyrosinase. The electron density of the NH group and/or the whole structure of the ring should also be important for the inhibitory activity.

The inhibition mechanism of **4**, **5**, and **6** for HMV-II tyrosinase was kinetically studied. The Lineweaver-Burk plots obtained with different concentrations of each compound cross on or very near the *Y*-axis (Fig. 3), indicating that these are

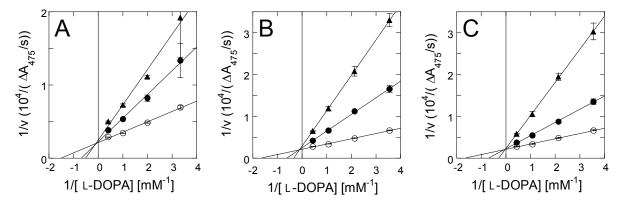


Fig. 3. HMV-II tyrosinase reaction kinetics in the presence of hydroxyindoles. (A) Lineweaver-Burk (L-B) plots for 4 [0 (\circ), 100 (\bullet), and 300 μ M (Δ)]. (B) L-B plots for 5 [0 (\circ), 10 (\bullet), and 30 μ M (Δ)]. (C) L-B plots for 6 [0 (\circ), 30 (\bullet), and 100 μ M (Δ)]. Every point is the mean \pm sd (n=3 or 4) and plot lines are drawn by least square linear regression.

competitive inhibitors ($K_i = 160, 6, \text{ and } 18 \,\mu\text{M}$ for 4, 5, and 6, respectively) in the L-DOPA oxidation by HMV-II tyrosinase. Tyrosinase contains one binuclear copper complex in the catalytic centre and the catechol hydroxy groups of the substrate are considered to be bound to the copper atoms in the enzymatic oxidation process (Chang, 2009; Muñoz-Muñoz et al., 2008). Phenolic hydroxy groups are generally capable of coordinating copper atoms so that they inhibit the enzyme in competition with the catechol substrate. The competitive inhibition by the hydroxyindoles may arise in such a way. However, the present phenolics other than the hydroxyindole compounds were inactive or only weak active in inhibiting HMV-II tyrosinase. Even 8-hydroxyquinoline (9) that is a strong copper-chelating ligand (Perrin, 1979) was almost inactive. Therefore, the copper-binding ability of a phenol group alone is not enough to perform the tyrosinase inhibition. This primary binding to one copper atom should be strengthened by additional interactions between the indole ring (e.g., the NH group and π electrons of the ring) and another copper atom or the amino acid residues of the enzyme's active site to accomplish the inhibition. Since 5,6-dihydroxyindole is an intermediate of melanin biosynthesis and is also a substrate of tyrosinase (Chang, 2009), it is likely that 6-hydroxyindole (5) has the sites that crucially interact with the enzyme. 6- (5) and 7-hydroxyindole (6) are reported to be useful in controlling hair colouration (i.e., as melanin precursors) (Cotteret and Audousset, 1993), but little is known about their tyrosinase inhibitory activity. The present study reveals that 5 is considerably a stronger inhibitor than the other hydroxyindoles 2, 3, 4, and 6 for HMV-II tyrosinase, showing that the position of the hydroxy group in a specific heterocyclic ring such as in indole is possibly optimized to produce more potent inhibitors against human tyrosinase.

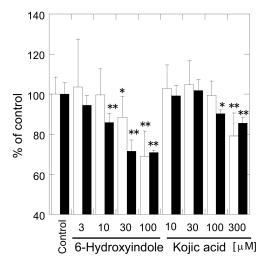


Fig. 4. Antimelanogenic activity and cytotoxicity of 6-hydroxyindole (5) and kojic acid (17) in a HMV-II cell culture. Values (% of control) are the means \pm sd (n=3-4 for melanin and n=8-10 for cell viability) in a representative experiment that was repeated with comparable results. Closed bars, melanin content; open bars, cell viability. ** and * represent p < 0.01 and p < 0.05 versus control by Student's t-test, respectively.

The antimelanogenic activity of **5** was evaluated in a culture of HMV-II melanoma cells. The melanin formation was suppressed by 14% with **5** at a concentration of $10 \,\mu\text{M}$, where the cell viability was not significantly different from the control (Fig. 4). On the other hand, $10 \,\mu\text{M}$ kojic acid (**17**) showed no antimelanogenic activity in the same experimental conditions as above. The melanin suppression by **5** increased up to 25-30% at 30 or $100 \,\mu\text{M}$, while the cell viability also significantly decreased. Thus, 6-hydroxyindole (**5**) is a potentially useful pharmacophore of antimelanogenic agents, but the solution of the problem of cytotoxicity, for example, by chemical modifications has to be studied in the future.

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