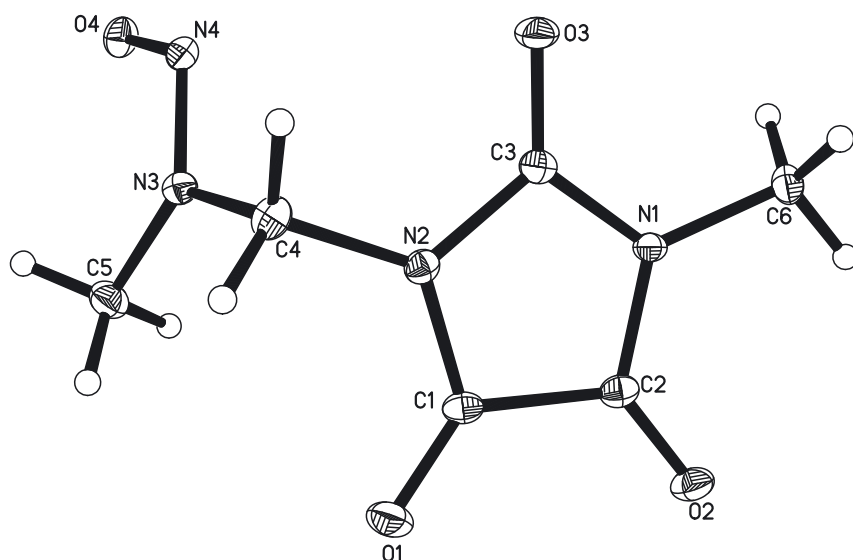


Wei Xiaolin, Zhou Jianhui, Geng Jinjun, Wen Zhiwei, Qu Manyi, Hou Xiaoting, Li Ruiqin, Feng Rui, Zhao Lukui, Lian Pengbao\* and Wang Jianlong

# The crystal structure of 1-methyl-3-(N-methylnitrous amide–N-methylene)imidazolidine-2,4,5-trione



<https://doi.org/10.1515/ncrs-2025-0013>

Received January 7, 2025; accepted February 4, 2025;

published online February 21, 2025

## Abstract

$C_6H_8O_4N_4$ , orthorhombic,  $Pna2_1$  (no. 33),  $a = 6.5487(4)$  Å,  $b = 17.5223(9)$  Å,  $c = 7.5728(4)$  Å,  $V = 868.97(8)$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{gt}(F) = 0.0740$ ,  $wR_{ref}(F^2) = 0.0761$ ,  $T = 106.0$  K.

CCDC no.: 2421335

The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of

**Table 1:** Data collection and handling.

|                                                       |                                                               |
|-------------------------------------------------------|---------------------------------------------------------------|
| Crystal:                                              | block                                                         |
| Size:                                                 | $0.40 \times 0.35 \times 0.15$ mm                             |
| Wavelength:                                           | MoK $\alpha$ radiation (0.71073 Å)                            |
| $\mu$ :                                               | 0.13 mm <sup>-1</sup>                                         |
| Diffractometer, scan mode:                            | Xcalibur, Eos, Gemini, $\varphi$ and $\omega$ scans           |
| $\theta_{max}$ , completeness:                        | 26.0°, 100 %                                                  |
| $N(hkl)_{measured}$ , $N(hkl)_{unique}$ , $R_{int}$ : | 3112, 1612, 0.026                                             |
| Criterion for $I_{obs}$ , $N(hkl)_{gt}$ :             | $I_{obs} > 2\sigma(I_{obs})$ , 1,530                          |
| $N(param)_{refined}$ :                                | 129                                                           |
| Programs:                                             | Rigaku, <sup>1</sup> Olex2, <sup>2</sup> SHELX <sup>3,4</sup> |

\*Corresponding author: Lian Pengbao, Shanxi North Xingan Chemical Industry CO. LTD, China North Industries Group Corporation Limited, Taiyuan 030008, Shanxi Province, P.R. China, E-mail: lianpengbao@163.com. <https://orcid.org/0000-0002-0443-8806>

Wei Xiaolin, Zhou Jianhui, Wen Zhiwei, Qu Manyi, Hou Xiaoting, Li Ruiqin, Feng Rui and Zhao Lukui, Shanxi North Xingan Chemical Industry CO. LTD, China North Industries Group Corporation Limited, Taiyuan 030008, Shanxi Province, P.R. China

Geng Jinjun, Inner Mongolia Aerospace Hongxia Chemical CO. LTD, Hohhot 010076, Inner Mongolia, P.R. China

Wang Jianlong, School of Chemistry and Chemical Engineering, North University of China, Taiyuan 030051, Shanxi Province, P.R. China

the atoms including atomic coordinates and displacement parameters.

## 1 Source of material

An amount of 0.2 g of 1-methyl-3-(N-methylnitrous amide–N-methylene)imidazolidine-2,4,5-trione was added to 5 ml of methanol and evaporated at room temperature to give colorless block crystals.

**Table 2:** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ ).

|     | <i>x</i>   | <i>y</i>     | <i>z</i>   | <i>U</i> <sub>iso</sub> */ <i>U</i> <sub>eq</sub> |
|-----|------------|--------------|------------|---------------------------------------------------|
| O1  | 0.4777 (3) | 0.98274 (10) | 0.2486 (3) | 0.0284 (5)                                        |
| O2  | 0.1086 (3) | 0.99297 (10) | 0.4813 (3) | 0.0249 (5)                                        |
| O3  | 0.5304 (3) | 0.80631 (9)  | 0.6911 (2) | 0.0263 (5)                                        |
| O4  | 0.7007 (3) | 0.67930 (10) | 0.2076 (2) | 0.0223 (5)                                        |
| N1  | 0.2743 (3) | 0.89228 (12) | 0.6179 (3) | 0.0161 (5)                                        |
| N2  | 0.5581 (3) | 0.88764 (12) | 0.4497 (3) | 0.0166 (5)                                        |
| N3  | 0.7308 (3) | 0.80105 (12) | 0.2525 (3) | 0.0167 (5)                                        |
| N4  | 0.7354 (3) | 0.73085 (12) | 0.3168 (3) | 0.0188 (5)                                        |
| C1  | 0.4395 (4) | 0.94426 (14) | 0.3765 (3) | 0.0189 (6)                                        |
| C2  | 0.2483 (4) | 0.94871 (15) | 0.4952 (4) | 0.0182 (6)                                        |
| C3  | 0.4619 (4) | 0.85584 (13) | 0.5981 (3) | 0.0181 (6)                                        |
| C4  | 0.7543 (4) | 0.86032 (16) | 0.3840 (4) | 0.0209 (6)                                        |
| H4A | 0.828936   | 0.902718     | 0.332792   | 0.025*                                            |
| H4B | 0.833870   | 0.840649     | 0.481958   | 0.025*                                            |
| C5  | 0.6891 (5) | 0.81782 (16) | 0.0680 (4) | 0.0250 (7)                                        |
| H5A | 0.755578   | 0.780663     | −0.005253  | 0.037*                                            |
| H5B | 0.739611   | 0.867779     | 0.039830   | 0.037*                                            |
| H5C | 0.544441   | 0.816083     | 0.047669   | 0.037*                                            |
| C6  | 0.1395 (5) | 0.87887 (15) | 0.7695 (4) | 0.0249 (7)                                        |
| H6A | 0.095577   | 0.826603     | 0.769584   | 0.037*                                            |
| H6B | 0.022570   | 0.911733     | 0.761351   | 0.037*                                            |
| H6C | 0.212121   | 0.889513     | 0.876860   | 0.037*                                            |

## 2 Experimental details

Hydrogen atoms were placed in their geometrically idealized positions and constrained to ride on their parent atoms.

## 3 Discussion

1-Methyl-2,4,5-trinitroimidazole is an energetic compound with good performance.<sup>5,6</sup> 1-Methyl-2,4,5-trinitroimidazole is obtained by nitration of 1-methyl-2,4,5-triiodoimidazole with a mixture of  $\text{HNO}_3$  and  $\text{H}_2\text{SO}_4$ .<sup>7–10</sup> The title compound was obtained by nitration of 1-methyl-2,4,5-triiodoimidazole with a mixture of  $\text{HNO}_3$  and  $\text{H}_2\text{SO}_4$  in DMF solution.

The crystal structure of the title compound was analyzed and refined by Shelxl program.<sup>1–4</sup> As shown in the

figure, the asymmetric unit contains one title molecule. The bond lengths and angles within these moieties are in the expected ranges.<sup>11,12</sup> The imidazole rings are planar, the carbon atoms of the methyl group (N1) and the methylene group (N2), the oxygen atom of the carbonyl group are nearly coplanar with the imidazole ring. However, all other atoms are twisted out of the imidazole ring.

**Acknowledgments:** We thank the Center of Testing and Analysis, Beijing University of Chemical Technology, for support.

## References

- Agilent Technologies. CrysAlis<sup>PRO</sup> Software System, Version 1.171.35.15; Agilent Technologies UK Ltd: Oxford, UK, 2011.
- Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX2: A Complete Structure Solution, Refinement and Analysis Program. *J. Appl. Cryst.* **2009**, *42*, 339–341.
- Sheldrick, G. M. SHELXT – Integrated Space-Group and Crystal-Structure Determination. *Acta Crystallogr.* **2015**, *A71*, 3–8.
- Sheldrick, G. M. Crystal Structure Refinement with SHELXL. *Acta Crystallogr.* **2015**, *C71*, 3–8.
- Lian, P. B.; Chen, J.; Chen, L. Z.; Zhao, C. Y.; Wang, J. L.; Shen, F. F. Preparation of 1-Methyl-2,4,5-Trinitroimidazole from Derivatives of 1-methylimidazole and its Oxidation under Nitration Conditions. *Chem. Heterocycl. Compd.* **2020**, *56*, 55–59.
- Jadhav, H. S.; Talawar, M. B.; Sivabalan, R.; Dhavale, D. D.; Asthana, S. N.; Krishnamurthy, V. N. Synthesis, Characterization and Thermolysis Studies on New Derivatives of 2,4,5-trinitroimidazoles: Potential Insensitive High Energy Materials. *J. Hazard. Mater.* **2007**, *143*, 192–197.
- Chen, L. Z.; Lian, P. B.; Wang, J. L. The Crystal Structure of 1-methyl-2,4-dinitro-5-iodoimidazole,  $\text{C}_4\text{H}_3\text{IN}_4\text{O}_4$ . *Z. Kristallogr. NCS* **2017**, *232*, 213–214.
- Ravi, P.; Tewari, S. P. Microwave-assisted Synthesis of 1-Methyl-2,4,5-Trinitroimidazole. *Propellants Explos. Pyrotech.* **2012**, *37*, 544–548.
- Lian, P. B.; Chen, L. Z.; Chen, J.; Wang, J. Q.; Wang, J. L.; Chen, J. The Nitration of 1-Methyl-2,4,5-Triiodoimidazole and its Oxidation By-Product under Nitration Conditions. *J. Energ. Mater.* **2022**, *40*, 46–60.
- Lian, P. B.; Zhang, Y.; Gu, Y. L.; Wu, J. J.; Zhao, L. K.; Cao, C.; Wang, J. L. Synthesis of 1-Methyl-2,4,5-Trinitroimidazole by Micro-channel Reaction Technology. *Chin. J. Explos. Pyrotech.* **2024**, *47*, 541–548.
- Yoshihara, R.; Hosomi, H.; Aoyama, H.; Ohba, S. Acta Crystallographica, Section C: Crystal. *Struct. Commun.* **1999**, *55*, 594.
- Lian, P.-B.; Yuan, Y.; Chen, J.; Chen, L.-Zh.; Wang, J.-L. *Chem. Heterocycl. Compd.* **2020**, *56*, 1010–1014.