

## Research Article

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# Physicochemical and microbiological characteristics of various stem bark extracts of *Hopea beccariana* Burck potential as natural preservatives of coconut sap

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**Abstract:** The bark of *Hopea beccariana* Burck is used to prevent damage to coconut sap. The purpose of this study was to determine the physicochemical and microbiological characteristics of various extracts of the bark of *H. beccariana* Burck as potential natural preservatives for coconut sap. The bark was extracted by maceration method for 24 h assisted by stirring using ethanol, methanol, *n*-hexane, and water at 60°C as solvent. The type of solvent used for extraction had a significant effect on the yield, total phenolic, total flavonoid, antioxidant activity, toxicity, and diameter of the inhibition zone against *Lactobacillus plantarum*. The highest yield was found in methanol extract at 22.34%, the highest total phenolic content was found in ethanol extract of 53.39 mg gallic acid equivalent/g extract, and the highest total flavonoid content was found in the aqueous extract at 60°C of 106.70 mg QE/g extract; all extracts have an IC<sub>50</sub> value of 80.28–91.80 ppm (very strong antioxidant) and ascorbic acid of 5.78 ppm. Methanol extract is classified as very toxic with an LC<sub>50</sub> of 38.21 ppm. The dominant compounds produced by gas chromatography–mass spectrometer were hexanedioic acid, bis(2-ethylhexyl) ester; ethyl oleate; 9-octadecenoic acid (Z)-, ethyl

ester; and hexadecanoic acid, ethyl ester. The diameter of the inhibition zone for *Saccharomyces cerevisiae* was 13.50–14.72 mm, *L. plantarum* was 10.31–17.72 mm, and *Leuconostoc mesenteroides* was 13.25–18.06 mm. All extracts had minimum inhibitory concentration values of 2.5 mg/mL and minimum bactericidal concentration values of 2.5 mg/mL for *n*-hexane extract and 60°C water, while ethanol and methanol extracts had MBC values >50 mg/mL.

**Keywords:** *Hopea beccariana* Burck, phenolic, antioxidant, inhibition zone diameter, fatty acid ester

## 1 Introduction

*Hopea beccariana* Burck is a member of the Dipterocarpaceae family whose regional names are Cengal Pasir (Malay and Sarawak), Jangkang, Merawan, Merawan Batu (Malay), and Selangan Penak (Sabah) [1], while the local name of the plant in Mandor Village, Landak Regency, West Kalimantan Province-Indonesia, is Resak Jawai. The stem bark of *H. beccariana* Burck in West Kalimantan is used in the processing of coconut sugar, while other countries use the stem bark of *Sacoglottis gabonensis* and *Alstonia boonei* in Ikot Ekpene, Akwa Ibom State, Nigeria [2], Payorm (*Shorea roxburghii* G. Don) in Thailand [3], and Kiam wood (*Cotylelobium lanceatum* Carth) in Thailand [4].

The bark of *H. beccariana* Burck, which is inserted into the coconut sap container during the sap-tapping process, aims to prevent the sap damage process caused by microbes while the sap is tapped. The presence of the bark of *H. beccariana* Burck added in the tapping process will produce juice that has good quality and can be processed into coconut sugar. It is suspected that the bark of *H. beccariana* Burck contains antimicrobial compounds, such as phenols and flavonoids. Based on existing references, the secondary metabolite content of plants from the Dipterocarpaceae family is very diverse and is

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believed to be an antimicrobial compound [5,6], such as phenols [7], oligostilbenoids (oligomer resveratrol) [8–11], flavonoids [12,13], and triterpenoid [14,15].

The active components in plants can be separated by extracting plant material using a solvent. The process of extracting active ingredients from plants has been carried out by several researchers [16–20]. Commonly used organic solvents to extract phytochemical content from plants are ethanol, methanol, *n*-hexane, and water. Methanol solvent was used to extract the phytochemistry compounds from the bark, fruit, and flowers of *Crotalaria retusa* L. [21]. Methanol and water solvents were used to determine the total phenol and flavonoid contents in the bark of *Ricinodendron heudelotii* [22]. Ethanol solvent was used to determine the total content of flavonoids and phenols in the leaves and roots of *Euphorbia hirta* [23]. Hot water was used to extract fresh *Dendrobium sonia* “Earsakul” orchids [24].

The stem bark of *H. beccariana* Burck extract obtained through the maceration process needs to be proven that the extract contains antimicrobial bioactive compounds, and then, the extract can be analyzed using gas chromatography–mass spectrometer (GC–MS) to see the secondary metabolite compounds. Several researchers have used GC–MS to determine the compound content of a plant [25–28]. For this reason, the role of secondary metabolite compounds that are antimicrobial and are believed to prevent the process of sap damage can be identified through the antimicrobial activity of the extracts obtained.

Currently, there is no scientific information related to the physicochemical characteristics of various types of *H. beccariana* Burck bark extract including yield, total phenol content, total flavonoid content, antioxidant activity based on the highest 50% inhibition concentration ( $IC_{50}$ ) value, toxicity based on lethality concentration 50% ( $LC_{50}$ ), bioactive compounds, minimum inhibitory concentration (MIC), and minimum bactericidal concentration (MBC). Therefore, in this study, it is necessary to extract the bark of *H. beccariana* Burck using various solvents of ethanol, methanol, *n*-hexane, and hot water. The purpose of this study was to determine the physicochemical and microbiological characteristics of various types of *H. beccariana* Burck extract, which have the potential as natural preservatives for coconut sap. Therefore, in this research, it is necessary to extract the bark of *H. beccariana* Burck using various solvents of ethanol, methanol, *n*-hexane, and hot water. The purpose of this study was to determine the physicochemical characteristics of various types of *H. beccariana* Burck extract, which have the potential as natural preservatives for coconut sap. It is hoped that the results of the study can provide information on the type of

solvent used to extract the bark of *H. beccariana* Burck so that in the future the extract can be used to prevent the process of damage to coconut sap caused by microbes.

## 2 Materials and methods

### 2.1 Materials

The material used in this study was stem bark of *H. beccariana* Burck, obtained from Mandor Village, Mandor District, Landak Regency, West Kalimantan Province, Indonesia. The sampling location with the coordinate points is E 109°21'35.73", N 00°18'26.50" altitude of 41 m above sea level. The characteristics of the stem bark of *H. beccariana* Burck taken are the outer bark patterned with a mixed color of dark brown, gray, and green, while the inner bark is light brown, the age of the tree is  $\pm$ 20 years, and the height of the tree is  $\pm$ 12 m. The solvents used for the extraction process were ethanol (99.9%, J.T. Baker Avantor), methanol (99.8%, J.T. Baker Avantor), *n*-hexane (98.5%, J.T. Baker Avantor), and water.

### 2.2 Sample preparation

The stem bark of *H. beccariana* Burck is dried naturally using the sun until the moisture content reaches 9.70%. The bark was crushed with a disk mill (Type FFC-23 Qingdao Dahua Double Circle Machinery Co., LTD, China) into powder form [29] and then sieved through an 80 mesh sieve [30]. The dry powder obtained was packed in a plastic jar, tightly closed, and stored in a refrigerator at a low temperature (+4°C) until waiting for further use [21].

### 2.3 Extraction of *H. beccariana* Burck stem bark

The process of extracting the stem bark of *H. beccariana* Burck uses the maceration method and is modified from a combination of procedures carried out by [31–33]. Samples of *H. beccariana* Burck bark powder of 20 g mixed with 100 mL of solvent each (ethanol grade 99.8%, methanol grade 99.8%, *n*-hexane, and hot water temperature 60°C) were macerated while stirring using a magnetic stirrer at room temperature for 24 h. The solution is filtered with filter paper Whatman No. 42 in a vacuum. Solvents are removed

using a rotary evaporator (Buchi Rotaevaporator R114, Switzerland) at 45°C for ethanol, methanol, and *n*-hexane solvents until 90% of the solvent is evaporated, while the water solvent is evaporated at a temperature of 60°C. The extract is stored at a temperature condition of +4°C until analysis.

## 2.4 Yield extract

The yield of the extract is the weight of the extracted material divided by the weight of the raw bark material multiplied by 100% [34].

## 2.5 Total phenolic content determination

The total phenolic content in the stem bark extract of *H. beccariana* Burck was determined by following the procedure from [35] using spectrophotometry by the Folin–Ciocalteu test method. The extract is made as much as 5 mg and aquadest are added until the volume reaches 5 mL in a measuring flask. The extract solution was pipetted 1 mL and added to 9 mL of distilled water in a volumetric flask (25 mL). The phenol reagent Folin–Ciocalteu has added as much as 1 mL, homogenized, and incubated for 5 min. A 7% solution of sodium carbonate ( $\text{Na}_2\text{CO}_3$ ) (weighed as much as 3.5 g of  $\text{Na}_2\text{CO}_3$  then dissolved with aquadest up to 50 mL) of 10 mL was added to the mixture. The volume of the solution is made up to 25 mL by adding aquadest. Gallic acid standard solutions were made in several concentrations, namely 20, 40, 60, 80, and 100  $\mu\text{g}/\text{mL}$ , in the same way as the extract sample preparation described previously. Samples of the extract and standard solution of gallic acid were incubated for 90 min at room temperature. The absorbance value of the test sample and the standard solution was determined against the blank reagent at 550 nm with an ultraviolet (UV)/visible spectrophotometer. The total phenol content is expressed as mg of gallic acid equivalent (GAE)/g extract.

## 2.6 Total flavonoid content determination

The total flavonoid content of the stem bark of *H. beccariana* Burck extract was measured following the procedure of [35]. The total flavonoid content was measured by the aluminum chloride colorimetric test. The extract was

made as much as 5 mg and added methanol until the volume reached 5 mL in a measuring flask. The extract solution was pipetted 1 mL and reacted with 4 mL of distilled water into a 10 mL volumetric flask. 0.3 mL of 5% sodium nitrite was added and left for 5 min, adding 0.3 mL of 10% aluminum chloride. After 5 min, 2 mL of 1 M sodium hydroxide was added to distilled water so that the volume becomes 10 mL. Standard solutions of quercetin were prepared at various concentrations (20, 40, 60, 80, and 100  $\mu\text{g}/\text{mL}$ ) in the same manner as previously described. Measuring the absorbance of the sample solution and standard solution based on the reagent blank at 510 nm using a UV/Visible spectrophotometer. The total flavonoid content in the extract was expressed as mg QE/g extract.

## 2.7 Antioxidant activity testing

Determination of the antioxidant activity of the bark extract of *H. beccariana* Burck using the DPPH (1,1-diphenyl-2-picrylhydrazyl) method following the procedure of [36,37] modified. The solution DPPH 0.4 mM is prepared by weighing DPPH 0.0158 g and dissolved with methanol p.a up to 100 mL in a measuring flask. A stock solution with a concentration of 1,000 ppm consisting of 50 mg of *H. beccariana* Burck stem bark extract dissolved in a measuring flask by adding up to 50 mL of methanol was made. On the absorption measurement of the blank DPPH solution, first pipetting 1 mL of 0.4 mM DPPH solution by adding methanol to 5 mL in the measuring flask. On the other hand, it prepares a series of extract solution concentrations from the stock that has been made (concentration of 1,000 ppm) for testing the free radical scavenging activity of DPPH with extract samples consisting of concentrations of 20, 40, 60, 80, 100, 120, and 140 ppm. Each concentration series plus 1 mL of DPPH 0.4 mM is then added methanol until the volume reaches 5 mL in the measuring flask. The solution of DPPH blanks and extract samples were homogenized and incubated for 30 min. Measure the uptake of the DPPH blank solution and extract the sample with UV–Vis spectrophotometry at a wavelength of 515 nm. As a comparison, ascorbic acid solutions were made with concentrations of 2, 4, 6, and 8 ppm, the work was the same as before.

The percentage of free radical scavenging is calculated by the formula:

$$\% \text{ Radical scavenging} = \frac{(\text{Absorbance blank} - \text{Absorbance sample})}{\text{Absorbance blank}} \times 100\%$$

and determination of  $IC_{50}$  value (50% inhibitory concentration) based on probit analysis from concentration log data with probit free radical scavenging percentage.

## 2.8 Toxicity ( $LC_{50}$ ) test of *Hopea beccariaan* Burck stem bark extract

Extract toxicity was carried out through the Brine Shrimp Lethality test (BSLT) [36]. *Artemia salina* L. eggs were incubated in a 1 L glass beaker filled with seawater and irradiated with a 40-W fluorescent lamp while being aerated for 48 h. The eggs hatch into nauplii and are ready to be tested. A sample stock solution with a concentration of 2,000 ppm (40 mg of bark extract in 20 mL of seawater) was made. Five sterile tubes were prepared, then add 1 mL of seawater and 10 nauplii into each tube, and sample solution from the stock solution of 2,000 ppm made into 10, 100, 500, and 1,000 ppm, while the control used 4 mL of seawater containing 10 nauplii without adding sample stock. The five tubes were incubated for 24 hours at room temperature. The mean percentage of nauplii mortality was plotted against the logarithm of the concentration of the bark extract of *Hopea beccariaan* Burck. The half maximal mortality concentration ( $LC_{50}$ ) was determined by calculating the antilogarithmic linear equation obtained from the curve of the relationship between the level of the bark extract and the mean percent mortality of *Artemia nauplii*.

## 2.9 Analysis of the content of bioactive compounds in extracts using GC-MS

GC-MS was used to identify the compounds present in the extract according to the procedure [38]. The analysis was carried out according to the specifications of the Shimadzu QP 2010 SE GC-MS equipment: column type is 5 ms, Restek Corp (30 m length). The condition of the injector is gas chromatography with a split ratio of 8.4, the sample flow rate is 21.8 mL/min, and the helium flow rate is 2.00 mL/min. The condition of the gas chromatographic column was that the initial column temperature was 60°C, the final column temperature was 290°C, the initial holding time was 5 min, the final holding time was 10 min, and the temperature rise rate was 6°C per minute. Mass spectroscopy conditions with interface temperature 250°C, electron ionization temperature 300°C, and detection range  $m/z$  at 45–500  $m/z$ , with retention time (Rt)

40 min. The carrier gas used is helium with an amount of 45–500 atomic mass units (AMU). Compound identification and structure determination were based on a comparison of mass spectra and fragmentation profiles using published data using Wiley 229, NIST 12, and NIST 62 Library software.

## 2.10 Antimicrobial activity test using disk diffusion method

The antimicrobial activity test was carried out using the agar diffusion method using disk paper according to the procedure [39]. The test microbes used were the yeast *Saccharomyces cerevisiae* and the bacteria *Lactobacillus plantarum* and *Leuconostoc mesenteroides*. Cultures of *S. cerevisiae* were first grown on potato dextrose agar (PDA) media [40]. One inoculating loop of *S. cerevisiae* culture was inoculated on 5 mL of sterile PDA media and then incubated at 30°C for 48 h [41], while *L. plantarum* and *L. mesenteroides* were grown on de Man, Rogosa and Sharpe (MRS) broth media [42], and one loop of bacterial culture was inoculated on 5 mL of sterile MRS broth media and then incubated at 30°C for 24 h. Microbial suspension of 300  $\mu$ L/Petri dish (for yeast) and 250  $\mu$ L/Petri dish (for bacteria) on the agar surface by the spread plate method, so that each cup contains about  $3 \times 10^6$  CFU. Disks (size 5 mm) were immersed in the bark extract of *H. beccariana* Burck at a concentration of 5% (250  $\mu$ g/50  $\mu$ L or 50  $\mu$ g/ $\mu$ L) which was prepared from 1 mg of extract dissolved in 200  $\mu$ L of sterile distilled water. The disks were placed on a medium that had been inoculated with yeast and bacteria aseptically, respectively. The positive control used fluconazole and streptomycin antibiotics each with a concentration of 250 ppm and the negative control used ethanol, methanol, *n*-hexane, and sterile distilled water as solvents. All plates were incubated at 30°C for 24 h. Observations were made on the diameter of the clear zone around the disk, which was recorded in millimeters (mm).

## 2.11 Determination of MIC and MBC

### 2.11.1 Preparation of growth media and microbial suspension

The yeast suspension of *S. cerevisiae* was grown on PDA media, while *L. plantarum* and *L. mesenteroides* were grown in MRS broth with a density of  $10^6$  CFU/mL. A

suspension was prepared based on the turbidity of the McFarland 0.5 standard made from 9.95 mL of 1% sulfate acid solution and 0.05 mL of 1.175% barium chloride solution, which is equivalent to the density of yeast and bacteria  $10^8$  CFU/mL [43,44]. A tube containing a standard solution of 0.5 McFarland was prepared. Yeast and bacterial suspensions were prepared by taking 4–10 oses of PDA and MRS broth media that had been incubated for 24 h, putting them in a tube containing 0.9% sodium chloride and then homogenized. The yeast and bacterial suspensions were equalized for turbidity with a standard solution of 0.5 McFarland. The suspension that has been made is then diluted by pipetting 0.1 mL of yeast and bacterial suspension ( $10^8$  CFU/mL) put into a sterile tube and added 9.9 mL of 0.9% sodium chloride solution so that the density of the yeast and test bacteria is  $10^6$  CFU/mL.

#### 2.11.2 Determination of MIC

The determination of the MIC was carried out according to the modified study [42,45]. A series of concentrations of *H. beccariana* Burck bark extract 2.5, 5.0, 10, 30, and 50 mg/mL, which has been diluted with 1% CMC solvent, were made. Sterile test tubes were prepared according to the number of treatments. The test tubes were filled with 8.8 mL of PDA and 8.8 mL of sterile MRS broth, respectively. Each reaction tube was added with 1 mL of *H. beccariana* Burck extract according to the predetermined concentration and 200  $\mu$ L of each culture of the yeast *S. cerevisiae* and *L. plantarum* and *L. mesenteroides* bacteria. All tubes were vortexed so that they were homogeneous. All reaction tubes were incubated for 24 h at 37°C. Determination of MIC based on the difference in optical density value after incubation minus before incubation resulted in the lowest negative concentration from the results of measuring using a spectrophotometer at a wavelength of 630 nm.

#### 2.11.3 Determination of MBC

Determination of MBC refers to research [42] by first diluting the PDA and MRS broth and preparing five sterile Petri dishes. The bark extract of *H. beccariana* Burck from the concentration series was determined during the MIC test and also the control was put into a tube containing PDA and MRS broth. The tube was added to each suspension of yeast and bacteria, then homogenized and then poured into a sterile Petri dish, and waited for the media

to solidify. After that, it was incubated at 37°C for 24 h. The incubation results can be seen by the presence or absence of colony growth on the media. The total bacterial colonies were calculated using the colony counter. Determination of MBC based on the absence of yeast and bacteria growth was not seen in the Petri dish.

## 2.12 Data analysis

The physicochemical and microbiological characteristics data were carried out in three replications and averaged by displaying the standard deviation (SD). The data obtained were analyzed by one-way analysis of variance (ANOVA) at a 5% level. If the treatment given has a significant effect, it is continued with the Tukey test at a 5% level. The SAS software version 9.4 was used to process ANOVA and Tukey test data, and Microsoft Excel 365 was used to collect and organize primary data.

## 3 Results

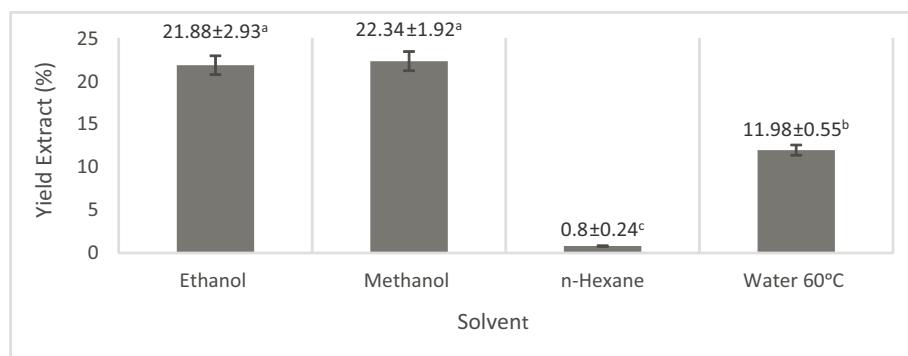
### 3.1 Physicochemical characteristics of *Hopea beccariana* Burck bark extract

#### 3.1.1 Yield extract of *H. beccariana* Burck stem bark

The results of the analysis of variance showed that the use of various types of solvents to extract the stem bark of *H. beccariana* Burck had a noticeable influence ( $\alpha = 5\%$ ) on the yield. Figure 1 shows that ethanol and methanol have higher yields of 21.88 and 22.34%, respectively, compared to using *n*-hexane and water of 0.80 and 11.98%, respectively.

#### 3.1.2 Total phenolic content of *H. beccariana* Burck stem bark extract

The results of the analysis of variance showed that the use of various types of solvents to extract the bark of *H. beccariana* Burck had a significant effect ( $\alpha = 5\%$ ) on the total phenolic content. Table 1 shows that the ethanol extract of the bark of *H. beccariana* Burck has the highest total phenolic content (53.39 mg GAE/g) and is significantly different from other extracts, while the lowest total phenolic content was found in the *n*-hexane extract at 18.56 mg GAE/g.



**Figure 1:** The yield of *H. beccariana* Burck stem bark extract from various solvents.

### 3.1.3 Total flavonoid content of *H. beccariana* Burck stem bark extract

The results of the analysis of variance showed that the use of various types of solvents to extract the bark of *H. beccariana* Burck had a significant effect ( $\alpha = 5\%$ ) on the total flavonoid content. The extract was obtained by maceration process for 24 h at room temperature with three repetitions using solvents of ethanol, methanol, *n*-hexane, and hot water at 60°C and added with stirring using a magnetic stirrer, the total flavonoid content varied. Based on Table 1, the highest total flavonoid content was found in the bark extract of *H. beccariana* Burck using hot water at 60°C, which was 106.70 mg QE/g, while the lowest total flavonoid content was found in extracts using the *n*-hexane solvent of 78.58 mg QE/g.

### 3.1.4 Antioxidant activity ( $IC_{50}$ ) of *H. beccariana* Burck stem bark extract

The results of the analysis showed that the use of various types of solvents to extract the stem bark of *H. beccariana* Burck had a significant effect ( $\alpha = 5\%$ ) on the  $IC_{50}$  value. Table 1 shows that all extracts of the bark of *H. beccariana* Burck, both ethanol, methanol, *n*-hexane, and water extracts are classified as having strong antioxidant

activity with  $IC_{50}$  of 80.28, 84.75, 81.83, and 91.80 ppm, respectively, while the  $IC_{50}$  value of ascorbic acid as a comparison is 5.78 ppm.

### 3.1.5 Toxicity test of *H. beccariana* Burck stem bark extract

The results of the toxicity test using the brine shrimp lethality test (BSLT) method using *Artemia salina* L. sea shrimp larvae from various types of *H. beccariana* Burck bark extract are presented in Figure 2. The concentration of *H. beccariana* Burck stem bark extract used was up to 1,000 ppm in the toxicity test resulting that the methanol and water extracts being classified as very toxic with  $LC_{50}$  of 38.21 and 75.74 ppm, respectively.

## 3.2 Content of bioactive compounds in *H. beccariana* Burck stem bark extract using GC-MS

### 3.2.1 *H. beccariana* Burck stem bark ethanol extract

The GC-MS chromatogram (Figure 3) shows 20 compounds identified from the ethanolic extract of the bark of *H. beccariana* Burck consisting of groups of alkaloids,

**Table 1:** Total phenolic content, total flavonoid, and antioxidant activity ( $IC_{50}$ ) of various types of *H. beccariana* Burck stem bark extract

Solvent	Total phenolic content $\pm$ SD* (mg GAE/g extract)	Total flavonoid content $\pm$ SD* (mg QE/g extract)	Antioxidant activity, $IC_{50}$ $\pm$ SD* (ppm)
Ethanol	53.39 $\pm$ 1.61 <sup>a</sup>	99.37 $\pm$ 6.78 <sup>a</sup>	80.28 $\pm$ 1.44 <sup>b</sup>
Methanol	30.47 $\pm$ 1.45 <sup>b</sup>	91.58 $\pm$ 9.90 <sup>ab</sup>	84.75 $\pm$ 1.77 <sup>ab</sup>
<i>n</i> -Hexane	18.56 $\pm$ 1.41 <sup>c</sup>	78.58 $\pm$ 3.30 <sup>b</sup>	81.83 $\pm$ 4.01 <sup>b</sup>
Water 60°C	33.94 $\pm$ 2.09 <sup>b</sup>	106.70 $\pm$ 0.99 <sup>a</sup>	91.80 $\pm$ 2.83 <sup>a</sup>
Tukey $\alpha = 5\%$	4.35	16.32	7.08
Ascorbic acid			5.78

\*Values with different notations in the same column show a significant difference in the 5% Tukey test.

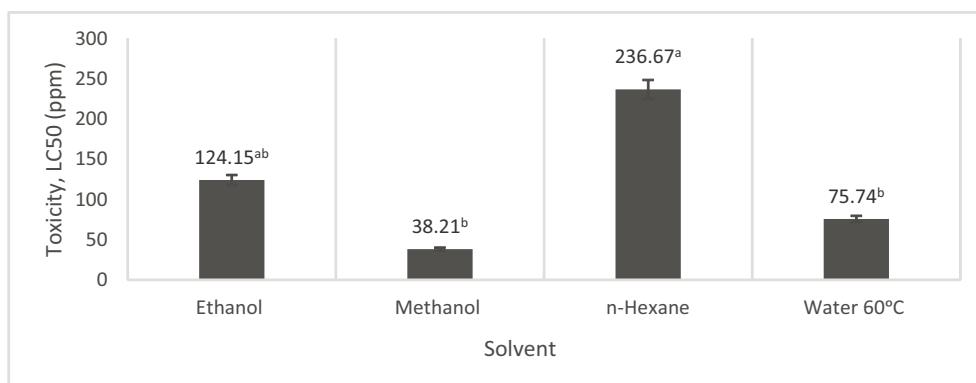


Figure 2: Toxicity LC<sub>50</sub> (ppm) of *H. beccariana* Burck stem bark extract from various solvents.

alcohols, phenolics, aromatics, alkanes, alkenes, and fatty acid esters (Table 2).

The group of fatty acid ester compounds was the largest identified. The percentage of the fatty acid ester compound group was 85.37 consisting of hexadecanoic acid, methyl ester (9.35%), hexadecanoic acid, ethyl ester (13.42%), 9-octadecenoic acid, methyl ester, (E)- (9.55%), 9-octadecenoic acid, methyl ester, (E)- (5.96%), octadecanoic acid, methyl ester (1.85%), ethyl oleate (15.52%), ethyl oleate (10.85%), heptadecanoic acid, ethyl ester (3.70%), ethyl linoleate (0.72%), hexanedioic acid, bis (2-ethylhexyl) ester (13.36%), and decanoic acid, 2-ethylhexyl ester (0.44%). The group of alkene compounds consists of 9-eicosene (E) (0.85%) and 1-pentadecene (0.47%). The group of alcohol compounds (4.14%) consisted of 2-furanmethanol (0.46%) and 1-hexanol, 2-ethyl (3.68%).

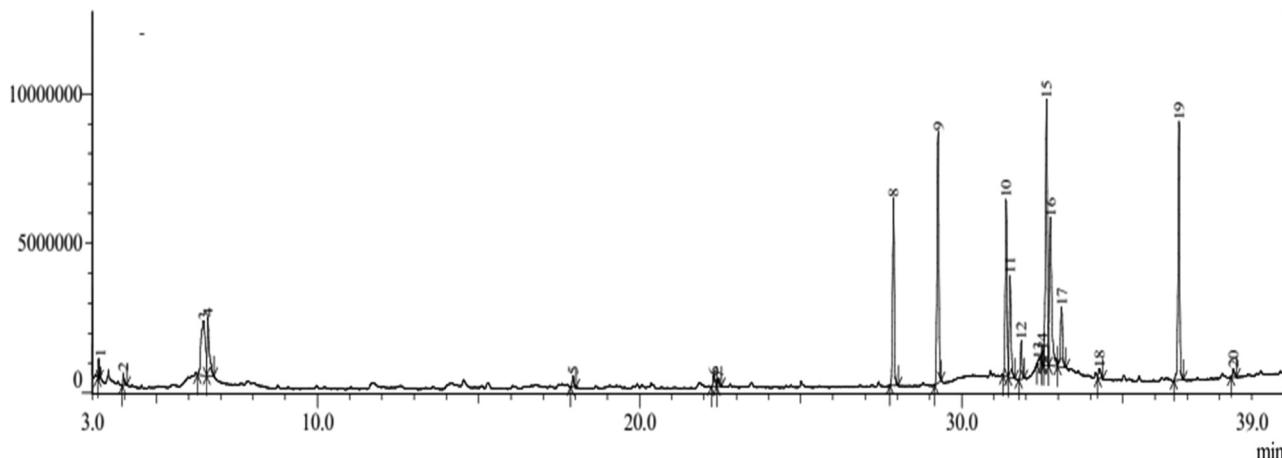
There was a group of compounds that identified only one compound, including groups of alkaloids, phenols, aromatics, and alkanes. The group of alkaloid compounds

is pyrrolidine-.alpha.,.alpha.,.alpha.,.alpha.-D4 (0.42%). The group of phenolic compounds is phenol (7.48%). The group of aromatic compounds is phenol, 4,4'-methylenebis- (0.67%). The group of alkane compounds is pentadecane (0.60%).

### 3.2.2 *H. beccariana* Burck stem bark methanol extract

The GC-MS chromatogram (Figure 4) shows 20 compounds identified from the methanol extract of the bark of *H. beccariana* Burck consisting of groups of ketones, aldehydes, alcohols, phenolics, aromatics, alkanes, alkenes, and fatty acid esters (Table 3),

The group of fatty acid ester compounds was the largest identified (Table 3). The percentage of fatty acid ester compound group was 84.78 consisting of propanoic acid, 2-hydroxy-, ethyl ester (0.47%), hexadecanoic acid, methyl ester (8.71%), hexadecanoic acid, ethyl ester (11.57%), 9-octadecenoic acid (Z)-, methyl ester (8.63%), 9-octadecenoic



**Table 2:** Identified compounds from ethanol extract of *H. beccariana* Burck stem bark based on GC-MS spectrum

No. peak	Retention time (min)	Compound name	Formula molecule	Compound group	Area (%)
1	3.201	Pyrrolidine-.alpha.,.alpha.,.alpha.,.alpha.-D4	C <sub>4</sub> H <sub>5</sub> D <sub>4</sub> N	Alkaloid	0.42
2	3.960	2-Furanmethanol	C <sub>5</sub> H <sub>6</sub> O <sub>2</sub>	Alcohol	0.46
3	6.455	Phenol	C <sub>6</sub> H <sub>6</sub> O	Phenol	7.48
4	6.600	1-Hexanol, 2-ethyl	C <sub>8</sub> H <sub>18</sub> O	Alcohol	3.68
5	17.924	Pentadecane	C <sub>15</sub> H <sub>32</sub>	Alkane	0.60
6	22.296	9-Eicosene, (E)	C <sub>20</sub> H <sub>40</sub>	Alkene	0.85
7	22.430	1-Pentadecene	C <sub>15</sub> H <sub>30</sub>	Alkene	0.47
8	27.875	Hexadecanoic acid, methyl ester	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid ester	9.35
9	29.263	Hexadecanoic acid, ethyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	13.42
10	31.381	9-Octadecenoic acid, methyl ester, (E)-	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	9.55
11	31.504	9-Octadecenoic acid, methyl ester, (E)-	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	5.96
12	31.854	Octadecanoic acid, methyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	1.85
13	32.385	Phenol, 4,4'-methylenebis-	C <sub>12</sub> H <sub>12</sub> O <sub>2</sub>	Aromatic	0.67
14	32.505	9,12-Octadecadienoic acid (Z, Z)-methyl ester	C <sub>19</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid ester	0.65
15	32.637	Ethyl oleate	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	15.52
16	32.757	Ethyl oleate	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	10.85
17	33.103	Heptadecanoic acid, ethyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	3.70
18	34.276	Ethyl linoleate	C <sub>20</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	0.72
19	36.744	Hexanedioic acid, bis(2-ethylhexyl) ester	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub>	Fatty acid ester	13.36
20	38.428	Decanoic acid, 2-ethylhexyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	0.44

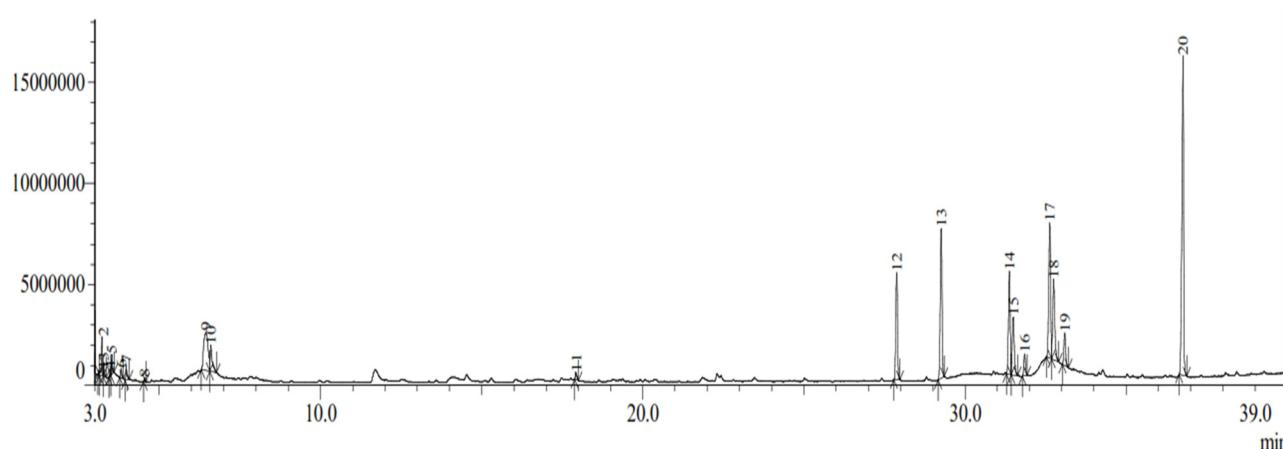
acid (Z)-, methyl ester (5.69%), octadecanoic acid, methyl ester (1.68%), 9-octadecenoic acid (Z)-, ethyl ester (11.52%), 9-octadecenoic acid (Z)-, ethyl ester (7.93%), heptadecanoic acid, ethyl ester (2.65%), and hexanedioic acid, bis(2-ethylhexyl) ester (25.93%).

Only one or two groups of compounds were identified including groups of aldehydes, aromatics, phenols, alkenes, alcohols, ketones, and alkanes (Table 3). The aldehyde group is 2-furancarboxaldehyde (1.06%). The aromatic group is *N*-methoxy formamide (0.40%). The phenol group is phenol (8.07%). The alkene group is pentadecane (0.48%). The alcohol group included 2-furanmethanol (0.78%) and

1-hexanol, 2-ethyl- (2.26%). The ketone group of 0.42% consisted of (2H)-furan-3-one (0.15%) and 2,5-furandione (0.27%). The alkane group of 1.77% consisted of ethane, 1,1',1''-[methylidynetris(oxy)]tris (1.62%), and ethane, 1,1',1''-[methylidynetris(oxy)]tris (0.15%).

### 3.2.3 *H. beccariana* Burck stem bark *n*-hexane extract

The GC-MS chromatogram (Figure 5) shows 20 compounds identified from the *n*-hexane extract of the bark of *H. beccariana* Burck consisting of groups of phenolic

**Figure 4:** Chromatogram of GC-MS methanol extract of *H. beccariana* Burck stem bark.

**Table 3:** Identified compounds from methanol extract of *H. beccariana* Burck stem bark based on GC-MS spectrum

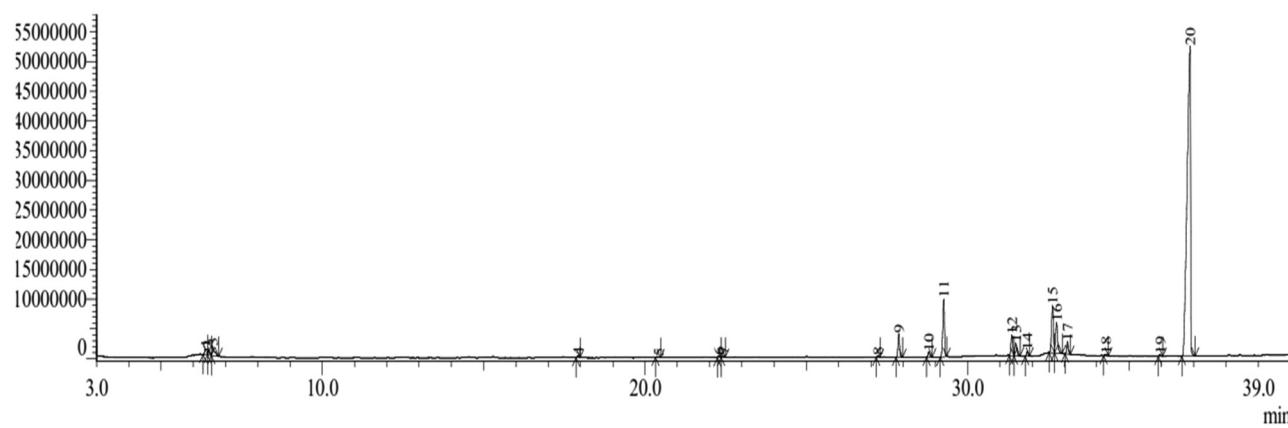
No. peak	Retention time (min)	Compound name	Formula molecule	Compound group	Area (%)
1	3.121	(2H)-Furan-3-one	C <sub>4</sub> H <sub>4</sub> O <sub>2</sub>	Ketone	0.15
2	3.223	Ethane, 1,1',1''-[methylidynetris(oxy)]tris	C <sub>2</sub> H <sub>16</sub> O <sub>3</sub>	Alkane	1.62
3	3.284	Propanoic acid 2-hydroxy-ethyl ester	C <sub>5</sub> H <sub>10</sub> O <sub>3</sub>	Fatty acid ester	0.47
4	3.480	2.5-Furandione	C <sub>4</sub> H <sub>2</sub> O <sub>3</sub>	Ketone	0.27
5	3.531	2-Furancarboxaldehyde	C <sub>5</sub> H <sub>4</sub> O <sub>2</sub>	Aldehyde	1.06
6	3.822	N-Methoxy formamide	C <sub>2</sub> H <sub>5</sub> NO <sub>2</sub>	Aromatic	0.40
7	3.975	2-Furanmethanol	C <sub>5</sub> H <sub>6</sub> O <sub>2</sub>	Alcohol	0.78
8	4.546	Ethane, 1,1',1''-[methylidynetris(oxy)]tris	C <sub>7</sub> H <sub>16</sub> O <sub>3</sub>	Alkane	0.15
9	6.454	Phenol	C <sub>6</sub> H <sub>6</sub> O	Phenol	8.07
10	6.602	1-Hexanol, 2-ethyl-	C <sub>8</sub> H <sub>18</sub> O	Alcohol	2.26
11	17.924	Pentadecane	C <sub>15</sub> H <sub>32</sub>	Alkene	0.48
12	27.875	Hexadecanoic acid, methyl ester	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid ester	8.71
13	29.254	Hexadecanoic acid, ethyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	11.57
14	31.382	9-Octadecenoic acid (Z)-, methyl ester	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	8.63
15	31.503	9-Octadecenoic acid (Z)-, methyl ester	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	5.69
16	31.854	Octadecanoic acid, methyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	1.68
17	32.634	9-Octadecenoic acid (Z)-, ethyl ester	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	11.52
18	32.755	9-Octadecenoic acid (Z)-, ethyl ester	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	7.93
19	33.103	Heptadecanoic acid, ethyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	2.65
20	36.765	Hexanedioic acid, bis(2-ethylhexyl) ester	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub>	Fatty acid ester	25.93

compounds, alcohols, alkanes, alkenes, aromatics, and fatty acid esters (Table 4).

The group of fatty acid ester compounds was the largest identified (Table 4). The percentage of fatty acid ester compound group is 95.84% consisting of 1-pentadecene (0.12%), hexanedioic acid, diethyl ester (0.14%), hexadecanoic acid, methyl ester (2.61%), hexadecanoic acid, ethyl ester (6.23%), 9-octadecenoic acid, methyl ester, (E)- (2.48%), 9-octadecenoic acid (Z)-, methyl ester (1.72%), octadecanoic acid, methyl ester (0.51%), ethyl oleate (5.96%), 9-octadecenoic acid (Z)-, ethyl ester (4.23%),

heptadecanoic acid, ethyl ester (1.38%), ethyl linoleate (0.21%), hexanedioic acid, bis(2-ethylhexyl) ester (0.18%), and Hexanedioic acid, bis(2-ethylhexyl) ester (70.07%).

Only one or two groups of compounds were identified (Table 4), including aromatic groups, alkenes, alcohols, phenols, and alkanes. The aromatic group is dibutyl phthalate (0.63%). The alcohol group is 1-hexanol, 2-ethyl- (1.32%). The alkene group is 9-octadecene, (E)- (0.20%). The phenol group (1.63%) consisted of phenol (0.99%) and phenol (0.64%). The alkane group (0.37%) consisted of pentadecane (0.23%) and hexadecane (0.14%).

**Figure 5:** Chromatogram of GC-MS *n*-hexane extract of *H. beccariana* Burck stem bark.

**Table 4:** Identified compounds from the *n*-hexane extract of *H. beccariana* Burck stem bark based on GC-MS spectrum

No. peak	Retention time (min)	Compound name	Formula molecule	Compound group	Area (%)
1	6.385	Phenol	C <sub>6</sub> H <sub>6</sub> O	Phenol	0.99
2	6.455	Phenol	C <sub>6</sub> H <sub>6</sub> O	Phenol	0.64
3	6.602	1-Hexanol, 2-ethyl-	C <sub>8</sub> H <sub>18</sub> O	Alcohol	1.32
4	17.919	Pentadecane	C <sub>15</sub> H <sub>32</sub>	Alkane	0.23
5	20.425	Hexadecane	C <sub>16</sub> H <sub>34</sub>	Alkane	0.14
6	22.288	9-Octadecene, ( <i>E</i> )-	C <sub>18</sub> H <sub>36</sub>	Alkene	0.20
7	22.428	1-Pentadecene	C <sub>15</sub> H <sub>30</sub>	Fatty acid ester	0.12
8	27.211	Hexanedioic acid, diethyl ester	C <sub>10</sub> H <sub>18</sub> O <sub>4</sub>	Fatty acid ester	0.14
9	27.862	Hexadecanoic acid, methyl ester	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid ester	2.61
10	28.792	Dibutyl phthalate	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>	Aromatic	0.63
11	29.256	Hexadecanoic acid, ethyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	6.23
12	31.368	9-Octadecenoic acid, methyl ester, ( <i>E</i> )-	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	2.48
13	31.493	9-Octadecenoic acid ( <i>Z</i> )-, methyl ester	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	1.72
14	31.846	Octadecanoic acid, methyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	0.51
15	32.635	Ethyl oleate	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	5.96
16	32.757	9-Octadecenoic acid ( <i>Z</i> )-, ethyl ester	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	4.23
17	33.095	Heptadecanoic acid, ethyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	1.38
18	34.280	Ethyl linoleate	C <sub>20</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	0.21
19	35.957	Hexanedioic acid, bis(2-ethylhexyl) ester	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub>	Fatty acid ester	0.18
20	36.890	Hexanedioic acid, bis(2-ethylhexyl) ester	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub>	Fatty acid ester	70.07

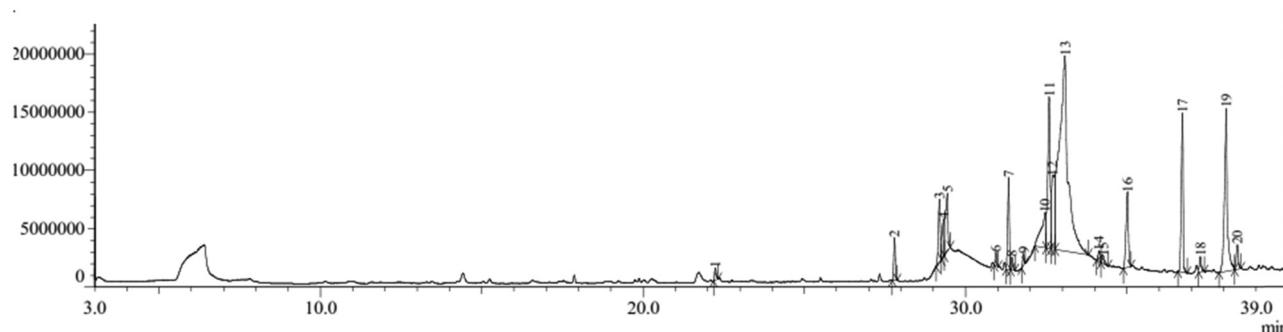
### 3.2.4 Water 60°C extract stem bark of *H. beccariana* Burck

The GC-MS chromatogram (Figure 6) shows 20 compounds identified from the 60°C aqueous extracts of the bark of *H. beccariana* Burck consisting of groups of alkenes, aldehydes, fatty acids and fatty acid esters, ethers, steroids, and alcohols (Table 5).

The group of fatty acid compounds and fatty acid esters was the largest identified (Table 5). The percentage of fatty acid compounds and fatty acid esters is 84.61 consisting of hexadecanoic acid, methyl ester (1.52%), hexadecanoic acid, ethyl ester (2.95%), *n*-hexadecanoic acid (3.80%), 9-octadecenoic acid, methyl ester, (*E*)- (3.62%), 9-octadecenoic acid (*Z*)-, methyl ester (0.60%),

octadecanoic acid, methyl ester (0.40%), ethyl linoleate (4.21%), 9-octadecenoic acid (*Z*)-, ethyl ester (8.54%), ethyl oleate (6.25%), oleic acid (40.46%), hexadecanoic acid, 2-hydroxy-1,3-propanediyl ester (4.31%), hexanedioic acid, bis(2-ethylhexyl) ester (6.76%), and hexadecanoic acid, 2-hydroxy-1,3-propanediyl ester (1.19%). The aldehyde group (3.40%) consisted of 9-octadecenal, (*Z*)- (1.89%), 9-tetradecenal, (*Z*)- (0.69%), and 9-octadecenal, (*Z*)- (10.85%).

The group of compounds identified only one compound, namely the ether group, steroids, alcohols, and alkenes (Table 5). The ether group is oxirane, tetradecyl- (0.51%). The steroid group was 14- $\beta$ -*H*-pregna (0.27%). The alcohol group is (*R*)-(-)-14-methyl-8-hexadecyn-1-ol (0.49%), and the alkene group is 9-octadecene, (*E*)- (0.48%).

**Figure 6:** Chromatogram of GC-MS water 60°C extract of *H. beccariana* Burck stem bark.

**Table 5:** Identified compounds from hot water 60°C extracts of *H. beccariana* Burck stem bark based on GC-MS spectrum

No. peak	Retention time (min)	Compound name	Formula molecule	Compound group	Area (%)
1	22.237	9-Octadecene, (E)-	C <sub>18</sub> H <sub>36</sub>	Alkene	0.48
2	27.796	Hexadecanoic acid, methyl ester	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid ester	1.52
3	29.188	Hexadecanoic acid, ethyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	2.95
4	29.315	9-Octadecenal, (Z)-	C <sub>18</sub> H <sub>34</sub> O	Aldehyde	1.89
5	29.431	n-Hexadecanoic acid	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	Fatty acid	3.80
6	30.936	Oxirane, tetradecyl-	C <sub>16</sub> H <sub>32</sub> O	Ether	0.51
7	31.328	9-Octadecenoic acid, methyl ester, (E)-	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	3.62
8	31.427	9-Octadecenoic acid (Z)-, methyl ester	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	0.60
9	31.791	Octadecanoic acid, methyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	0.40
10	32.455	Ethyl linoleate	C <sub>20</sub> H <sub>36</sub> O <sub>2</sub>	Fatty acid ester	4.21
11	32.585	9-Octadecenoic acid (Z)-, ethyl ester	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	8.54
12	32.690	Ethyl oleate	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	Fatty acid ester	6.25
13	33.073	Oleic acid	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	Fatty acid	40.46
14	34.127	14-β-H-Pregna	C <sub>21</sub> H <sub>36</sub>	Steroid	0.47
15	34.260	(R)-(-)-14-Methyl-8-hexadecyn-1-ol	C <sub>17</sub> H <sub>32</sub> O	Alcohol	0.49
16	35.017	Hexadecanoic acid, 2-hydroxy-1,3-propanediyl ester	C <sub>35</sub> H <sub>68</sub> O <sub>5</sub>	Fatty acid ester	4.31
17	36.726	Hexanedioic acid, bis(2-ethylhexyl) ester	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub>	Fatty acid ester	6.76
18	37.276	9-Tetradecenal, (Z)-	C <sub>14</sub> H <sub>26</sub> O	Aldehyde	0.69
19	38.076	9-Octadecenal, (Z)-	C <sub>18</sub> H <sub>34</sub> O	Aldehyde	10.85
20	38.425	Hexadecanoic acid, 2-hydroxy-1,3-propanediyl ester	C <sub>35</sub> H <sub>68</sub> O <sub>5</sub>	Fatty acid ester	1.19

### 3.3 Microbiological characteristics of *H. beccariana* Burck stem bark extract

#### 3.3.1 Inhibition zone diameter

The results of the analysis of variance showed that the use of various solvents to extract the bark of *H. beccariana* Burck had a significant effect ( $\alpha = 5\%$ ) on *L. plantarum*, while for *S. cerevisiae* and *L. mesenteroides* it had an insignificant effect ( $\alpha = 5\%$ ). The data on the diameter of the inhibition zone are presented in Table 6.

The diameter of the fluconazole antibiotic inhibition zone (positive control) against the yeast *S. cerevisiae* was between 9.83 and 27.33 mm (Table 7), while the antimicrobial activity against the yeast *S. cerevisiae* in the negative control was only shown in ethanol and methanol solvents with the

diameter of the inhibition zones 10.82 and 10.55 mm, respectively. Compared with the results of the study (Table 6), extracts of ethanol, methanol, *n*-hexane, and water at 60°C had the same inhibitory ability against yeast *S. cerevisiae* with inhibition zone diameters 14.28, 14.72, 13.50, and 14.56 mm, respectively. This means that the antimicrobial activity of the stem bark extract has the same ability as fluconazole antibiotics and is better than the negative control.

The diameter of the zone of inhibition of streptomycin antibiotics (positive control) against *L. plantarum* bacteria ranged from 10.89 to 13.56 mm and *L. mesenteroides* bacteria between 10.16 and 18.42 mm. In the negative control, the antimicrobial activity against *L. plantarum* and *L. mesenteroides* bacteria was only shown in ethanol and methanol solvents with inhibition zone diameters 10.39, 8.51, 9.42,

**Table 6:** Diameter of inhibition zones of various types of *H. beccariana* Burck stem bark extract

Extracts	Inhibition zone diameter (mm) $\pm$ SD		
	<i>S. cerevisiae</i> *	<i>L. plantarum</i> *	<i>L. mesenteroides</i> *
Ethanol	14.28 $\pm$ 3.29 <sup>a</sup>	17.72 $\pm$ 4.28 <sup>a</sup>	13.25 $\pm$ 3.44 <sup>a</sup>
Methanol	14.72 $\pm$ 0.92 <sup>a</sup>	15.67 $\pm$ 2.68 <sup>ab</sup>	15.22 $\pm$ 3.59 <sup>a</sup>
<i>n</i> -Hexane	13.50 $\pm$ 0.00 <sup>a</sup>	10.31 $\pm$ 1.63 <sup>b</sup>	13.83 $\pm$ 2.47 <sup>a</sup>
Water 60°C	14.56 $\pm$ 1.75 <sup>a</sup>	13.56 $\pm$ 1.86 <sup>ab</sup>	18.06 $\pm$ 8.22 <sup>a</sup>
Tukey 5%	—	7.34	—

\*Values with different notations in the same column show significant differences in the 5% Tukey test.

**Table 7:** Inhibition zone diameter of positive control and negative control

Kontrol	Jenis Ekstrak	Diameter zona penghambatan (mm) ± SD		
		<i>S. cerevisiae</i>	<i>L. plantarum</i>	<i>L. mesenteroides</i>
<b>Positive control</b>				
Fluconazole antibiotics	Ethanol	12.33 ± 1.84	nt	nt
	Methanol	27.33 ± 0.34	nt	nt
	<i>n</i> -Hexane	9.83 ± 0.60	nt	nt
	Water 60°C	10.28 ± 0.35	nt	nt
Streptomycin antibiotics	Ethanol	nt	13.56 ± 4.88	11.36 ± 1.13
	Methanol	nt	12.05 ± 0.25	11.25 ± 1.09
	<i>n</i> -Hexane	nt	11.34 ± 0.76	10.16 ± 0.17
	Water 60°C	nt	10.89 ± 0.63	18.42 ± 9.05
<b>Negative control</b>	Ethanol	10.82 ± 0.02	10.39 ± 0.38	9.42 ± 0.51
	Methanol	10.55 ± 0.15	8.51 ± 0.50	9.37 ± 0.38
	<i>n</i> -Hexane	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00
	Water 60°C	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00

0.00 – does not show antimicrobial activity; nt – not tested.

and 9.37 mm. respectively. Compared with the data shown in Table 6, the extract had an inhibitory zone diameter exceeding that of the positive control against *L. plantarum* and had the same strength against the positive control of *L. mesenteroides* and better than the negative control. This means that the resulting extract has antimicrobial compounds that can inhibit the growth of *L. plantarum* and *L. mesenteroides* bacteria.

### 3.3.2 MIC and MBC values

The concentrations of the tested *H. beccariana* Burck stem bark extract ranged from 2.5, 5.0, 10, 30, and 50 mg/mL. The data on the results of the minimum

inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) tests are shown in Table 8.

Table 8 shows that the MIC values of all extracts with solvents of ethanol, methanol, *n*-hexane, and water were at the smallest concentration of 2.5 mg/mL for each microbe used. In the MBC test, especially the ethanol and methanol extracts with a concentration range of 2.5–50 mg/mL were still unable to kill the microbes tested for *S. cerevisiae*, *L. plantarum*, and *L. mesenteroides*. The *n*-hexane extract was unable to kill *S. cerevisiae* too; thus, they were requiring a concentration of >50 mg/mL to kill the test microbes used. The *n*-hexane extract was able to kill only *L. plantarum* and *L. mesenteroides* with a concentration of 2.5 mg/mL.

## 4 Discussion

### 4.1 Yield extract of *H. beccariana* Burck stem bark

The solvent used in the maceration process has a different degree of polarity, so the yield of the extract produced is also different. More polar solvents in the extraction process will produce more yields than using non-polar solvents. This is evidenced by [46] that extraction in highly polar solvents (ethanol and methanol) yields high extract yields compared to non-polar solvents such as *n*-hexane. Methanol solvent has easy properties to form hydrogen and water in plant tissue cells and can dissolve

**Table 8:** MIC and MBC of various types of stem bark extract of *H. beccariana* Burck against *S. cerevisiae* yeast and *L. plantarum* and *L. mesenteroides* bacteria

Extract	Types of microbes	MIC (mg/mL)	MBC (mg/mL)
Ethanol	<i>S. cerevisiae</i>	2.5	>50
	<i>L. plantarum</i>	2.5	>50
	<i>L. mesenteroides</i>	2.5	>50
Methanol	<i>S. cerevisiae</i>	2.5	>50
	<i>L. plantarum</i>	2.5	>50
	<i>L. mesenteroides</i>	2.5	>50
<i>n</i> -Hexane	<i>S. cerevisiae</i>	2.5	>50
	<i>L. plantarum</i>	2.5	2.5
	<i>L. mesenteroides</i>	2.5	2.5
Water 60°C	<i>S. cerevisiae</i>	2.5	2.5
	<i>L. plantarum</i>	2.5	2.5
	<i>L. mesenteroides</i>	2.5	2.5

polar organic compounds [47]. Results were also reported by [48] that the best results were obtained when using ethanol as a solvent for the extraction of kinnow (*Citrus reticulata* L.) peel. [49] reported that the yield of *n*-hexane extract from *Dillenia suffruticosa* leaves was smaller (5.77%) compared to the yield of methanol extract (9.27%).

## 4.2 Total phenolic content of *H. beccariana* Burck stem bark extract

The use of *n*-hexane as a solvent resulted in lower total phenolic compared to ethanol, methanol, and water at 60°C. These results indicate that most of the phenolic compounds in the bark of Dipterocarpaceae are soluble in polar solvents. The extraction of phenolic compounds in plants is very suitable using polar solvents [50]. This study also used hot water at 60°C with a total phenolic content of 33.94 mg GAE/g extract. These results are the same as those of [51] that the total phenol content produced from a crude extract of Neem leaves using water as a solvent is higher than that of methanol and *n*-hexane. The release of hydrophilic phenolic compounds in plant cells will increase if given heat treatment [52].

Each *H. beccariana* Burck bark extract contains phenolic compounds that have the potential to prevent microbial damage to coconut sap. Phenolic compounds obtained from plant extracts can function as antimicrobials [7,53,54]. The phenolic extract obtained from honey functions as an antimicrobial and can inhibit the growth of gram-negative bacteria (*Pseudomonas aeruginosa*) and gram-positive bacteria (*Enterococcus faecalis* and *Staphylococcus aureus*) [55].

## 4.3 Total flavonoid content of *H. beccariana* Burck stem bark extract

The total flavonoid content obtained in the extract in this study was influenced by the type of solvent used for the extraction process. The levels of flavonoid compounds produced will vary because they are influenced by the source of the raw material or the type of plant, the type and condition of the solvent [56,57], and the type of extraction used [58,59].

The total flavonoid content (Table 2) produced using a solvent temperature of 60°C was higher than that of using ethanol, methanol, and *n*-hexane as solvents. Although statistically, the total levels of flavonoids produced by water

solvents were not significantly different from those using ethanol and methanol solvents. The high level of total flavonoids is influenced by the water solvent, which is given heat treatment. However, the yield (Figure 1) and total phenol content (Table 1) were lower than ethanol and methanol without heat treatment. The results are in line with the research of [60] that the heat treatment given to the extraction process of *Calophyllum inophyllum* leaves using 80% methanol as a solvent at temperatures ranging from 30°C to a maximum temperature of 60°C increased the total flavonoid content but could increase the extraction causing the yield and total phenol content to be small. Water solvent is used in the extraction process and assisted by an increase in temperature; the total flavonoid content in the extract will increase [18].

In terms of the potential content of flavonoid compounds in each *H. beccariana* Burck bark extract. The extracted material can function as a natural ingredient to prevent the process of damage to coconut sap caused by microbes. Flavonoids from natural ingredients have the potential to affect antimicrobial activity [61–63]. The antimicrobial activity test shown in Table 6 stated that the stem bark extract of *H. beccariana* Burck has an inhibitory effect on the sap-destroying microbe, such as *S. cerevisiae* yeast and *L. plantarum* and *L. mesenteroides* bacteria. Some researchers state that flavonoid compounds from plants function as antimicrobials [12,13,64] and have antibacterial properties [65,66] because the hydroxyl group on the aromatic ring of flavonoids increases the activity of flavonoid compounds [67].

## 4.4 Antioxidant activity (IC<sub>50</sub>) of *H. beccariana* Burck stem bark extract

The antioxidant activity shown in Table 1 has an IC<sub>50</sub> value that is almost the same as the ethanol extract of *Eucheuma Cotoni* [68], namely 90.10 ppm. The lower the IC<sub>50</sub> measurement value, so the stronger the antioxidant activity. The strong antioxidant activity is reflected in the presence of phenolic and flavonoid compounds in the extract. The presence of total phenolic and flavonoid levels in plant extracts will indicate antioxidant activity [69,70]. The phenolic compounds in a plant material act as antioxidants because phenolic compounds have redox properties, have hydroxyl groups, and can inhibit free radicals [71]. The conformation of structural antioxidants is influenced by antioxidants in the extract and DPPH radicals, so the number of antioxidant hydroxyl groups in the extract will be the same as the number of

reduced DPPH radical molecules [72]. Very high levels of flavonoid compounds correlated with having strong antioxidant activity in *Hypericum perforatum* extract [59].

#### 4.5 Toxicity test of *H. beccariana* Burck stem bark extract

The division of toxicity test categories with the BSLT method was  $LC_{50} < 500$  g/mL (1 g/mL = 1 ppm) classified as relatively toxic and  $LC_{50} < 100$  g/mL classified as very toxic [73]. The  $LC_{50}$  value of <200 g/mL of the compounds tested by the BSLT method is classified as toxic compounds and has the potential to be used as an anticancer [74]. Compounds that have high toxicity activity have the potential for anticancer compounds [75]. Crude methanol extract of the bark and leaves of *Diospyros mespiliformis* was safe to use for 28 days for therapeutic purposes through acute and sub-chronic toxicity testing in rats [76]. However, it is necessary to do further research *in vivo* on the bark extract of *H. beccariana* Burck to determine the actual dose for safe use.

#### 4.6 Content of bioactive compounds of *H. beccariana* Burck bark extract

##### 4.6.1 *H. beccariana* Burck stem bark ethanol extract

The most dominant compounds found in the ethanol extract of the stem bark of *H. beccariana* Burck are classified as fatty acid ester compounds (Table 5) consisting of ethyl oleate (15.52 and 10.85%), hexadecanoic acid, ethyl ester (13.42%), and hexanedioic acid, bis(2-ethylhexyl)ester (13.36%). The group of fatty acid ester compounds includes antimicrobial compounds. Oleic compounds are fatty acids that can be produced by *S. aureus* with antimicrobial functions [77]. In addition to fatty acid ester compounds, phenolic compounds have the potential as antimicrobial compounds. The phenolic group bound to the fatty acid chain is antimicrobial and can inactivate gram-positive bacteria, such as *Listeria innocua* [78].

##### 4.6.2 *H. beccariana* Burck stem bark methanol extract

The most dominant compounds found in the ethanol extract of the bark of *H. beccariana* Burck (Table 6) are

hexanedioic acid, bis(2-ethylhexyl)ester (25.93%), hexadecanoic acid, ethyl ester (11.57%), and 9-octadecenoic acid (Z)-ethyl ester (11.57 and 7.93%). These compounds are classified as fatty acid ester compounds. Hexanedioic acid, bis(2-ethylhexyl)ester compounds were also identified in the methanol:water extract of *Senegali gaumeri* leaf, which has anthelmintic activity against eggs and larvae of *Haemonchus contortus* [79]. Phenol compounds were also identified in the methanol extract of about 8.07%. These compounds function as antimicrobials [80].

##### 4.6.3 *H. beccariana* Burck stem bark *n*-hexane extract

The most dominant compounds found in the *n*-hexane extract of the bark of *H. beccariana* Burck (Table 7) were hexanedioic acid, bis(2-ethylhexyl)ester (70.07 and 0.18%) and hexadecanoic acid, ethyl ester (6.23%). These compounds are classified as fatty acid ester compounds. Hexanedioic acid, bis(2-ethylhexyl)ester compound is found in the biosynthesis of *Streptomyces* sp. TN262 strain and has antimicrobial properties [81]. On the other hand, the GC-MS results also identified hexadecane compounds (0.14%), although the concentration of these compounds was small. Fatty acid compounds such as hexadecane compounds function as antimicrobials obtained from isolates of D-3 actinomycetes [82].

##### 4.6.4 *H. beccariana* Burck stem bark water 60°C extract

The most dominant compounds found in the aqueous extract of the bark of *H. beccariana* Burck (Table 8) were 9-octadecenoic acid (Z)-ethyl ester (8.54%) and hexanedioic acid, bis(2-ethylhexyl)ester (6.76%). These compounds are classified as fatty acid ester compounds. The oleic acid compound identified was suspected to be antimicrobial. Oleic acid has antibacterial activity against *S. aureus* [83]. Hexanedioic acid, bis(2-ethylhexyl)ester is also present in the seed extract of *Foeniculum vulgare* Mill. functions as antimicrobial, anticancer, diuretic and anti-inflammatory [84] and is found in the wood extracts of *Populus lasiocarpa* and *Populus tomentosa* [85]. On the other hand, 14- $\beta$ -H-pregna compounds were identified in the aqueous extract of the bark of *H. beccariana* Burck. The compound 14- $\beta$ -H-pregna is a compound that has the prevention of diabetic retinopathy and this compound is found in the essential oils of the plants *Scutellaria multicaulis* and *Scutellaria bornmuelleri* [86].

## 4.7 Microbiological characteristics of *H. beccariana* Burck stem bark extract

### 4.7.1 Antimicrobial activity

The results shown in Table 6 that the bark extract of *H. beccariana* Burck used ethanol, methanol, *n*-hexane, and water at 60°C proved to be able to inhibit the growth of *S. cerevisiae* yeast and *L. plantarum* and *L. mesenteroides* bacteria. This is evident from the results shown in the GC-MS that each extract identified antimicrobial compounds with the largest group of compounds being fatty acid esters, followed by phenolic compounds. Groups of fatty acid compounds and fatty acid esters show their ability as antimicrobial compounds [87,88]. Phenol compounds can inhibit the growth of *Lactobacillus* bacteria [53]. Groups of compounds, such as hexanedioic acid, bis(2-ethylhexyl) ester, which are included in the group of fatty acid ester compounds, have antimicrobial properties. Hexanedioic acid, bis(2-ethylhexyl)ester compound is found in the methanol extract of Marine sponges (phylum Porifera), which functions as an antibacterial [89].

### 4.7.2 MIC and MBC values

Based on Table 8, the results of the MIC test were at 2.5 mg/mL against the *S. cerevisiae* yeast and *L. plantarum* and *L. mesenteroides* bacteria, which caused damage to coconut sap. These results indicate that the bark extract of *H. beccariana* Burck is still better than the study [42], which reported that propolis alcohol extract had an MIC at 50 mg/mL and propolis water extract had an MIC of 3.12–25 mg/mL of bacteria that cause spoilage in fish. The MIC value of the *H. beccariana* Burck extracts shown in Table 8 have an inhibitory concentration that is not much different from that of the methanol extract of the leaves of *Bridelia micrantha* (Hochst.) Baill had MICs against *Streptococcus pyogenes*, *Salmonella typhi*, and *Candida albicans* of 2.5, 1.25, and 2.5 mg/mL, respectively [90].

In the MBC test results, there were still extracts that were not able to kill the test microbes in the concentration range of 2.5–50 mg/mL. It is suspected that the bioactive components present in the extract have not been able to damage the cell membranes of yeasts and test bacteria, with the concentration of the extract given still inhibiting growth or being bacteriostatic and not yet capable of being bactericidal at the given concentration meaning that it has not been able to kill all

the microbes present. An antibiotic is bacteriostatic if it only suppresses microbial growth and is bactericidal if it can kill microbes [91]. Extracts that have MBC test results of 2.5 mg/mL against yeast and bacteria mean that the extract has bioactive components that can damage the cell membranes of yeasts and bacteria with a small concentration of 2.5 mg/mL. The bioactivity of active compounds is produced due to the interaction between active components that are antagonistic so that antimicrobial properties are formed and can damage microbial cell membranes [92].

## 5 Conclusion

The use of various solvents for the maceration extraction process gave different results in terms of yield, total phenol content, total flavonoid content, antioxidant activity, and toxicity of *Hopea beccariana* Burck bark extract. Ethanol and methanol solvents can produce higher yields than using *n*-hexane and water at 60°C. The highest total phenol content was obtained in the use of ethanol as solvent, followed by water at 60°C, methanol, and *n*-hexane. The identified bioactive compounds using GC-MS are supporting materials to strengthen and provide inhibitory power against sap-destroying microbes. The use of water as a solvent resulted in a higher total flavonoid content for the bark extraction of *H. beccariana* Burck. On the other hand, water solvents are safe to use, environmentally friendly, inexpensive, affordable, and very suitable for reacting with other plant materials needed for the food and pharmaceutical fields. The extract can inhibit the growth of *S. cerevisiae* yeast and *L. plantarum* and *L. mesenteroides* bacteria, so it can be used as a natural preservative for coconut sap.

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