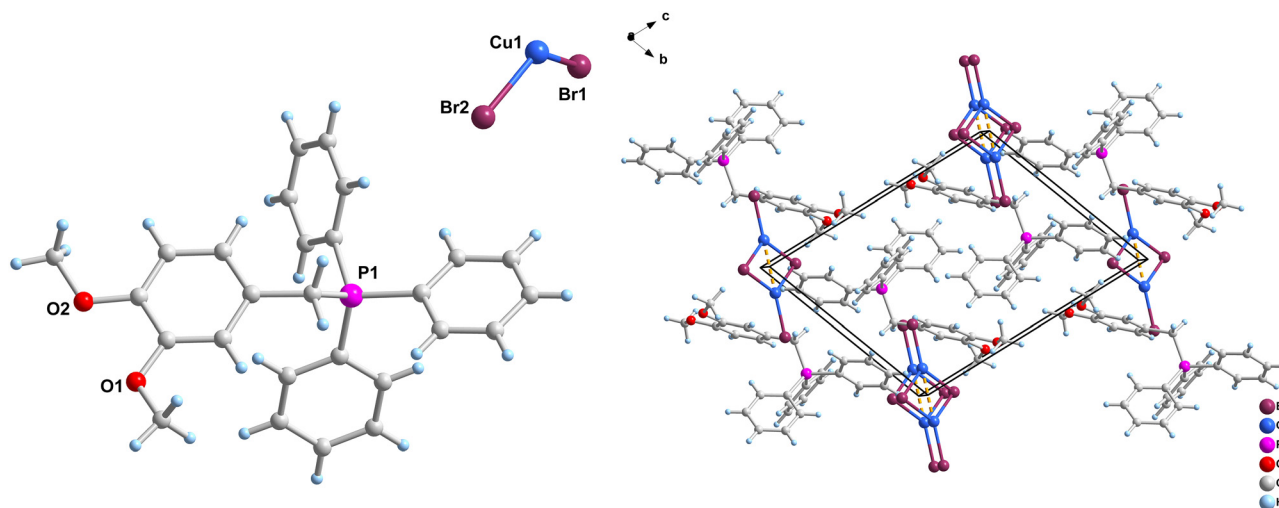


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Crystal structure of bis[(3,4-dimethoxybenzyl)triphenylphosphonium]di- μ -2-bromido-dibromidodicopper(I)



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Abstract

$C_{54}H_{52}Br_4Cu_2O_4P_2$, triclinic, $P\bar{1}$ (no. 2), $a = 9.9855(4)$ Å, $b = 10.5557(4)$ Å, $c = 13.3955(5)$ Å, $\alpha = 69.662(3)^\circ$, $\beta = 78.316(3)^\circ$, $\gamma = 79.167(3)^\circ$, $V = 1285.81$ Å³, $Z = 1$, $R_{gt}(F) = 0.0446$, $wR_{ref}(F^2) = 0.1110$, $T = 305.00(10)$ K.

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Table 1 contains the crystallographic data. The list of the atoms including atomic coordinates and displacement parameters can be found in the cif-file attached to this article.

1 Source of materials

The (3,4-dimethoxybenzyl)triphenylphosphonium bromide (0.5 mmol) and CuBr (0.5 mmol) were placed into a clean

Table 1: Data collection and handling.

Crystal:	Block, clear light colourless
Size:	0.12 × 0.1 × 0.1 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	4.04 mm ⁻¹
Diffractometer, scan mode:	XtaLAB Synergy R, ω -scans
θ_{max} , completeness:	25°, >99 %
$N(hkl)_{measured}$, $N(hkl)_{unique}$, R_{int} :	13810, 4543, 0.032
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 3,632
$N(param)_{refined}$:	300
Programs:	CrysAlis ^{PRO} , ¹ SHELX, ^{2,3} OLEX2 ⁴

glass vial. Subsequently, 2 mL of ethanol, 100 μ L of hydrobromic acid, and 200 μ L of hypophosphorous acid were added, and the mixture was stirred for 15 min. The CH₂Cl₂ (2 mL) was then introduced to form a clear solution. The resulting solution was filtered and left at room temperature to allow for slow evaporation. After five days, colorless transparent single crystals were obtained.

2 Experimental details

Single-crystal X-ray diffraction data were collected using a Rigaku diffractometer.¹ The structure was solved with the SHELXT program and refined using SHELXL within the Olex2.^{2–4}

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The H atoms were placed in geometrically idealized positions.

3 Comment

Among zero-dimensional metal halides, copper(I)-based halides have emerged as promising competitors for optoelectronic functional materials, owing to their abundant coordination configurations, high photoluminescence quantum yields (PLQYs), and exceptional structural designability, etc.^{5,6} In this work, one zero-dimensional copper(I)-based halide was synthesized. As shown in left part of the picture, its asymmetric unit contains one (3,4-dimethoxybenzyl)triphenylphosphonium (DOMBTP) cation and one half of a $Cu_2Br_4^{2-}$ anion. The bonds lengths of Cu1–Br1, Cu1–Br2 are 2.305, 2.412 Å, falling within the normal range reported in the literature.^{7,8} As shown in right part of the picture, its packing structure is zero-dimensional, containing $[Cu_2Br_4]^{2-}$ clusters and DOMBTP organic ligands.

Author contributions: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Conflict of interest: The authors declare no conflicts of interest regarding this article.

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References

1. Rigaku Oxford Diffraction. *CrysAlispro*; Rigaku Oxford Diffraction Ltd: Yarnton, Oxfordshire, England, 2015.
2. Sheldrick, G. M. *SHELXT*-Integrated Space-Group and Crystal-Structure Determination. *Acta Crystallogr.* **2015**, A71, 3–8.
3. Sheldrick, G. M. Crystal Structure Refinement with *SHELXL*. *Acta Crystallogr.* **2015**, C71, 3–8.
4. Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX2: A Complete Structure Solution, Refinement and Analysis Program. *J. Appl. Crystallogr.* **2009**, 42, 339–341.
5. Wu, J. -J.; Qi, J. -L.; Guo, Y.; Yan, S. -F.; Liu, W. -L.; Guo, S. -P. Reversible Tri-state Structural Transitions of Hybrid Copper(I) Bromides toward Tunable Multiple Emissions. *Inorg. Chem. Front.* **2024**, 11, 156–163.
6. Niu, C. -Y.; Jiang, H. Crystal Structure of $\{[(4\text{-fluorophenyl)methyl}]\text{triphenylphosphonium}\}\text{Dibromocopper(I)}$, $[C_{25}H_{21}FP]^+[CuBr_2]^-$. *Z. Kristallogr. – N. Cryst. Struct.* **2025**, 240, 167–169.
7. Wang, Q. -Q.; Li, H. -B.; Tong, H.; Zhu, J. -L.; Fan, J. -L.; Pan, S. -F.; Zhou, Z. -N.; Liu, W.; Ouyang, G. -F. Copper Mixed-Halide Hybrids with Intrinsic Broadband White-Light Emission for WLED Application with Ultrahigh Color Rendering Index. *Small* **2025**, 21, 2500660.
8. Lin, N.; Li, Y. -X.; Liu, Y.; Sun, K. -Q.; Zhang, H. -Y.; Lei, X. -W.; Chen, Z. -W. Broadband Blue Light Emissions of One-Dimensional Hybrid Cu(I) Halides with Ultrahigh Anti-water Stability. *Dalton Trans.* **2024**, 53, 16577–16584.