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Crystal structure of 4-cyclohexyl-5-(thiophen-2-yl)-2,4-dihydro-3H-1,2,4-triazole-3-thione, $C_{12}H_{15}N_3S_2$

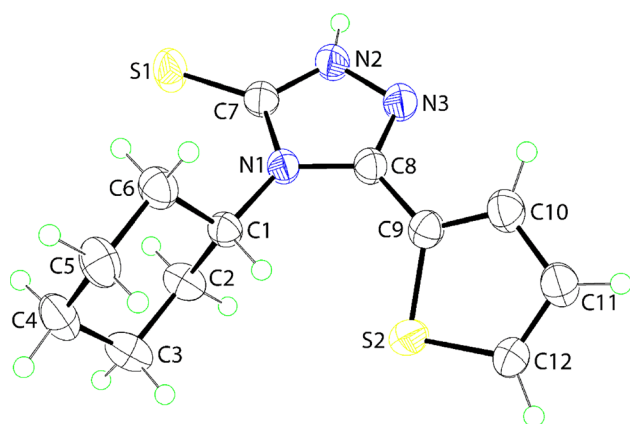


Table 1: Data collection and handling.

Crystal:	Colourless prism
Size:	0.22 × 0.11 × 0.05 mm
Wavelength:	CuK α radiation (1.54184 Å)
μ :	3.46 mm ⁻¹
Diffractometer, scan mode:	Rigaku Synergy, ω scan
$\theta_{\max}$, completeness:	74.5°, 100 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	14587, 2725, 0.051
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2,403
$N(\text{param})_{\text{refined}}$:	196
Programs:	Rigaku, ¹ SHELX, ^{2,3} WinGX ⁴

<https://doi.org/10.1515/ncrs-2025-0245>

Received May 22, 2025; accepted June 23, 2025;

published online June 27, 2025

Abstract

$C_{12}H_{15}N_3S_2$, monoclinic, $P2_1/n$ (no. 14), $a = 5.51310(10)$ Å, $b = 15.4525(4)$ Å, $c = 15.9261(4)$ Å, $\beta = 99.844(2)^\circ$, $V = 1336.79(5)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0359$, $wR_{\text{ref}}(F^2) = 0.1009$, $T = 160$ K.

CCDC no.: 2450625

1 Source of material

A mixture of thiophene-2-carbohydrazide (1.42 g, 0.01 mol) and cyclohexyl isothiocyanate (1.41 g, 0.01 mol), in ethanol

(10 mL), was heated under reflux for 30 min, and the solvent was then distilled off under vacuum. Sodium hydroxide in a 10 % aqueous solution (8 mL) was then added to the residue, and the mixture was heated under reflux for 2 h, then filtered hot. On cooling, the mixture was acidified with hydrochloric acid (pH 1–2) and the precipitated crude product was filtered, washed with water, dried and crystallised from aqueous ethanol to yield 2.28 g (86 %) of the title compound as colourless prisms. Single crystals suitable for X-ray diffraction were obtained by slow evaporation of a solution of the title compound in EtOH/CHCl₃ (2:1, v/v) held at room temperature for 2 days. **M. pt:** 469–471 K (uncorrected). ¹H NMR (CDCl₃, 500.13 MHz): δ 13.6 (s, 1H, NH), 6.98 (t, 1H, Thiophene–H, $J = 4.2$ Hz), 7.10 (d, 1H, Thiophene–H, $J = 4.2$ Hz), 7.18 (d, 1H, Thiophene–H, $J = 4.2$ Hz), 2.54–2.68 (m, 1H, Cyclohexyl–H), 1.50–1.72 (m, 4H, Cyclohexyl–H), 1.12–1.44 (m, 6H, Cyclohexyl–H). ¹³C NMR (CDCl₃, 125.76 MHz): δ 169.26 (C=S), 148.66 (Triazole–C5), 124.62, 126.04, 127.88, 128.0 (Thiophene–C), 24.44, 25.98, 30.88, 66.62 (Cyclohexyl–C).

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2 Experimental details

The C-bound H atoms were geometrically placed (C–H = 0.95–1.00 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The N-bound H atom was refined freely. The S2-containing thiophene-2-yl ring was disordered over two co-planar orientations, each of which was refined independently. The major component of the disorder had a site occupancy factor = 0.544(2).

3 Discussion

The 1,2,4-triazole scaffold is recognised as an essential component of several biologically interesting drugs and exhibits a wide range of chemotherapeutic properties.^{5,6} Marketed triazole-based drugs are extensively used as advantageous anti-fungal,⁷ anti-cancer,⁸ anti-bacterial⁹ and anti-viral¹⁰ agents. In addition, the thiophene heterocycle constitutes a main building block of many drugs.¹¹

The molecular structure of the title triazole-thiophene hybrid molecule, (I), is shown in figure (50 % probability ellipsoids; only the major component of the disordered S2-containing thiophen-2-yl ring is shown for reasons of clarity). The molecule comprises a central 1,2,4-triazole-3-thione group connected at the 4-position to a cyclohexyl ring, and at the 5-position to a thiophen-2-yl ring; the dihedral angle between the two five-membered rings is 35.9(3)° consistent with a splayed orientation. Within the 1,2,4-triazole ring, the lengthening of the formally C8–N3 [1.3041(19) Å] double bond and the shortening, for example, of the formally N2–N3 [1.3640(18) Å] and C8–N1 [1.3854(18) Å] single bonds indicate significant delocalisation of π -electron density over the constituent atoms.

In the crystallographic literature, there are two related N-alkyl structures available for comparison, namely alkyl = ethyl¹² and cyclohexyl,¹³ hereafter (II) and (III), respectively. Each of (II) and (III) exhibit similar patterns in the bond lengths within the 1,2,4-triazole ring as for (I). The relative orientations of the five-membered rings differ, however. Thus, in (II) and (III), the dihedral angles are 13.53(12) and 6.19(7)°, respectively, indicating closer to coplanar orientations than for (I). This difference is likely to arise due to steric considerations in (I). The separation between the methine–C–H and the thiophen-2-yl–C–H atoms of the disordered component of the ring is 2.25 Å which would contract if a more co-planar orientation was adopted.

In the molecular packing, the most prominent intermolecular interactions correspond to amine–N–H...S(thione) hydrogen-bonding [N2–H2n...S1ⁱ: H2n...S1ⁱ = 2.38(2) Å, N2...S1ⁱ = 3.2511(14) Å with angle at H2n = 169.1(19)° for symmetry operation (i) 2 – x, 1 – y, 1 – z]. The hydrogen bonds occur within a centrosymmetric aggregate through an eight-membered {...HNCS}₂ synthon. Similar synthons are noted in the crystals of (II) and (III).

Author contributions: All authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Conflict of interest: The authors declare no conflicts of interest.

Research funding: This study was financially supported by the the Princess Nourah bint Abdulrahman University Researchers Supporting Project No. PNURSP2023R3, from the Princess Nourah bint Abdulrahman University, Riyadh, Saudi Arabia.

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