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Sex Pheromone of *Tortrix viridana*: (Z)-11-Tetradecenyl Acetate as the Main Component

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In electroantennographic studies on *Tortrix viridana* males, (E)-11-tridecenyl acetate displayed the highest activity of all straight-chain alkenyl acetates. However, evidence obtained by gas chromatography coupled to an electroantennographic detector or a mass spectrometer indicated that the ubiquitous pheromone compound (Z)-11-tetradecenyl acetate is the main component in the T. viridana female secretion. This compound and both (Z)- and (E)-11-tridecenyl acetate were found to be attractants for T. viridana males in the field. (Z)-9- and (E)-11-tetradecenyl acetate were not attractive and reduced male catches when added to either (Z)-11-tetradecenyl acetate or (E)-11-tridecenyl acetate. No evidence for synergistic effects was obtained.

(Z)-11-Tetradecenyl acetate (Z11-14Ac) has been identified in the sex pheromone secretion of many species of Lepidoptera, notably of the tortricid family, and is an important component of many sex attractants discovered by field screening [1]. We have now obtained evidence that the chemical is also the main component of the sex pheromone of the European oak leaf roller or green tortrix moth, Tortrix viridana L. (Lepidoptera: Tortricidae), and a strong attractant for males of the species.

Outbreaks of the leaf roller moth near Freiburg/Br. permitted field collection of substantial numbers of insects in the pupal stage for electrophysiological and chemical investigation. Electroantennogram (EAG) and single receptor responses were recorded as reported in earlier studies with tortricid species [2]. Of the series of monounsaturated straight-

chain alcohols, aldehydes and esters tested, acetates of a 13 or 14 carbon chain with a double bond in position 11 gave the highest responses. (E)-11-Tridecenyl acetate (E11-13Ac) was the most effective test chemical, followed by E11-14Ac (Table I). The findings were confirmed in gas chromatograms of acetate standards with electroantennographic detection [3] which showed the prominent signal obtained with E11-13Ac at the expected retention time. In field tests in June 1977 at Freiburg/Br. 10 and $100 \,\mu g$ E11-13Ac caught 7.0 and 44.3 T. viridana males per trap, respectively, while E11-14Ac at the same test amounts was not attractive. These results suggested that this unusual C_{13} compound could be a key substance in the sex pheromone of T. virida-

Extracts for chemical analysis were made by collecting calling females in methylene chloride and removing the bodies within a few seconds to avoid extracting too much fatty material. A temperature drop was found to be essential to induce calling in *T. viridana* females. At the end of a 16 h light phase the temperature was lowered from 20 to 15 ° and a light of 4 lux was maintained for observation. During the following 3 h, a substantial number of females was observed in calling position, and a wash of 76 females could be obtained in one session.

Gas chromatography of the female wash was carried out on a Silar 10 C glass capillary column; the effluent gas was split in a 10:1 ratio between the flame ionization (FID) and the electroantennograph-

Table I. Electroantennogram response values of male *Tortrix viridana* to (*Z*)- and (*E*)-alkenyl acetates. Values indicate equipotent stimulus amounts in a half log scale (0.01 represents a range from 0.0056 to 0.018 μ g; 0.03, from 0.018 to 0.056 μ g; etc.).

Z8 - 12 Ac	1	E8 - 12 Ac	0.3
Z9 - 12 Ac	0.3	E9 - 12 Ac	0.3
Z10 - 12 Ac	0.1	E10 - 12 Ac	0.1
Z9 - 13 Ac	0.3	E9 - 13 Ac	0.3
Z10 - 13 Ac	0.1	E10 - 13 Ac	0.1
Z11 - 13 Ac	0.03	E11 - 13 Ac	0.003
Z9 – 14Ac	0.3	E9 – 14 Ac	0.3
Z10 – 14Ac	0.1	E10 – 14 Ac	0.1
Z11 – 14Ac	0.03	E11 – 14 Ac	0.01
Z12 – 14Ac	0.1	E12 – 14 Ac	0.03
Z 10 - 15 Ac	0.3	E 10 – 15 Ac	0.1
Z 11 - 15 Ac	0.1	E 11 – 15 Ac	0.03
Z 12 - 15 Ac	0.1	E 12 – 15 Ac	0.1
Z11 – 16Ac	1	E 11 – 16 Ac	0.3
Z12 – 16Ac	1	E 12 – 16 Ac	1

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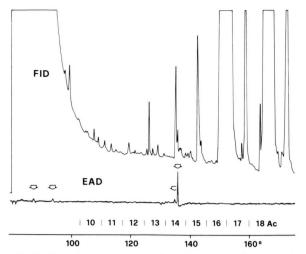


Fig. 1. Gas chromatogram of a *Tortrix viridana* female wash, recorded with flame ionization (FID)and electroantennographic detection (EAD). Arrows point at EAD peaks reproducible in multiple runs. Elution ranges of straightchain acetates are indicated. Conditions: Silar 10 C, $22 \text{ m} \times 0.3 \text{ mm}$ glass capillary column. Injection of 1 female equivalent at 70 °, after 2 min heated to 90 ° at 30 °/min, then heated at 6 °/min to end. The dip in the EAD trace following the main peak is caused by the electronic filter used to compensate for baseline drift.

ic detector (EAD) [3]. The filtered extract was introduced by splitless injection without prior cleanup. EAD recordings consistently showed a main peak at a retention time of Z11-14Ac which coincided with a small but sharp peak in the FID trace corresponding to an amount of a few nanograms per female (Fig. 1). The relative heights of the peaks obtained with the two detectors was approximately the same for the female wash and synthetic Z11-14Ac. Some chromatograms also showed a small EAD peak eluting just before Z11-14Ac, at a retention time similar to E11-14Ac or Z9-14Ac (Fig. 1). This peak appeared with the front slope of a major peak on the FID trace. Two other small EAD peaks were observed during elution of the solvent. These results indicate that other, unidentified compounds beside Z11-14Ac may be involved in the sex pheromone of T. viridana. No other EAD activity was observed within the retention range of 10 to 18 carbon acetates.

Further information on the main component was obtained by gas chromatography/mass spectrometry on a Finnigan 4000 GC-MS instrument equipped with a data system. Complete mass spectra (m/e 80 - 450, 1.4 sec/scan) were recorded using chemical ionization with iso-butane. After data acquisition,

the spectra were searched for tri- and tetradecenyl acetates. The mass chromatogram at m/e 255 contained a major peak whose retention time and mass spectrum (M + 1 = 255, M + 1 - CH₃COOH = 195) were as expected for a tetradecenyl acetate. The amount was estimated at 4 ng per female. No isomers were found at neighbouring retention times; if present, their amount would have been less than 5% of the main peak. No tridecenyl acetate was detect-

To allow an isomer assignment, mass fragmentograms were carried out at a slower temperature program (2 °/min), using the fragment of m/e 194 (M+-CH₃COOH) obtained in electron impact ionization. Its main peak coincided with that of synthetic Z11-14Ac on two different columns. The Silar 10 C column separated Z11-14Ac from all other tetradecenyl acetates except from E12-14Ac and 13-14Ac; separation of these isomers from Z11-14Ac was achieved on OV-101. The above evidence indicated that Z11-14Ac is the main component of the T. viridana sex pheromone.

Extensive attraction studies were conducted in June 1978 near Freiburg/Br. in the same test area as in 1977. Since the middle of the 19th century the riparian woodland in the Upper Rhine Valley between Basel and Freiburg has changed into incomplete brushwood by ground water subsidence. Only a few solitary oaks (*Quercus robur* L.) overtop the xerophilous shrubs; their crowns often extend to the ground and are periodically infested with *T. viridana*.

The field tests included Z11 - 14 Ac as the identified component, E11 - 13Ac as an attractant discovered in 1977, Z9- and E11 - 14 Ac as candidate secondary components, and the analogues Z11 - 13 Ac and Z11 - 14OH. Test compounds were at least 99% pure and contained less than 0.1% of the opposite geometric isomer. The chemicals were applied to the cavity of serum bottle caps (Tellergummikappen No. 90142, Auer Bittmann Soulié AG, Zürich) from hexane solutions. Tetra Traps with flaps [4] were hung from oak tree branches at eye level. Each test included a comparison of 6 treatments per tree in 6 replicates. Distances between treatments were 2-3 m, between replicates 20 - 200 m. Trap positions were systematically changed for each replicate. Catches were recorded weekly and the figures subjected to $\log (x+1)$ transformation followed by two-way analysis of variance.

E11 - 13 Ac, the compound showing highest EAG activity, was found to be a good attractant for T. viri-

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Table II. Tortrix viridana catches with (E)- and (Z)-11-tridecenyl and tetradecenyl acetates, Freiburg, 13 to 20 June 1978. Totales of 6 replicates. Numbers in the same column followed by the same letter are not significantly different at the 95% probability level as indicated by Duncan's multiple range test.

Chemical and amount per dispenser		Test number			
		1	2	3	4
untreated		6 d	3 c	6 c	2 c
E11 – 13 Ac	1 μg 10 μg 100 μg 1 mg	59 c 96 ab 185 a 105 ab	111 b 179 b	356 b 728 a	84 b 121 b
Z11 – 13 Ac	100 μg 1 mg 10 mg		372 a 575 a 537 a		
E11 – 14 Ac	100 μg 1 mg 10 mg			11 c 12 c 14 c	
Z11 – 14Ac	100 μg 1 mg				809 a 984 a

dana males (Table II), confirming the 1977 field results. The attractant threshold was below 1 μ g per dispenser. Z11-13 Ac and Z11-14 Ac, which had not been tested previously, both attracted significantly more males than E11-13 Ac when compared at a dispenser load of $100 \,\mu$ g. No significant catch improvement was obtained at dispenser loads above $100 \,\mu$ g with any of these three compounds. The other test compounds, E11-14 Ac (Table II), Z9-14 Ac and Z11-14 OH (not shown) were not attractive to T. viridana males when tested at $100 \,\mu$ g to $10 \,m$ g. The alcohol attracted large numbers of *Choristoneura sorbiana*, confirming other reports [5, 7].

When $100 \,\mu g \, E11-13 \, Ac$ was tested in combination with 2, $10 \, \text{or} \, 50 \, \mu g \, Z11-13 \, Ac$, no difference in attraction compared to the single compounds was observed (A 9:1 mixture of these two compounds, recently found to be attractive to T. viridana in field screening [6], was not included in our tests). Also, no synergistic or inhibitory effects were obtained with varying combinations of $E11-13 \, Ac$ and $Z11-14 \, Ac$.

Binary mixtures of Z9-14Ac and E11-14Ac with attractants were tested since the GC-EAD data suggested their possible involvement as secondary components. E11-14Ac showed no synergistic effect when added to either E11-13Ac (Table III) or Z11-14Ac (Table IV); at 3% of the main product

and above, it was inhibitory. Similarly, no synergism at low level, and a significant inhibition at 3% and above, was found for Z9-14Ac when combined with Z11-14Ac (Table IV). Thus, the specific attraction of T. viridana males to mixtures of Z11-14Ac and Z9-14Ac [7, 8] was not confirmed. A detailed report of the field data will be published elsewhere [9].

The detection of Z11 - 14 Ac in the T. viridana female secretion represents the first sex pheromone identification in the Tortricini tribe of the subfamily Tortricinae. Synthetic sex attractants, found by field screening for various other Tortricini spp., include

Z11 - 14 Ac for Tortrix sinapina [11],

E 11–14Ac for Croesia holmiana [10, 6], C. bergmanniana [6] and Acleris paradesana [11],

E11- and Z11 – 14Ac 9:1 to 8:2 for Acleris rhombana [5, 6],

Z11- and Z9 – 14Ac 5:5 for Spatalistis bifasciana [11],

Table III. Tortrix viridana catches with (E)-11-tridecenyl acetate alone and in combination with (E)-11-tetradecenyl acetate, Freiburg, 13 to 20 June 1978. Totals of 6 replicates. Numbers followed by the same letter are not significantly different at the 95% probability level as indicated by Duncan's multiple range test.

Amounts per dispenser			
E11 – 13 Ac [μg]	E11 – 14 Ac [μg]		
0	0	3 d	
10	0	117 b	
100	0	238 a	
100	3	85 bc	
100	10	57 c	
100	100	5 d	

Table IV. Tortrix viridana catches with (Z)-11-tetradecenyl acetate (100 µg per dispenser) in combination with (E)-11 or (Z)-9-tetradecenyl acetate, Freiburg, 20 to 28 June 1978. Totals of 6 replicates. Numbers followed by the same letter are not significantly different at the 95% probability level as indicated by Duncan's multiple range test.

Amount added	Chemical added to $Z11 - 14$ Ac		
	E11-14 Ac	Z9 – 14 Ac	
0	189 a	190 a	
0.1 µg	210 a	194 a	
0.3 μg	137 a	250 a	
lμg	115 a	242 a	
$3 \mu g$	26 b	53 b	
10 μg	2 c	29 c	

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Z11-14Ac and Z11-14-aldehyde and 5:5 to 1:9 for *Croesia ascoldana* and *C. conchyloides* [11], and E11- and Z11-14-aldehyde 8:2 to 9:1 for *Croesia semipurpurana* [12].

Our results indicate that either the pheromone component Z11-14Ac or the parapheromones E11-and Z11-13Ac can be used as single compounds to monitor the T. viridana flight by sex traps.

- Y. Tamaki, Chemical Control of Insect Behavior (H. H. Shorey and J. J. McKelvey, eds.), p. 253, Wiley-Interscience, New York 1977.
- [2] E. Priesner, Chemical Ecology: Odour Communication in Animals (F. J. Ritter, ed.), p. 57, Elsevier, Amsterdam 1979.
- [3] H. Arn, E. Städler, and S. Rauscher, Z. Naturforsch. **30 c**, 722 (1975).
- [4] H. Árn, S. Rauscher, and A. Schmid, Mitt. Schweiz. Ent. Ges. 52, 49 (1979).
- [5] C. Descoins and B. Frérot, Chemical Ecology: Odour Communication in Animals (F. J. Ritter, ed.), p. 181, Elsevier, Amsterdam 1979.
- [6] S. Voerman, Chemical Ecology: Odour Communication in Animals (F. J. Ritter, ed.), p. 353, Elsevier, Amsterdam 1979.

An unusual discovery is the outstanding EAG activity of $E11-13\,\mathrm{Ac}$, a chemical not found in the female moth. EAG activities of the test compounds (Table I) will be discussed further in the context of single receptor types in the olfactory system of the T. viridana male antenna [13].

- [7] B. Frérot, C. R. Réunion sur les Phéromones Sexuelles des Insectes, Publ. INRA, p. 47, Montfavet 1977.
- [8] C. Descoins, personal communication.
- [9] H. Bogenschütz et al., in preparation.
- [10] H. Arn, C. Schwarz, H. Limacher, and E. Mani, Experientia 30, 1142 (1974).
- [11] T. Ando, H. Kuroko, S. Nakagaki, O. Saito, T. Oku, and T. Takahashi, Agric. Biol. Chem. 42, 1081 (1978).
- [12] J. Weatherston, G. G. Grant, L. M. MacDonald, D. Frech, R. A. Werner, C. C. Leznoff, and T. M. Fyles, J. Chem. Ecol. 4, 543 (1978).
- [13] E. Priesner et al., in preparation.