

Inhibition of Electron Transport on the Oxygen-Evolving Side of Photosystem II by an Antiserum to a Polypeptide Isolated from the Thylakoid Membrane

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Dedicated to Prof. Dr. J. Straub on the Occasion of His 65th Birthday

Thylakoid Membrane, Polypeptide, Photosystem II, Antiserum

A polypeptide fraction with the apparent molecular weight 11000 was isolated from stroma-freed chloroplasts from *Antirrhinum majus*. An antiserum to this polypeptide fraction inhibits photosynthetic electron transport in chloroplasts from *Nicotiana tabacum*. The relative degree of inhibition is pH dependent and has its maximum at pH 7.4. The maximal inhibition observed was 93%. The dependence of the inhibition on the amount of antiserum yields a sigmoidal curve which hints at a cooperative effect. A calculation of the Hill interaction coefficient gave the value of 10. The inhibition occurs on the water splitting side of photosystem II between the sites of electron donation of tetramethyl benzidine and diphenylcarbazine. Tetramethyl benzidine donates its electrons before the site where diphenylcarbazine feeds in its electrons. Analysis of the steady state level of the variable fluorescence also indicates that the inhibition site is on the water splitting side of photosystem II. Tris-washed chloroplasts are equally inhibited by the antiserum and the inhibition is also observed in the presence of an inhibitor of photophosphorylation like dicyclohexyl carbodiimide and in the presence of the uncoupler carbonylcyanide *m*-chlorophenyl hydrazine (CCCP) which means that the inhibitory action is directed towards the electron transport chain. Valinomycin which is supposed to affect the cation permeability of the thylakoid membrane has no influence on the inhibitory action of the antiserum. The same is valid for gramicidin. Methylamine on the other hand can induce a state in the thylakoids in which the antiserum is not effective. If the antibodies are already adsorbed prior to the methylamine addition then the high inhibitory effect by the antiserum remains unchanged upon addition of methylamine.

From the experiments it follows that a component from the vicinity of photosystem II is accessible to antibodies that is, the component is located in the outer surface of the thylakoid membrane. It appears that the inhibitory effect is produced in the course of the light reaction.

In previous publications we have reported on the isolation of different polypeptide fractions from the thylakoid membrane^{1,2}. Antisera to these polypeptides affected photosynthetic electron transport in different ways. Among these, an antiserum to the polypeptide with the molecular weight 11000 inhibited photosynthetic electron transport on the O₂-evolving side of photosystem II which obviously means that this polypeptide is a component from the vicinity of photosystem II and that this component is located in the outer surface of the thylakoid membrane. Earlier investigations with antibodies to a membrane particle fraction, obtained by deoxycholate treatment, and to lutein and neo-

xanthin also showed an inhibition of electron transport on the oxygen-evolving side of photosystem II³⁻⁵. However, in these cases the degree of inhibition was generally low and found to be between 15–20%. With the antiserum to polypeptide 11000 we observed much higher degrees of inhibition which offered us the opportunity to characterize the inhibition site of this antiserum more in detail.

Materials and Methods

The isolation of the polypeptide fraction 11000 from stroma-freed chloroplasts of *Antirrhinum majus* was described previously².

Gel electrophoreses were carried out according to Weber and Osborn⁶. In the dodecyl sulfate phosphate gels the acrylamide concentration was 10% whereas in the dodecyl sulfate Tris gels the acrylamide concentration of the separating gel was 12.5%. The gels were stained with Coomassie blue and scanned at $\lambda = 558$ nm on a Zeiss PMQ II spectrophotometer fitted with a DISC Ansatz ZK 4.

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Abbreviations: Polypeptide 11000 stands for a polypeptide fraction with the apparent molecular weight 11000; DPIP, 2,6-dichlorophenol indophenol; PMS, phenazine methosulfate; DCMU, dichlorophenyl dimethyl urea.

The isolation of photochemically active stroma-free swellable chloroplasts from *Nicotiana tabacum* var. John Williams Broadleaf was also described earlier^{7,8}.

Light reactions were mainly carried out as reported earlier⁵. A large part of the electron transport reactions was carried out as anthraquinone-2-sulfonate Mehler reaction by measuring the oxygen consumption resulting from the autoxidation of the reduced electron acceptor anthraquinone-2-sulfonate in a Warburg apparatus at 20 °C. The assay composition was mainly as described by Harth *et al.*^{9,10}. The reaction mixture contained: in the main vessel compartment 1.8 ml 0.075 M tricine 0.2 M KCl pH 7.6, 0.1 ml 0.1 M MgCl₂, 0.1 ml 0.05 M K₂HPO₄, 0.1 ml 0.05 M ADP, chloroplasts corresponding to 200 µg chlorophyll, antiserum or control serum as indicated, 0.1 ml 3×10^{-3} M anthraquinone-2-sulfonate and 0.1 ml 10^{-3} M NaCN; in the side arm 0.1 ml 0.1 M ascorbate and 0.1 ml 0.005 M tetramethyl benzidine or benzidine and in the center well 0.1 ml 30% KOH. The final volume was 3 ml. The reaction vessels were illuminated from below with $30000 \text{ ergs} \cdot \text{sec}^{-1} \cdot \text{cm}^{-2}$ of red light $580 \text{ nm} < \lambda < 700 \text{ nm}$, transmitted through a plexiglas filter.

Proton translocation and photophosphorylation if not measured *via* ³²P-incorporation as described earlier⁷, were measured according to Dilley¹¹, using a pH-meter model PHM 26 from Radiometer, Copenhagen equipped with a Servogor S Model RE541 recorder.

The preparation of the antiserum to the polypeptide 11000 was described previously as well as the agglutination reactions of this antiserum with chloroplast preparations².

Fluorescence measurements were carried out with a modified Perkin-Elmer Fluorescence Spectrophotometer MPF-3 equipped with a Siemens oscillograph model Oscillar MO 7114.

Results

1. Characterization of the polypeptide fraction 11000

The polypeptide fraction which served for the immunization migrated in the dodecyl sulfate polyacrylamide gel electrophoresis as a single narrow band provided Tris-buffer pH 8.8 was used. Fig. 1 shows an optical scan of the electropherogram after staining of the gel with Coomassie blue. If the electrophoresis was carried out in phosphate buffer at pH 7.1 the polypeptide migrated as a broad diffuse

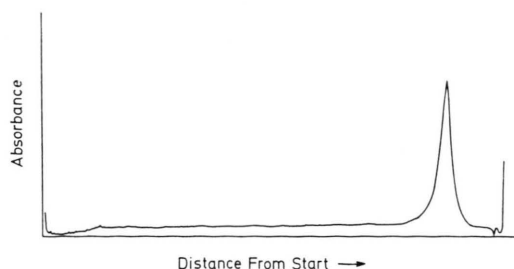


Fig. 1. Optical scan of the dodecylsulfate polyacrylamide gel electropherogram of the polypeptide 11000. Buffer pH 8.8.

band. Orienting determinations of the amino terminal amino acids revealed that the preparation consists of several polypeptide species which all exhibit the apparent molecular weight of about 11000. The immunological characterization by immune electrophoresis and Ouchterlony double diffusion also hints at more than one component.

2. Determination of optimal experimental conditions

The photoreduction of anthraquinone-2-sulfonate with water as the electron donor in stroma-free swellable chloroplasts from tobacco is considerably enhanced if benzidine/ascorbate or tetramethyl benzidine/ascorbate are added to the reaction mixture. The anthraquinone-2-sulfonate Hill reaction as well as the photoreduction of anthraquinone-2-sulfonate with tetramethyl benzidine/ascorbate as the electron donors are inhibited by the antiserum to the polypeptide 11000. This inhibition has a marked pH dependence which is shown in Fig. 2. The

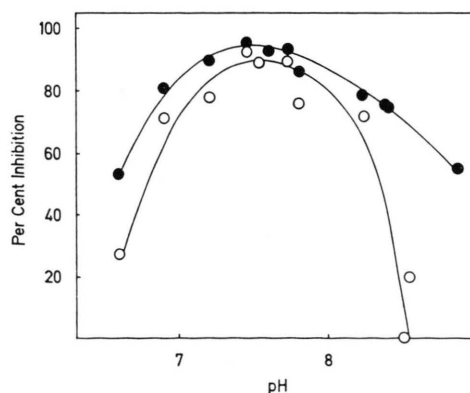


Fig. 2. pH-Dependence of the degree of inhibition of photosynthetic electron transport in tobacco chloroplasts by the antiserum to polypeptide 11000. ●, Reaction in the system tetramethyl benzidine/ascorbate → anthraquinone-2-sulfonate. ○, Reaction with water as the electron donor and anthraquinone-2-sulfonate as the electron acceptor.

highest inhibition was at pH 7.4 and the maximal inhibition was 93%. The effect of the antiserum on the Hill reaction lies in a narrower pH range than that of the photooxidation of tetramethyl benzidine. The dependence of the inhibition at this optimal pH of 7.4 on the amount of antiserum added yields a sigmoidal curve shape, which hints at a cooperative effect (Fig. 3). A preliminary calculation of

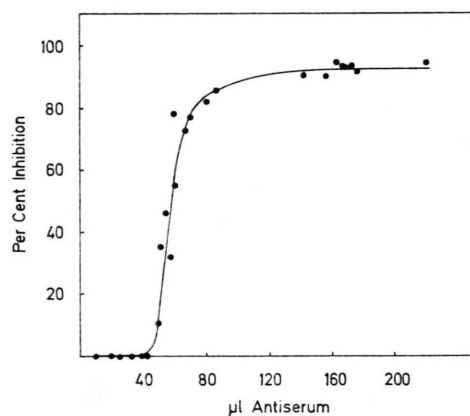


Fig. 3. Dependence of the degree of inhibition on the amount of added antiserum to polypeptide 11000 at pH 7.4. Reaction in the system tetramethyl benzidine/ascorbate $\rightarrow$ anthraquinone-2-sulfonate.

the Hill interaction coefficient yields 10. After the determination of the optimal experimental conditions all reactions described from here onward were carried out with chloroplasts corresponding to 200 μ g total chlorophyll at pH 7.4–7.5 and in the presence of a saturating amount of antiserum (0.1 ml).

3. Localization of the inhibition site by the antiserum

From a number of experiments the reaction sequence shown in Fig. 4 gives the localization of the inhibition site by the antiserum. The assay was carried out in two Warburg vessels each of which had 3 side arms. One Warburg vessel contained the antiserum and the other the control serum. Two side arms contained each the electron donor to photosystem II namely tetramethyl benzidine and diphenylcarbazide; the third side arm contained 2,6-dichlorophenol indophenol as electron donor to photosystem I. Anthraquinone-2-sulfonate served as an electron acceptor. Before addition of an artificial electron donor to photosystem II water serves as an electron donor. This Hill reaction is clearly

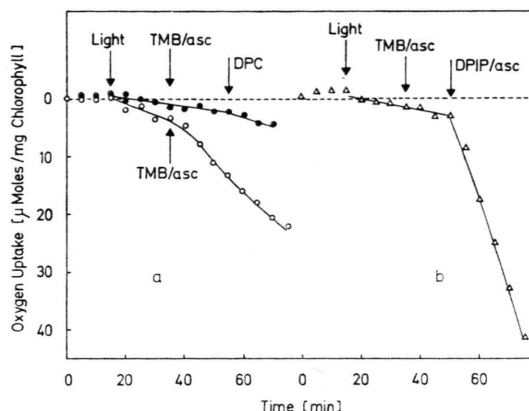


Fig. 4. Localization of the inhibition site of the antiserum in the photosynthetic electron transport chain. a. ●, Reaction of tobacco chloroplasts in the system $H_2O \rightarrow$ anthraquinone-2-sulfonate in the presence of antiserum. The arrows indicate the addition tetramethyl benzidine/ascorbate (TMB/Asc) and diphenylcarbazide (DPC). ○, Reaction in the same electron transport system but in the presence of control serum. The arrow indicates the addition of TMB/Asc. b. ▲, Same reaction as ● in Fig. 4a in the presence of antiserum. The arrows indicate the addition of TMB/Asc and 2,6-dichlorophenol indophenol (DPIP)/ascorbate.

inhibited by the antiserum. Addition of tetramethyl benzidine/ascorbate at the time indicated in Fig. 4 does not influence anymore the reaction rate in the sample with antiserum whereas the same sample responds to the addition of diphenylcarbazide. However, the control assay in the presence of the same amount of control serum reacts upon addition of tetramethyl benzidine/ascorbate. From this it follows that the inhibition site is between the sites of electron donation of tetramethyl benzidine/ascorbate and that of diphenylcarbazide. Tetramethyl benzidine thus feeds in the electrons before diphenylcarbazide. The localization of the inhibition site in the region of photosystem II is further supported by the fact that after inhibition of electron transport by the antiserum the typical donor system to photosystem I 2,6-dichlorophenol indophenol/ascorbate yields high rates of anthraquinone-2-sulfonate photoreduction (Fig. 4). In other experiments we were always able to show that the antiserum does not affect photosystem I reactions (Table I, Fig. 13). These results are schematically summarized in Fig. 5.

As a general observation it must be added that despite the fact that agglutination of the chloroplasts occurs immediately upon addition of the antiserum, the inhibition of electron transport becomes only manifest in the course of the illumination period.

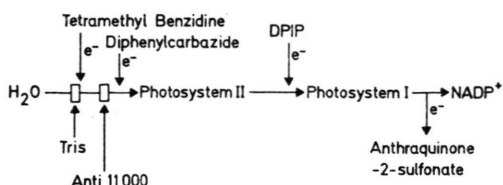


Fig. 5. Scheme to illustrate the inhibition site of the photosynthetic electron transport by the antiserum to polypeptide 11000.

This becomes especially obvious in the following fluorescence measurements.

The influence of light on the inhibitory action of the antiserum and the above described localization of the inhibition site is further strengthened by

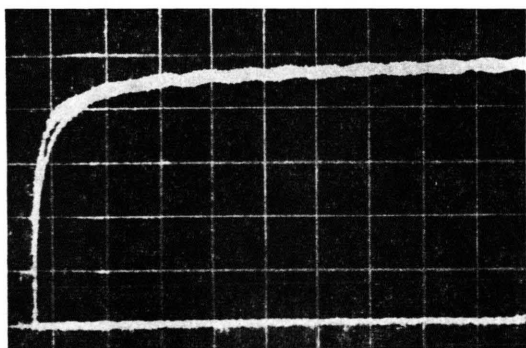


Fig. 6. Fluorescence rise in a chloroplast preparation in the assay system tetramethyl benzidine/ascorbate $\rightarrow$ anthraquinone-2-sulfonate without illumination prior to the fluorescence assay. One assay with antiserum to polypeptide 11000, the other assay with control serum. Excitation wavelength 440 nm, excitation slit 40 nm; emission wavelength 685 nm, emission slit 34 nm. One grade of the abscissa represents 0.2 sec.

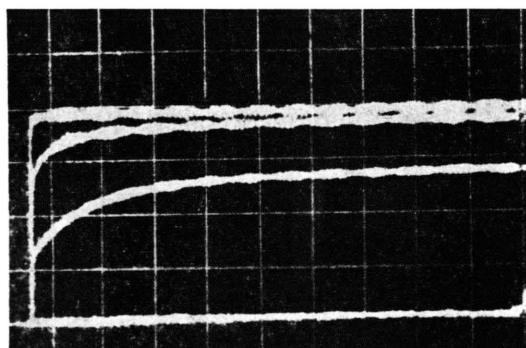


Fig. 7. Fluorescence rise curve of the same assay as in Fig. 6. The assay mixture was shaken in the Warburg apparatus at $30000 \text{ ergs} \cdot \text{sec}^{-1} \cdot \text{cm}^{-2}$ of red light $580 \text{ nm} < \lambda < 700 \text{ nm}$ until the inhibitory effect reached by the antiserum was 93% inhibition. Then the assay was kept in the dark for 20 min prior to the fluorescence measurement. Lower rise curve in the presence of antiserum; middle curve in the presence of control serum; upper curve in the presence of control serum plus $10 \mu\text{mol DCMU}$.

measurements of the steady state level of the variable fluorescence. The assay system contained tetramethyl benzidine/ascorbate as the electron donor and anthraquinone-2-sulfonate as the acceptor. A dark adapted mixture which has received no light prior to fluorescence excitation shows no influence of the antiserum on the steady state level of fluorescence (Fig. 6). If, however, the reaction mixture has been illuminated with red light until the inhibitory effect by the antiserum was established then the fluorescence level was lower in the presence of antiserum than in the presence of control serum (Fig. 7). This shows according to the literature that the inhibition site of the antiserum is on the water splitting side of photosystem II¹³. If inhibition strikes on the oxygen-evolving side of photosystem II then the fluorescence quencher Q is not or only slowly reduced. In this condition electrons are faster removed by photosystem I than they are supplied by photosystem II. This means that the quencher Q is preponderantly in the oxidized state in which the fluorescence is quenched and consequently low. If the drain of electrons through photosystem I is prevented by the addition of DCMU, fluorescence is immediately high because the quencher Q stays in the reduced state (Fig. 7). Not shown in Fig. 7 is the effect of DCMU on the assay with the antiserum. The addition of DCMU blocks the reoxidation of the reduced quencher which also in the presence of antiserum makes fluorescence rise up again.

As the chloroplasts are agglutinated by the antiserum they scatter stronger than in the presence of control serum. Therefore, an antiserum to a polypeptide 24000 which strongly agglutinates chloroplasts but which does not affect electron transport was tested for fluorescence with and without preillumination in the electron transport system TMB/Asc $\rightarrow$ anthraquinone-2-sulfonate. No effect of the antiserum on the steady state level of variable fluorescence was observed which excludes scattering effects as an error source.

Tris-washed chloroplasts show also in the fluorescence assay that electron transport in the system tetramethyl benzidine/ascorbate $\rightarrow$ anthraquinone-2-sulfonate is inhibited by the antiserum on the donor side of photosystem II.

4. Influence of uncouplers and inhibitors

The high degree of inhibition at first let us suspect that the antibody action might be somehow

connected with the proton or cation translocation or with some kind of permeability change. The inhibitor of photophosphorylation dicyclohexyl carbodiimide (DCCD) applied in concentrations which entirely suppress the formation of the proton gradient and photophosphorylation (Table I) does not remove the inhibitory effect by the antiserum on the electron transport reaction (Fig. 8). The observed

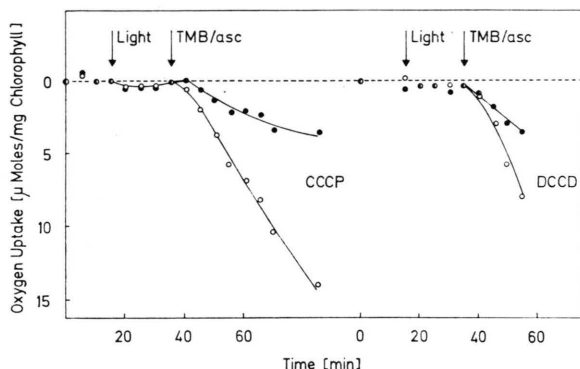


Fig. 8. Influence of the uncoupler carbonylcyanide *m*-chlorophenyl hydrazone (CCCP) and of the photophosphorylation inhibitor *N,N'*-dicyclohexyl carbodiimide (DCCD) on the inhibition degree of photosynthetic electron transport in tobacco chloroplasts caused by the antiserum in the electron transport system $\text{H}_2\text{O} \rightarrow \text{anthraquinone-2-sulfonate}$. The assay mixture was 3.3×10^{-6} M in CCCP or 10^{-4} M in DCCD. ●, Assay in the presence of antiserum, ○, in the presence of control serum.

degree of inhibition by the antiserum is still 65%. The reaction rate itself is lower due to the presence of DCCP in the reaction mixture. This leads to the conclusion that primarily the electron transport and not the photophosphorylation is inhibited by the antiserum. The uncoupler carbonylcyanide *m*-chlorophenyl hydrazone applied in concentrations which already inhibit electron transport does also not remove the inhibitory effect of the antiserum (Fig. 8). Again, this is taken to mean that the effect of the antiserum is primarily directed towards electron transport and not towards photophosphorylation.

The antibiotic valinomycin also does not influence the inhibitory action of the antiserum on electron transport. The antibiotic is supposed to increase the cation permeability of the membrane¹². It is seen from Fig. 9 that in the presence of valinomycin the Hill reaction rate of the control decreases in the light until the rate is zero. Upon addition of tetramethyl benzidine the electron transport rate is fully restored. The relative degree of inhibition of

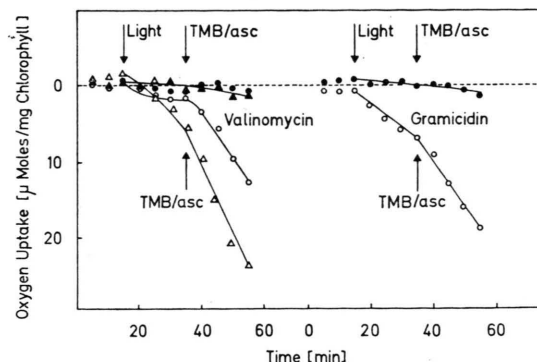


Fig. 9. Influence of the antibiotics valinomycin and gramicidin on the degree of inhibition by the antiserum. The antibiotic concentration in the reaction mixture was $0.6 \mu\text{g}$ gramicidin/ml or $0.1 \mu\text{M}$ valinomycin. Otherwise conditions as in Fig. 8. ▲, △, Controls without antibiotic addition; △, in the presence of control serum; ▲, in the presence of antiserum.

the electron transport rate by the antiserum in the system tetramethyl benzidine/ascorbate $\rightarrow$ anthraquinone-2-sulfonate is not changed by the presence of valinomycin (Fig. 9). The antibiotic gramicidin which affects the proton permeability and thereby inhibits photophosphorylation does not alter the influence of the antiserum on electron transport (Fig. 9). The relative degree of inhibition is essentially the same in the presence and absence of gramicidin at pH 7.5. Again, no effect by either antibiotic, namely valinomycin and gramicidin, was observed on the effect of the antiserum on the electron transport rate.

The effect of the uncoupler methylamine is clearly different from that of CCCP. This is demonstrated in Fig. 10. If methylamine is added to the assay after the establishment of inhibition by the antiserum no relieve is observed. If, however, methylamine is added to the chloroplast suspension before serum addition the antiserum has no effect at all on the reaction rate. From this we infer that methylamine changes the morphological structure of the lamellar system^{14,15} and possibly also the state of the thylakoid membrane to a condition in which the antigen is not accessible anymore to antibodies or where binding of antibodies does not induce inhibition (Fig. 10).

Tris-washing of the chloroplasts abolishes the capacity for oxygen evolution¹⁶. Addition of the artificial electron donor couple tetramethyl benzidine/ascorbate restores electron transport to NADP^+ or to an artificial electron acceptor such as

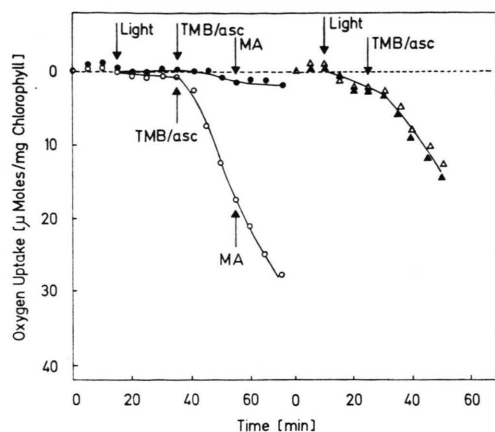


Fig. 10. Characterization of the influence of methylamine on the degree of inhibition by the antiserum. ●, Electron transport reaction in the presence of antiserum in the system $\text{H}_2\text{O} \rightarrow$ anthraquinone-2-sulfonate. The arrows indicate the addition of TMB/Asc and methylamine. ○, Same reaction in the presence of control serum. ▲, Electron transport reaction in the system TMB/Asc $\rightarrow$ anthraquinone-2-sulfonate in the presence of antiserum. Here methylamine was given to the assay before antiserum addition. △, Same reaction as (▲) but in the presence of control serum.

anthraquinone-2-sulfonate. In this condition the antiserum also exerts its inhibitory effect (Fig. 11). It appears that the relative degree of inhibition is lower than in unwashed chloroplasts. If instead of tetramethyl benzidine the artificial donor diphenylcarbazide is added then the antiserum does not influence the electron transport rate (Fig. 11). This also clearly shows that the inhibition site of the

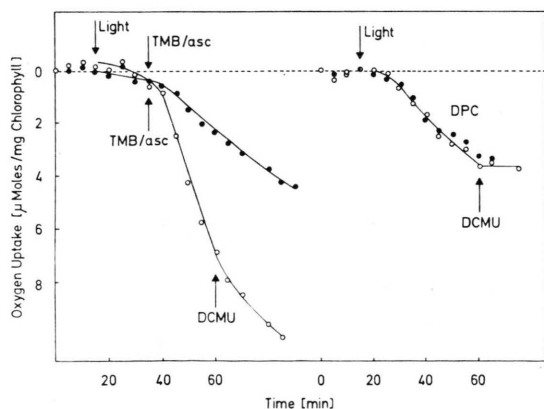


Fig. 11. Effect of the antiserum on the electron transport in the Tris-buffer washed chloroplasts from tobacco. The electron transport system is TMB/Asc $\rightarrow$ anthraquinone-2-sulfonate or diphenylcarbazide (DPC) $\rightarrow$ anthraquinone-2-sulfonate. ●, In the presence of antiserum, ○, in the presence of control serum. Concentration of DCMU after addition $3.3 \times 10^{-6} \text{ M}$ in the TMB experiment and 2×10^{-5} in the DPC-assay.

antiserum lies between the site of electron donation of tetramethyl benzidine and that of diphenylcarbazide. Occasionally, it was observed that the antiserum to polypeptide 11000 did not inhibit anymore the electron transport rate in Tris-treated chloroplasts in the presence of the donor couple tetramethyl benzidine/ascorbate. This led us to the idea that Tris-buffer treatment might wash out the polypeptide 11000. This is not the case. By means of our different antisera we have found that amongst other proteins, coupling factor, reductase and plastocyanin were washed out. In contrast to these antisera the antiserum to the polypeptide 11000 gave no precipitation with the supernatant from the Tris-wash. The reason for the loss of inhibition might be that a too complete removal of the above named proteins leads to an altered molecular structure which does not permit access of the antibodies to the antigen or that the active component has lost its functionality. In this latter case antibodies might well become adsorbed. This leads to the formal explanation that the site of inhibition of the antiserum is closely behind the Tris-block. Furthermore, in Tris-washed chloroplasts manganese II ions can serve as electron donor to photosystem II. In the electron transport system $\text{Mn}^{2+} \rightarrow$ anthraquinone-2-sulfonate the antiserum inhibits electron

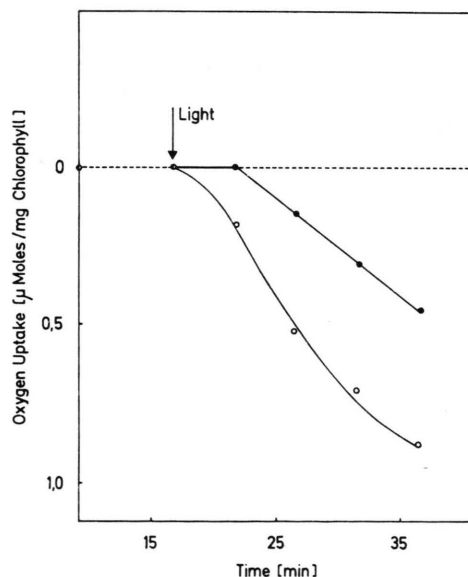


Fig. 12. Electron transport to anthraquinone-2-sulfonate in Tris-washed chloroplasts from tobacco with manganese II as the electron donor. ●, Reaction in the presence of antiserum, ○, reaction in the presence of control serum.

flow, showing that the site of electron donation of manganese is before the inhibition site by the antiserum (Fig. 12).

5. Effect of the antiserum on photophosphorylation reactions

Good photophosphorylation rates are only obtained if the illumination period is short. Under such conditions photophosphorylation is affected to a lower degree than is electron transport under the above described experimental conditions. Phenazine methosulfonate-mediated cyclic photophosphorylation which is a photosystem I reaction is not inhibited by the antiserum (Table I). In line with

Table I. Effect of the antiserum to polypeptide 11000 on photophosphorylation reactions in Tris-buffer washed chloroplasts from wild type tobacco.

Additions	$\mu\text{mol}[^{32}\text{P}]\text{ATP}$ formed $\cdot (\text{mg}$ Chl) $^{-1} \cdot \text{h}^{-1}$	ATP/2e
PMS + antiserum to polypeptide 11000	205	—
PMS + control serum	179	—
Tetramethyl benzidine/ascorbate/ anthraquinone-2-sulfonate		
+ antiserum to polypeptide 11000	2.9	0.58
+ control serum	9.00	0.57
+ control serum + DCCD	0	
+ control serum + DCMU	0.002	

The PMS mediated photophosphorylation assay was illuminated with 120000 lx white light. The other assays were illuminated for 4 min with 30000 ergs $\cdot \text{sec}^{-1} \cdot \text{cm}^{-2}$ of red light $580 \text{ nm} < \lambda < 700 \text{ nm}$ at $14-15^\circ \text{C}$. The electron transport rate for TMB/Asc + control serum $\rightarrow$ anthraquinone-2-sulfonate was $15.7 \mu\text{mol O}_2 \text{ uptake} \cdot (\text{mg Chlorophyll})^{-1} \cdot \text{h}^{-1}$ and $5.0 \mu\text{mol O}_2 \text{ uptake} \cdot (\text{mg Chlorophyll})^{-1} \cdot \text{h}^{-1}$ in the presence of antiserum. DCCD, dicyclohexyl carbodiimide was 10^{-4} M in the assay; DCMU 10^{-5} M in the PMS assay and where indicated.

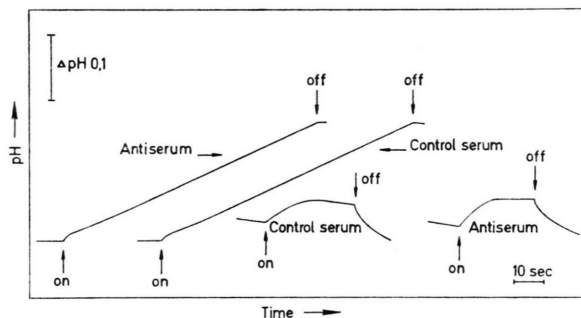


Fig. 13. Effect of the antiserum to polypeptide 11000 on proton translocation and photophosphorylation in tobacco chloroplasts in the presence of PMS.

this observation the rate of proton translocation with PMS as mediator is also not affected by the antiserum (Fig. 13).

In Tris-buffer washed chloroplasts the electron transport system tetramethyl benzidine/ascorbate $\rightarrow$ anthraquinone-2-sulfonate supports noncyclic photophosphorylation which is DCMU-sensitive and is consequently a photosystem II reaction (Table I). This photophosphorylation reaction is inhibited by the antiserum. The inhibition is a consequence of the coupling of photophosphorylation to electron transport which is clearly seen from the ATP/2e-ratio which is the same in the presence of antiserum and control serum.

Discussion

In the present paper we have described the effects of an antiserum to a polypeptide with the apparent molecular weight 11000. This antiserum inhibits photosynthetic electron transport between the sites of electron donation of tetramethyl benzidine and diphenylcarbazide. According to the literature, these donors are supposed to feed in their electrons on the oxygen evolving side of photosystem II^{10,16}. From all hitherto investigated antisera to purified polypeptide fractions this is the only one with such properties, whereas all the others inhibit either photosystem I reactions or inhibit between the two photosystems^{1,2}. Consequently, the polypeptide participates in reactions which occur on the water splitting side of photosystem II. This does not mean, however, that polypeptide 11000 is necessarily an electron transport component. The inhibition could also be due to a conformational change within the polypeptide 11000 itself or due to a conformational change of a protein molecule from which polypeptide 11000 is a part. Moreover, the inhibition can be due to an alteration of the molecular structure of the membrane, which would be the change of positions of proteins in relation to each other.

The fact that a component which plays a role in the vicinity of photosystem II and in the water splitting reaction is located in the outer surface of the thylakoid membrane is at first glance opposed to concepts advanced in the literature¹⁷. According to this view photosystem II is located on the inner surface of the thylakoid membrane. However, we should like to emphasize at this point that our ob-

servation does not permit a conclusion as to the location where the splitting of the water molecule really occurs. Also it must be borne in mind that the water splitting reaction and photosystem II need not be located in the membrane at the same site. In addition, we have shown earlier that antisera to a membrane particle fraction and to lutein and neoxanthin inhibit electron transport on the water splitting side of photosystem II³⁻⁵. Therefore, these components must be located in the outer surface of the thylakoid membrane. Braun and Govindjee also observed an inhibition on the water splitting side of photosystem II by an antiserum to a chemically undefined antigen mixture^{18,19}. The degree of inhibition was in all these cases low and found to be around 15% whereas here in the case of the antibodies to polypeptide 11000 the degree of inhibition is 93%. At first glance this could mean that the polypeptide 11000 is preponderantly located in the intergrana region of the lamellar system which is accessible to antibodies. This, however, is rather improbable, because until now it always appeared that the capacity for oxygen evolution was somehow connected with the presence of stacked thylakoids and that antibodies did not enter into partitions^{15,20}. At this point a satisfactory explanation for the high effectiveness of the antiserum cannot be given. One possible interpretation would be that under our experimental conditions the polypeptide 11000, originally located in the partitions, becomes accessible to antibodies due to a rearrangement. This rearrangement could be induced or favored by light.

Another, not easily explainable effect, is the influence of light on the onset of inhibition. This influence of light is best seen in the fluorescence assay and was originally found in the ferricyanide Hill reaction². A minimum reaction time in the light of two minutes was necessary until the effect just began to show up. A mere preillumination in the presence of antiserum but in the absence of the electron transport components of the assay system and subsequent completion of the assay gave no or only a low inhibition.

It looks as if the complete assay system establishes in the light a membrane condition, possibly a redox state, which represents the condition in which inhibition becomes possible. Here, the observation by Giaquinta *et al.* would fit into ours²¹. These authors observed that diazonium benzene [³⁵S]sulfonate was incorporated into the lamellar system when electron flux through photosystem II and plastoquinone occurred. Also, in this case a rearrangement of the membrane or in the membrane might be induced by light.

Finally, it should be noted that, as shown earlier and again in this paper, antisera to the proteins of the thylakoid membrane are at best only slightly species specific.

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¹ W. Menke, F. Koenig, A. Radunz, and G. H. Schmid, *FEBS Letters* **49**, 372 [1975].

² F. Koenig, G. H. Schmid, A. Radunz, B. Pineau, and W. Menke, *FEBS Letters* **62**, 342 [1976].

³ F. Koenig, W. Menke, H. Craubner, G. H. Schmid, and A. Radunz, *Z. Naturforsch.* **27b**, 1225 [1972].

⁴ A. Radunz and G. H. Schmid, *Z. Naturforsch.* **28c**, 36 [1972].

⁵ A. Radunz and G. H. Schmid, *Z. Naturforsch.* **30c**, 622 [1975].

⁶ K. Weber and M. Osborn, *The Proteins* (H. Neurath and R. L. Hill, eds.), p. 179, Acad. Press, New York, San Francisco, London 1975.

⁷ G. H. Schmid, A. Radunz, and W. Menke, *Z. Naturforsch.* **30c**, 201 [1975].

⁸ P. H. Homann and G. H. Schmid, *Plant Physiol.* **42**, 1619 [1967].

⁹ E. Harth, S. Reimer, and A. Trebst, *FEBS Letters* **42**, 165 [1974].

¹⁰ E. Harth, W. Oettmeier, and A. Trebst, *FEBS Letters* **43**, 23 [1974].

¹¹ R. A. Dilley, *Methods in Enzymology*, **Vol. XXIV** (A. San Pietro, ed.), pp. 68-74, Acad. Press, New York and London 1972.

¹² S. J. D. Karlish, N. Shavit, and M. Avron, *Europ. J. Biochem.* **9**, 291 [1969].

¹³ H. G. Aach, U. F. Franck, and R. Bauer, *Naturwissenschaften* **58**, 525 [1971].

¹⁴ S. Izawa and N. E. Good, *Plant Physiol.* **41**, 544 [1966].

¹⁵ G. H. Schmid and A. Radunz, *Z. Naturforsch.* **29c**, 384 [1974].

¹⁶ T. Yamashita and W. Butler, *Plant Physiol.* **44**, 435 [1969].

¹⁷ Govindjee, *Bioenergetics of Photosynthesis*, Cell Biology. Acad. Press, New York, San Francisco, London 1975.

¹⁸ B. Z. Braun and Govindjee, *FEBS Letters* **25**, 143 [1972].

¹⁹ B. Z. Braun and Govindjee, *Plant Science Letters* **3**, 219 [1974].

²⁰ W. Menke, *Physiol. Vég.* **II**, 231 [1973].

²¹ R. T. Giaquinta, D. R. Ort, and R. A. Dilley, *Biochemistry* **14**, 4392 [1975].