# Crystal Structure of the New Ternary Antimonide Ho<sub>5</sub>GaSb<sub>3</sub>

Iryna Antonyshyn<sup>a</sup>, Olga Zhak<sup>a</sup>, Stepan Oryshchyn<sup>a</sup>, Volodymyr Babizhetskyy<sup>b</sup>, Constantin Hoch<sup>b</sup>, and Lev Aksel'rud<sup>a</sup>

<sup>a</sup> Department of Analytical Chemistry, Ivan Franko National University of L'viv, Kyrylo & Mefodij Str., 6, 79005 L'viv, Ukraine

<sup>b</sup> Max-Planck-Institut für Festkörperforschung, Heisenbergstrasse, 1, Postfach 800665, 70569 Stuttgart, Germany

Reprint requests to V. Babizhetskyy. E-mail: v.babizhetskyy@fkf.mpg.de

Z. Naturforsch. 2009, 64b, 909 – 914; received May 29, 2009

The crystal structure of the new ternary antimonide  $Ho_5GaSb_3$  has been determined from X-ray single-crystal data: space group *Pnma*, a = 7.9667(8), b = 15.128(2), c = 7.9616(8) Å, V = 959.5(3) Å<sup>3</sup>, Z = 4,  $R_F = 0.059$ ,  $R_W = 0.066$  for 9020 reflections. The crystal structure of  $Ho_5GaSb_3$  is a ternary derivative of the  $Sm_5Ge_4$  structure type with partially ordered distribution of gallium and antimony atoms.

Key words: Crystal Structure, Ternary Holmium Antimonide

## Introduction

The isothermal sections of phase diagrams for the ternary systems RE-Ga-Sb (where RE = rare-earth metal) have not been reported in literature, but several selected ternary compounds in these systems have been investigated for some time.

Mills and Mar reported on the existence of ternary compounds, such as Pr<sub>12</sub>Ga<sub>4</sub>Sb<sub>23</sub> and the isotypic compounds  $RE_{12}Ga_4Sb_{23}$  (RE = La, Ce, Sm),  $La_{13}Ga_8Sb_{21}$  and  $REGaSb_2$  (RE = La-Nd, Sm) [1,2]. The crystal structure of Pr<sub>12</sub>Ga<sub>4</sub>Sb<sub>23</sub> [1] is related to that of La<sub>6</sub>MnSb<sub>15</sub> [3], in which the position 4c is tetrahedrally coordinated by Sb atoms and only partially occupied by Mn atoms. In the structure of Pr<sub>12</sub>Ga<sub>4</sub>Sb<sub>23</sub>, atoms Ga1, Ga2 and Sb1 (positions 4i, 4h, 8l, respectively) are also partially occupied. La<sub>13</sub>Ga<sub>8</sub>Sb<sub>21</sub> [1] (the precise composition is La<sub>12.85</sub>Ga<sub>7.56</sub>Sb<sub>21</sub>) crystallizes with hexagonal symmetry. The peculiarity of the crystal structure of this compound is a partial occupancy of Ga1, Ga2, and La3 (positions 12n, 4h, 1a, respectively). The orthorhombic structures of the compounds Pr<sub>12</sub>Ga<sub>4</sub>Sb<sub>23</sub> and La<sub>13</sub>Ga<sub>8</sub>Sb<sub>21</sub> are closely related to the hexagonal structures of a large family of metal-rich compounds, generally phosphides and silicides, composed of differently sized triangular arrangements of trigonal prisms [4, 5].

For the layer compounds *REGaSb*<sub>2</sub> two structure types have been described. The SmGaSb<sub>2</sub> type (space

group  $C222_1$ , also adopted by RE = La) represents a stacking variant of the NdGaSb<sub>2</sub> type (space group  $I4_1/amd$ , also adopted by RE = Ce-Nd) [2].

Other authors [6] noticed the existence of the ternary compound  $SmGaSb_2$ , which was found during the investigation of the respective ternary system in the section of 50 at.-% of antimony. The X-ray powder pattern of the sample containing 50% SmSb differs from the powder patterns of other samples also containing binary SmSb. This fact was taken as evidence for the formation of a new ternary compound. Later it was found that this compound belongs to the NaCl structure type (space group  $Fm\bar{3}m$ , a = 6.0896 Å) [6].

The ternary compound  $Yb_{11}GaSb_9$  belongs to the  $Ca_{11}InSb_9$  structure type. It is one of the examples of Zintl phases with a high content of rareearth metal [7]. The structure is built up from Gacentered  $[GaSb_4]^{9-}$  tetrahedra,  $[Sb_2]^{4-}$  dimers and isolated  $[Sb]^{3-}$  anions, separated by  $Yb^{2+}$  cations:  $[Yb^{2+}]_{11}$   $[GaSb_4]^{9-}$   $[Sb_2]^{4-}$   $[Sb^{3-}]_3$ . Eu<sub>7</sub>Ga<sub>6</sub>Sb<sub>8</sub> also belongs to the Zintl phases and crystallizes with its own orthorhombic structure type [8]. In this structure the anionic framework consists of infinite chains  $[Ga_6Sb_8]^{14-}$ , arranged together with  $Eu^{2+}$  cations to  $[(Eu^{2+})_7(Ga_6Sb_8)^{14-}]$  sheets, which are separated by  $Eu^{2+}$  cations. A charged-balanced formula is best represented as  $[Eu^{2+}]_7$   $[(Ga_4)^{6+}(Ga_2)^{4+}-(Sb^{3-})_8]^{14-}$ .

The continuation of the crystal structure investigation of the ternary compounds of gallium and antimony with rare-earth metals is worthwhile from a crystal chemistry point of view. The main goal of our work was to determine the crystal structure of the new ternary compound Ho<sub>5</sub>GaSb<sub>3</sub>, which was identified during the systematic investigation of interactions in the ternary system Ho-Ga-Sb in the region with a high content of holmium.

#### **Experimental Section**

Synthesis

The sample for the crystal structure investigation of the new ternary compound  ${\rm Ho_5GaSb_3}$  was prepared from pieces of holmium, gallium and antimony (all with minimum purities  $> 99.5\,\%$ ). A stoichiometric mixture of these pieces (total amount 1 g) was melted several times in an arc furnace on a water-cooled copper bottom under purified argon atmosphere to improve homogeneity. Weight losses were found to be within 1 % of the original mass. The sample was subsequently wrapped in tantalum foil and annealed in a silica tube under an argon atmosphere for 1000 h at 500 °C, followed by quenching in cold water without breaking the vacuum ampoule. The sample is stable in air over months in powdered as well as in polycrystalline form. Single crystals of  ${\rm Ho_5GaSb_3}$  were isolated from the crushed sample after the thermal treatment.

### X-Ray powder diffraction

All samples were characterized by X-ray powder diffraction performed with a DRON-3M diffractometer ( $CuK_{\alpha}$  radiation,  $\lambda = 1.54185$  Å). Phase analysis was performed using the WINCSD program package [9]. The lattice parameters of  $Ho_5GaSb_3$  were refined by least-squares fitting of the powder data with LaB<sub>6</sub> as an internal standard (a = 7.966(2), b = 15.089(4), c = 7.954(3) Å).

# Single-crystal X-ray diffraction

Single-crystal intensity data of  $\text{Ho}_5\text{GaSb}_3$  were collected at r.t. on a Stoe IPDS II image plate diffractometer with monochromatized  $\text{Mo}K_\alpha$  radiation by oscillation of the crystal around the  $\omega$  axis. Crystallographic data and information on data collection and evaluation are listed in Table 1. All structure calculations were performed using the WINCSD software package [9].

## Energy-dispersive X-ray spectroscopy analysis

Energy-dispersive X-ray spectroscopy analysis (EDXS) of the polished sample and the single crystal by a scanning electron microscope (TESCAN 5130MM with Oxford

Table 1. Crystal data, intensity collection and refinement for Ho<sub>5</sub>GaSb<sub>3</sub>.

110) 04003.	
Composition	$\text{Ho}_5\text{Ga}_x\text{Sb}_{4-x} \ (x = 0.99(1))$
Structure type	ordered superstructure
	of the Sm <sub>5</sub> Ge <sub>4</sub> type
Crystal system	orthorhombic
Space group	<i>Pnma</i> (no. 62)
Pearson symbol	oP36
a, Å	7.9667(8)
b, Å	15.128(2)
c, Å	7.9616(8)
$V$ , $\mathring{A}^3$	959.5(3)
Number of formula units per cell, Z	4
Calculated density, g cm <sup>-3</sup>	8.72
Radiation and wavelength	$MoK_{\alpha}$ , $\lambda = 0.71069 \text{ Å}$
Diffractometer	Stoe IPDS II
Linear absorption coefficient, cm <sup>-1</sup>	534.5
$2\theta$ range, deg	2.86 - 64.80
h, k, l	$-9 \le h \le 9$ ,
	$-18 \le k \le 18$ ,
	$-9 \le l \le 9$
Reflections collected	15222
Independent reflections	9902
$R_{\rm int}$ / $R_{\sigma}$	0.072 / 0.057
Reflections used in refinement	9020
$[F_{hkl} \ge 4\sigma(F_{hkl})]$	
Number of variable parameters	48
Mode of refinement	$F_{ m hkl}$
$R_{ m F}$	0.058
$R_{ m w}$	0.065
Goodness of fit	1.09
Largest diff. peak / hole <sup>a</sup> , e Å <sup>-3</sup>	5.4 / -2.5

<sup>&</sup>lt;sup>a</sup> Largest residual electron density found close (< 1.0 Å) to Ho3.

Si-detector) confirmed the presence of only holmium, gallium and antimony. The nominal composition in atomic percentages measured by EDXS on several crystals was found to be Ho: Ga: Sb = 52.5:9.9:37.6 (standard deviations estimated to be about 1.5 at.-%).

Further details of the crystal structure investigation may be obtained from Fachinformationszentrum Karlsruhe, 76344 Eggenstein-Leopoldshafen, Germany (fax: +49-7247-808-666; e-mail: crysdata@fiz-karlsruhe.de, http://www.fiz-informationsdienste.de/en/DB/icsd/depot\_anforderung.html) on quoting the deposition number CSD-420712.

## **Results and Discussion**

During the systematic investigation of the system Ho-Ga-Sb at 500 °C a new ternary phase with the approximate composition  $\sim$ Ho<sub>55</sub>Ga<sub>15</sub>Sb<sub>30</sub> was found. The X-ray and EDXS analysis of the samples of different compositions revealed the existence of this ternary compound in the phase equilibrium with the binary compounds HoGa, HoSb and Ho<sub>5</sub>Ga<sub>3</sub>, and the ternary phase Ho<sub>5</sub>(Ga,Sb)<sub>3</sub> [10]. A single crystal of the new

Table 2. Atomic positional and displacement parameters (B, Å<sup>2</sup>) for Ho<sub>5</sub>GaSb<sub>3</sub>.

Atom	Site	x/a	y/b	z/c	$B_{ m eq}$	$B_{11}$	$B_{22}$	$B_{33}$	$B_{12}$	$B_{13}$	$B_{23}$
Ho1	8 <i>d</i>	0.05946(5)	0.11174(3)	0.32580(6)	0.921(9)	1.15(2)	0.75(1)	0.86(2)	0.140(5)	0.171(5)	0.033(5)
Ho2	8d	0.10360(5)	0.62117(3)	0.15847(6)	0.794(8)	1.06(2)	0.51(1)	0.81(2)	-0.000(5)	-0.181(5)	0.039(5)
Ho3	4c	0.22806(6)	1/4	0.00414(6)	0.49(1)	0.66(2)	0.42(2)	0.39(2)	0	-0.025(7)	0
Sb1	8d	0.26323(6)	0.54511(4)	0.48180(7)	0.74(1)	0.78(2)	0.54(2)	0.91(2)	-0.016(6)	-0.117(7)	0.042(7)
$M1^a$	4c	0.0947(1)	1/4	0.6256(1)	0.78(2)	0.89(3)	0.70(3)	0.74(4)	0	-0.02(1)	0
$M2^a$	4c	0.3649(1)	1/4	0.3583(2)	0.77(2)	0.93(3)	0.54(3)	0.83(2)	0	-0.08(1)	0

<sup>&</sup>lt;sup>a</sup> Occupation: M1 = 30.7(2) % Ga + 69.3(2) % Sb; M2 = 68.8(1) % Ga + 31.2(1) % Sb.

Table 3. Comparison of the Ho<sub>5</sub>GaSb<sub>3</sub> and Sm<sub>5</sub>Ge<sub>4</sub> [11] structures<sup>a</sup>.

Chemical composition		Но	5GaSb3		Sm <sub>5</sub> Ge <sub>4</sub>				
Space group		Ì	Рпта		Pnma				
a, Å		7	7.9667		7.75				
b, Å		1	5.128	14.94					
c, Å		7	7.9616		7.84				
Wyckoff positions	Atom	х	у	Z	Atom	х	у	Z	
8 <i>d</i>	Ho1	0.05946	0.11174	0.32580	Sm1	0.1205	0.1157	0.3388	
8 <i>d</i>	Ho2	0.10360	0.62117	0.15847	Sm2	0.0253	0.6004	0.1781	
4 <i>c</i>	Ho3	0.22806	1/4	0.00414	Sm3	0.2880	1/4	0.0024	
8d	Sb	0.26323	0.54511	0.48180	Ge1	0.2794	0.0449	0.0312	
4 <i>c</i>	M1 <sup>b</sup>	0.0947	1/4	0.6256	Ge2	0.1761	1/4	0.6333	
4 <i>c</i>	M2 <sup>b</sup>	0.3649	1/4	0.3583	Ge3	0.41320	1/4	0.3885	

<sup>&</sup>lt;sup>a</sup> Atomic coordinates were standardized using the program STRUCTURE TIDY [25]; <sup>b</sup> M1 and M2 are statistical mixtures of Ga and Sb atoms

ternary antimonide was isolated from a heat-treated sample in which the unknown ternary phase was the main phase. Pre-screening of the single crystal revealed orthorhombic symmetry with the lattice parameters listed in Table 1. The systematic extinctions (0kl, k+l = 2n+1; hk0, h = 2n+1, where n = 0, 1,2, 3,...) are characteristic for the space groups *Pnma* and Pna21. The solution by Direct Methods was successful in the centrosymmetric space group Pnma. The model was refined by full-matrix least-squares methods on F. The final values of the atomic coordinates and anisotropic displacement parameters for all atoms are listed in Table 2. During data reduction and refinement the crystal was supposed to be pseudomerohedrally twinned. The relation of two twin domains can be expressed by the matrix refined twin volume ratio is 0.829(1): 0.171(1).

The crystal structure refinement of Ho<sub>5</sub>GaSb<sub>3</sub> revealed that antimony and gallium atoms statistically occupy two crystallographic positions 4c: M1 = 30.7(2)% Ga + 69.3(2)% Sb and M2 = 68.8(1)% Ga + 31.2(1)% Sb. All other sites are fully occupied by either holmium or antimony atoms (Table 2). The final composition of the new compound can be described as Ho<sub>5</sub>Ga<sub>x</sub>Sb<sub>4-x</sub> [x = 0.99(1)].

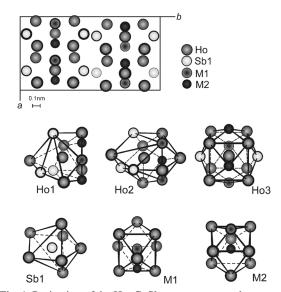
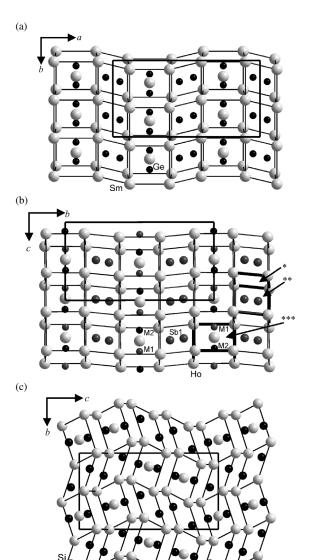


Fig. 1. Projection of the Ho<sub>5</sub>GaSb<sub>3</sub> structure onto the crystal-lographic *ab* plane and coordination polyhedra of the atoms.

A projection of the  $Ho_5GaSb_3$  structure onto the crystallographic ab plane and the coordination polyhedra for all atoms are shown in Fig. 1. The structure of  $Ho_5GaSb_3$  is a ternary derivative of the  $Sm_5Ge_4$  type (space group Pnma, a = 7.75, b = 14.94, c = 7.84 Å) [11]. In the structure of  $Sm_5Ge_4$ , the Ge atoms

Table 4. Interatomic distances (d, Å) for Ho<sub>5</sub>GaSb<sub>3</sub>.

Atoms	d	Atoms	d	Atoms	d	Atoms	d	Atoms	d	Atoms	d
Ho1-M2	2.9877(9)	Ho2-M1	3.0414(8)	Но3-М2	3.023(1)	Sb1-Ho2	3.0710(7)	M1-M2	3.027(2)	M2-2Ho1	2.9877(9)
Sb1	3.1317(7)	Sb1	3.0710(7)	M2	3.094(1)	Ho2	3.0932(7)	2Ho2	3.0414(8)	Ho3	3.023(1)
Sb1	3.1578(7)	Sb1	3.0932(7)	M1	3.098(1)	Ho3	3.1055(5)	Ho3	3.098(1)	M1	3.027(2)
M1	3.1862(9)	M2	3.094(1)	2Sb1	3.1055(5)	Ho1	3.1317(7)	2Ho2	3.1053(8)	Ho3	3.094(1)
M2	3.2191(9)	M1	3.1053(8)	M1	3.195(1)	Ho2	3.1502(7)	2Ho1	3.1862(9)	2Ho2	3.094(1)
Sb1	3.2423(7)	Sb1	3.1502(7)	2Ho2	3.5293(6)	Ho1	3.1578(7)	Ho3	3.195(1)	2Ho1	3.2191(9)
Ho3	3.5689(6)	Ho3	3.5293(6)	2Ho1	3.5689(6)	Ho1	3.2423(7)				
Ho3	3.6302(6)	Ho3	3.6292(6)	2Ho2	3.6292(6)						
Ho2	3.7737(6)	Ho1	3.7737(6)	2Ho1	3.6302(6)						
Ho2	3.7832(6)	Ho1	3.7832(6)								
Ho2	4.0709(7)	Ho2	3.8980(5)								
		Ho1	4.0709(7)								



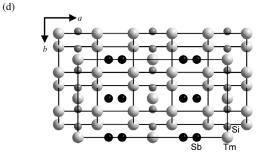


Fig. 2. Fragments of the structures of  $Sm_5Ge_4$  (a),  $Ho_5GaSb_3$  (b),  $Zr_5Si_4$  (c), and  $Tm_5Sb_2Si_2$  (d). The  $\alpha$ -Po\*,  $CuAl_2**$ , and  $U_3Si_2***$  type slabs are emphasized.

occupy two 4c and one 8d sites. In the Ho<sub>5</sub>GaSb<sub>3</sub> structure, the 8d site is occupied by Sb only, and two 4c positions are statistically occupied by Ga and Sb atoms. A comparative analysis of these structures is given in Table 3. Replacement of germanium positions by antimony and gallium atoms does not change the coordination numbers and coordination polyhedra of the atoms (Fig. 1). In the Ho<sub>5</sub>GaSb<sub>3</sub> structure, the Ho<sub>1</sub> and Ho2 atoms are coordinated by polyhedra with 11 and 12 vertices, respectively, whereas the Ho3 atoms center slightly distorted hexa-capped cubes [coordination number 14(8+6)]. The Sb atoms are situated in the centers of polyhedra with 7-vertices. The coordination polyhedra around the positions of the statistical mixtures of Sb and Ga atoms are tricapped trigonal prisms (Fig. 1).

Interatomic distances between the atoms in  $\text{Ho}_5\text{GaSb}_3$  are listed in Table 4. The distances are in good agreement with the respective sums of atomic radii of the components ( $r_{\text{Ho}} = 1.74 \text{ Å}$ ,  $r_{\text{Ga}} = 1.35 \text{ Å}$ ,  $r_{\text{Sb}} = 1.41 \text{ Å}$  [12]). The shortest distances in this structure are between Ho1 and M2 (2.9877(9) Å) and Ho3 and M2 (3.023(1) Å). Deviations of these

interatomic distances from the sum of their atomic radii are smaller than 5%.

The  $Sm_5Ge_4$  structure type [11] is adopted by many rare-earth silicides and germanides ( $RE_5Si_4$  for RE = Y, Sm, Gd–Er;  $RE_5Ge_4$  for RE = all rare-earth metals, except Pm, Eu) [13–16]. Exceptions are  $RE_5Si_4$  for RE = La–Nd, which crystallize in the tetragonal  $Zr_5Si_4$  structure type [17], and RE = Lu, which crystallizes in its own monoclinic structure [13]. For these parent types several superstructures are known, which can be derived by splitting or combining the Wyckoff sites.

According to ref. [18] the structure of Tm<sub>5</sub>Sb<sub>2</sub>Si<sub>2</sub> is a ternary version of the binary structure type Eu<sub>5</sub>As<sub>4</sub> [19] with a fully ordered distribution of antimony and silicon atoms on the crystallographic positions 8*d* and 8*f*. For the Zr<sub>5</sub>Si<sub>4</sub> type only one ternary superstructure has been described, namely Sc<sub>2</sub>Re<sub>3</sub>Si<sub>4</sub> [20]. For the Sm<sub>5</sub>Ge<sub>4</sub> type two ternary superstructures are known: Ce<sub>2</sub>Sc<sub>3</sub>Si<sub>4</sub> [21] and LiTm<sub>4</sub>Ge<sub>4</sub> [22]. The relations between these structure variants are close, but they can be addressed neither as isotypical nor as isostructural.

Besides the Ho<sub>5</sub>GaSb<sub>3</sub> structure with partially ordered gallium and antimony atomic positions, the structure of Sm<sub>5</sub>SiGe<sub>3</sub> with the Sm<sub>5</sub>Ge<sub>4</sub> type was ob-

served also with disordered non-metal atomic positions [23]. In the investigation of the pseudo-binary system  $Gd_5Si_4$ - $Gd_5Ge_4$  the appearance of an intermediate ternary phase  $Gd_5(Si_2Ge_2)$  with a monoclinic crystal structure was observed [24]. The formation of this intermediate phase is probably due to the large difference in bonding characteristics of Si and Ge in the compounds  $Gd_5Si_4$  and  $Gd_5Ge_4$  which crystallize in the orthorhombic  $Sm_5Ge_4$  type. Fig. 2 shows these related structures and also the relations between the new ternary compound  $Ho_5GaSb_3$  and  $Sm_5Ge_4$ . All these related structures contain slabs of the  $CuAl_2$ ,  $U_3Si_2$  and  $\alpha$ -Po types, as emphasized in Fig. 2.

### Conclusion

During the systematic investigations of the holmium-rich part of the ternary system Ho-Ga-Sb a new ternary antimonide  $Ho_5GaSb_3$  has been synthesized at 500 °C, and its crystal structure was studied using single crystal X-ray data.  $Ho_5GaSb_3$  is a ternary derivative of the  $Sm_5Ge_4$  structure type with a partially ordered distribution of gallium and antimony atoms.

Acknowledgement

The authors thank Viola Duppel for EPMA analysis.

- [1] A. Mills, A. Mar, *Inorg. Chem.* **2000**, *39*, 4599 4607.
- [2] A. Mills, A. Mar, J. Am. Chem. Soc. 2001, 123, 1151 1158.
- [3] O. Sologub, M. Vybornov, P. Rogl, K. Hiebl, G. Cordier, P. Woll, J. Solid State Chem. 1996, 122, 266-272.
- [4] Yu. Kuz'ma, S. Chykhrij, in *Handbook on the Physics and Chemistry of Rare Earths*, Vol. 23 (Eds.: K. A. Gschneider, Jr., L. Eyring), Elsevier, Amsterdam, 1996, pp. 285–434.
- [5] Ya. Yarmoljuk, L. Aksel'rud, Kristallografiya 1983, 28, 1111 – 1117.
- [6] A. Kuliev, G. Safaraliev, G. Guseinov, *Izv. Akad. Nauk SSSR*, *Neorg. Mater.* 1990, 26, 500 503.
- [7] S. Bobev, V. Fritsch, J.D. Thompson, J.L. Sarrao, B. Eck, R. Dronskowski, S.M. Kauzlarich, J. Solid State Chem. 2005, 178, 1071 – 1079.
- [8] S.-M. Park, S.-J. Kim, M. Kanatzidis, J. Solid State Chem. 2004, 177, 2867 – 2874.
- [9] L. G. Aksel'rud, P. Yu. Zavalii, Yu. Grin, V. K. Pecharsky, B. Baumgartner, E. Wölfel, WINCSD, Universal Program Package for Single Crystal and/or Powder Structure Data Treatment. See: *Mater. Sci. Forum*, 1993, 133–136, 335–342.

- [10] P. Villars, L. D. Calvert, Pearson's Handbook of Crystallographic Data for Intermetallic Phases, ASM International, Materials Park, Ohio, 1997, pp. 2147 – 2168.
- [11] G. S. Smith, Q. Johnson, A. G. Tharp, *Acta Crystallogr*. 1967, 22, 269 – 272.
- [12] N. Wiberg, Holleman-Wiberg Lehrbuch der anorganischen Chemie, Walter de Gruyter, Berlin-New York, 1995, pp. 1838 – 1841.
- [13] G. S. Smith, A. G. Tharp, Q. Johnson, *Acta Crystallogr*. 1967, 22, 940 – 943.
- [14] M. Pani, A. Palenzona, J. Alloys Compd. 2003, 360, 151-161.
- [15] U. Ch. Rodewald, B. Heying, D. Johrendt, R.-D. Hoff-mann, R. Pöttgen, Z. Naturforsch. 2004, 59b, 174–181.
- [16] S. Misra, G. Miller, J. Am. Chem. Soc. 2008, 130, 13900 – 13911.
- [17] H.-U. Pfeifer, K. Schubert, Z. Metallkd. 1966, 57, 884 888.
- [18] A. Kozlov, V. Pavlyuk, V. Davydov, *Intermetallics* 2004, 12, 151–155.
- [19] Y. Wang, L.D. Calvert, E.J. Gabe, J.B. Taylor, *Acta Crystallogr. B* 1978, 34, 1962–1965.
- [20] V. K. Pecharskii, O. I. Bodak, E. I. Gladyshevskii,

- Dopov. Akad. Nauk Ukr. RSR, Ser. A **1978**, 40, 758–762
- [21] I. R. Mokraya, O. I. Bodak, E. I. Gladyshevskii, *Kristallografiya* **1979**, *24*, 1274 1276.
- [22] V. V. Pavlyuk, O. I. Bodak, V. E. Zavodnik, *Dopov. Akad. Nauk Ukr. RSR*, *Ser. B* **1990**, *12*, 29–31.
- [23] K. Ahn, V. K. Pecharsky, K. A. Gschneidner, Jr., Phys. Review B 2007, 76, 014415-1-13.
- [24] V. K. Pecharsky, K. A. Gschneidner, Jr., J. Alloys Comp. 1997, 260, 98 – 106.
- [25] E. Gelato, E. Parthé, STRUCTURE TIDY, Program to Standardize Structure Data, University of Geneva, Geneva (Switzerland) 1986; see: L. M. Gelato, E. Parthé, J. Appl. Crystallogr. 1987, 20, 139–143.