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The crystal structure of 3-(1-fluoro-2-(naphthalen-2-yl)-2-oxoethyl)-2-methoxy-3,4-dihydroisoquinolin-1(2H)-one, C₂₂H₁₈FNO₃

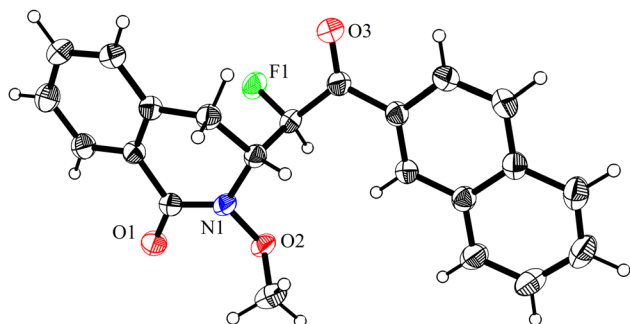


Table 1: Data collection and handling.

Crystal:	Clear light white plate
Size:	0.20 × 0.03 × 0.01 mm
Wavelength:	Cu K α radiation (1.54184 Å)
μ :	0.82 mm ⁻¹
Diffractometer, scan mode:	Rigaku Synergy, ω scans
$\theta_{\max}$, completeness:	78.6°, 100 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	11495, 3518, 0.029
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2967
$N(\text{param})_{\text{refined}}$:	245
Programs:	BRUKER, ¹ SHELX, ² OLEX2 ³

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Abstract

C₂₂H₁₈FNO₃, monoclinic, $P2_1/n$, (no. 14), $a = 11.0201(3)$ Å, $b = 7.3436(2)$ Å, $c = 21.6386(6)$ Å, $\beta = 94.376(2)^\circ$, $V = 1746.05(8)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0391$, $wR_{\text{ref}}(F^2) = 0.1090$, $T = 213$ K.

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The molecular structure is shown in the figure. Table 1 contains the crystallographic data and the list of the atoms including atomic coordinates and displacement parameters can be found in the cif-file attached to this article.

1 Source of materials

N-Methoxybenzamide (30.2 mg, 0.2 mmol), 2,2-difluoro-1-(naphthalen-2-yl) but-3-en-1-one (69.6 mg, 0.3 mmol), Rh cat.

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(8.3 mg, 5 mol%), sodium dihydrogenphosphate (31.2 mg, 0.2 mmol), and MeOH (2 mL) were placed in a 10 mL glass tube. The mixture was stirred at 50 °C. After 24 h the mixture was concentrated *in vacuo*. The residue was purified through flash chromatography. Finally, the crude product was recrystallized from dichloromethane and hexane (2 mL/4 mL). Clear light white plate, yield 80.2 %.

2 Experimental details

The SHELXL² refinement package was employed to refine the crystal structure. The Olex2³ was used to visualize the crystal structure. All hydrogen atoms were placed in idealized positions. Their U_{iso} values were set to $1.2U_{\text{eq}}$ of the parent atoms.

3 Comment

Fluorine chemistry is widely prevalent in the fields of organic chemistry and pharmaceutical chemistry. The introduction of fluorine atoms or fluorine-containing groups into biologically active compounds often alters their biological activity and physicochemical properties.^{4,5} This unique characteristic of fluorine means that introducing fluorine atoms or fluorine-containing groups into molecular structures can significantly alter relevant molecular properties, including physical, chemical, and biological characteristics.⁶ There are two main basic strategies for forming C–F bonds in organic fluorine compounds: fluorination and building block methods.⁷ In addition, dihydroisoquinolinone

series compounds exhibit numerous important biological activities such as antimalarial,⁸ antitussive,⁹ and anti-parkinsonian properties,¹⁰ and so on. We have enriched the library of fluorine-containing compounds by using a rhodium-catalyzed fluorine-modification method for the synthesis of dihydroisoquinoline derivatives.

In this study, the crystal structure was determined to gain a better understanding of its molecular conformation and intermolecular interactions. The title molecule, is shown in the figure. The bond distance of F1–C11 is 1.3926(15) Å. The bond lengths of O1–C1, O2–N1 and O3–C12 are 1.2284(16), 1.4025(14) and 1.2113(17) Å. The N1–O2–C10, and O2–N1–C9 angles are 110.32(11) and 112.69°, respectively, while those for C1–N1–C9 is 127.67(11)°. All of the hydrogen atoms were placed in the calculated positions. The bond lengths and angles within the molecule are within the expected ranges, indicating a well-ordered structure.^{11,12}

Author contribution: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Conflict of interest: The authors declare no conflicts of interest regarding this article.

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