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# First-principles study of structures, elastic and optical properties of single-layer metal iodides under strain

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Abstract: The structure, elastic, electronic and optical properties of two-dimensional (2D) MI<sub>2</sub> (M = Pb, Ge, Cd) under strain are systematically studied by the first-principles method. It is proved that the monolayer structure of 2D-MI<sub>2</sub> is stable by phonon spectra. Moreover, the large ideal strain strength (40%), the large range of strain and the elastic constants of far smaller than other 2D materials indicate that the single-layer PbI2 and GeI2 possess excellent ductility and flexibility. By applying appropriate strain to the structure of 2D-MI<sub>2</sub>, the band gaps of single-layer MI<sub>2</sub> can be effectively controlled (PbI<sub>2</sub>:  $1.04 \sim 3.03$  eV, GeI<sub>2</sub>:  $0.43 \sim 2.99$  eV and CdI<sub>2</sub>:  $0.54 \sim 3.36$  eV). It is found that the wavelength range of light absorbed by these three metal iodides is 82–621 nm, so 2D-MI<sub>2</sub> has great absorption intensity for ultraviolet light in a large wavelength range, and the strain of structure can effectively regulate the optical parameters.

**Keywords:** elastic properties; first-principles; metal iodide (MI<sub>2</sub>); optical properties.

#### 1 Introduction

For two-dimensional (2D) materials, the movement of electrons is limited in the plane, so they have some physical and chemical properties that block materials do not have. These unique properties make 2D materials develop rapidly. With the development of graphene, boron nitride (h-BN), transition metal sulfides (TMDs), monoene, transition metal carbides, transition metal oxides and other 2D

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materials, some drawbacks have gradually emerged [1–9], such as the band gap loss of graphene, the low carrier concentration of TMDs and so on. In order to meet the specific needs of production, some new 2D materials, such as metal iodide, have been explored [10–13].

For single-layer metal iodides (MI<sub>2</sub>), the most representative materials are 2D lead iodide (2D-PbI<sub>2</sub>), germanium iodide (2D-GeI<sub>2</sub>) and cadmium iodide (2D-CdI<sub>2</sub>). There are many kinds of phase, among which the most stable is 2H phase, which is a sandwich structure stacked in the ABC model [14]. The layer of MI<sub>2</sub> where the metal atom M is located is sandwiched between two I layers, and there is a van der Waals force between the layers [15]. Stable structure, band gap comparable to third-generation semiconductors, excellent physical and chemical properties are the characteristics of metal iodides. Single-layer PbI2 can be used to make radiation-proof devices because of its high atomic number and density [16-18]; single-layer GeI<sub>2</sub> has a unique light absorption range and can be used to prepare photoelectric converter devices [19, 20]; single-layer CdI<sub>2</sub> is a good material for making heterojunction and heterojunction arrays [21, 22]. According to previous reports, the single-layer metal iodide  $MI_2$  (M = Pb, Ge, Cd) has been successfully stripped and grown in the laboratory.

There are many methods to obtain single-layer MI<sub>2</sub>, including mechanical peeling, solution method and physical vapour deposition [23]. Most single-layer structures can be obtained by mechanical peeling, but the production efficiency of this method is relatively low. The sample prepared by this method has small transverse size and poor thickness control, so it is impossible to prepare a large area of single-layer structure. Impurities are easily introduced in the preparation of monolayer structures using solution method, thus affecting the properties. In contrast, the yield of vapour deposition method is higher and the quality of single-layer structure produced is better [24]. Monolayer structure of metal iodides has been prepared on a large scale in experiments, but the number of layers, phases, densities of defects and edge morphology of samples is uncertain due to limitations of experimental equipment. The physical properties of 2D metal iodides cannot be accurately studied. Therefore, the theoretical calculation of single-layer metal iodides is very necessary.

In 2015, Zhou et al. [25] explored the structure, stability, electronic and optical properties of single-layer PbI<sub>2</sub> and showed some significant results. In our previous work, the single-layer structure, electronic, elastic and thermal transport properties of PbI<sub>2</sub> have been explored and some important results have been obtained [26]. However, previous studies have been limited to PbI2, which is not representative and does not take into account the properties under strain.

In this work, the structure, elastic and optical properties of single-layer metal iodide MI<sub>2</sub> (M = Pb, Ge, Cd) under strain are first investigated by using the first-principles method. Firstly, it is proved that the monolayer structure of MI<sub>2</sub> is stable by the phonon spectra. Then, the stress-strain curve, elastic constant. Young's modulus, layer modulus and Poisson's ratio of 2D-MI<sub>2</sub> are obtained. Comparing with other 2D materials, it can be concluded that 2D-MI<sub>2</sub> has excellent flexibility and ductility, as well as good ability to adjust strain. Considering the spin-orbit coupling (SOC) effect, the band structure and optical parameters in the strain ranging from -20 to 20% were calculated by using the hybrid functional Heyd-Scuseria-Ernzerhof (HSE06). Our results show that the strain can effectively regulate the band gap and optical absorption properties of 2D-MI<sub>2</sub>.

# 2 Theoretical methods and calculation details

The Vienna Ab initio Simulation Package (VASP) based on first-principles is applied to all our calculations [27–29]. The Perdew-Burke-Ernzerhof method in the generalized gradient approximation model is used in the structural optimization and elastic calculations [30, 31]. For the calculation of electronic structure, we use the hybrid functional (HSE06) approximation and consider the influence of SOC [32-34]. After the convergence tests, the plane wave cut-off energies were set to 440, 500 and 400 eV for single-layer PbI<sub>2</sub>, GeI<sub>2</sub> and CdI<sub>2</sub>, respectively. The forces on all relaxed atoms were less than 10<sup>-4</sup> eV/Å, and the energy convergence criterion was  $10^{-6}$  eV per unit cell. The k-point sampling grids in the Brillouin zone were chosen as  $15 \times 15 \times 1$  for geometry optimization and  $21 \times 21 \times 1$  for the calculation of elastic properties.

In general, the elastic constants of materials are derived from the relationship between stress and strain. The relationship between stress and strain can be expressed as follows [35]:

$$\sigma_{ij} = C_{ijkl} \, \varepsilon_{kl} \tag{1}$$

The metal iodide studied in this paper belongs to hexagonal system, so there are only six independent elastic constants.

$$C_{\text{hexal}} = \begin{pmatrix} C_{11} & C_{12} & C_{13} & & & \\ \cdot & C_{11} & C_{13} & & & \\ \cdot & \cdot & C_{33} & & & \\ & & & C_{44} & & \\ & & & & C_{66} \end{pmatrix}$$
 (2)

In fact, there are only five independent elastic constants because for the hexagonal structure  $C_{66}$  is obtained from  $C_{11}$ and  $C_{12}$  ( $C_{66} = (C_{11} - C_{12})/2$ ) [36]. The above elastic theory can also be applied to 2D materials, but some details are different. Different from bulk material, the strain of 2D material is in plane. We know that bulk modulus represents the resistance of bulk materials to external forces. Similarly, the ability of 2D materials to resist external forces (in-plane) can be expressed by layer modulus y [36], which is given as follows:

$$\gamma = -A \frac{\partial F}{\partial A} = A \frac{\partial}{\partial A} \left( \frac{\partial E}{\partial A} \right) = A \frac{\partial^2 E}{\partial A^2}$$
 (3)

In addition, some other parameters which can be used to express the elastic properties of 2D materials can also be calculated by elastic constants. Young's modulus Y and Poisson's ratio  $\nu$  can be expressed as follows [36]:

$$Y_{[10]}^{2D} = \frac{C_{11}C_{12} - C_{12}^2}{C_{22}} \tag{4}$$

$$Y_{[01]}^{2D} = \frac{C_{11}C_{12} - C_{12}^2}{C_{11}}$$
 (5)

$$v_{[10]}^{2D} = \frac{C_{12}}{C_{22}} \tag{6}$$

$$v_{[01]}^{2D} = \frac{C_{12}}{C_{11}} \tag{7}$$

The optical properties of materials are represented by some optical parameters, such as reflectivity, refractive index, absorption coefficient, energy loss and so on. The interaction between light and the electrons and atoms it passes through produces light absorption, which reveals the optical properties of materials and also contains the information of electronic structure. To explore the optical properties of materials, we need to calculate the complex dielectric function, which is expressed as follows [37]:

$$\varepsilon(\omega) = \varepsilon_1(\omega) + i\varepsilon_2(\omega) \tag{8}$$

The imaginary part of the complex dielectric function  $\varepsilon_2$ can be obtained by the sum of occupied and unoccupied wave functions, as follows [37]:

$$\varepsilon_{\alpha\beta}(\omega) = \frac{4\pi^{2}e^{2}}{\Omega} \lim_{q \to 0} \frac{1}{q^{2}} \sum_{c,v,k} 2w_{k} \delta(\varepsilon_{ck} - \varepsilon_{vk} - \varepsilon) \times \langle u_{ck+\varepsilon_{a}q} | u_{vk} \rangle$$

$$\langle u_{ck+\varepsilon_{\beta}q} | u_{vk} \rangle^{*}$$
(9)

Among them,  $e_x(x = \alpha, \beta, \gamma)$  represents the unit vector in Cartesian direction,  $u_k$  represents the periodic part of the primitive cell of pseudo wave function, c and v are the conduction band and the valence band, respectively. The real part of the complex dielectric function  $\varepsilon_1$  can be deduced by Kramer-Kronig relation on the basis of the imaginary part, as follows [38]:

$$\varepsilon_1(\omega) = 1 + \frac{2}{\pi} P \int \frac{\omega' \varepsilon_2(\omega')}{{\omega'}^2 - \omega^2} d\omega'$$
 (10)

Then, the reflectivity and absorption coefficient can be obtained by the following formula [38]:

$$R(\omega) = \left| \frac{\sqrt{\varepsilon_1(\omega) + j\varepsilon_2(\omega)} - 1}{\sqrt{\varepsilon_1(\omega) + j\varepsilon_2(\omega)} + 1} \right|^2 \tag{11}$$

$$\alpha = \sqrt{2\left[-\varepsilon_1(\omega) \pm \sqrt{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)}\right]}$$
 (12)

### 3 Results and discussion

#### 3.1 Structure and elasticity

The most stable phase of bulk  $MI_2$  (M = Pb, Ge, Cd) is the semiconducting 2H phase, whose unit cell consists of two iodide atoms and a metal atom. Figure 1a shows the bulk structure. Recleave surface was performed to achieve a monolayer crystal structure, that is, the bulk phase of  $MI_2$  was cleaved along (0 0 1) plane. After convergence test, we found that the vacuum layer of 8 Å is enough

when eliminating the interaction between layers from Figure 1b. In addition, the lattice constants, bond lengths and bond angles of the optimized monolayer  $\mathrm{MI}_2$  are shown in Table 1. Our previous work has proved the stability of single-layer  $\mathrm{PbI}_2$  [26]. It can be seen in Figure 2 that all phonon modes in monolayer  $\mathrm{MI}_2$  have positive frequency indicating the stability of this structure.

After the stable single-layer structure is obtained, the strain-stress curve is calculated to explore the ability of 2D-MI<sub>2</sub> to resist external strain. Figure 3 depicts the stress per unit length  $(N \cdot m^{-1})$  as a function of strain (%). It can be seen from Figure 3 that the strain-stress curves of singlelayer PbI<sub>2</sub> and GeI<sub>2</sub> are relatively similar. The maximum stresses are 10.6 and 12.11 N·m<sup>-1</sup>, respectively, and the corresponding ideal strains are all around 40%, indicating that their structures can stretch to 40% of the original. The stress-strain curve of CdI2 is different from those of PbI2 and GeI<sub>2</sub>. Its maximum stress is 18.81 N·m<sup>-1</sup>, and the corresponding ideal strain is 24%, which is obviously lower than the former two. Compared with the ideal strain strength of other 2D materials, such as MoS<sub>2</sub> (20%) and graphene (24%) [39–42], the ideal strain strength of single-layer PbI<sub>2</sub> and GeI<sub>2</sub> as high as 40% makes them show excellent strain capacity and flexibility. On the one hand, this is due to their fold structures. For example, the surfaces of graphene and h-BN are flat, and there are no folds in z direction; some 2D materials have fold structures but close to plane structures. These 2D materials have no space to coordinate the strain when they are stressed, either they have high hardness and are not easy to deform or they have general hardness and their structures are easily damaged. What is more is that this phenomenon can also be explained by elasticity theory.

As it is known, another parameter that can indicate the resistance to compression is the elastic constant.

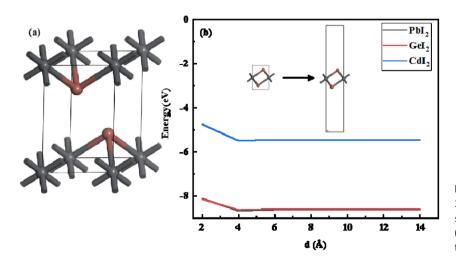


Figure 1: (Colour online) (a) Unit cell of 2H-metal iodide. Grey and red balls represent metal atoms and I atoms, respectively; (b) The energy dependences of the height of the vacuum layer.

**Table 1:** The lattice constants, I–M bond lengths and M–I–M bond angles of monolayer MI<sub>2</sub> (M = Pb, Ge, Cd).

	Single-layer PbI <sub>2</sub>	Single-layer Gel <sub>2</sub>	Single-layer CdI <sub>2</sub>
Lattice constants (Å)	4.66	4.30	4.34
I–M bond lengths (Å)	3.23	3.04	3.04
M-I-M bond angles	89.84°	90.31°	91.10°

The value of elastic constant ( $C_{11}$ ), Young's modulus ( $Y^{2D}$ ) and shear modulus ( $G^{2D}$ ) is listed in Table 2. Compared with other 2D materials, the Young's modulus of 2D-MI<sub>2</sub> (M = Pb, Ge, Cd) is far lower than that of h-BN (289 N·m<sup>-1</sup>), 2D TMDs like  $MoS_2$  (199  $N \cdot m^{-1}$ ),  $WS_2$  (272  $N \cdot m^{-1}$ ) and  $TcS_2$ (93 N·m<sup>-1</sup>) but similar to that of phosphorene (armchair)  $(44 \text{ N} \cdot \text{m}^{-1})$  [36, 43–45]. Therefore, the 2D-MI<sub>2</sub>, which is more flexible, also has potential applications in flexible optoelectronics. Nevertheless, although both single-layer PbI2 and GeI2 are metal iodides, the Young's modulus of PbI<sub>2</sub> is nearly 40% smaller than that of GeI<sub>2</sub>. To further explore why single-layer PbI<sub>2</sub> is more flexible, the Bader charge analysis has be done. The I-M bond is a transitional form between the covalent bond and ionic bond. In general, the stronger the ionicity, the weaker the resistance of the chemical bond to an external force and therefore the smaller the elastic constant. Each I atom gets 0.45 e transferred from the Pb atom when Pb atom is bonding with I atom, but instead, each I atom gets 0.36 e transferred from the Ge atom when Ge atom is bonding

with I atom. So single-layer  $PbI_2$  is more ionic, which leads to a smaller Young's modulus. What is more is that it can be seen from Table 2 that the ratio of layer modulus and sheer modulus  $(\gamma/G^{2D})$  of  $GeI_2$  is less than 1.75, indicating its brittleness [46], whereas that of  $MI_2$  (M = Pb, Cd) is greater than 1.75, showing good ductility.

By summarizing the above calculation, it can be concluded that  $2D-MI_2$  has low hardness, excellent

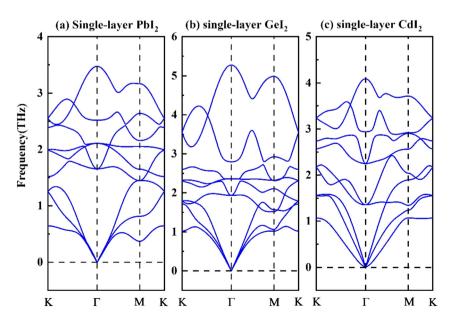
**Table 2:** Calculated elastic constants ( $C_{11}$  and  $C_{12}$  in N·m<sup>-1</sup>), sheer modulus ( $G^{20}$  in N·m<sup>-1</sup>), Young's modulus ( $Y^{20}$  in N·m<sup>-1</sup>) and layer modulus ( $Y^{20}$  in N·m<sup>-1</sup>).

	<b>C</b> 11	<b>C</b> <sub>12</sub>	<b>G</b> <sup>2D</sup>	<b>Y</b> <sup>2D</sup>	γ	γ/ <b>G</b> <sup>2D</sup>
Monolayer PbI <sub>2</sub>	14.69	3.95	5.30	13.63	9.35	1.80
Monolayer Gel <sub>2</sub>	23.76	5.36	9.20	22.55	14.56	1.58
Monolayer CdI <sub>2</sub>	22.85	8.26	7.29	19.86	15.56	2.13

flexibility and the ability to coordinate strain, so it can be utilized to synthesize some semiconductor devices that require extensive strain.

#### 3.2 Electronic structure

Materials will be deformed due to the action of external forces in the process of practical application, so it is meaningful to explore the change of electronic structure in the case of strain. The electronic structure of monolayer  $PbI_2$  and  $GeI_2$  under strain is systematically studied in this section. Firstly, the strain structure is optimized, and then, the electronic structure is calculated. All kinds of lattice



**Figure 2:** (Colour online) The phonon spectra of (a) 2D-Pbl<sub>2</sub> [26], (b) 2D-Gel<sub>2</sub> and (c) 2D-Cdl<sub>2</sub>.

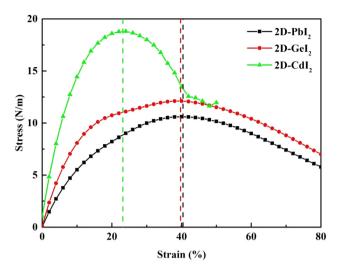


Figure 3: (Colour online) The stress-strain curves for single-layer  $MI_2$  (M = Pb, Ge, Cd).

parameters along lattice vector of a/b direction are chosen. A biaxial strain is defined as  $\varepsilon = \Delta a/a_0$ , where  $\Delta a$  is the difference between the frozen and optimized lattice constant along lattice vector a/b direction and  $a_0$  is the equilibrium lattice constant.

The electronic structure of single-layer  $MI_2$  (M = Pb, Ge, Cd) under different strain is shown in Figure 4a-c. In the calculation process, the top of valence band is Fermi surface. It can be clearly seen that the conduction band of single-layer PbI<sub>2</sub> decreases continuously and the valence band increases continuously with the stretching of the structure. In this process, the position of the bottom of the conduction band is always at the  $\Gamma$ -point, but the position of the top of the valence band changes from the position between Γ-point and M-point in equilibrium state to the position of M-point (the  $\varepsilon$  is more than 15%). For singlelayer GeI<sub>2</sub>, the variation trend of its valence band and conduction band is basically the same as that of singlelayer PbI<sub>2</sub>. The only difference is the position of the conduction band bottom when  $\varepsilon = -10\%$ . The change of electronic structure of CdI2 under strain is different from that of PbI<sub>2</sub> and GeI<sub>2</sub>. With the compression of the structure, the bottom of the conduction band decreases gradually. In this process, the position of the bottom of the conduction band and the top of the valence band has not changed. When the structure is stretched, the conduction band rises slowly to  $\varepsilon = 5\%$  and then drops sharply. At the same time, the position of the bottom of the conduction band changes from M-point ( $\varepsilon$  = 0 ~ 5%) to Γ-point ( $\varepsilon$  > 10%), and the position of the top of the valence band changes from  $\Gamma$ -point to the position between Γ-point and M-point. The band gap of single-layer MI2 is always indirect, no matter in tension or compression.

The curve of band gap calculated by the HSE06 + SOC method with strain is shown in Figure 5. It can be seen that the band gaps of single-layer PbI2 and single-layer GeI2 continue to increase with the compression strain of the structures. When they increase to 3.03 and 2.89 eV (compression strain to -10%), respectively, the band gap drops sharply. As the tensile strain of the structure goes on, the band gap decreases. The curve of single-layer CdI<sub>2</sub> is different from the former two. With the compression of the structure, the band gap of the singlelayer CdI<sub>2</sub> decreases gradually. With the structure stretching, the band gap increases slowly at first and suddenly drops sharply when it increases to 3.46 eV (tensile strain is around 5%).

Through the above analysis, we can conclude that compared with compression strain, stretching can regulate the band gap of single-layer CdI<sub>2</sub> more efficiently, while single-layer PbI<sub>2</sub> and GeI<sub>2</sub> are the opposite, because compression has a more significant role in regulating the band gap of both. In addition, in the process of strain changing from -20 to 20%, the effect of strain on the band gap of single-layer MI<sub>2</sub> is very obvious (CdI<sub>2</sub>:  $0.54 \sim 3.46 \text{ eV}$ , PbI<sub>2</sub>: 1.04 ~ 3.03 eV, GeI<sub>2</sub>: 0.43 ~ 2.99 eV). Therefore, the structural strain can effectively control the band gap of single-layer MI<sub>2</sub>. In the application, the appropriate strain can be applied to the single-layer MI<sub>2</sub> to obtain the desired band gap so that it can be used in the photoelectric devices that need specific band gap.

#### 3.3 Optical properties

Now, we calculate the optical properties of the monolayer metal iodide MI<sub>2</sub> (M = Pb, Ge, Cd). We first calculate the real part  $\varepsilon_1$  and the imaginary part  $\varepsilon_2$  of the dielectric function. Then, we use Eqs. (8) and (9) to get the absorption curve and reflection curve. The real part  $\varepsilon_1$  and imaginary part  $\varepsilon_2$  of dielectric function, absorption coefficient, reflectivity of the single-layer MI2 as a function of photon energy are shown in Figure 6. It can be seen from Figure 6a and b that the real part and imaginary part of 2D-MI<sub>2</sub> have a maximum value in the low energy region (infrared, visible and near ultraviolet). Clearly, we can see from Figure 6c that PbI<sub>2</sub> and GeI<sub>2</sub> monolayers have two similar absorption bands. The absorption curve of singlelayer CdI<sub>2</sub> is different from that of single-layer PbI<sub>2</sub> and GeI<sub>2</sub>, which is due to the fact that Cd atom is not in the same group as Pb and Ge atoms. For single-layer PbI<sub>2</sub>, GeI<sub>2</sub> and CdI<sub>2</sub>, the absorptions begin when the photon energy is zero, and the energies at the absorption edge are basically consistent with the corresponding band gaps.

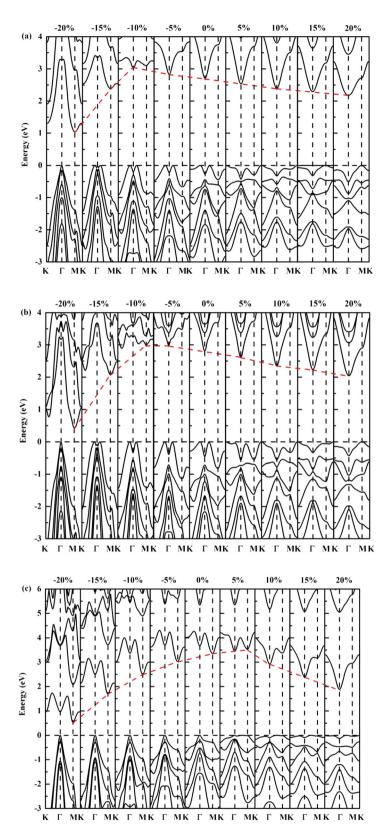


Figure 4: (Colour online) The electronic structure of single-layer  $MI_2$  (M = Pb, Ge, Cd) under different strains.

However, it can be found that the energy at the absorption edge is slightly lower than that in the band gap because the energy required for free carrier absorption is lower than that for the intrinsic absorption, and free carrier absorption occurs when the energy is lower than the forbidden band energy. With the increase of photon

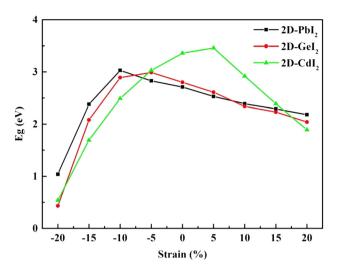


Figure 5: (Colour online) The curves of band gaps of single-layer  $MI_2$  (M = Pb, Ge, Cd) calculated by the HSE06 + SOC method.

energy, the  $\varepsilon_2$  of  $MI_2$  rises rapidly in the visible light region and then the peak appears, indicating the absorption part of the visible light. The first peak appeared in the absorption spectrum of  $MI_2$  near 4 eV, and the light absorption of single-layer  $PbI_2$  and  $GeI_2$  reached

790,000  $\text{cm}^{-1}$  and 750,000  $\text{cm}^{-1}$ , respectively, indicating that the single-layer PbI<sub>2</sub> and GeI<sub>2</sub> have a strong absorption capacity for ultraviolet light, while CdI<sub>2</sub> has relatively weak absorption capacity for UV light. However, the absorption spectrum of single-layer MI<sub>2</sub> in the range of UV light appears several peaks continuously, and the peak value is also very large. To sum up, we can conclude that the monolayer MI<sub>2</sub> has a strong absorption capacity for ultraviolet light in a wide range of wavelengths. It can be seen from Figure 6d that the first peak appears in the visible light region for the reflection, and then many peaks appear in the ultraviolet light region, with the highest reflectivity around 50, 47 and 34% for single-layer PbI<sub>2</sub>, GeI<sub>2</sub> and CdI<sub>2</sub>, respectively. In general, there are many peaks in the reflection curve of single-layer MI<sub>2</sub>, which are concentrated in the UV region, and the peak value is small, showing that the main reflective parts of the 2D-MI<sub>2</sub> are visible light and ultraviolet light, but the reflectivity is not high. However, they have strong absorption capacity for UV light and can be used to make UV detection devices or anti-UV radiation devices.

In view of the strong ability of 2D-MI<sub>2</sub> to absorb UV light in the equilibrium state, the optical absorption

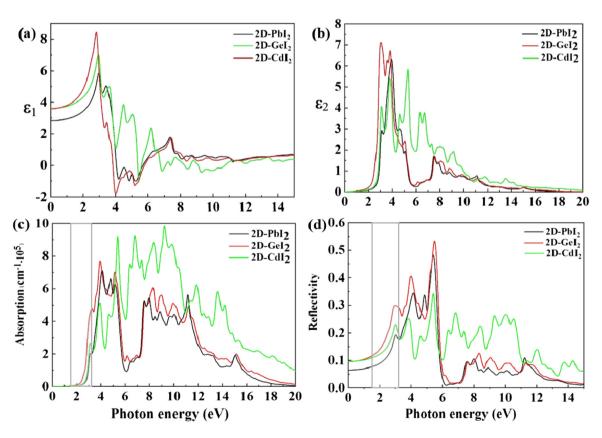


Figure 6: (Colour online) (a) The real part  $\varepsilon_1$  and (b) imaginary part  $\varepsilon_2$  of dielectric function, (c) absorption coefficient, (d) reflectivity of single-layer MI<sub>2</sub> (M = Pb, Ge, Cd) as a function of photon energy.

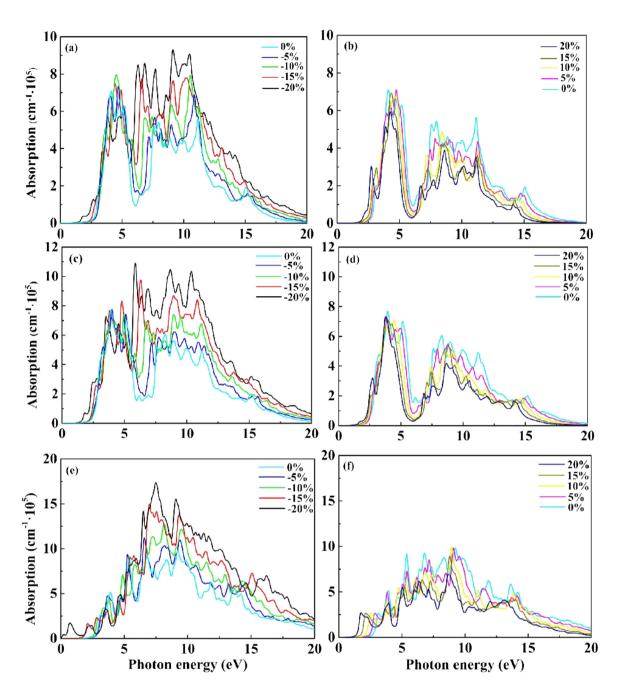


Figure 7: (Colour online) Absorption spectra of 2D-Pbl<sub>2</sub> (a, b), 2D-Gel<sub>2</sub> (c, d) and 2D-Cdl<sub>2</sub> (e, f) under compressions ( $-20 \sim 0\%$ ) and tensions ( $0 \sim 20\%$ ).

properties of single-layer  $MI_2$  under strains are further calculated, and the results are shown in Figure 7. Among them, Figure 7 (a), (c) and (e) show the absorption spectra of singe-layer  $PbI_2$ ,  $GeI_2$  and  $CdI_2$  under compressions ( $\varepsilon = -20 \sim 0\%$ ), respectively, while (b), (d) and (f) show the absorption spectra of the three under tensions ( $\varepsilon = 0 \sim 20\%$ ). It can be seen that the light absorptions still start at 0 eV even though the structures are strained. When the structure is compressed, the absorption edge value

moves to the left, while the absorption edge value moves to the right when the structure is stretched. This is due to the change of band gap. The absorption band of 2D-MI $_2$  becomes narrower and lower with the tensile strain. On the other hand, the compressive strain can improve the optical absorption of 2D-MI $_2$  expanding the range of absorbing ultraviolet. Therefore, the strain can effectively regulate the light absorption ability of single-layer MI $_2$  so that it can be used in different photoelectric detection devices.

# 4 Conclusions

The structural, elastic and optical properties of 2D metal iodide MI<sub>2</sub> (M = Pb, Ge, Cd) in equilibrium and strain are systematically studied by using the first-principles method. It is proved that the monolayer structure of  $MI_2$  (M = Pb, Ge, Cd) is stable, and it presents a fold structure similar to "sandwich". Because of this fold structure, single-layer PbI<sub>2</sub> and GeI<sub>2</sub> have large ideal strain strength (40%), indicating that they have a good strain capacity and excellent flexibility. In addition, the layer modulus y, Young's modulus  $Y^{2D}$ , Poisson's ratio  $\nu$  and shear modulus  $G^{2D}$  are calculated. The results show that the layer modulus of single-layer MI2 is much smaller than other twodimensional materials, showing the characteristics of small hardness and good flexibility. The excellent flexibility, ductility and strong ability of coordinating strain all make the 2D-MI<sub>2</sub> be used to manufacture photoelectric devices with large strain range.

The effect of strain on the band of 2D-MI<sub>2</sub> was investigated using the HSE06 + SOC method. It is found that the effect of structural compression on the band gap of singlelayer PbI2 and GeI2 is more significant, but structural stretching can change the band gap of single-layer CdI<sub>2</sub> more efficiently. From  $\varepsilon = -20\%$  to  $\varepsilon = 20\%$ , the change of the band gap of single-layer MI<sub>2</sub> is very obvious (CdI<sub>2</sub>: 0.54 ~ 3.46 eV; PbI<sub>2</sub>: 1.04 ~ 3.03 eV; GeI<sub>2</sub>:0.43 ~ 2.99 eV). Therefore, the strain can effectively control the band gap of single-layer MI<sub>2</sub>. By exploring the optical properties, it is found that the single-layer MI<sub>2</sub> has strong absorption ability for ultraviolet light over a large wavelength range. Moreover, strain can not only effectively change the band gap but also effectively control the light absorption ability.

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