

Supporting Information

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Microwave synthesis of a blue luminescent silver(I) coordination polymer with a rigid tris-triazole ligand

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Table S1 The effect of reaction parameters on the yield of complex **1**.

Entry	Temperature (°C)	Irradiation time (min)	Microwave power (W)	Yield (%)
1	100	30	300	56
2	110	30	300	81
3	120	30	300	73
4	130	30	300	68
5	110	15	300	72
6	110	45	300	81
7	110	30	200	39
8	110	30	250	52
9	110	30	350	79

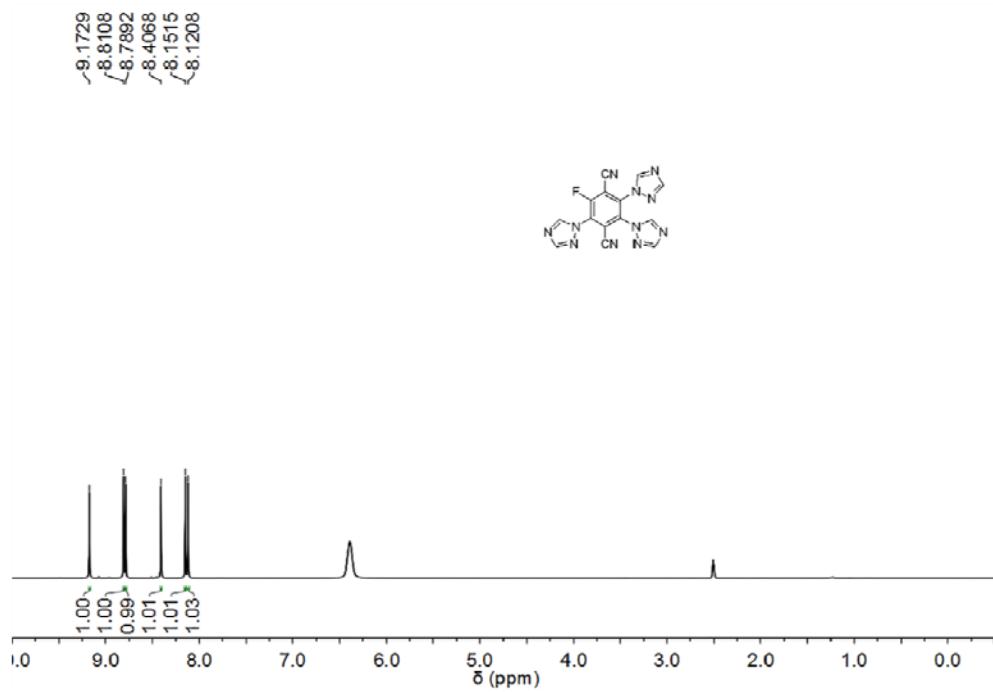


Fig. S1 ^1H NMR spectrum of
2-fluoro-3,5,6-tri(1H-1,2,4-triazol-1-yl)-terephthalonitrile (L^1).

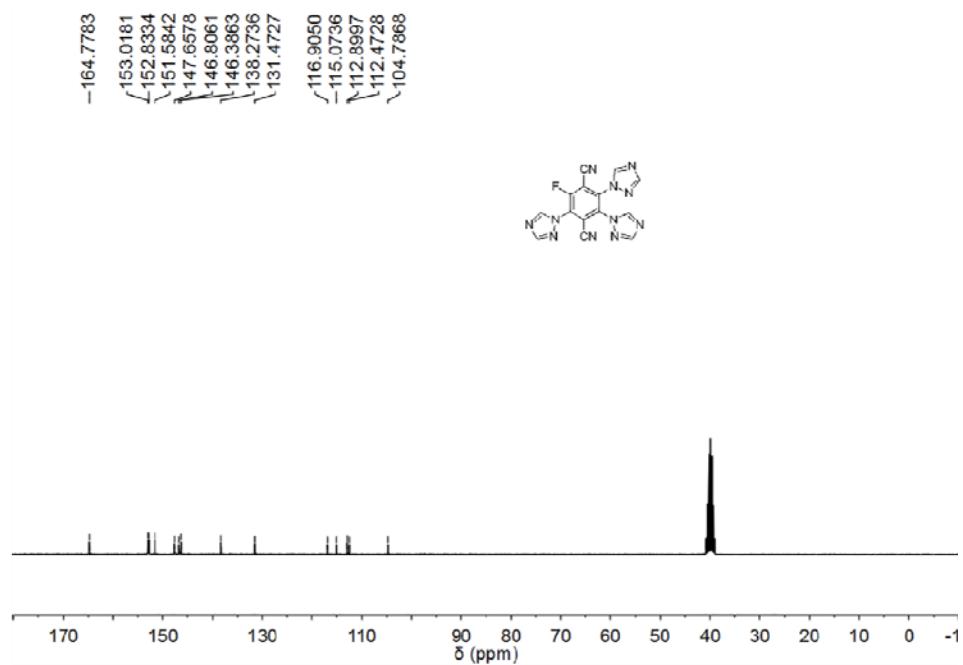


Fig. S2 ^{13}C NMR spectrum of
2-fluoro-3,5,6-tri(1H-1,2,4-triazol-1-yl)-terephthalonitrile (L^1).

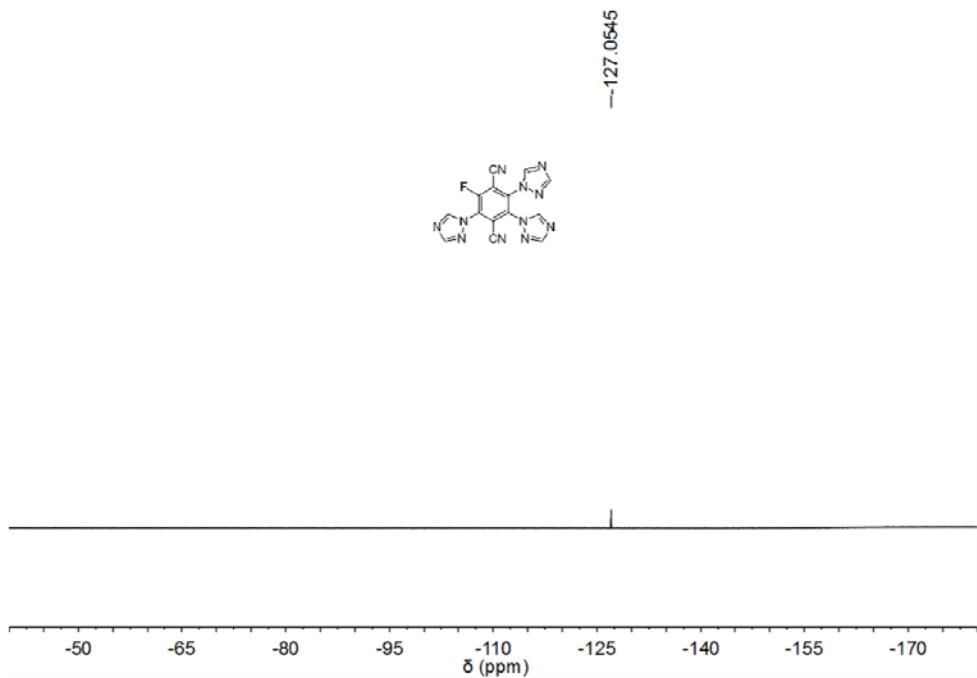


Fig. S3 ^{19}F NMR spectrum of
2-fluoro-3,5,6-tri(1*H*-1,2,4-triazol-1-yl)-terephthalonitrile (L^1).

~ 9.0210
 ~ 8.7637
 ~ 8.7266
 ~ 8.2677
 ~ 8.0985
 ~ 8.0564

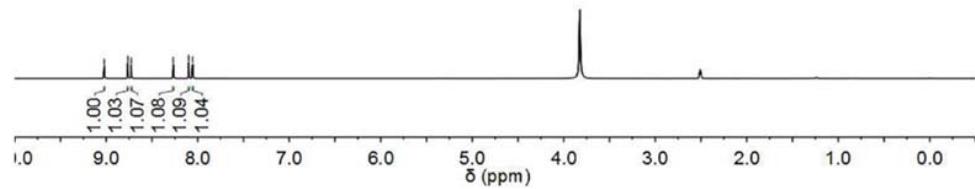
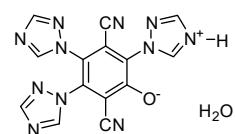


Fig. S4 ^1H NMR spectrum of $HL^2 \cdot H_2O$.

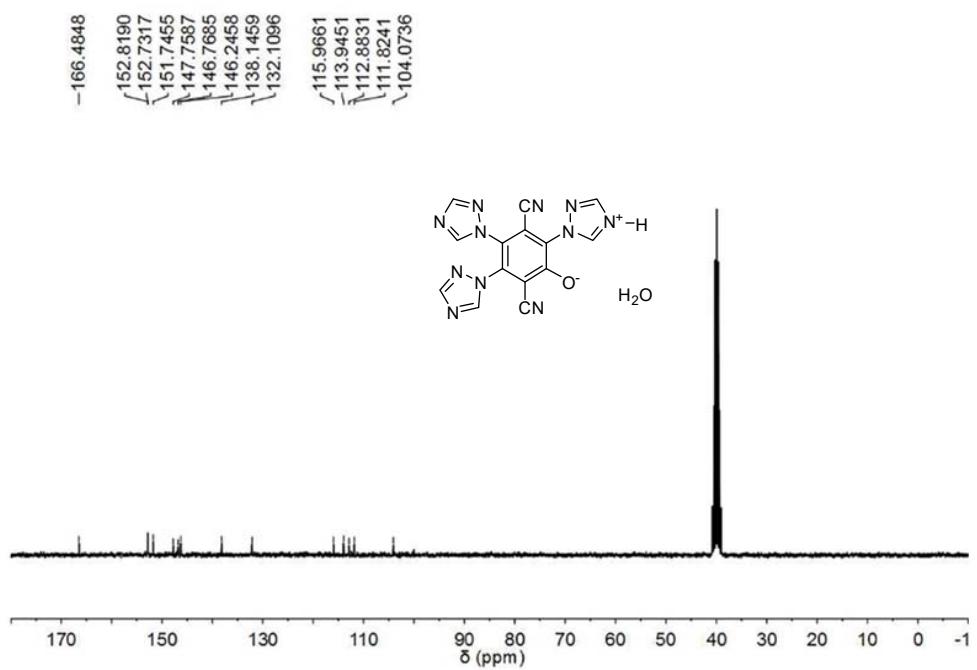


Fig. S5 ^{13}C NMR spectrum of $\text{HL}^2\cdot\text{H}_2\text{O}$.

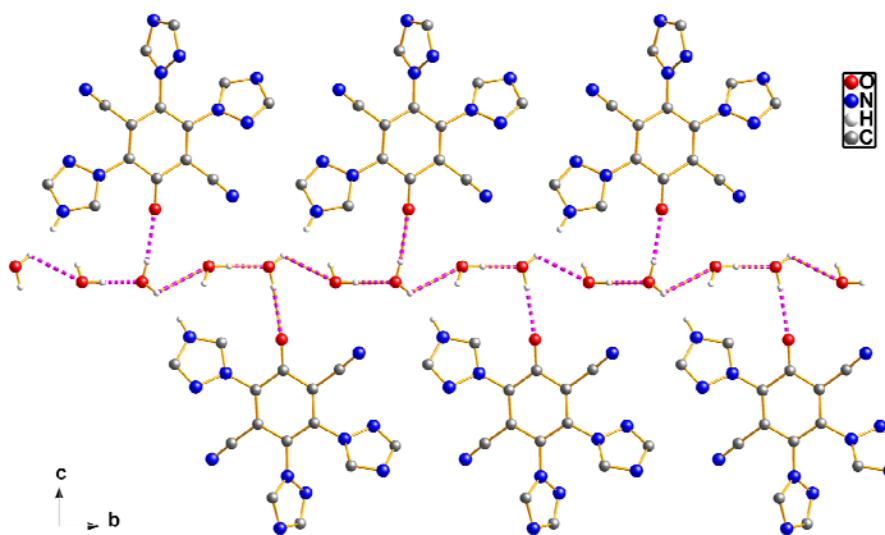


Fig. S6 View of the $\text{O}_{\text{water}}\cdots\text{O}1$ interactions between the heavily disordered lattice water molecules and phenolate oxygen atom in crystals of $\text{HL}^2\cdot\text{H}_2\text{O}$ and chain formation of the water molecules.

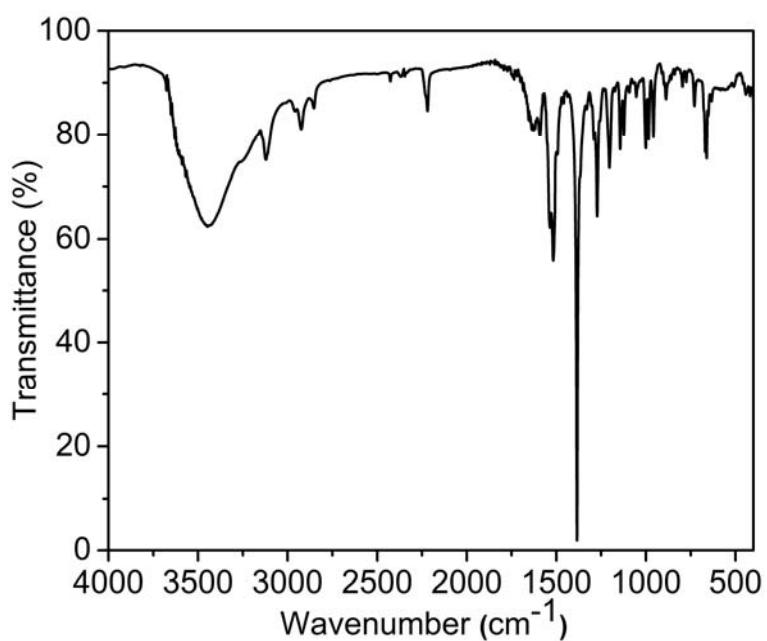


Figure S7: FT-IR spectrum of complex **1**.

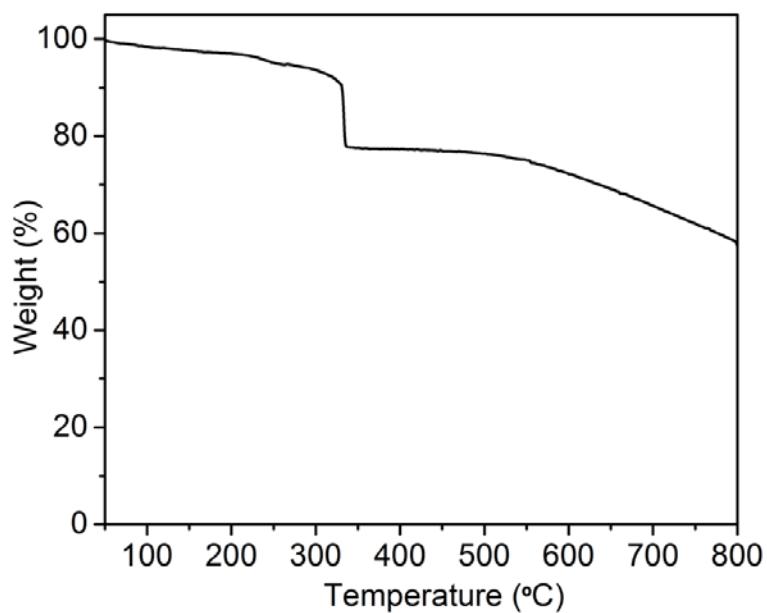


Figure S8: Thermogravimetric profile for complex **1**.