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Crystal structure of (E)-2-(((5-chloro-3-methyl-1-phenyl-1H-pyrazol-4-yl)methylene)amino)-3',6'-dihydroxyspiro[isoindoline-1,9'-xanthen]-3-one – ethanol (1/2), $C_{35}H_{33}ClN_4O_6$

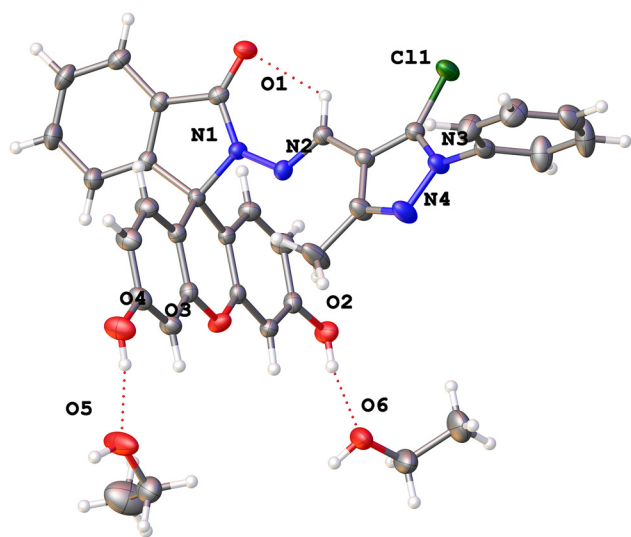


Table 1: Data collection and handling.

Crystal:	Light colourless block
Size:	0.28 × 0.15 × 0.10 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	0.17 mm ⁻¹
Diffractometer, scan mode:	Bruker APEX-II, φ and ω scans
$\theta_{\max}$, completeness:	27.5°, 99 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	27117, 7343, 0.030
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 6,152
$N(\text{param})_{\text{refined}}$:	422
Programs:	Bruker, ¹ Olex2, ^{2,3} SHELX ^{4,5}

The molecular structure is shown in the figure. Table 1 contains the crystallographic data and the list of the atoms including atomic coordinates and displacement parameters can be found in the cif-file attached to this article.

1 Source of material

The title compound is a natural product, which has been obtained by extraction and isolation from fungal fermentation products.⁶ All chemicals were of analytical grade and used as received without further purification. The preparation of the title compound is similar to the previously reported literature procedure.⁶ Fluorescein hydrazide (200 mg, 577.46 μmol) and 5-chloro-3-methyl-1-phenyl-1H-pyrazole-4-carbaldehyde (127.42 mg, 577.46 μmol) were mixed in ethanol (15 mL) with addition of a catalytic amount of acetic acid. The reaction mixture was refluxed at 78 °C for 2.5 h and the reaction was shown to be completed by thin layer chromatography (TLC). After cooling to room temperature, the solution was filtered, and the colorless crystals were obtained by slow evaporating in ethanol at room temperature.

2 Experimental details

The structure was treated with the Olex2 crystallographic software package,^{2,3} solved with the SHELXT structure

<https://doi.org/10.1515/ncrs-2025-0236>

Received May 20, 2025; accepted July 21, 2025;

published online July 31, 2025

Abstract

$C_{35}H_{33}ClN_4O_6$, monoclinic, $P2_1/c$ (no. 14), $a = 18.2808(8)$ Å, $b = 9.0510(4)$ Å, $c = 19.4323(7)$ Å, $\beta = 93.406(10)^\circ$, $V = 3209.6(2)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.046$, $wR_{\text{ref}}(F^2) = 0.129$, $T = 120.0$ K.

CCDC no.: 2446999

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solution program and refined with the SHELXL refinement package.^{4,5} Carbon-bound hydrogen atoms were placed in calculated positions and refined with riding coordinates, with $U_{\text{iso}}(\text{H})$ fixed at 1.2 times of $U_{\text{eq}}(\text{C})$ (Tables 1 and 2).

3 Comment

Fluorescein serves as an ideal fluorescent reporter for chemosensors due to its superior photophysical properties.⁷ Moreover, certain fluorescein hydrazones derived from substituted benzaldehydes have demonstrated significant biological activity, by functioning as non-intercalative topoisomerase catalytic inhibitors.⁸ Notably, fluorescein hydrazone derivatives incorporating aldehyde moieties have been developed as effective fluorescent probes for the detection of Hg²⁺, Zn²⁺ or Cu²⁺ ions.^{9–11} Herein, we report the crystal structure of the title compound derived from fluorescein hydrazone and a substituted pyrazole-4-aldehyde, namely 5-chloro-3-methyl-1-phenyl-1H-pyrazole-4-carbaldehyde.

The title compound C₃₅H₃₃ClN₄O₆ was crystallized in $P2_1/c$ with one C₃₁H₂₁ClN₄O₄ and two solvent molecules C₂H₆O in its asymmetric unit. The molecule structure comprises one xanthene ring and one 1-phenyl-1H-pyrazole moieties. The geometry of the title structure was characterized with the bond angles and lengths. In particular, the bond angles for C23...N4...N3, O1...C20...N1, and C22...C24...Cl1 are 106.14 (13)°, 125.74(14)° and 129.00(12)°, respectively. For another, the bond lengths for N2...C30, N2...N1, O1...C20, Cl1...C24, N4...C23 and C22...C24, are 1.283(2) Å, 1.3661(17) Å, 1.2306(19) Å, 1.6973(16) Å, 1.324(2) Å and 1.377(2) Å. All bond angles and lengths of the title structure are within the expected range and comparable with those of reported fluorescein hydrazones^{12–14} and a directly related spiro compound.¹⁵ In the crystal, hydrazine molecules are arranged in layers by ethanol molecules via hydrogen bonds O–H...O and O–H...N, with max D–A distance 2.9 Å and minimum angle 120°.

Acknowledgments: This work was supported by Jiangsu Provincial Drug Administration's Scientific Research Program for Drugs (No. 202367), the Scientific Research Personnel Training Program of Kangda College of Nanjing Medical University (No. KD2022KYRC007), Science & Technology Funds of Kangda College of Nanjing Medical University (No. KD2023KYJJ017, KD2023KYJJ018 & KD2023KYJJ019), and the Science & Technology Funds of Lianyungang (No. JCYJ2320), Jiangsu Student's foundation for innovation and entrepreneurship training program (No. 202413980013Y & 202413980053Y).

Author contributions: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Conflict of interest: The authors declare no conflicts of interest regarding this article.

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